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NASA CONTRACTOR REPORT 166494



Laboratory Infrared Studies of Molecules of Atmospheric And Astrophysical Interest



K. Narahari Rao

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Laboratory Infrared Studies of Molecules of Atmospheric And Astrophysical Interest

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Prepared for Ames Research Center Under Grant NSG-2175



National Aeronautics and Space Administration

Ames Research Center Moffett Field, California 94035



The Ohio State University

Department of Physics

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Columbus, Ohio 43210

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Re: Laboratory Infrared Studies of Molecules of Atmospheric and Astrophysical Interest

Contract Number: NSG 2175

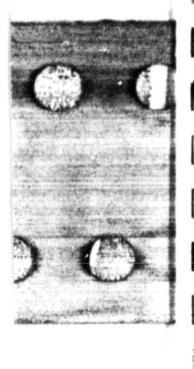
Duration: October 1, 1976--October 1, 1982

This NASA grant administered by Moffett Field (Supervisor, Dr. R. Boese) has provided funding for a period of three years. During the no-cost extension period, all research done has been published and the report is a compilation of nineteen reprints on the molecular species:

NO₂ \vee_2 \leq 2 \vee_2 ; H₂O (6050-9730 cm⁻¹; O₃ \vee_2 ; C₂H₂ \vee_5 and 6680-6460 cm⁻¹; CH₃D \vee_3 \leq \vee_6 ; CH₃I \vee_6 ; NH₃ \vee_2 , 2 \vee_2 , \vee_4 ; ND₃ \vee_2 ; PH₃ \vee_1 \leq \vee_3 ; OCS \vee_1 ; GeH₄ \vee_4 .

Much of the work was done by using the Dopplerlimited resolution provided by diode lasers. The diode laser was used as a source to a grating spectrometer which has been used earlier for high resolution studies. This technique provided many advantages. Wherever possible, the studies have been directed to intensity determinations of infrared bands.

> K. Marahari Rao Principal Investigate





JOURNAL OF MOLECULAR SPECTROSCOPY 88, 251-258 (1981)

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Diode Laser Measurements of Intensities, N_2 -Broadening, and Self-Broadening Coefficients of Lines of the ν_2 Band of $^{14}N^{16}O_2$

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A diode laser spectrometer (resolution 0.0013 cm⁻¹) was used to record, in the 12- μ m region, high-quality spectra of the ν_2 band of NO₂. Using these spectra, it was possible to obtain the N₂-broadening coefficients and an average self-broadening coefficient from measurements made for sever lines of this band. In addition, 30 single spin-component line intensities were measured. From them, through a least-squares fit, the purely vibrational transition moment of the ν_2 band, as well as two correcting rotational terms involved in the expansion of the transition moment operator, were obtained. These results led to the determination of the dipole moment derivative $\partial \mu_x/\partial q_2 = -0.0604_1 \pm 0.0037$ D. It was also demonstrated that there is good consistency between the correcting terms deduced from the observed intensities and their theoretical estimates. Finally, a complete spectrum of the ν_2 band of NO₂ was computed, providing a total band intensity $S_v(\nu_2) = 0.542 \times 10^{-18}$ cm⁻¹/ molecule cn⁻² at 296 ν .

1. INTRODUCTION

Many high-resolution studies concerning the line positions of nitrogen dioxide in the infrared have been reported but much less work has been performed on NO_2 line intensities and widths (I-4). These line parameters are of importance for precise determinations of the concentration of nitrogen dioxide in the atmosphere and in combustion products. As a natural continuation of our study of the line positions of the ν_2 band of NO_2 (5), we present here results on intensities, N_2 -broadening, and self-broadening coefficients of lines belonging to the ν_2 band of this molecule.

II. EXPERIMENTAL DETAILS

The diode laser spectrometer used in this investigation has been described earlier (6, 7). A commercial sample of NO₂ was used. Although it was stated to

¹ Present address: NOAA-NESS, FB4, S/RE21, Washington, D. C. 20233.

² Present address: Tachisto Inc., 13 Highland Circle, Needham, Mass. 02194.

be 99% pure, our experience indicated that it contained a significant amount of NO as an impurity. CO₂ is also believed to be another minor impurity in this gas. In order to purify the sample, NO₂ at about 100 Torr was introduced into a glass bulb connected to a mechanical pump and a diffusion pump. Using a mixture of methanol and dry ice the sample was then frozen to -78°C. This frozen solid usually had a bluish tinge indicating the presence of NO. In order to remove this NO, the frozen solid at -78° C was pumped for an hour, then warmed to room temperature, refrozen, and pumped again. This procedure removed the blue tinge completely, leaving behind a pure white solid. This procedure also eliminated CO₂. For line strength measurements, two Pyrex cells of lengths 101.6 and 201.6 cm were used. Since NO₂ reacts with H₂O to form HNO₃, which can cause interference, the cells and the gas-handling systems were pumped to pressures less than 10⁻⁵ Torr. KRS-5 windows were used for the cells. Ordinarily, the amount of dimer of NO2, namely, N2O4, depends on the total gas pressure and temperature. If P_{NO_2} and $P_{N_2O_4}$ denote the pressures of NO₂ and N₂O₄ (in atmospheres), then, the equilibrium constant $k_P(T)$ for the reaction (8)

$$N_2O_4 \rightleftharpoons 2NO_2$$

is given by

$$\frac{(P_{\text{NO}_2})^2}{P_{\text{NoO}_4}} = k = 1.08 \times 10^{13} \times T^{-1.304} \exp\left(-\frac{14520}{1.987 \, T}\right) = \frac{(P_{\text{NO}_2})^2}{P - P_{\text{NO}_2}} \,, \quad (1)$$

where $P = P_{\rm NO_2} + P_{\rm N_2O_4}$. For 296 K and a total pressure of 1 Torr, the above equation was solved numerically. It was found that only 1% of the gas was in the form of N₂O₄. Therefore, for all line strength measurements, pressures were kept below 1 Torr. The scans were taken within 1 hr after filling the cells and several repetitive scans were made on each line for various pressure × pathlength values. Pressures were measured with MKS Baratron gauges with an estimated uncertainty of less than 1% and the cell temperatures were monitored during each scan by precision thermistor probes. The dispersion was calculated from the wavenumbers of NO₂ lines recorded simultaneously with the fringes obtained with a 1-in.-long germanium etalon. The analog records were digitized with a Bendix Datagrid digitizer for processing. For self-broadening measurements we used three Pyrex cells 6.3, 11.3, and 21.7 cm in length and the pressure of NO₂ ranged from 12 to 30 Torr. For nitrogen-broadening experiments, a Pyrex cell, 201.6 cm long with total pressures up to 30 Torr and NO₂ pressures less than 1 Torr, was used.

III. MEASUREMENTS OF LINE INTENSITIES

We have selected 30 well-isolated lines, the equivalent widths of which have been measured several times at different pressures and path lengths. The intensity k_{σ}^{P} (in cm⁻² atm⁻¹) of each line has been extracted from the equivalent width assuming a value of 0.112 cm^{-1} atm⁻¹ for the self-broadening coefficient and taking into account wing corrections. The results are given in Table I together with their statistical errors (one standard deviation).

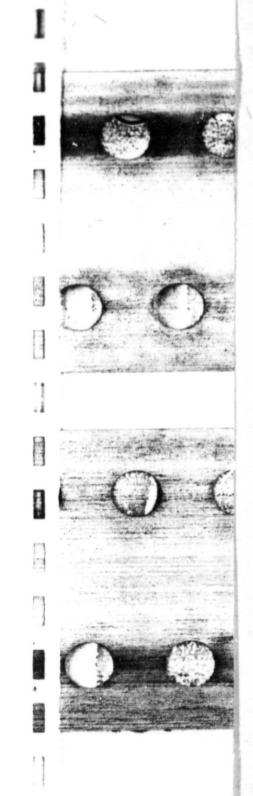


TABLE I $\label{eq:table_sure}$ Measured and Calculated Line Intensities of the ν_3 Band of $^{14}N^{16}O_2$

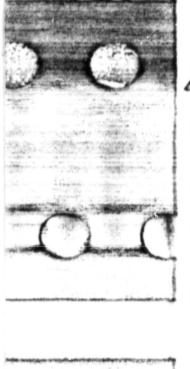
SIGMA	N. K. K.	N K . K c	i =N	kp (obs)	hp (calc)	$\frac{k_{\sigma}^{p}(obs) - k_{\sigma}^{p}(cetc)}{k_{\sigma}^{p}(obs)}$
(cm ⁻¹)				(10 ⁻³ cm ⁻² atm ⁻¹)	(10 ⁻³ ce ⁻² ete ⁻¹)	
825.335	35 5 31	35 4 32	-1/2	2.067 ± 0.11	1.984	48
825.345	36 5 31	36 4 32	+1/2	1.812 # 0.14	1.812	ox
825.357	34 5 29	34 4 30	-1/2	2.174 # 0.11	2.219	21
825.381	35 5 31	35 4 32	+1/2	2.060 # 0.12	2.041	11
825.386	33 5 29	33 4 30	-1/2	2.349 ± 0.15	2.472	5 %
825.476	29 5 25	29 4 26	-1/2	3.643 # 0.18	3.595	1 %
825.488	31 5 27	31 4 28	+1/2	3.252 ± 0.14	3.124	4 X
825.495	28 5 23	28 4 24	-1/2	4.206 ± 0.18	3.917	7%
835.806	12 5 7	11 4 8	-1/2	6.819 ± 0.39	7.116	4 %
836.648	13 5 9	12 4 8	-1/2	7.099 ± 0.44	7.116	ox
36.761	13 5 9	12 4 8	+1/2	7.548 + 0.63	7.711	5.8
37.488	14 5 9	13 4 10	-1/2	7.422 ± 0.44	7.116	41
37.594	14 5 9	13 4 10	+1/2	7.283 ± 0.32	7.637	31
39.166	16 5 11	15 4 12	-1/2	7.075 # 0.19	6.967	2 %
39.259	16 5 11	15 4 12	+1/2	7.063 ± 0.24	7.413	5 X
843.705	25 6 20	25 5 21	+1/2	3.505 # 0.20	3.446	5.2
843.719	24 6 18	24 5 19	+1,2	3.547 # 0.14	3.645	31
343.732	23 6 18	23 5 19	+1/2	3.705 # 0.16	3.843	4 %
343.745	22 6 16	22 5 17	+1/2	3.833 ± 0.23	4.017	5 X
343.758	21 6 16	21 5 17	91/2	4.471 ± 0.19	4.190	6%
843.770	20 6 14	20 5 15	+1/2	4.229 ± 0.18	4.314	2.8
843.783	19 6 14	19 5 15	+1/2	4.265 # 0.19	4.438	4%
843.794	18 6 12	18 5 13	+1/2	4.367 = 0.20	4.512	3 X
343.806	17 6 12	17 5 13	+1/2	4.358 ± 0.17	4.537	4%
843.817	16 6 10	16 5 11	+1/2	4.348 = 0.19	4.537	4%
61,929	28 7 21	28 6 22	+1/2	1.766 ± 0.074	1.778	1 %
861.965	25 7 19	25 6 20	+1/2	2.115 4 0.10	2.189	4%
862.032	19 7 13	19 6 14	+1/2	3.247 ± 0.15	2.777	15%
862.065	16 7 9	16 6 10	+1/2	2.937 ± 0.10	2.802	5 %
862.100	13 7 7	13 6 8	+1/2	2.579 # 0.12	2.529	2 %

IV. MEASUREMENTS OF LINEWIDTHS

The instrumental function of the diode laser spectrometer has been shown to be Gaussian with a half-width at half-maximum (HWHM) $\gamma_I = 0.65 \times 10^{-3}$ cm⁻¹ (9). We have measured the half-width $\gamma^{\rm obs}$ of the observed line absorption profile at $I_{\rm o}(I_{\rm min}/I_{\rm o})^{1/2}$, where $I_{\rm o}$ is the measured continuum or baseline determined from the region of no absorption and $I_{\rm min}$ is the measured absorption at the center of the line. This would give the true half-width at half-maximum of the absorption coefficient assuming no contribution from the apparatus function. As a first approximation, where γ_I is small compared to the half-width $\gamma_{\rm V}$ of the Voigt profile, the Lorentzian half-width $\gamma_{\rm L}$ can be obtained from $\gamma^{\rm obs}$ assuming $\gamma^{\rm obs} \simeq \gamma_{\rm V}$ through the relation (10)

$$\gamma_{\rm L} = \gamma_{\rm V} \left(7.7254 - 6.7254 \left(1 + 0.3195 \left(\frac{\gamma_{\rm D}^{\rm eff}}{\gamma_{\rm V}} \right)^2 \right)^{1/2} \right) ,$$
 (2)

where $\gamma_D^{\text{eff}} = (\gamma_I^2 + \gamma_D^2)^{1/2}$. Then, using a program which explicitly performs





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TABLE II N_2 -Broadening Coefficients for Seven Lines of the ν_2 Band of NO_2

								7002- 05 (cm-1 atm -1)
SIGMA	N.	К,	к,	N	К,	Кс	J -N	measured (this work)
835.806	12	5	7	11	4	8	- 1	0.079 ± 0.009
836.648	13	5	9	12	4	8	- ž	0.078 + 0.004
836.761	13	5	9	12	4	8	+ 2	0.076 ± 0.005
837.488	14	5	9	13	4	10	- ž	0.078 ± 0.007
837.594	14	5	9	13	4	10	* 2	0.076 ± 0.006
839.166	16	5	11	15	4	12	- 1	0.075 ± 0.007
839.259	16	5	11	15	4	12	+ 2	0.075 ± 0.005

the convolution of the absorption profile with the Gaussian instrumental function (characterized by γ_l), we refined the value of γ_l in order to obtain the best agreement between the calculated half-width at $I_o(I_{\min}/I_o)^{1/2}$ and γ^{obs} . The N₂-broadening coefficients $\gamma_{No_2-N_2}^0$ measured in this work are gathered in Table II. It appears that the calculated values³ are, on the average, 12% lower than the measured values but a slight decrease in the widths with N is predicted in the calculations. It has not been possible to detect any noticeable difference between the widths of the two spin components for a given transition. Since our set of measured values is limited to a few lines, no general statement about the variation with the quantum numbers of the N₂-broadening coefficients can be made based on these data.

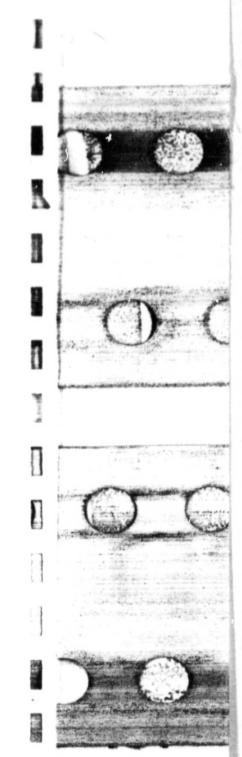
We have attempted to determine the self-broadening coefficients for the same lines using the method described above; but this determination much more difficult because of the dimenization of NO2 into N2O4. Under these conditions, the quantity actually deduced from the experiment is $\gamma_L = \gamma_{NO_2-NO_2}^0 P_{NO_2}$ + $\gamma_{NO_2-N_2O_4}^0 P_{N_2O_4}$. Because the only measured pressure is the total pressure P, we used the equilibrium constant $k_{\rho}(T)$ (8) of the reaction $2NO_2 \implies N_2O_4$ in order to obtain the values of (P_{NO_2}) and $(P_{N_2O_4})$ and thus any error on $k_P(T)$ would produce errors on these two pressures. Moreover, in our experiment, the maximum pressure of NC_2 (P_{NO_2}) was 30 Torr, corresponding to a maximum pressure of N_2O_4 ($P_{N_2O_4}$) of 10 Torr; consequently, if one assumes that $\gamma_{NO_2-NO_2}^0/\gamma_{NO_2-N_2O_4}^0$ \geq 3, the contribution of $(\gamma_{NO_2-N_2O_4}^{0}P_{N_2O_4})$ to γ_L would be less than 10%; that is the order of magnitude of the experimental error on γ_L . For all these reasons it has only been possible to determine an average value $\gamma_{\text{NO}_2-\text{NO}_2}^0 = 0.112 \pm 0.013 \text{ cm}^{-1}$ from the seven lines we have studied. Some experiments which have been performed at a temperature of 45° C, where the influence of N_2O_4 is negligible, confirm the value of $\gamma_{NO_2-NO_2}^0$ that we have found.

V. CALCULATION OF LINE INTENSITIES

The intensity of a line in (cm⁻¹/molecule) cm⁻² can be written as

$$k_{\sigma}^{N} = \frac{8\pi^{3}}{3hc4\pi\epsilon_{0}} \left(1 - \exp\left(-\frac{hc\sigma}{kT}\right)\right) \frac{g_{A}}{Z(T)} \exp\left(-\frac{E_{A}}{kT}\right) R_{A}^{B}$$
(3)

³ G. D. T. Tejwani and E. S. Young, Iowa State University, Ames (unpublished report (1975)).



with

$$R_A^B = \sum_{a} \sum_{b} 3 |\langle a | \mu_Z' | b \rangle|^2,$$

where

-A and B are respectively the lower and upper levels of the transition,

 $-\sigma = (E_B - E_A)/hc$ is the wavenumber of the transition,

-Z(T) is the partition function at the temperature T,

 $-g_A$ is the degeneracy due to the nuclear spin of the level A,

 $-|a\rangle$ and $|b\rangle$ represent single states corresponding to the M_J degeneracy of the levels A and B.

To calculate the vibration-rotation energies (5) we have used a transformed Hamiltonian and determined its eigenstates $|v_1v_2v_3NK_aK_c\rangle$. For this reason, although the transformed dipole moment μ_Z is more complicated than the non-transformed dipole moment μ_Z , we calculated the matrix elements of μ_Z instead of those of μ_Z .

For NO₂, the single states $|a\rangle$ and $|b\rangle$ are the $|v_1v_2v_3NK_aK_cSJM_J\rangle$ states which can be expanded as

 $|v_1v_2v_3NK_aK_cSJM_J\rangle$

$$= \sum_{M_S + M_N = M_J} \langle NM_N SM_S | JM_J \rangle | v_1 v_2 v_3 NK_a K_c M_N \rangle | SM_S \rangle. \quad (4)$$

From

$$R_{B}^{A} = 3 \sum_{k_{\perp}} |\langle v_{1}v_{2}v_{3}NK_{a}K_{c}SJM_{J}| \mu_{Z}' |v'_{1}v'_{2}v'_{3}N'K'_{a}K'_{c}SJ'M_{J}\rangle|^{2}$$
 (5)

after some calculations, one obtains

$$R_{A}^{B} = (2J + 1)(2J' + 1) \begin{cases} J' & 1 & J \\ N & S & N' \end{cases}^{2} \left| \langle v_{1}v_{2}v_{3}NK_{a}K_{c} | \mu_{Z}' | v_{1}'v_{2}'v_{3}'N'K_{a}'K_{c}' \rangle \right|^{2}, \quad (6)$$

where { } represents the usual 6 j coefficient.

Under these conditions, the transition strength R_A^B is the product of the spin-free transition strength $|\langle v_1v_2v_3NK_aK_c|\mu_Z'|v_1'v_2'v_3'N'K_a'K_c'\rangle|^2$ times the relative intensity g(N,J,N',J') of the spin-rotation components (allowed or forbidden). Using the relation

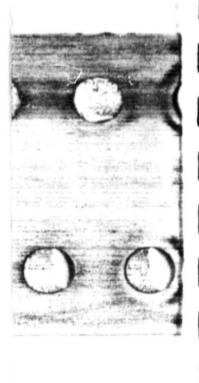
$$\sum_{\alpha} (2\alpha + 1) \begin{Bmatrix} a & b & \alpha \\ c & d & f \end{Bmatrix} \begin{Bmatrix} a & b & \alpha \\ c & d & g \end{Bmatrix} = \frac{1}{2f + 1} \delta_{fg}, \tag{7}$$

it is easy to deduce that $\sum_{J,J'} g(N,J,N',J') = 2S + 1$; consequently, for a spin 1/2, one has $\sum_{J,J'} g(N,J,N',J') = 2$. The different values of the coefficients g(N,J,N',J') for the different possible transitions are gathered in Table III.

To evaluate the matrix element of μ'_z , we will follow the general theory developed in Ref. (11) and we recall here only the relevant notations and formulas.

The wavefunctions $|v_1v_2v_3NK_aK_c\rangle$ have been obtained as a byproduct of the diagonalization of Watson's Hamiltonian for the (000) and the (010) vibrational states (5). One has

$$\left| v_1 v_2 v_3 N K_a K_c \right\rangle \; = \; \left| v \right\rangle \; \sum_K \, C_K^{\, \mathrm{r}} \left| N K \gamma \right\rangle$$





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N *	Spin coaponent	21	ı	G(N, J, N', J')
N' = N - 1	ri - ri ri - r2	31 - N1+3 31 - N1-3 31 - N1+3	J - N-3	$ \frac{2(N+1)}{2N+1} $ $ \frac{2(N-1)}{2N-1} $ $ \frac{2}{(2N-1)(2N+1)} $
N' - N	Fi - F1 F2 - F2 F2 - F1 F1 - F2	31 = M1+会 31 = M1-会 31 = M1-会 31 = M1+会	J = N-ģ	$\frac{2N(2N+3)}{(2N+1)^{\frac{2}{2}}}$ $\frac{2(N+1)(2N-1)}{(2N+1)^{\frac{2}{2}}}$ $\frac{2}{(2N+1)^{\frac{2}{2}}}$
N' = N+1	Fi - Fi Fi - F2 Fi - F2	31 = N1+2 31 = N1-2 31 = N1-2	J = N+2 J = N+2	2(N +2) 2N +3 2N +1 2N +1 2 (2N +1)(2N+3)

with

$$v = 0$$
 for (000), $v = 1$ for (010), (8)

where the rotational functions $|NK\gamma\rangle$ are the usual Wang functions. The general form of the transformed dipole moment operator is

$$\mu_{Z}' = \sum_{v,v'} |v\rangle^{vv'} \mu_{Z}' \langle v'|$$
 (9)

and its relevant part to be used for the ν_2 band is the transformed transition moment operator $^{01}\mu_Z'$ with

$${}^{01}\mu'_{Z} = {}^{01}\mu'_{1}\varphi_{x} + {}^{01}\mu'_{4}\{i\varphi_{y}, J_{z}\} + {}^{01}\mu'_{5}\{\varphi_{z}, iJ_{y}\} + \cdots,$$
 (10)

where the transition moment constants ${}^{01}\mu'_j$ are to be determined from measured intensities. In Eq. (10), we have kept only terms which proved to be significant in the least-squares fit but the general form of ${}^{rr'}\mu'_z$ as well as the matrix elements of the rotational operators involved in its expansion can be found in Ref. (11).

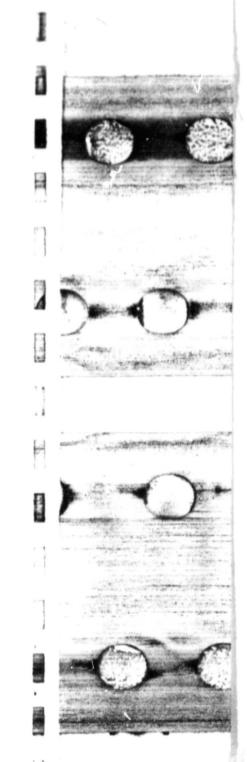


TABLE IV

Constants Involved in the Expansion of the Transformed Transition Moment Operator of the ν_2 Band of NO_2

Operator	Constant	Value in D
$\varphi_{\mathbf{k}}$	014;	(-0.427 ₂ = 0.026) x 10 ⁻¹
[10, 12]	014	(0.40 ₃ ± 0.25) × 10 ⁻³
(10,11 _y)	0145	(0.108 ₃ = 0.028) × 10 ⁻³

The errors quoted are 95% statistical confidence intervals

Using the 30 measured intensities as input parameters in a least-squares fit, we were able to deduce the values of the $^{01}\mu_j$'s quoted in Table IV. These constants were then used to compute the whole spectrum of the ν_2 band of NO₂ leading to a total band intensity (i.e., the sum of individual line strengths) of $S_r(\nu_2) = 0.542 \times 10^{-18} \text{ cm}^{-1}/\text{molecule cm}^{-2}$ at 296 K.

In fact, since the intensity of a line is proportional to the square of the matrix element of the dipole moment, from the least-squares fit of the measured intensities it is only possible to determine the relative signs of the coefficients $^{01}\mu'_{J}$. The choice actually made is discussed in the next section.

VI. THEORETICAL DISCUSSION

The theory of the contact transformation used to obtain the transformed dipole moment has been developed by Legay (12). For molecules with a permanent dipole moment, one has, to the second order of approximation,

$${}^{01}\mu'_{1} = \frac{1}{(2)^{1/2}} {}^{2}\mu_{x},$$

$${}^{01}\mu'_{4} = 2 \cos \gamma \left(\frac{A}{\omega_{2}}\right)^{3/2} {}^{0}\mu_{x},$$

$${}^{01}\mu'_{5} = \frac{C \zeta_{23}^{y}(2\omega_{2}\omega_{3})^{1/2}}{\omega_{3}^{2} - \omega_{2}^{2}} {}^{3}\mu_{\lambda} + 2\zeta_{13}^{y} \left(\frac{C}{\omega_{2}}\right)^{3/2} {}^{0}\mu_{x}.$$
(11)

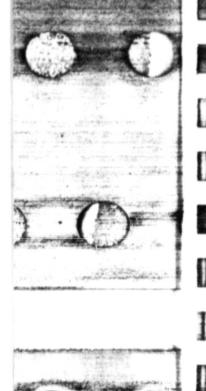
Using the equilibrium geometry and force constants of Ref. (13) (leading to $\cos \gamma = 0.767$ and $\sin \gamma = 0.642$) one obtains

$$^{01}\mu_{1}^{\prime} = \frac{1}{(2)^{1/2}} \,^{2}\mu_{x},\tag{12}$$

$$^{01}\mu_4' = 0.1648 \times 10^{-2} \,^{0}\mu_x,$$
 (13)

$$^{01}\mu_5' = -0.2656 \times 10^{-3} \, ^{3}\mu_z + 0.1151 \times 10^{-4} \, ^{6}\mu_x.$$
 (14)

Assuming that the sign of ${}^{0}\mu_{x}$ corresponds, as shown by ab initio calculations (14),



to the polarity $N^+ - O_2^-$ and that its value is not very different from the ground-state dipole moment $\mu_x(000)$ given in Ref. (15), we can take the value $^0\mu_x = 0.316$ D.

Then, Eq. (13) implies that ${}^{01}\mu_4$ must be positive. As we already said since the relative signs of the constant ${}^{01}\mu'_3$ are determined from the least-squares fit, the signs of these constants with respect to the sign of ${}^{0}\mu_x$ are then fixed.

Moreover, from Eq. (13) we obtain for $^{01}\mu'_4$ the calculated value 0.574×10^{-3} D, which compares well with the value $(0.403 \pm 0.25) \times 10^{-3}$ D we have obtained. Using Eq. (12), one gets

$$^{2}\mu_{x} = (-0.604_{1} \pm 0.037) \times 10^{-1} \text{ D}.$$

Finally, Eq. (14) allows us to deduce from $^{01}\mu'_5$ the value

$$^{3}\mu_{z} = -0.39_{4} \pm 0.11 \text{ D}.$$

This value is to be compared to the value of ${}^3\mu_z$ deduced from the intensity of the ν_3 band. From the two values 1430 \pm 300 cm⁻² atm⁻¹ at 40°C (2) and 2059 cm⁻² atm⁻¹ at 25°C (1), assuming that the ν_3 band can be treated in the rigid rotor approximation, one obtains

$$^3\mu_z = -0.711 \pm 0.043 \text{ D}$$

and

$$^{3}\mu_{z} = -0.483 \text{ D}.$$

It is of interest to note that the measurements of Goldman *et al.* (2) provide a better estimate of the ν_3 band intensity because they lead to a value of ${}^3\mu_z$ comparable to the present work.

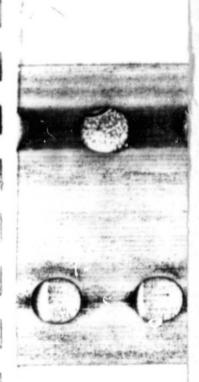
ACKNOWLEDGMENTS

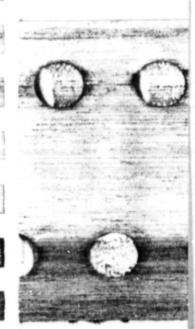
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Diode Laser Spectra of the ν_2 Band of $^{14}N^{16}O_2$: The (010) State of NO_2

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Diode laser spectra of the ν_2 band of NO₂ are recorded and analyzed. Due to the very high resolution (0.002 cm⁻¹), the Q branches are completely resolved. Although the whole ν_2 band is not completely covered, the quality of the spectra led to an improved set of rotational and spin-rotation constants. The band center is found to be $\nu_2 = 749.6541 \pm 0.0012$ cm⁻¹.

INTRODUCTION

The importance of nitrogen dioxide in the photochemistry of the atmosphere has stimulated many high-resolution studies of the infrared spectra of this molecule. Moreover, a good knowledge of the lowest vibrorotational levels of the electronic ground state is very helpful to assign visible fluorescence or absorption spectra involving higher vibronic states.

This paper reports the analysis of diode laser spectra of the ν_2 band of $^{14}N^{16}O_2$. Previous studies of this band using spectra recorded on grating spectrometers (resolution =0.040 cm⁻¹) have been performed by Hurlock *et al.* (1) and by Cabana *et al.* (2). The resolution of the diode laser spectra (=0.002 cm⁻¹) allows a much better analysis, particularly in the Q branches, which are now completely resolved. However, it has not been possible with the diode used in this study to cover continuously the whole extent of the ν_2 band. Nevertheless, the positions of about 380 lines were precisely determined and from them, improved band center, rotational, and spin-rotation constants for the (010) vibrational state were obtained.

EXPERIMENTAL DETAILS

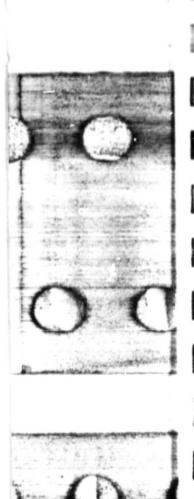
The experimental setup used in this study has been described elsewhere (3). Two diode lasers operating between 687-720 and 820-865 cm⁻¹ were used. Only

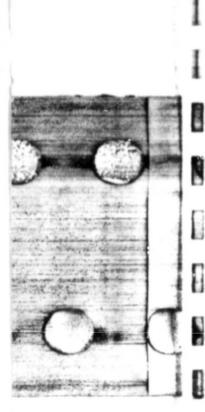
¹ Financial support from the University Pierre and Marie Curie (Paris) is acknowledged.

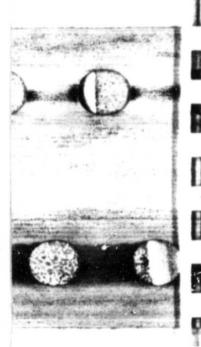
² The support extended this research by the National Aeronautics and Space Administration is gratefully acknowledged.

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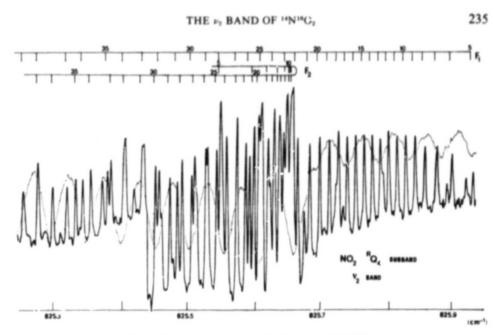


Fig. 1. Example of a resolved Q branch of 14N16O2

those modes were scanned where suitable calibration lines were available. In the $14-\mu m$ region ν_2 and $\nu_1 - \nu_2$ band lines (4) of $^{12}C^{16}O_2$, ν_2 band lines of $H^{12}C^{14}N$ (3) and in the $12-\mu m$ region ν_1 band lines of OCS (5) were used for determining the absolute positions of the lines. A commercial sample of NO_2 , at pressures ranging from 0.5 to 1 Torr, was contained in a multipass cell with a total path length of 12 m. Some impurity lines, especially CO_2 and HNO_3 , were observed. A 33-cm-lcng air-spaced etalon with 0.015-cm⁻¹ fringe spacing was used in the $14-\mu m$ region while a 2.54-cm (1-in)-long germanium etalon with 0.049-cm⁻¹ fringe separation was employed in the $12-\mu m$ region for producing wavenumber scales on the spectra. The relative accuracy of our measurements was estimated to be about ± 0.0005 cm⁻¹ and the absolute accuracy is believed to be about ± 0.002 cm⁻¹.

To show the quality of the spectra, we present a reproduction of the ${}^{R}Q_{4}$ branch around 825.500 cm⁻¹ (Fig. 1). On the high wavenumber edge of this region, the lines corresponding to the F_{1} component appear clearly whereas around 825.650 cm⁻¹ they are superimposed on the lines corresponding to the F_{2} component. On the low wavenumber eage of the region, the F_{1} and F_{2} series are again relatively well separated. At high K_{a} , near the origins of the Q branches the spin splitting is expected to be very large. This can be clearly seen in the spectrum of ${}^{R}Q_{5}$ displayed in Fig. 2. For example, for N=20, the splitting between the F_{1} and F_{2} components is as much as ~ 0.1 cm⁻¹. The observation of such splittings helped us to determine the spin-rotation interaction constants more precisely than before.

DATA ANALYSIS AND RESULTS

 NO_2 is an asymmetric top molecule belonging to the C_{2v} group with the complication that the spin-rotation interaction must be considered. Under these

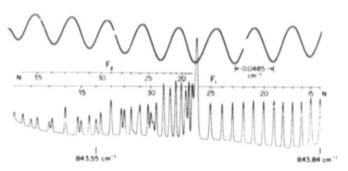


Fig. 2. ${}^{R}Q_{5}$ subband of ν_{2} of ${}^{14}N^{16}O_{2}$.

conditions, the energy of a level can be expressed as the sum of a rotational energy E_r , and of a contribution due to the spin-rotation interaction. Treating this interaction up to second order of perturbation (2, 6, 7) one obtains:

$$-\text{For } J = N + 1/2$$

$$F_1 = \frac{E_r}{hc} + \frac{1}{2(N+1)} \left[(\epsilon_{aa} - \tilde{\epsilon}) K_a^2 + \tilde{\epsilon} N(N+1) + \gamma \frac{(\epsilon_{bb} - \epsilon_{cc})}{4} \delta_{1,K_a} N(N+1) + \eta_{aaaa} K_a^4 - \frac{(\epsilon_{aa} - \tilde{\epsilon})^2}{4\tilde{B}} K_a^2 \left(1 - \frac{K_a^2}{(N+1)^2} \right) \right],$$

$$-\text{For } J = N - 1/2$$

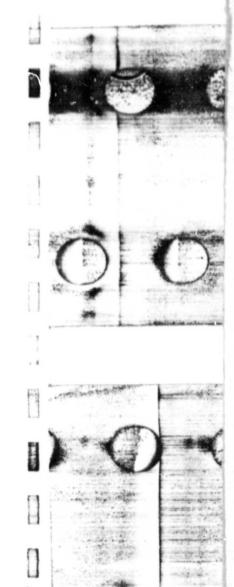
$$\begin{split} F_2 &= \frac{E_r}{hc} - \frac{1}{2N} \bigg[(\epsilon_{aa} - \tilde{\epsilon}) K_a^2 + \tilde{\epsilon} N(N+1) + \gamma \frac{(\epsilon_{bb} - \epsilon_{cc})}{4} \, \delta_{1,K_a} N(N+1) \\ &+ \gamma_{aaaa} K_a^4 - \frac{(\epsilon_{aa} - \tilde{\epsilon})^2}{4\tilde{B}} \, K_a^2 \bigg(1 - \frac{K_a^2}{N^2} \bigg) \bigg] \;, \end{split}$$

In these equations $\tilde{\epsilon} = (1/2)(\epsilon_{bb} + \epsilon_{cc})$, $\tilde{B} = (1/2)(B + C)$, and

$$\gamma = -1$$
 for N odd,
 $\gamma = +1$ for N even.

The rotational energy E_r is calculated by diagonalizing a Watson-type Hamiltonian (8)

$$\begin{split} H_{vv} &= \frac{E_v}{hc} + [A^v - (1/2)(B^v + C^v)]P_x^2 + (1/2)(B^v + C^v)P^2 + (1/2)(B^v - C^v)P_{xy}^2 \\ &= \Delta_K^v P_z^4 - \Delta_{NK}^v P_z^2 P^2 - \Delta_N^v (P^2)^2 - \delta_K^v \{P_z^2, P_{xy}^2\} - 2\delta_N^v P_{xy}^2 P^2 + H_K^v P_z^6 \\ &+ H_{KN}^v P_z^4 P^2 + H_{NK}^v P_z^2 (P^2)^2 + H_N^v (P^2)^3 + h_K^v \{P_z^4, P_{xy}^2\} \\ &+ h_{NK}^v \{P_z^1, P_{xy}^2\} P^2 + 2h_N^v P_{xy}^2 (P^2)^2 + L_K^v P_z^8, \end{split}$$
 where $P_{xy}^2 = P_x^2 - P_y^2, \, \{A, B\} = AB + BA, \, \text{and} \, (P_x \mp i P_y) |NK\rangle = (N(N+1) - K(K\pm1))^{1/2} |NK\pm1\rangle. \end{split}$



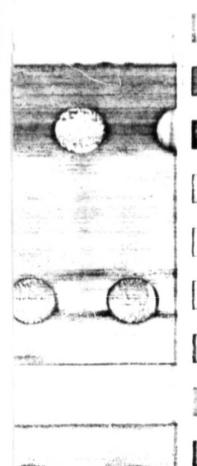
THE V2 BAND OF 14N16O2

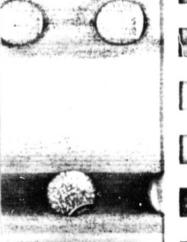
TABLE I Rotational Constants (in cm $^{-1}$) of $^{14}N^{16}\mathrm{O}_2$

	(000) ^a	(010)
E _v		749.654 ₁ ± 0.0012 ^b
A ^V	8.0023657	8.3736611 : 0.00010
в	0.43370797	0.4336271 ₅ ± 0.0000067
c*	0.41044536	0.40957705 ± 0.0000050
△ _K	0.26881 x 10 ⁻²	(0.34139 ₅ ± 0.00042) × 16"
NK.	-0.19499 x 10 ⁻⁴	$(-0.2160_5 \pm 0.0010) \times 10^{-4}$
△N	0.0068 x 10 ⁻⁶	$(0.3071_6 \pm 0.0028) \times 10^{-6}$
δĸ	0.413×10^{-5}	$(0.46_3 \pm 0.40) \times 10^{-5}$
δN	0.3179 × 10 ⁻⁷	$(0.308_2 + 0.022) \times 10^{-7}$
н <mark>v</mark>	0.2946 × 10 ⁻⁵	$(0.3948_2 + 0.0053) \times 10^{-5}$
H _{KN}	-0.197 × 10 ⁻⁷	$(-0.322_0 + 0.022) \times 10^{-7}$
H _{NK}	-0.370 x 10 ⁻⁹	
H _N	0.53×10^{-12}	0.53 x 10 ^{-12^c}
h _K	-0.78 × 10 ⁻⁷	
h _N	0.22×10^{-12}	
L _K ^v	-0.355 × 10 ⁻⁸	

 $^{^{\}mathbf{a}}\mathbf{G}\mathbf{round}$ state constants taken from Lafferty and Sams (9).

As a starting point in the analysis of the diode laser spectra, we computed a theoretical spectrum of the ν_2 band of NO₂ using the rotational and spin-rotation constants of the (010) and (000) states given in Refs. (2, 9, 10). With this calculation, we assigned the most intense lines lying in the available spectral intervals. The upper-state energies deduced from these transitions and from the ground-state energy levels calculated with the constants of Ref. (9) (see Tables I and II) were introduced in a nonlinear least-squares fit to obtain refined molecular constants for the (010) vibrational state. We repeated this procedure up to the point where all the NO₂ lines were assigned and the set of constants determined from the final fit is given in Table I for the rotational constants and in Table II for the spin-rotation





bUncertainties cited are one standard deviation.

^CFixed to the ground state value.

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TABLE II
Spin-Rotation Constants (in cm⁻¹) of ¹⁴N¹⁶O₉

	(000)	(010)						
$\epsilon_{aa} - \frac{1}{2} (\epsilon_{bb} + r_{cc})$	0.181644 ^a	0.199308	±0.00070°					
1/2(* bb + * cc)	-0.0014608 ^a	-0.00151934	± 0.000011					
•bb - •cc	0.0034319 ^a	0.00384 ₀	±0.00022					
n _{aaaa}	-0.1.8 x 10 ^{-3b}	(-0.20014	±0.017)×10 ⁻³					

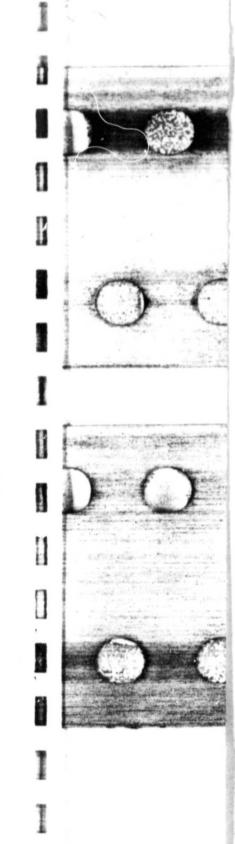
^{&#}x27;Taken from microwave work of Lees et al. [10].

constants. An improved band center was obtained: $\nu_2 = 749.6541 \pm 0.0012$ cm⁻¹. With this final set of constants, we calculated the whole spectrum of the ν_2 band and Table III presents the experimental and calculated wavenumbers together with the assignments of the lines.

DISCUSSION

A few comments are to be made about the assignment problem. First, only stretches of the whole band were recorded with the diode laser spectrometer. Except for the ${}^{R}Q_{4}$ subband which is completely covered (see Fig. 1), only parts of the ${}^{p}Q_{5}$, ${}^{p}Q_{4}$, ${}^{p}Q_{3}$ subbands as well as parts of the ${}^{R}Q_{5}$ and ${}^{R}Q_{6}$ subbands were available together with less crowded intervals. Second, some interferences with impurities (particularly HNO₃ and CO₂) complicated the assignment. However, because a good self-consistency in the calibration of wavenumbers was achieved, a very satisfactory fit was finally obtained (the unweighted standard deviation of 0.0018 cm⁻¹ is to be compared with the estimated precision of the position of the lines, i.e., 0.002 cm⁻¹). From the set of energy levels of (010) deduced in this work, it was not possible to determine the value of the octic constant L_{κ} . Indeed the highest K'_a value reached in our study is 7. However, our calculations reproduce very satisfactorily the positions of the ${}^{R}Q_{7}$ $(K'_{a}=8)$ and ${}^{R}Q_{8}$ $(K'_{a}=9)$ subbands observed at lower resolution in Ref. (1). Indeed, in the ${}^{R}Q_{7}$ subband, the calculated intensities peak for J = 17, 18, 19 at 880.367 cm⁻¹ in the F_1 component and at 880.199 in the F_2 component and these calculated positions compare well with the maxima observed in Ref. (1) at 880.339 and 880.186 cm⁻¹. The same agreement was obtained for the ${}^{R}Q_{8}$ subband: calculated maxima (J = 18, 19, 20) at 898.680 and 898.496 cm⁻¹ are to be compared with the observed ones at 898.604 and 898.455 cm⁻¹.

It is important to emphasize that both the rotational and spin-rotation constants of the (010) vibrational state are highly dependent on the parameters assumed



bTaken from Cabana et al. [2].

Cuncertainties cited are one standard deviation.

TABLE III List of the Observed ν_2 Lines (in vac.-cm $^{-1}$) of $^{14}N^{16}O_2$

σ_{obs}	σ_{calc}	N'K' K'	$NK^{9}K^{C}$	J-N	σ_{obs}	$\sigma_{\rm calc}$	M.K. K.	NK _a K _c	J-11
687.3857	.387	37 4 34	35 5 33	+1/2	700.3829	.384	7 3 5	7 4 4	-1/:
.3936	.395	38 4 34	38 5 33	-1/2	•4056	.406	6 3 3	6 4 2	-1/2 -1/2
.4188	.421	37 4 34	37 5 33	-1/2	•4341	.435	5 3 3	5 4 2	-1/2
.4260	.428	36 4 32	36 5 31	+1/2	.4781	.477	4 3 1	4 4 0	-1/2
-4533	•455	35 4 32	35 5 31	+1/2	.6399	.640	15 4 12	14 5 9	-1/2 -1/2 +1/2
.4601 .4883	•463 •491	36 4 32 35 4 32	36 5 31 35 5 31	-1/2	.6651	.665	14 2 12	15 3 13	+1/2
•4905	493	34 4 30	35 5 31 34 5 29	+1/2	.7089 .7386	.709	14 2 12 15 4 12	15 3 13 14 5 9	-1/2
.5189	.521	33 4 30	33 5 29	+1/2	. / 300	. /33	15 4 12	14 5 9	-1/2
.5237	.526	14 3 11	15 4 12	+1/2	706.4127	.414	22 4 18	21 5 17	+1/2
.5260	.529	34 4 30	34 5 29	-1/2	.4292	.432	36 5 31	35 6 30	+1/2
•5547	.556	32 4 28	32 5 27	+1/2	•4740	.477	36 5 31	35 6 30	-1/2
•5557 •5827	•559	33 4 30	32 5 29	-1/2	.4783 _b	• 479	22 4 18	21 5 17	-1/2
.5827	.583	14 3 11	15 4 12	-1/2	•5773	• 575	7 2 6	8 3 5	+1/2
•5936	•585 •595	31 4 28 32 4 28	31 5 27	+1/2		•577	48 0 48	49 1 49	-1/2
•6178	.618	30 4 26	32 5 27 30 5 25	-1/2	.6156	.615	48 0 48	49 1 49	+1/2
.6246	.625	31 4 28	31 5 27	-1/2	.6516 .9796	.653	7 2 6 8 3 5	8 3 5	-1/2
.6466	.646	29 4 26	29 5 25	-1/2 +1/2	707.1472	.148	8 3 5	7 4 4	+1/2
.6521	.654	13 5 9	12 6 6	49/2	1-1-1-12		0 3 3	/ 4 4	-1/2
.6574	.659	30 4 26	30 5 25	-1/2	712.7576	.759	25 2 24	25 3 23	+1/2
.6757	.677	28 4 24	28 5 23	+1/2	.7760	.759 .782	40 0 40	41 1 41	41/2
.6870	.689	29 4 26	29 5 25	-1/2	• 7907	• 790	25 2 24	25 3 23	-1/2
.7041	.705	27 4 24	27 5 23	+1/2	.8364	.838	15 3 13	14 4 10	+1/2
.7194	.720	28 4 24	28 5 23	-1/2	.8798	.884	23 2 22	23 3 21	-1/2 +1/2 +1/2
•7323	.733	26 4 22	26 5 21	+1/2	•9177	.916	15 3 13	14 4 10	-1/2 -1/2
• 7484 • 7590	.750	27 4 24 25 4 22	27 5 23	-1/2	.9195b	.917	23 2 22	23 3 21	-1/2
. 7590	.760	25 4 22	25 5 21	+1/2	.9876	.945	30 4 26 21 2 20	29 5 25	+1/2
000 0010	.819	24 2 20	04 4 00	/ .	.9908	.992	30 4 26	21 3 19 29 5 25	+1/2
.8530	.853	31 3 29 31 3 29	31 4 28 31 4 28	+1/2 -1/2 +1/2	713.0226	.025	21 2 20	21 3 19	-1/2
.8784	.876	29 3 27	29 4 26	+1/2	.0762b	.077	19 2 18	19 3 17	+1/2
.8959	.894	34 3 31	34 4 30	+1/2	·1172	.116	19 2 18	19 3 17	-1/2
*****	.891	36 3 33	36 4 32	41/2	.1477	.148	17 2 16	17 3 15	+1/2
.9005	.902	38 3 35	38 4 34	+1/2		.141	18 1 17	19 2 18	+1/2
.9071	.908	32 3 29	32 4 28	+1/2		.146	18 1 17	19 2 18	-1/2
.9120	.912	29 3 27	29 4 26	. 1/2	• 1934	.192	17 2 16	17 3 15	-1/2
.9170	.917	14 4 10	13 5 9	1/2	.2070 .2548	.205	15 2 14 15 2 14	15 3 13 15 3 13	+1/2
.9237	.920	34 3 31	34 4 30	-1/2	• 2540	.250	13 2 12	15 3 13 13 3 11	+1/2
.9311	.930	36 3 33 27 3 25	36 4 32 27 4 24	-1/2	.2840	.284	11 2 10	11 3 9	+1/2
. 9311	.930	30 3 27	30 4 26	+1/2 +1/2 -1/2	·2929b	. 291	3 2 2	3 3 1	+1/2
	.930	38 3 35	38 4 34	-1/2	• 3061 D	.307	13 2 12	13 3 11	-1/2
	.929	40 3 37	40 4 36	+1/2		.307	9 2 8	9 3 7	+1/2
.9391	.940	32 3 29	32 4 28	-1/2	• 3120	. 311	4 2 2	4 3 1	+1/2
• 9579	.955	40 3 37	40 4 36	-1/2 +1/2	.3207	.322	12 2 10	12 3 9	+1/2
.9310	.958	28 3 25	28 4 24	+1/2		.319	7 2 6	7 3 5	+1/2
.9646 .9677	.964	30 3 27 27 3 25	30 4 26	-1/2 -1/2 +1/2 +1/2	• 3245	.324	5 2 4	5 3 3	+1/2
•9777	.975	27 3 25 42 3 39	27 4 24 42 4 38	-1/2	• 364)	.325	6 2 4	6 3 3	+1/2
.9802	.979	25 3 23	25 4 22	+1/2	• 3274	. 327	14 2 12	14 3 11	+1/2
. 3002	.979	28 5 23	27 6 22	+1/2		.326	8 2 6	8 3 5	+1/2
.9922	.990	26 3 23	26 4 22	+1/2	.3423	. 342	16 2 14	16 3 13	+1/2
.9943	.994	28 3 25	28 4 24	-1/2	• 3498	.350	11 2 10	11 3 9	-1/2
700.0192	.019	25 3 23	25 4 22	-1/2	.3716	• 371	18 2 16	18 3 15	+1/2
.0261	.025	23 3 21	23 4 20	+1/2	. 3784	. 380	14 2 12	14 3 11	-1/2
	.025	24 3 21	24 4 20	+1/2	.3822	. 384	12 2 10	12 3 9	-1/2
.0284	.029	26 3 23	26 4 22	-1/2	.3868	.388	16 2 14 9 2 8	16 3 13	-1/2
.0356	.037	25 1 25	26 2 24	+1/2	.3958	.388	10 2 8	9 3 7	-1/2
	.033	42 1 41 28 5 23	43 2 42 27 6 22	-1/2	•4110	.413	18 2 16	18 3 15	-1/2
.0600b	.061	28 5 23	27 6 22 22 4 18	-1/2 +1/2	•4170	. 41	8 2 6	8 3 5	-1/2
.0687	.068	21 3 19	21 4 18	+1/2	•4230	419	20 2 18	20 3 17	41/2
/	.067	24 3 21	24 4 20	-1/2		.423	7 2 6	7 3 5	-1/2
	.069	23 3 21	23 4 20	-1/2	•4454	.445	6 2 4	6 3 3	-1/2
.0780	.779	25 1 25	26 2 24	-1/2	• 4550	.457	20 2 18	20 3 17	-1/2 -1/2 -1/2
.0966	.096	20 3 17	20 4 16	+1/2	•4632	.463	5 2 4	5 3 3	-1/2
.1074	.106	22 3 19	22 4 18	-1/2	•4946	.492	22 2 20	22 3 19	+1/2
.10/4						400	4 0 0		4
,1163	.107	19 3 17 21 3 19	19 4 16 21 4 18	+1/2		· 493	4 2 2	4 3 1 45 2 44	-1/2

Note. The different diode laser spectra are separated. The superscript b after an observed wavenumber means that the line is blended with an interfering line (CO_2, HNO_3) . When several lines are assigned to the same observed line, they are listed in the order of decreasing intensity.

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TABLE III—Continued

o _{obs}	$\sigma_{\rm calc}$	N, K,K,	NKa Kc	J-N	σ_{obs}	$\sigma_{\rm calc}$	N. K. K.	NK _a K _c	J-N
713.5219	•526	22 2 20	22 3 19	-1/2	825.5966	•595	26 5 21	26 4 22	+1/
•5411	•541	3 2 2	3 3 1	-1/2 +1/2	. 6024	.601	21 5 17	21 4 18	-1/
.3946	.592	24 2 22	24 3 21	+1/2		.599	6 5 1	6 4 2	-1/
.6209	.624	24 2 22	24 3 21	-1/2		.602	40 3 37	39 2 38	-1/
					.6101	.612	20 5 15	20 4 16	-1/
17.0916	.081	34 0 34	35 1 35	-1/2		.608	21 4 18	20 3 17	+1/
.1009	.107	34 0 34	35 1 35	+1/2	.6150	.614	25 5 21	25 4 22	+1/
.1928	.192	12 1 11	13 2 12	+1/2	.6247	.622	19 5 15	19 4 16	-1/
.2136	.213	12 1 11	13 2 12	-1/2		.624	7 5 3	7 4 4	-1/
2000	700					.621	40 3 37	39 2 38	+1/
117.8002	.799	21 3 19 36 4 32	20 4 16	+1/2	.6332	.631	18 5 13	18 4 14	-1/
- 341	.854	36 4 32 21 3 19	35 5 31 20 4 16	-1/2		.632	24 5 19	24 4 20	+1/
0.0 3441	.861	39 1 39	20 4 16	-1/2 $-1/2$.6422	.639	17 5 13	17 4 14	-1/
.9008	.901	37 1 37	32 2 36	+1/2	4 40 4	.639	8 5 3	8 4 4	-1/
.9497	.952	37 1 37	37 2 36	-1/2	.6486	.645	16 5 11	16 4 12	-1/
• 343/	. 376	3/ 1 3/	3/ 2 30	-1/2	.6528	.651	15 5 11	15 4 12	-1/
19.4295	.428	38 4 34	37 5 33	+1/2		.649	23 5 19	23 4 20	+1/
.4450	. 444	23 3 21	22 4 18	41/2	.6589	.655	9 5 5	9 4 6	-1/
. 4656	.465	36 4 34	37 5 33	-1/2	.0309	.655	10 5 5	10 4 6	-1/
.4946	. 494	23 3 21	22 4 18	-1/2	.6611	.658	13 5 9	13 4 10	-1/
.8151	.817	33 1 33	33 2 32	+1/2	.0011	.658	11 5 7	11 4 8	-1/
.8492	.844	30 0 30	31 1 31	-1/2		.659	12 5 7	12 4 8	-1/
	.851	37 1 37	38 0 38	+1/2	.6693	.665	22 5 17	22 4 18	-1/
.8614b	.866	30 0 30	31 1 31	+1/2	.6849	.681	21 5 17	21 4 18	+1/
.8646	.865	33 1 33	32 2 32	-1/2	.6970	.696	20 5 15	20 4 16	+1/
					.7112	.711	19 5 15	19 4 16	+1/
24.7863	. 789	20 4 16	19 3 17	+1/2	.7248	.725	18 5 13	18 4 14	+1/ +1/ +1/ +1/
25.0677	.071	21 6 16	22 5 17	-1/2	.7377	.738	17 5 13	17 4 14	+1/
.0921	.088	42 5 37	42 4 38	-1/2	.7507	.751	16 5 11	16 4 12	+1/
.1091	.109	43 5 39	43 4 40	+1/2	.7628	.764	15 5 11	15 4 12	+1/
.1276 _b	.126	42 5 37	42 4 38	+1/2	.7753	.776	14 5 9	14 4 10	+1/
• 1509	• 146	41 5 37	41 4 38	-1/2	.7883	.789	13 5 9	13 4 10	+1/
.1694	.166	40 5 35	40 4 36	-1/2	.8001	.801	12 5 7	12 4 8	+1/
.1792	.181	21 6 16	22 5 17	+1/2	.8126	.814	11 5 7	11 4 8	+1/
. 1864	•185	41 5 37	41 4 38	+1/2	.8263	.827	10 5 5	10 4 6	+1/
.2079	.206	40 5 35	40 4 36	+1/2	.8418	.842	9 5 5	9 4 6	+1/
.2170	.214	39 5 35	39 4 36	-1/2	.8577	.858	8 5 3	8 4 4	+1/
.2370	.236	38 5 33	38 4 34	-1/2	.8774	.877	7 5 3	7 4 4	+1/
• 2575	. 256	39 5 35	39 4 36	+1/2	.9007	.901	6 5 1	6 4 2	+1/
.2788	• 277	37 5 33	37 4 34	-1/2	.9225	.922	20 6 14	21 5 17	-1/
2000	. 279	38 5 33	38 4 34	+1/2	.9317	.933	5 5 1	5 4 2	+1/
. 3002	. 300	36 5 31	36 4 32	-1/2	826.0396	.038	20 6 14	21 5 17	+1/
· 3213 · 3348	. 321	37 5 33 35 5 31	37 4 34	+1/2	829.3173	.316	16 6 10	17 5 13	-1/
.3452	.334		35 4 32	-1/2	.4665	.465	16 6 10	17 5 13	+1/
. 3566	345357	36 5 31 34 5 29	36 4 32	+1/2	920 1000	4.00		4.4.0	4.4
.3808	.381		34 4 30	-1/2	830.1059	.105	5 5 1	4 4 0	+1/
.3861	. 38 7	35 5 31 33 5 29	35 4 32	+1/2 $-1/2$.1624 .3234	.162		16 5 11 16 5 11	-1/3 +1/3
.4056	.405	34 5 29	33 4 30 34 4 30	+1/2	. 4741	.472	15 6 10 27 4 24	16 5 11 26 3 23	-1/3
.4073	.408	34 5 29	32 4 38	+1/2	•5158	.515	27 4 24	26 3 23	+1/
.4342	.434	31 5 27	31 4 28	-1/2	• 51 50	1717	-/ 4 -4	20 3 23	+1/
.4363	.436	33 5 29	33 4 30	+1/2	830.9327	.934	6 5 1	5 4 2	+1/
.4543	. 454	30 5 25	30 4 26	-1/2	831.0068	.006	14 6 8	15 5 11	-1/
•4599	.46C	32 5 27	32 4 28	+1/2	.1790	.179	14 6 8	15 5 11	+1/
. 4763	.476	29 5 25	29 4 26	-1/2	.3292	.329	28 4 24	27 3 25	-1/
.4876	.487	31 5 27	31 4 28	+1/2	.3704	.370	28 4 24	27 3 25	+1/
. 4948	• 495	28 5 23	28 4 24	-1/2	•5718	.572	7 5 3	6 4 2	-1/
.5099	.509	30 5 25	30 4 26	+1/2	.7648	.764	7 5 3	6 4 2	+1/
.5143	.514	27 5 23	27 4 24	-1/2	.8506	.849	13 6 8	14 5 9	-1/
.5313	•531	26 5 21	26 4 22	-1/2	.9050	.905	46 3 43	45 2 44	-1/
.5336	•533	29 5 25	29 4 26	+1/2	.9196	.921	46 3 43	45 2 44	+1/
. 5475	•547	25 5 21	25 4 22	-1/2	832.0357	.037	13 6 8	14 5 9	+1/
.5538	.552	21 4 18	20 3 17	-1/2	832.0357b	.092	29 4 26	28 3 25	-1/
	•554	28 5 23	28 4 24	+1/2	.1330	.132	29 4 26	28 3 25	+1/
	.557	5 5 1	5 4 2	-1/2	•4233	.423	8 5 3	7 4 4	-1/2
.5625	.562	24 5 19	24 4 20	-1/2			- , ,	, , ,	/ .
•5775	.576	23 5 19	23 4 20	-1/2	832.8943	.895	12 6 6	135 9	+1/
			27 4 24	+1/2	.9698	.967	30 4 26	29 3 27	-1/3
	.576	27 5 23	2/4/24	+1/2	• 2020				+1/2

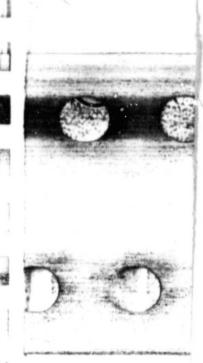
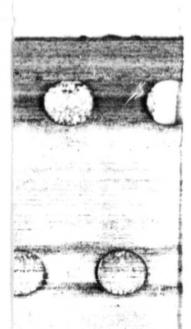
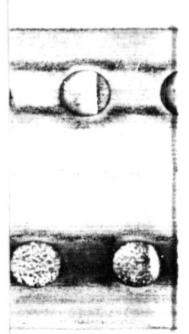




TABLE III - Continued

σ _{obs}	σ_{calc}	N. KIKI	NKa Lc	j =14	σ_{obs}	$\sigma_{\rm calc}$	N. K.K.	NKa Kc	J-11
833.0674 ^b	.074	33 7 27	34 6 28	-1/2	843.7827	.785	19 6 14	19 5 15	+1/2
.1547	.156	33 7 27	34 6 28	+1/2	•7943	•797	18 6 12	18 5 13	+1/2
.2725	.272	9 5 5	8 4 4	-1/2 +1/2	.8062	.808	17 6 12	17 5 13	+1/2
.4272	.428	9 5 5	8 4 4	+1/2	.8173b	.820	16 6 10	16 5 11	+1/2
.5280	.527	11 6 6	12 5 7	-1/2	·£ 399	.832 .843	15 6 10 14 5 8	15 5 11 14 5 9	+1/2
835.9300 _b	.928	12 5 7	11 4 8	+1/2	.6514	.855	13 6 8		+1/2
836.2339b	.232	34 4 30	33 3 31	-1/2	.8638	.868	12 6 6	13 5 9 12 5 7	+1/2
·2615	.266	34 4 30	33 3 31	+1/2	.8808	.881	11 6 6	11 5 7	+1/2
				,	.8955	.896	10 6 4	10 5 5	+1/2
836.6488	.648	13 5 9 13 5 9	12 4 8	-1/2	.9103	.912	9 6 4	9 5 5	+1/2
.7621	.761	13 5 9	12 4 8	+1/2	.9300	.931	8 6 2	8 5 3	+1/2
				. /-	•9534	•955	7 6 2	7 5 3	+1/2
837.4403	•439	28 7 21	29 6 24	+1/2	.9832	.986	6 6 0	6 5 1	+1/2
.4869	.488	14 5 9	13 4 10	-1/2	844.1706	.171	22 5 17	21 4 18	-1/2
•5942	.594	145 9	13 4 10	<1/2	0				
0.00	0.00	26 4 20	05 0 00	. /-	847.4858	.487	26 5 21 16 7 9	25 4 22 17 6 12	-1/2
837.8640	.860	36 4 32	35 3 33	-1/2	•5256	.524	16 7 9	17 6 12	-1/2
.8989	.892	36 4 32 27 7 21	35 3 33 28 6 22	+1/2	•5460	•546	26 5 21 16 7 9	25 4 22	+1/2
38.1936	.193	27 7 21	28 6 22	-1/2	•7106	.706	16 7 9	17 6 12	+1/2
39.1520	.150	26 7 19	27 6 22	+1/2	856.3734	.374	15 6 10	14 5 9	10
.1659	.165	16 5 11	15 4 12	-1/2	• 4981	.496	15 6 10	14 5 9	-1/2 +1/2
.2604	.259	16 5 11	15 4 12	+1/2	*4901	.496	15 6 10 37 5 33	36 4 32	-1/2
		,	17 4 16	4176	•5397	.537	37 5 33	36 4 32	-1/2 +1/2
839,9584	.961	39 4 36	38 3 35	-1/2	.,,,,,,		5, , 55	30 4 32	+1/2
	.990	39 4 36	38 3 35	+1/2	861.9155	.918	29 7 23	29 6 24	+1/2
40.0015	.002	17 5 13	16 4 12	-1/2	.9286	.931	28 7 21	28 6 22	+1/2
.0972	.091	17 5 13	16 4 12	-1/2	•9416	.943	27 7 21	27 6 22	+1/2
					•9534	.955	26 7 19	26 6 20	+1/2
843.5509	.550	34 6 28	34 5 29	+1/2	.9650	.967	25 7 19	25 6 20	+1/2
.5565	.557	30 6 24	30 5 25	-1/2	.9770	.979	24 7 17	24 6 18	+1/2
•5705	.572	29 6 24	29 5 25	-1/2	.9877	.990	23 7 17	23 6 18	+1/2
-0	.570	33 6 28	33 5 29	+1/2	862.0001	.001	22 7 15	22 6 16	+1/2
.5860	.586	28 6 22	28 5 23	-1/2	.0107	.012	21 7 15	21 6 16	+1/2
.5892	.590	32 6 26	32 5 27	+1/2	.0215	.023	20 7 13	20 6 14	+1/2
conab	.588	6 6 0	6 5 1	-1/2	.0318	.033	19 7 13	19 6 14	+1/2
.6027b	.599	27 6 22 26 6 20	27 5 23	-1/2	•0425	.044	18 7 11	18 6 12	+1/2
.6000	.608	26 6 20 31 6 26	26 5 21 31 5 27	-1/2	•0539	.054	17 7 11	17 6 12	+1/2
.6231	.622	25 6 20		+1/2	.0650 .0752	.065	16 7 9	16 6 10	+1/2
.6261	.626	30 6 24	25 5 21 30 5 25	+1/2	.0868	.087	15 7 9 14 7 7	15 6 10 14 6 8	+1/2
.0201	.626	7 6 2	7 5 3	-1/2	.1002	.099	13 7 7	13 6 8	+1/2
.6327	.633	24 6 18	24 5 19	-1/2	•1111	.112	12 7 5	12 6 6	+1/2
.6426	.643	23 6 18	23 5 19	-1/2	.1264	.127		11 6 6	+1/2
.0420	.643	29 6 24	29 5 25	+1/2	.1450	.143	11 7 5	10 6 4	+1/2
.6513	.652	22 6 16	22 5 17	-1/2	.1604	.162	9 7 3	9 6 4	+1/2
	.650	8 6 2	8 5 3	-1/2	.1767	.183	44 5 39	43 4 40	+1/2
.658€	.660	21 6 16	21 5 17	-1/2	.1822	.185	8 7 1	7 6 3	+1/2
	.660	28 6 22	28 5 23	+1/2	.2121	.214	7 7 1	762	+1/2
.6676	.667	20 6 14	20 5 15	-1/2					
	.666	9 6 4	9 5 5	-1/2					
.6744	.674	19 6 4	19 5 15	-1/2					
	.676	27 6 22	27 5 23	+1/2					
.6782	.679	18 6 12	18 5 13	-1/2					
	.677	10 6 4	10 5 5	-1/2					
.6858	.684	17 6 12	17 5 13	-1/2					
cn 0 -	.684	11 6 6	11 5 7	-1/2					
.6891	.687	16 6 10	16 5 11	-1/2					
	.688	12 6 6	12 5 7	-1/2					
	.690	15 6 10	15 5 11	-1/2					
	.690	13 6 8	13 5 9	-1/2					
	.691	14 6 8		-1/2					
.7050	•691 •706	26 6 20 25 6 20		+1/2					
.7050	.706			+1/2					
.7322	720	24 6 18 23 6 18	24 5 19	+1/2					
•7322	·734			+1/2					
•7453	.747	22 6 16 21 6 16	22 5 17 21 5 17	+1/2					
• 7704		20 6 14		+1/2					
. / / 04	•773	20 0 14	20 5 15	+1/2					





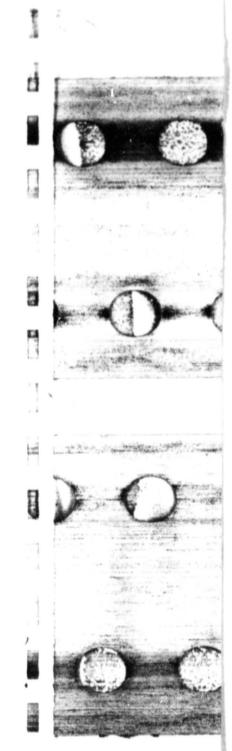
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for the (900) state. This is because the observed energy levels of (010) used as input in the fit were obtained by adding the observed line positions to the calculated energy levels of the ground state.

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Spectrum of Water Vapor between 8050 and 9370 cm⁻¹

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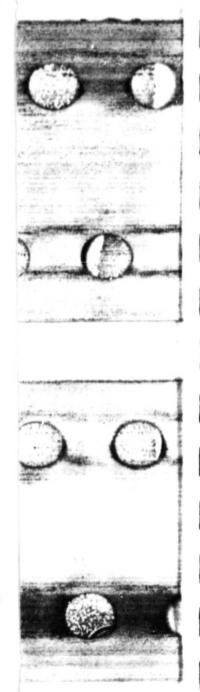
J.-P. MAILLARD

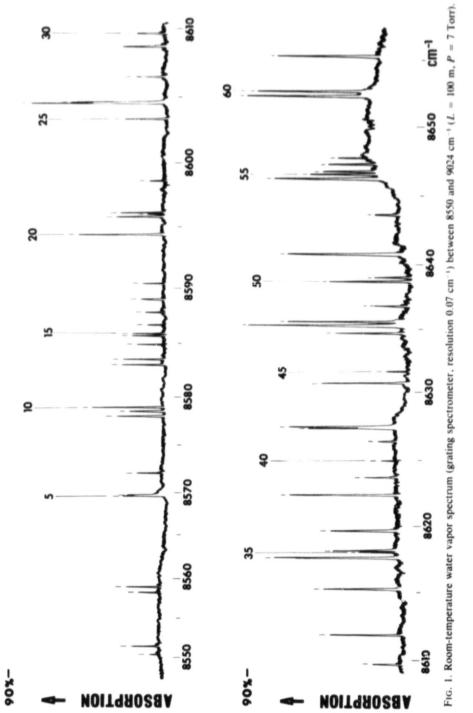
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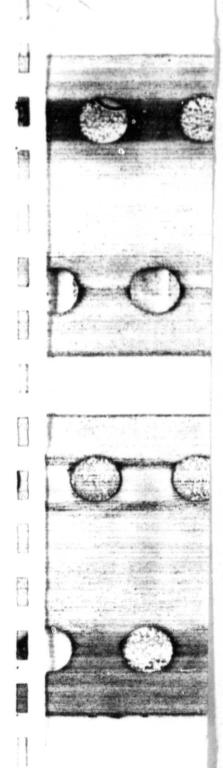
Measurements of the line positions of $H_2^{16}O$ in the 8050 to 9370 cm⁻¹ region have been performed at a spectral resolution of 0.07 cm⁻¹. A grating spectrum of room-temperature water vapor and a Fourier transform spectrum of heated water vapor (T = 60°C) were used in the interpretation. A careful analysis of the bands $\nu_1 + 3\nu_2$, $3\nu_2 + \nu_3$, $2\nu_1 + \nu_2$, $\nu_1 + \nu_2 + \nu_3$, and $\nu_2 + 2\nu_3$ has led to a set of rotational levels belonging to the vibrational states (130), (031), (210), (111), and (012). Many vibrorotational resonances were detected.

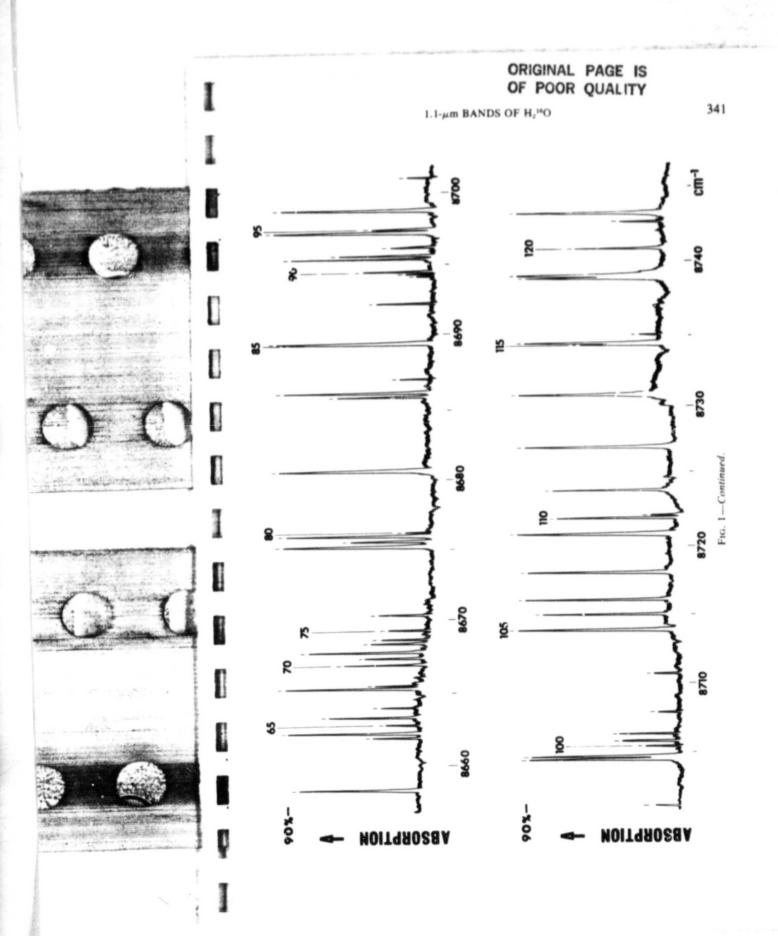
INTRODUCTION

The vibration-rotation spectrum of the water molecule begins to be understood in great detail in the 6.3- μ m (1-3), 2.7- μ m (4-6), 1.9- μ m (1,6,7), and 1.4- μ m (8,9) regions. Toward the photographic infrared and the visible, the situation is not so favorable because the bands are weak and also because of the possibility that multiple resonances will make both the assignment and the calculation of spectra much more difficult. However, it is important to improve our knowledge of these spectral regions from both practical considerations (atmospheric studies) and theoretical reasons (determination of the potential function). A systematic effort to analyze the H₂O spectra in these regions was performed earlier by Benedict (10). In the present paper, we report an experimental investigation of the spectrum of water vapor from 8050 to 9370 cm⁻¹; we have been able to assign about 1200 vibration-rotation lines belonging to the ν_1 + $3\nu_2$, $3\nu_2$ + ν_3 , $2\nu_1$ + ν_2 , ν_1 + ν_2 + ν_3 , and ν_2 + $2\nu_3$ bands. From the observed line positions, an extensive set of experimental rotational levels has been obtained for the vibrational states (130), (031), (210), (111), and (012).

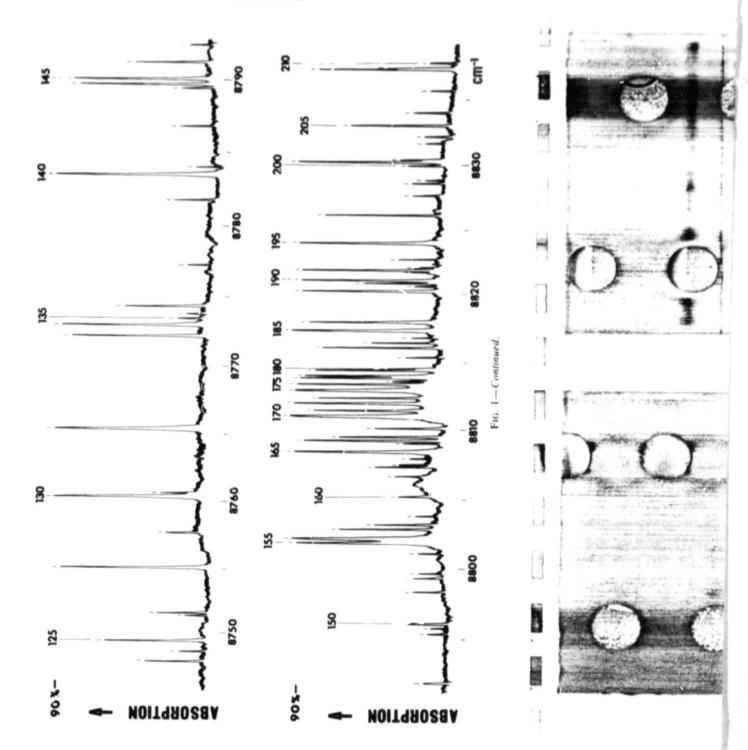




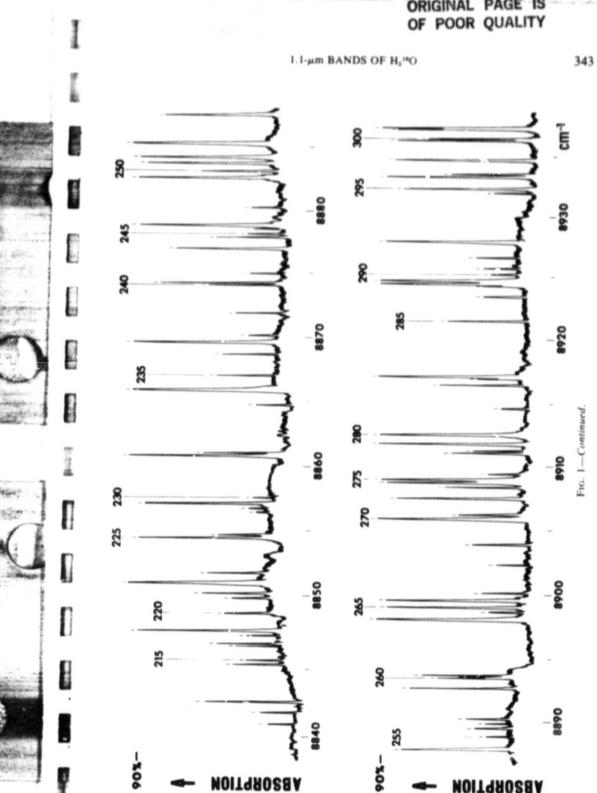


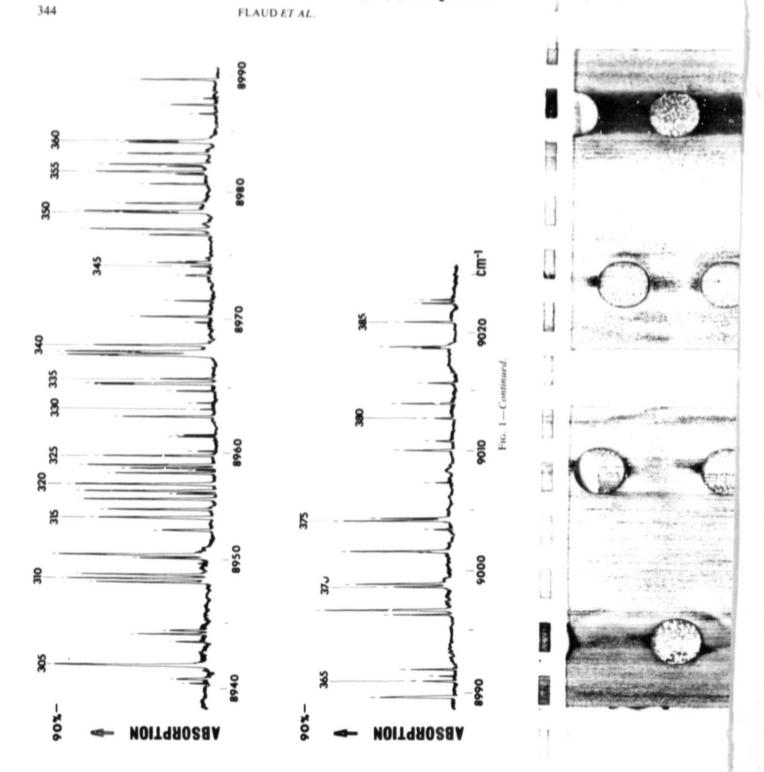


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EXPERIMENTAL DETAILS

For this study, we analyzed two different spectra. For the strongest lines, we used the line positions given by a room-temperature spectrum recorded from 8550 to 9024 cm⁻¹. The positions of the weaker lines were deduced from a Fourier transform spectrum of heated water vapor extending from 8050 to 9370 cm⁻¹. The characteristics of the two spectra are given below:

Room-Temperature Spectrum

A 10-m focal length Czerny-Turner vacuum spectrometer at the Ohio State University was used to record the spectrum. This spectrometer employed a 40 × 20-cm² (16 × 8-in.²) echeile with 79 grooves per millimeter on its surface and blazed at an angle of about 63°. The source of continuous infrared radiation was a carbon rod furnace working at an input power of 3.4 kW (10 V at 340 A). A description of this spectrometer was given previously (11). The infrared radiation was detected with two detectors—InSb for the CO calibrating lines and a liquid-nitrogen-cooled photomultiplier for the water vapor spectrum. The signal was amplified by a Princeton Applied Research Model HR-8 phase-sensitive amplifier and then recorded on a strip chart recorder. The resolution was about 0.07 cm⁻¹ and the precision of the positions of the lines is approximately =0.005 cm⁻¹. Figure 1 shows a reproduction of this spectrum. The water vapor partial pressure used was 7 mm Hg, and the length of the absorption path was 100 m.

Fourier Transform Spectrum

The water vapor was introduced into a heated multiple-path cell ($T \approx 60^{\circ}$ C, $P \approx 90$ Torr, total length L = 40 m) and the spectrum was recorded by one of us (J.-P.M.) with a resolution $\delta\sigma = 0.070$ cm⁻¹. Under these conditions, the experimental linewidth at half height was approximately 0.11 cm⁻¹, and the precision of the positions of the lines was ± 0.005 cm⁻¹. Figure 2 shows a reproduction of the observed spectrum.

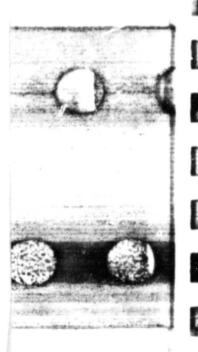
For unblended a⁻¹ unsaturated lines common to both spectra we have noticed that the line positions agree to within the stated experimental uncertainties, that is ± 0.005 cm⁻¹.

DATA ANALYSIS AND RESULTS

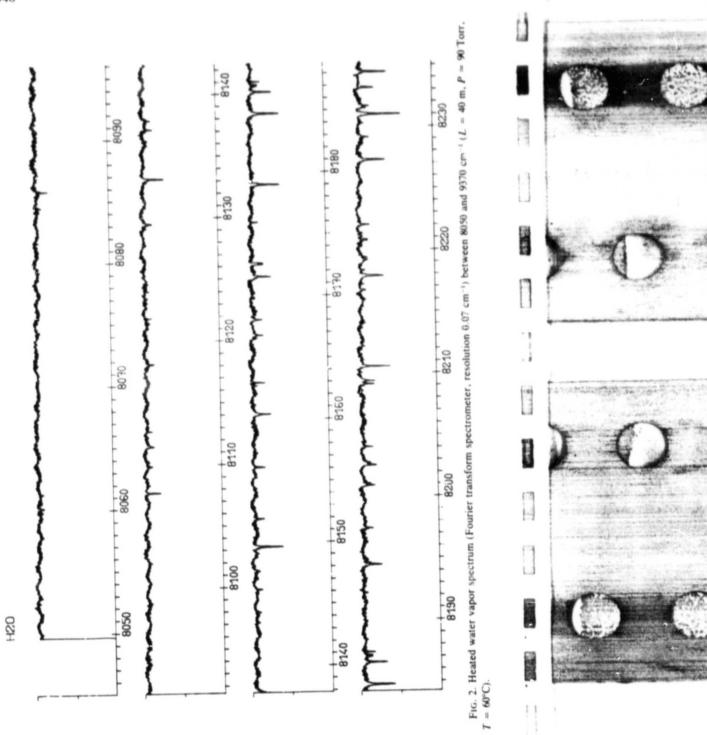
The identification of the lines of $H_2^{16}O$ was facilitated by the previous work of Benedict (10) and with the help of the available rotational energy levels for the (000) and (111) vibrational states of $H_2^{16}O$ (12,13).

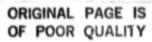
In the 1.1- μ m region, the A-type band $\nu_1 + \nu_2 + \nu_3$ is much stronger (see the strong Q branch at 8807 cm⁻¹) than the bands $\nu_1 + 3\nu_2$, $3\nu_2 + \nu_3$, $2\nu_1 + \nu_2$, and $\nu_2 + 2\nu_3$. Consequently, the number of levels determined for the (111) state is larger than the number determined for the (130), (031). (210), and (012) states. Also, since the $\nu_1 + \nu_2 + \nu_3$ band center (8807.000 cm⁻¹) is close to the $2\nu_1 + \nu_2$ band center (8761.579 cm⁻¹), the lines of this weak band are often blended with lines of the strong band, which complicated the interpretation.





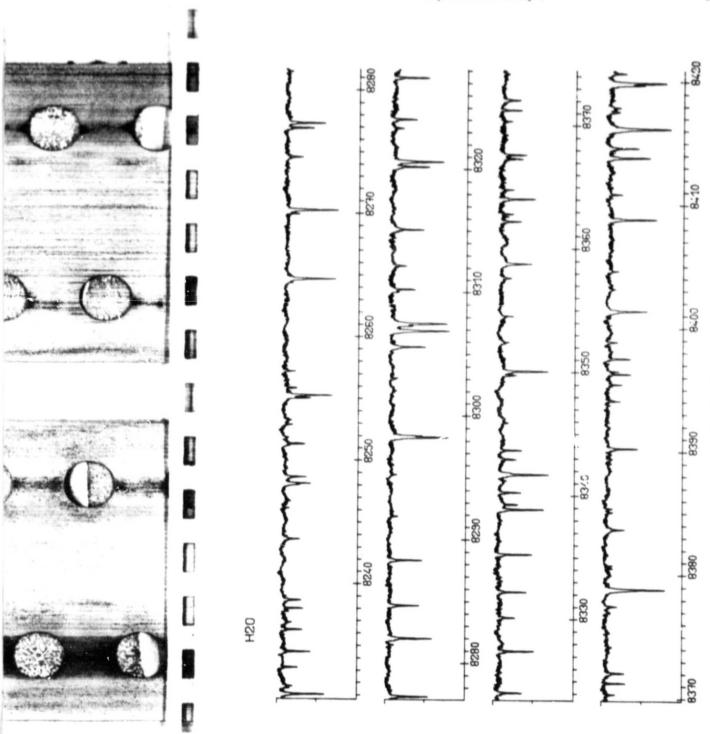
FLAUD ET AL.

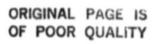


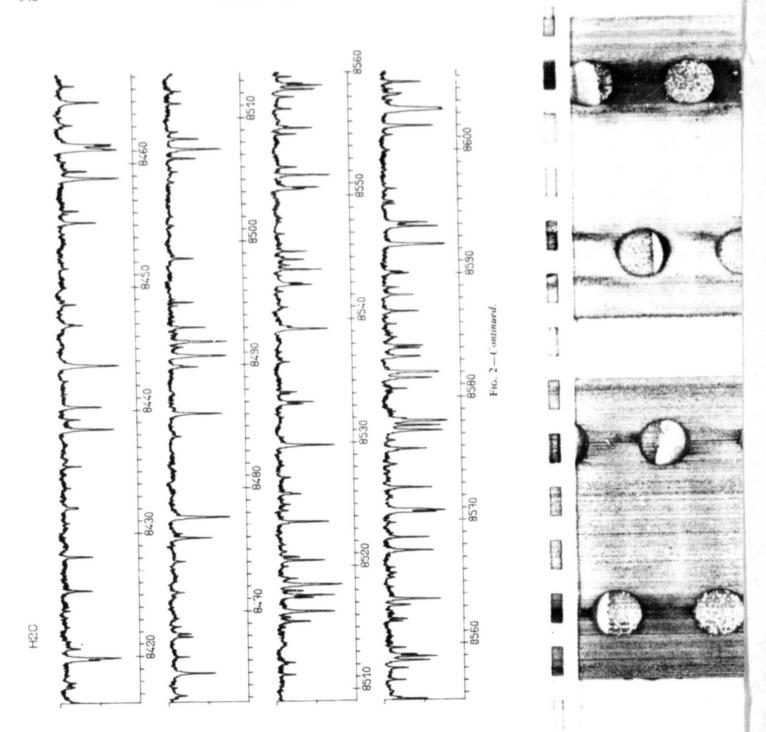


1.1-µm BANDS OF H₂16O

347

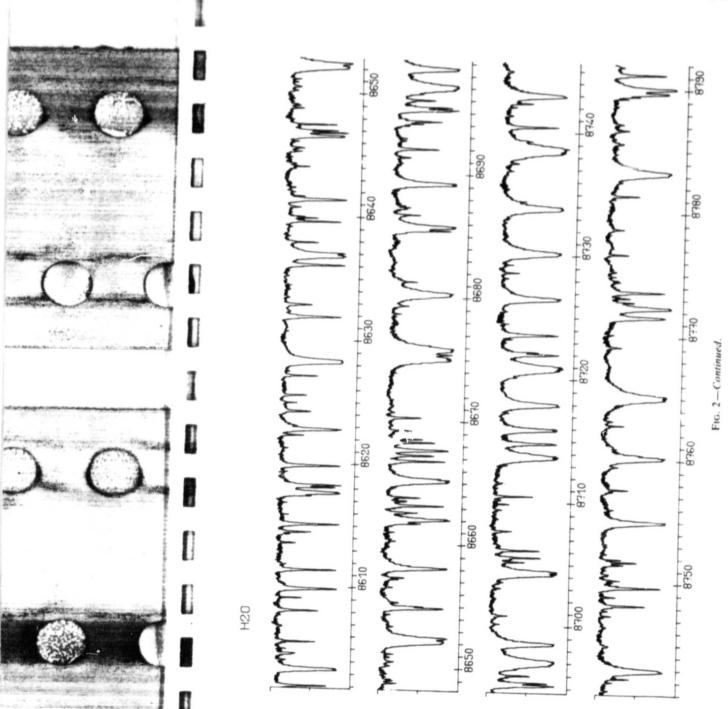


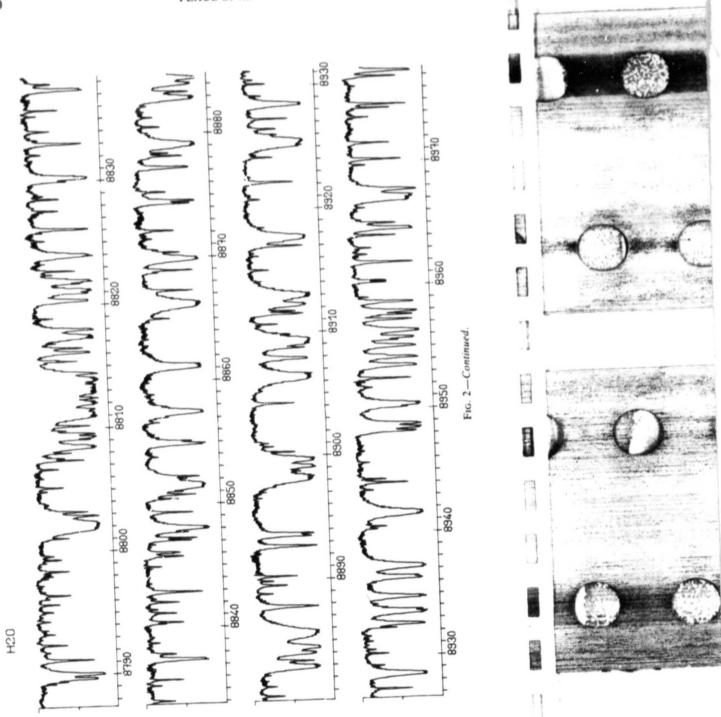




1.1-μm BANDS OF H₂¹⁶O

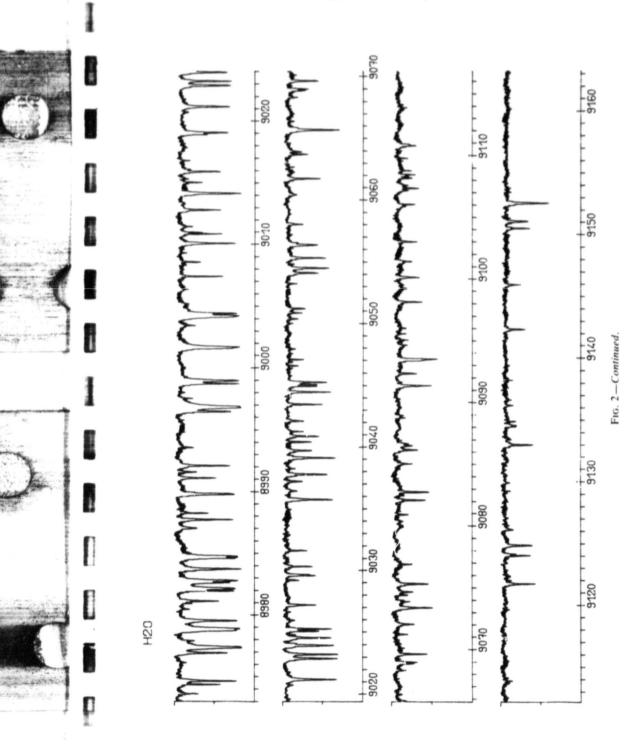
349

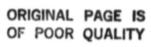




1.1-µm BANDS OF H₂16O

351





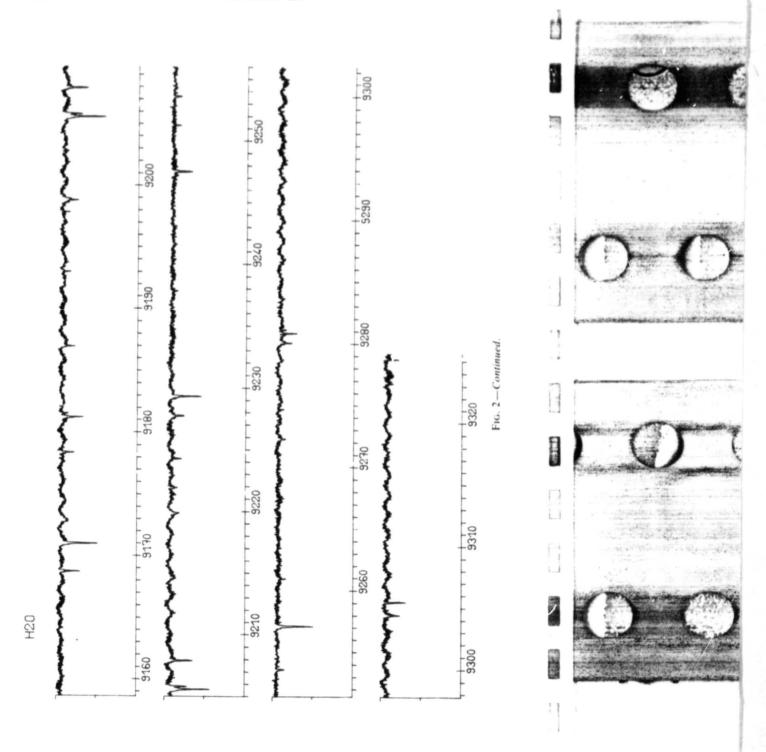


TABLE I

List of the Lines of Water Vapor between 8050 and 9370 cm^{-1 a}

_	<u> </u>	SIGMA	1.KW.KC.		PI	.5	G		٧.	*	SIGMA	J.KW.KC. J KW KC	-1	25		1
	0	8040.1451	: 0 :	10 110	ï		1 00	0	12	0	8200.9067	4 9 4 . 3 1 3	17		100	
8	ŏ	1062.1210		7 4 3	3		100	ő	10	0	5201.0501 8202.5530	111 111	. 1		100	
	0	9284.5096	6 2 4 6 0 8 6 1 8	9 1 9	•		1 00	ŏ	16	ŏ	8202.7505	8 2 6 9 2 7	12		100	
		4085.8634	9 1 2	9 9 9	3.1		1 00	9	15		8204.0090	111 111	13		100	
	0	8107.6925 9108.3160	7 0 7	9 2 8	19		100	0	10	9	6209.1095	111 111	17		100	
i	ŏ	8109.8407	7 1 7	* 6 8	- 2		1 00	ő	10	۰	8209.3604 8210.0465	707	19		100	
	ō.	7111.4434	4 2 3	5 3 2	10		1 00	ō	16		8210.5492	717	39		100	
	0	8112.1967	11 111	18 112	•		10	0	15	•	8212-1622	131 151			100	
	0	8112.6795 8118.1377	2 2 2	2 3 7	10		1 00	0	10		8216.7291		11		100	
	ŏ	8121.8469		1 1 1			100	0	15	2	6217.6633	101 212	*0		100	
	ě i	8124.0256	1 7 7	9 1 6	•		100	ŏ	15	ă	8215.1798	101 115			100	
	0	8127.8380	6 1 5	7 2 6			. 0	ō	15	ō	8219.3844	827 928	٠		100	
		4129-1353		2 7 2	. 2		10	0	15	0	6220.8081	4 2 3 5 1 4	13		100	
	0	8129.4951	6 1 6	7 1 7	32		100	0	16	9	8221 -0917				. 1	
		9137.1884	: 1 :	6 2 5	12		100	ő	16	ä	6222.0319	111 111	18		100	
	0 1	91 37.8942	10 010	11 011			100	ö	15	ŏ	8227.2969	212 303	37		100	
	0	6140.5570	3 2 2	: ? ?				0	15	۰	8229 - 1556	212 221	14		100	
		5143.4910	3 7 7		2		10	0			8230.2770	9 3 7 10 3 6			- 1	
		9140.1997			2		1 00	0	16	9	8231.0007	6 0 6 7 0 7	**		100	
		9187.0199	4 3 1	:::	- 3		100	ő	16	ŏ	8231.6626	2 0 2 3 2 1	10		100	
	0 1	8147.5560	4 1 3	9 2 4	2		10	ő	10	ŏ	8233.1343	616 717	2.7		100	
		81 49.0786		5 2 3	. 5		10	0	15	0	8234 5410	615 716	44		100	
		8149.6333	5 0 5	4 3 2	42		10	0	13	9	8234.4566	615 624			100	
		8151.9096	źîż	3 2 1	7.5		100	0	15	0	8236.2366	303 314	35		100	
	0 (5155.2790	4 1 3	5 3 2			· -ī	ö	15	ŏ	8237.7323	113 111	7		10	
		4156.1007	5 1 5	4 0 5	. 1		100	ě.	15	0	8237.9891	7 2 6 6 2 7	30		100	
		9160.4005	3 1 2	. 2 3	23		1 00	0	15	0	8238.6632	634 725	28		100	
		9161.1416		10 010	- 2		100	0	15	0	8243.5967		-2		100	
		1162.9312	9 1 9	10 110	16		100	ŏ	16	ŏ	6243.6306	7 3 4 8 3 5	5.5		100	
	0 1	9166.5175	6 1 6	6 2 5			10	0	16	0	8244.1766	734 635			iõ	
		1166.7729	6 2 5	7 1 6	12		100	0	18	•	6246.0951	211 220	7		- 1	
		1168.0421	2 2 1	3 3 0	27		100	0	15	0	8246 . 7045	3 5 1 3 3 0			!	
		9172.5373	5 5 0	9 9 1	19		1 00	ő	12	ŏ	8248.0105	202 211	37		100	
1	0 1	4172.6222	3 3 0		19		10	ŏ	16	ŏ	6246.1019	413 422	37		٠,	
		81 75.0359	211	3 2 2	10		10	0	15	0	6246.6071	3 1 2 3 2 1	32		100	
		8179.0216 8184.7649	111	5 0 5	45		1 00	0	16	0	8249.4038	8 3 6 9 3 7				
		5196.5422	3 0 3	9 0 9	30		100	0	16	0	(250.4520	111 119	. ?		0	
	ě i	91 97.0720		9 1 9	10		10	ŏ	15	ŏ	8251.2964	3 2 2 4 1 3	31		100	
		9197.1289	īiī	2 2 0			100	ō	15	ō	8252.8472	5 2 3 5 3 2	11		100	
	0 '	1147.3722	* * 1	5 5 0	1.7		10	0	15	ō	6257.2146	111 202			10	
		1197.3722	5 2 4	2 5 1	13			0	12	0	6255.0084	101 110	37		100	
		1194.4527	5 2 4	2 2 1	23		1 00	9	16	0	0251.2305	212 010	74		100	
	0 1	1194.8145	9 1 8	10 1 9	**			ő	16	ŏ	8255.8285	5 2 3 6 2 4	17		100	
	0 6	1137.4419	5 0 5	5 1 4	13		10	ŏ	16	ŏ	8257.1756	6 2 5 7 2 6	19		100	

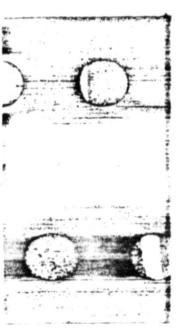
- * The columns mean:
 - v', v Labels of the upper and lower vibrational states of the transition with the correspondence:

$$0 = (000),$$
 $1 = (010),$ $20 = (060),$ $15 = (130),$ $16 = (031),$ $17 = (210),$ $18 = (111),$ $19 = (012),$ $24 = (121).$

Sigma Observed wavenumber of the transition (in cm⁻¹)

- J'K'aK'cJKaKc Rotational quantum numbers of the upper and lower levels
 - P_1, P_2 Percentages of absorption at the center of the line for the heated and room-temperature spectra
 - G Statistical weight of the line
 - I Isotopic species with the notation $0 = H_2^{16}O$, $1 = H_2^{17}O$, $2 = H_2^{18}O$





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TABLE I-Continued

٧.	V 51544	J.KW.KC. J KW KC MI	P2 G [Y' V SIGMA	J'KA'KC'J KA KC P1 P2 G [
10	0 4257.1460 0 4264.1324 0 8244.4485 0 9248.2340		100 0	16 0 8345-8825 16 0 9349-7551 16 0 9350-0540 15 0 8352-3907	6 4 3 7 4 4 9 100 0 3 1 3 3 1 2 21 10 0 0 0 0 1 0 1 76 100 0 4 0 4 3 1 3 10 100 0 3 1 3 2 0 2 19 100 0
4 4 6 5	0 4264.1940 0 4269.0060 0 4269.9243 0 4270.2154	4 3 1 4 4 0 2 7 3 5 6 3 6 17 4 0 4 5 0 5 61 5 2 3 6 2 4 5	0 0 0 0 1 00 0 1 00 0	15 0 8354-5115 15 0 8357-5961 15 0 8357-5961 15 0 8352-7588 8361-4660	4 2 2 4 1 3 14 10 0 4 1 3 3 2 2 14 1 0 3 2 1 3 1 2 46 1 0
16	0 5274.5200 0 5275.5598 0 5277.2556 5281.4498 0 5281.5061	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	100 0	15 0 8362.2005 15 0 8362.5975 16 0 8362.5069 16 0 8364.0108 8364.5500	5 2 3 5 1 4 30 100 0 2 2 0 2 1 1 7 100 0 3 3 0 4 3 1 22 100 0 3 3 1 4 3 2 52 100 0
16	0 4243.7940 0 4243.7940 0 4256.71/2	4 2 2 5 2 7 65 2 2 1 3 1 2 3 3 1 2 4 1 3 47 1 0 7 4 0 4 40 6 3 4 7 3 5 9	100 0 100 0 100 0	15 0 8364.7978 8365.3277 18 0 8366.3533 16 0 8367.2785 15 0 8367.6039	7 1 7 6 3 6 7 100 0 2 1 2 2 1 1 30 100 0 4 1 4 3 0 3 3 6 100 0
16	0 0290.2275 0 0290.2722 0 0290.3933 0 0303.0455 0 0305.5629	3 1 3 4 1 4 66 4 2 3 5 2 4 33 5 1 5 5 1 4 9 1 1 0 1 0 1 57	100 0	18 0 8369.2174 16 0 8370.3158 16 0 8371.2303 15 0 8372.0486 15 0 8374.1664	8 3 5 9 5 4 6 100 0 5 4 1 6 4 2 12 100 0 5 4 2 6 4 3 31 100 0 5 0 5 4 1 4 25 100 0 6 2 4 6 1 5 10 1 0
16	0 6305.5629 0 6306.0431 0 6307.3052 0 6307.4731	5 3 2 6 3 7 57 2 0 2 1 1 1 6 2 0 2 3 0 3 94 3 2 1 4 1 4 1 9 7 1 1 3 1 2 92	10 0 100 0 10 0	15 0 5377.4527 16 0 5378.8137 16 0 5360.5259 15 0 5360.7406 16 0 6363.6615	2 2 1 2 1 2 1 9 100 0 1 1 1 1 1 0 91 100 0 7 2 6 6 5 5 10 10 0 5 1 5 4 0 4 10 100 0 5 2 4 5 2 3 2 9 100 0
15	0 4310.0322 0 4310.2176 0 4312.1377 0 6313.4250 0 6314.1953	4 3 2 5 2 3 10 3 2 1 4 2 2 36 2 1 1 2 0 2 21 7 4 3 6 4 4 4 5 2 3 6 1 6 7	100 0	15 0 5365.0536 15 0 5365.3536 15 0 5369.3450 15 0 5369.7460 16 0 5390.2273	7 8 2 3 1 3 29 100 0 7 8 3 8 8 4 10 100 0 6 0 6 5 1 8 6 1 0 8 1 4 4 2 3 6 1 0 1 1 0 1 1 1 48 100 0
16	0 4120.141A 0 6120.5900 0 4123.2401	9 3 3 6 3 4 67 2 1 2 3 1 3 92 3 7 2 4 2 3 79 1 1 1 1 0 0 0 1 13 3 1 2 3 0 3 36	100 0 100 0 100 0	15 0 6394.0514 15 0 6394.4267 6395.6740 16 0 6394.2571	8 3 6 9 6 5 7 10 0 6 1 6 5 0 7 21 100 0 6 2 3 6 1 6 21 100 0
16	9 9329.0071 9 9329.0071 9 9330.1027	1 0 1 2 0 2 53 4 1 4 4 1 3 9 3 0 3 2 1 2 24 1 1 0 2 1 1 42 4 3 1 5 3 2 52	100 0 100 0 100 0 100 0	16 0 8396.4706 16 0 8397.4866 8398.0921 16 0 8398.8356 8400.2293 16 0 8401.3447	4 4 0 5 4 1 33 100 0 4 4 1 5 4 2 12 100 0 1 0 1 0 0 0 39 100 0 4 2 3 4 2 2 14 100 0
	4334.4675 4339.2737 4340.2577	2 2 0 3 2 1 66 4 3 2 9 3 3 24 2 1 2 1 0 1 34 7 2 6 7 2 9 6	1 00 0 1 00 0 1 00 0	15 0 8403.1271 8403.4784 15 0 8404.5999 15 0 8407.1648	2 1 1 2 1 2 59 100 0 4 2 2 5 0 5 10 100 0 7 0 7 6 1 6 1 4 100 0 7 1 7 6 0 6 5 0 0 5 2 4 5 1 5 6 100 0
15	9341.9130 9342.9104 9343.9610	4 1 3 4 0 4 0 6 4 2 7 4 1 24 2 2 1 3 2 2 32 7 7 1 4 2 2 4	100 0 100 0 100 0	15 0 8408.4438 16 0 8408.8001 16 0 8410.8527 16 0 8410.8164 16 0 8412.6770	5 2 4 5 1 5 6 100 0 5 2 7 5 3 6 100 0 6 5 2 7 5 3 6 100 0 6 5 1 7 5 2 18 100 0 6 3 3 7 1 6 5

TABLE I-Continued

19 0 8414 - 5786 2 1 1 0 42 100 0 16 0 8463 - 1277 5 3 2 5 3 3 2 100	۸.	٧	SIGMA	1.KW.KC	. 1 KW	KC	P1	92 9	- 1		•	٧	SIGMA	J.KW.KC.	J KA KC	PI	PZ	6	ı
19 0 8414 - 5786 2 1 1 0 42 100 0 16 0 8463 - 1277 6 2 9 7 3 10 100		•		2 2 1	2 2	•	66	10			•	0		3 3 0	3 3 1	••		10	
16 0 8616.1197 2 2 0 0 2 2 1 100 100 0 17 0 8663.7726 6 2 9 7 3 4 10 100 0 16 0 8664.9862 3 1 2 2 1 1 61 100 1 16 0 8664.9862 3 1 2 2 1 1 61 100 1 16 0 8664.9862 3 1 2 2 1 1 61 100 1 16 0 8664.9862 3 1 2 2 1 1 61 100 1 16 0 8664.9862 3 1 2 2 1 1 61 100 1 16 0 8664.9862 3 1 2 2 1 1 61 100 1 16 0 8664.9862 3 1 2 2 1 1 6 1 100 1 16 0 8664.9862 3 1 2 2 1 1 1 61 100 1 16 0 8664.9862 3 1 2 2 1 1 1 61 100 1 16 0 8664.9862 3 1 2 2 1 1 1 61 100 1 16 0 8664.9862 3 1 2 2 1 1 1 61 100 1 16 0 8664.9862 3 1 2 2 1 1 1 61 100 1 16 0 8664.9862 3 1 2 2 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1				5 5 0		3		-			۰	۰	6461.5281		* 3 2				•
19 0						0					٠	9		9 3 2	9 3 3				•
16 0 8417.6634 3 1 2 3 1 3 16 100 0 16 0 8664.9682 3 1 2 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1			8416.1197								7	0			7 3 4	10			0
10				7 1 2		7	12					9		2 1 2	2 1 2	. •			0
16 0 8619-7005 3 2 1 0 1 93 100 0 16 0 8466-2700 15 013 16 014 20 10 15 0 9620-0248 2 0 2 1 0 1 93 100 0 15 0 8466-2700 15 013 16 014 20 10 15 0 9620-0248 2 2 0 0 1 1 1 2 1 100 0 15 0 8466-2707 6 3 4 6 2 8 24 100 11 0 9620-0248 2 2 0 0 1 1 1 2 1 100 0 15 0 8466-2703 6 2 4 4 7 4 3 1 / 100 11 0 9620-0248 2 2 0 0 1 1 1 2 1 100 0 15 0 8466-2703 6 2 4 4 7 4 3 1 / 100 11 0 9620-0248 2 2 0 1 1 1 1 2 1 100 0 15 0 8466-2703 6 2 4 4 7 4 3 1 / 100 11 0 9620-0248 2 2 0 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	10	9		3 1 2	, ,	,								. 3 1 . 2	. # 1.1				
16 9 8619.0838 2 0 2 1 0 1 93 100 0 16 0 8467.0043 6 3 6 5 3 6 28 24 100 17 0 8460.2305 6 4 3 7 6 2 5 100 17 0 8460.2305 6 4 3 7 6 2 5 100 18 0 8468.2305 6 4 3 7 6 3 1 100 18 0 8468.2305 6 4 3 7 6 3 1 1 100 18 0 8468.2305 6 4 3 7 6 3 1 1 100 18 0 8468.2305 6 4 3 7 6 3 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		•					-4							13 113	12 112			10	2
15 9 6820 6248												×	8447 - 804 3	13 013	12 812				ö
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1				6 1 3	4 2		- 6					ă			7 1 1				ŏ
1				3 3 3		2	7					ě			7 5 2				ŏ
16 0 8625 3221 2 1 2 1 1 1 4 3 100 0	1 .	0	6423-1957					1.0			0	ō		6 1 6	7 4 3				ā
16 0 6425-3221 2 1 2 1 1 4 3 100 0 8475-4033		0	8423.7345	6 2 5	0 1			10	0 0	í	5	0	8471 . 5144	6 2 5	5 1 4	17		10	ō
16 0 84278.1111	16			2 1 2	1 1	1		10					8472.4033					100	
15 0 6436.2030 5 9 3 2 5 2 3 23 1 1 0 1 5 0 6476.2026 6 6 0 7 6 1 10 10 15 0 6436.2026 1 6 6 1 7 6 2 10 1 10 15 0 6436.2026 6 6 6 1 7 6 2 10 1 10 15 0 6436.2026 6 6 6 1 7 6 2 10 1 10 15 0 6436.2026 6 7 6 1 10 1 10 15 0 6436.2026 6 7 6 1 10 1 10 15 0 6436.2026 6 7 6 1 10 1 10 1 10 1 10 1 10 1 10 1				5 0 5	0 3	٠						٥							۰
15 0 8432.0814	16	9		4 2 2	* Z	3												100	
14 0 4334 6710		_				-	- 9	10							7 6 1			10	0
14 0 4334 6710	! 2	2		2 3 2	2.3	3	23						8476.2926		7 6 2			1	0
16 0 6433.*022 7 3 5 7 3 4 10 100 0 16 0 8486.0306 3 2 2 2 1 79 100 16 0 8487.4675 5 5 1 0 16 0 8487.3686 3 2 1 2 2 0 5 10 0 16 0 8487.3686 3 2 1 2 2 0 5 10 0 16 0 8487.3686 3 2 1 2 2 0 5 10 0 16 0 8487.3686 3 2 1 2 1 1 1 3 112 15 1 1 1 1 3 112 15 1 1 1 3 112 15 1 1 1 3 112 15 1 1 1 1 3 112 15 1 1 1 1 3 1 1 1 1 1 1 1 1 1 1 1 1 1	17	•		3 Z Z	2 1	ı						9			1 1 1				0
16 0 6435.4675 9 5 1 6 5 2 25 10 0 16 0 8490.2730 3 2 1 2 2 0 5 100 16 0 8490.2730 3 2 1 2 2 0 5 100 16 0 8490.2730 12 111 13 112 15 1 1		•					. 2					9		2 6 8	3 0 3	1.2			0
18-36-0150				4 4 7			10					ä			5 5 1	72			0
18-36-0150	12			2 2 1	2 2	1	22					×				. 2		100	9
19 7 8436-5910 14 014 15 015 6 0 0 18 0 8491-3881 12 21 13 212 5 1 1 0 1 0 6 0 6 6 1 6 17 1 0 1 0 6 0 8 6 1 6 17 1 0 1 1 0 1 0 6 1 0 8 6 1 6 1 7 1 0 1 1 0 1 0 1 1 0 1 1 0 1 1 0 1 1 0 1 1 0 1 1 1 0 1 1 1 0 1 1 1 0 1 1 1 0 1 1 1 0 1 1 1 0 1 1 1 0 1 1 1 0 1 1 1 0 1 1 1 0 1 1 1 0 1 1 1 0 1		•				•						=							ŏ
1 0 6 6 3 6 5 6 5 6 1 1 1 1 0 6 1 0 0 0 1 6 0 8 4 9 1 4 4 6 1 2 2 2 1 1 1 1 0 8 1 1 0 0 0 1 1 8 0 8 4 9 1 4 4 6 1 2 2 2 2 1 1 1 1 0 8 1 1 0 0 0 1 1 8 0 8 4 9 1 4 4 6 1 2 2 2 2 2 1 1 1 1 0 8 1 1 0 0 0 1 1 8 0 8 4 9 1 4 4 6 1 2 2 2 2 2 1 1 1 1 0 8 1 1 0 0 0 1 1 8 0 8 4 9 1 4 4 6 1 2 2 2 2 2 2 1 1 1 1 0 8 1 1 0 0 0 1 1 8 0 8 4 9 1 4 8 1 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	1.8	•		14 014	15 0	1.5	- 7								1 3 212				ŏ
16 0 8438.4632 28 1 1 1 1 0 8 1 100 0 18 0 8491.4861 5 2 3 6 4 2 17 10 16 0 84491.4861 3 2 3 6 4 2 17 10 16 0 84491.4863 0 6 0 6 5 0 5 9 6 100 15 0 84402.2861 3 0 3 2 0 2 63 100 0 16 0 8492.5530 4 2 3 4 0 4 10 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1				14 114	15 1	15										17		10	ŏ
16 0 6439.2697	16	0						10				ō		5 2 3	4 4 2			io	õ
15							26	10	0 0	í	6	ō	8491.8960	6 0 6	5 0 5	96			ō
15 0 8441.3503 4 3 1 4 2 2 9 100 0 15 0 8403.3252 8 3 9 8 3 6 7 100 16 0 8403.3252 8 3 9 8 3 6 7 100 16 0 8403.3252 8 3 9 8 3 6 7 100 16 0 8403.3252 8 3 9 8 3 6 7 100 16 0 8403.3252 8 3 9 8 3 6 7 100 16 0 8403.6160 3 1 3 2 1 2 95 100 0 15 0 8404.7874 12 112 13 113 14 100 16 0 8404.7874 12 112 13 113 14 100 16 0 8404.7874 12 112 13 113 14 100 16 0 8404.7874 12 112 112 112 112 113 113 14 100 16 0 8404.7874 12 112 112 112 112 112 112 112 112 112	1 6	0	8440.2801	3 0 3	2 0	2	63	10	0 0			0	8492.7530	4 2 3	4 0 4	10		1	
1			8441.1443									ō	6493.0373	6 1 6	5 1 5	7		100	0
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1.1-µm BANDS OF H216O

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TABLE I-Continued

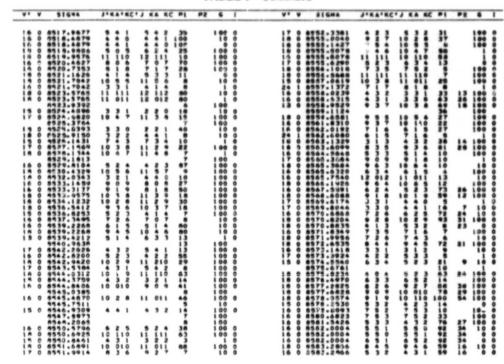
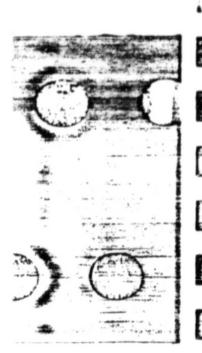


TABLE I-Continued

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۷٠.	V SIGNA	J.KW.KC.) KW KC b	1 P2	6		A. A SIGNY	J.KA.KC. J KA KC PI	**	•
7	0 8583.973 0 8584.167 0 8584.687 0 8584.859 0 9585.170	8 7 6 9 3 7 6 7 3 1 3 4 2 2 7 7 3 5 8 2 4 1 8 10 010 10 2 9 1	•	100	0000	15 0 8619-8257 8620-9466 17 0 8621-7255 16 0 8622-2576 18 0 8622-7556	191 714 st	10	100
	0 8595.909 0 8586.298 0 8535.889 0 8535.889		. 10	100	000	19 0 8623.0134 19 0 8623.0134 24 1 0624.1926 18 0 8624.1926 8624.9221	181 111 1	12	100
	0 8590.167 0 8590.896 0 8592.365 0 8593.833 0 8594.034	8 1 8 8 2 7 8 1 7 8 1 8 10 8 2 7 8 2 8 10	27	100	000000	16 0 6625-5127 16 0 6625-5127 16 0 6627-0696 8627-4304 18 0 8628-1426 16 0 8628-3136	735 634 5	::	100
:	0 8595.644 0 8595.841 0 8595.377 0 8596.898		6	100	0	17 0 8630.6644 17 0 8631.847 8632.268 16 0 8632.825 18 0 8633.398	111 111 4	**	100
	8598.398 3 8599.876 0 8599.999 0 8601.245 0 8601.987		5 0 4 0 27 0 70	100	0	16 0 8634.0 694 8635.140 16 0 8636.0 88 18 0 8636.760 16 0 6636.760	171 771.	**	100
:	0 5603.400 1 6604.020 8604.534 0 8604.578	111 111	53	10	0	18 0 8637 8747 18 0 8638 3676 18 0 8639 1367 18 0 8639 868 17 0 8639 7164		27	100
4 8 6	0 5605.529 1 8606.797 0 8607.679 0 8609.067		0 22 0 22 0 3 28 6 10	100	0000	16 0 8639-861 16 0 8639-886 8640-307 16 0 8640-714 18 0 8641-3369	414 417 20	**	100
	0 8609.827 0 8611.383 0 8611.828 0 8613.528	7 3 5 8 3 6 16	0 61 3 42 0	100		18 0 8641.5983 18 0 8641.5983 24 1 8641.9304 8642.3004		19	100
	0 8614.417 0 8615.070 9 8615.841 0 8615.841	2 2 1 3 3 0 4 7 1 0 8 1 7 10 1 8 7 1 9 7 2 1	9 21 0 56 0	100	0000	15 0 8644.0253 15 0 8644.2536 24 1 8645.2216 24 1 8645.2216 16 0 8646.0236			100
	0 8617.382 0 8617.585 0 8617.641 0 8618.059	7 5 2 6 5 3 3 6 2 4 7 2 5 7 7 2 6 6 2 7 1 6 7 5 3 6 5 4 7	3 11 8 63 0 61	10	000	18 0 8646.4654 18 0 8646.772 18 0 8646.9464 16 0 8646.9464	1 1 1 7 1 2 73 1 1 1 7 1 2 73	**	100
	8619.296	3 6 1 6 7 2 5 10	0 48	100		16 0 8647.439	660 661 71	23	19





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TABLE 1-Continued

4. 4	51544	1.44.46.	1 KA 40 PI	0.5	G	1	۸. ۸	,	SIGMA	J. KA	* 4.0	.,,	A 4	C #1	4.5	4	T
14 0	9647.6100 9648.7474 9648.9980 9649.2471 9649.2471	* 2 2 1 1 3 6 3 6 6 0 5 7 5 3	2 1 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2		100	200200	18 17 17 17 17 17 17 17 17 17 17 17 17 17	2	8675.0480 8675.3975 8675.3975 8675.7797 5676.1490 9078.4166	5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	5 0 5 3	**	9 6 2 5 1 6 6 6	100	13 53 99	100	0 0 0 0
19 0	#54 9. 90 45 954 9. 65 74 954 90. 15 34 959 9. 9 325 959 9. 9 325 959 9. 9 9. 9 9. 9 9. 9 9. 9 9. 9 9. 9	7 7 2	7 2 7 100		100	0000 00	17 17 19 19 18		5678.6395 5679.2915 6680.2593 5681.2614 5681.9122 5685.4213	* 1		***	21710	100	**	100	000000
16 0	4662.1944 4641.1427 4644.3164 4644.8239 4654.9601	4 1 4 7 7 2 6 6 0	7 1 7 100 9 1 4 10 9 0 9 11 4 3 9 100 7 6 9 36 7 6 1 38	41	100	000000	15 6		8686 - 574 1 8686 - 66 38 8687 - 0194 8687 - 1448 8687 - 2497 8687 - 3833	7 0	2 2 2	2 3 6	9 3 9	36	18	100	000 00
1 2 0	9696.9007 9696.9007 9697.1496 9697.1496 9697.2897 9698.0776	8 2 5 5 6 1 4	3 1 2 14 8 4 5 12 7 4 6 15 6 3 6 15 7 3 6 30 6 1 5 100	11	100	000000	18 6		8688-6043 8689-1910 8689-2177 8690-8701 8691-3091 8691-9912	10 2	3 6 7 5		3 3 4 7 2 3 4 7		37	100	200000
17 0	9560.2754 1560.3754 1560.9714 5560.9714 6561.6993 4662.0152	2 1 2	3 3 0 14 5 2 4 19 7 2 4 19 9 1 4 50 7 2 4 100 7 2 4 49	29	100	0 00000	10 10 10 10 10 10 10 10 10 10 10 10 10 1		8592.6678 8692.8594 8693.9054 8594.0690 8694.2668 9695.1602	3 1	ě	:	18144	31 68 100 76	1 0 32 02	100	
14 0	4661.1525 4661.5076 8661.5076 8661.5076 4661.5076 4665.1729	2 0 2	3 2 1 64 22 22 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	10	100	000000	19 18 18 18 18 18 18 18 18 18 18 18 18 18		5695 -8518 5695 -9768 5696 -4682 5696 -9875 5697 -1970 8697 -0998	9 2 6 1 6 4 7 1	9	5 6 7 5 4 3	3 2 5 5 5 5 5 5 5 5 5 5 5	100		100	000000
100	#465.4421 #466.7624 #467.2144 #467.4409 #468.2422	9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9	6 4 2 64 6 1 4 70 6 4 1 100 7 1 6 71	35	100	000000	10 0		8697.9098 8698.5247 4700.1275 8700.1275 8700.6325 8700.8018	* * *		37 50 60 4	9 1 9 0 1 3 4	100	**	100	
17 0	/ 646.7469 /669.1921 6669.6146 4649.6146 4470.2463 4670.4965	3 0 3 5 2 4 4 0 4 1 1 0 2 1 2	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	26	1.0	00000	19 (17 (18 (17 (;	8703-0176 8703-5188 8704-1744 8764-3898 8705-1591 8705-1781	3 1 2 1 2 1 2 1 2 1 2 1 2 1 2 1 2 1 2 1	ž	7 3 4 4 2 2	0 3 2 2 2	100	96 95 27	100	000000
19 0	4671.7047 4671.7047 4671.4559 4675.0079	5 0 5	9 9 4 9		100	000	17		6706.0945 6706.4383 6707.6510	9 0 4 3 2 2 5 1	900	3	2000	27	34	100	0 2 0

TABLE I-Continued

•	٧	815#4	1.44.76.) KA	« C	-1	92	G	1	۸,	٧	SIGMA	1.1	A.K	2.1	KA KE	PI	P 2	G	_
		3709.6333						100		19	0	8747.1495	10	110		911	17		100	
	_	9709. 1897				12		1 00		24		8747.7507	•	. 0	:	* 1	9		10	
•	0	8709.9441	9 7 1	2 ;	ř	5.2		10	0	24	à	5747.7507 6748.1755	:	2 4	:	: :	47	34	100	
•	0	4710.1515	0 0 0	1 0		22		10	ő	19	ő	8748.8765	:	4 7	3	5 6	20	10	100	
	ò	9710.4766	7 7 1	7 7	ò	25	12	10	ŏ	ió	ŏ	8749.6548	- 1	7 4		7.3	90	63	100	
7	ň	4710.4766	7 7 0	, ,		55	12	.,	ŏ	16	ő	8749.7977	7	9 2	- 6	ė i	- 6			
	0	9711.1327	2 2 1	3 2				10	5	15	ŏ	9750.0697				3 3	7		10	
Ä	ö	8711.5296	3 0 2	3 0	*	14		100	2	1.5	0	8750.2357		3 6		3 5	28	10	100	
4	0	8712.901A		7 .				1	0	24	- 1	8750 . 6371	2		2	1 5	7			
٩.	3	0713.4555	3 2 2	4 2		100	1 20	100	0	17	. 0	8750.7132		2 2		1 7	7		. 1	
	0	4 "14. "567	3 3 0	. 1		100	* 3	100	0	19	0	9751.0862	3	0 3		3 6			10	
4	0	4715.8799	3 3 1	. ,		100	96	100	0	19	0	8751.5048		3 5			30	10	100	
,	0	9717.2447	4 1 7			?		1 00	0	17	0	8751-7166	3	3 1	•	5 2	55	30	100	
;	0	9717.9115	3 0 3	* 0	:	100	99	100	0	16	9	5751.9172	- 6	3 1		3 5	- 2		100	
'n	ö	9720.1071	5 7 5	: :		100	1 00	100	ŏ	24	i	6752.3506		5 0		5 1	- 1			
9	ŏ	4720.4239	7 2 6	7 3			1 00	100	ŏ		•	4752.9511					- 7		100	
ä	ŏ	9721.9730		4 2		97	78	100	ŏ			0753.1231					- 6		100	
	ŏ	8722.1474	9 3 7	9 1		*1	25	1.0	ō			6752.8941					6		1	
9	ŏ	4727.3196	5 4 2	6 5			-	10	9	1.6	0	8754.9316	1	1 0	2	1 1	100	94	100	
9		4 72 3 . 21 12	5 4 1	4 4	2	1.3		100		1.6	0	8751.9653		0 0		0 1	10		100	
	15	4721.5736	5 1 5	5 1		90	75		0	2.4	٠,	4757.2675		0 1		0 0				
,	e	4724.4427	0 0 0	1 1	1	11		1 00	σ	17	0	8757.4013		5 3		2 3	1.0		10	
_	_	9725.4 161			_	2		1 00		17	9	8757.5596	•	3 0		2 3	32	17	100	
7	0	9726.4714	2 2 1	3 1		100	96	1 00		1.6	0	8760 - 1 386	1	0 1		0 2	100	97	100	
٠	c	4727.4960	1 1 0	2 1	•	•		100	5	17	ő	5760 -4229	- 2	2 0	2		100	22	100	
6	0	4728.1267	5 5 2			- 2		100	٥	1 "	٠	2762.7618		2 0			23	**	100	
2	ŏ	9729.5721	6 5 1	7 7		26	11	100		19	0	8763.1262			10	2 9	12		169	
	ĭ	4729.2114	1 1 1	1 1		14	io	1 00		19	ã	8763.4 394	ě	2 8	10	1 9			1	
	à	9730.1293	2 1 1	3 i	2	1 20	100	100		1.6	ā	9765.0393	- 1	1 1	. 5	1 2	100	100	100	
		9731.5100						1				8768 . 1 349							100	
		47 42 . 6415				•		1 00		15	0	8769 . 1185		5 0		• 1	20		100	
4	0	4772.9797	1 9 1	2 0	5			100				8769.5668	_				. 3		10	
٠	•	9733.6642	3 0 3	3 5		?	50	10	0	19	0	8770 - 1 287	7	3 3		3 0	. 12		100	
2	0	9711.9264	2 2 9	1 2	1	1 00	1 00	10	0	10	0	8771.5278	3	2 4		2 3	100	89	100	
*	0	9736.8376	2 0 2	2 1	9	23	12	1 00		19	0	8772.2915	- 4			4 1	28	1.6	100	
;	ò	4737.43A7	2 2 1	4 5		17		100		19	ŏ	6772.1106	- 3	3 2	10	110	24	10	100	
á	ě	9737.0190	7 7 1	2 1	7	14			š	19	ă	8773.2179	- 6	ĭ ø		010	~7		100	
ä	ĕ	A738.3312	2 2 1	3 2		100	93	1 60	0	1.6	ă	A773.5903	Ŷ	3 4		3 4	90	53	100	1
•	ŏ	9778.5468	2 0 2	3 0		100		1 20		1.6	ō	8774 . 4156	,	2 2	3	2 1	17		100	
	ĭ	4719.2230	2 2 0	2 2	i	23		1 00	9	19	Ö	A774.5258		3 1		4 2	12			
٠	٥	4740.4098	5 0 5	2 2		0.3	54	. 00		1.6	0	8774.8231		1 1		1 0	11		100	
٠	1	9741 - 31 91	* 3 1	. 1					0	24		8775.2076	2	1 2		1 1	10		100	
*	0	4741.9956	3 3 1			1.2		10	9	1.0	0	4776 - 6010					35	10	100	
7	?	A742,2442	1 0 1	1 !	0	37	71	100				9778.2395					.:		100	
•	1	4742.4130	3 3 1	3 3	0	100	1 00	100		16	å	8778.6572	2	2 1	1 2	2 0	21		100	
;	0	9744.9148	5 1 1	6 4	ż	100	1 00	1 00	9	1 7		8781.3610	- 1	1 0		0 1	47	22	100	
i	0	A764.314A	3 1 3	3 1		12		ò		19	ŏ	8782.4985	å	2 4	,	3 5			100	
Ä	č	4746.4565	4 2 7	7 4				1	ő	ie	ŏ		ő	0 0		0 1	100	100	100	
9		4747.2317	10 010	11 1	11	ě		10		1.5			2	2 0		2 i	17	10	100	

1.1-µm BANDS OF H₂16O

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TABLE 1-Continued

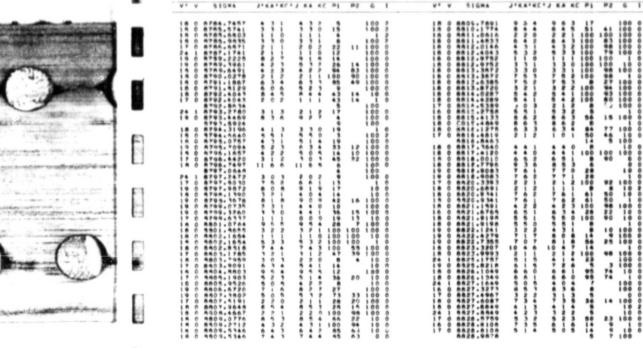
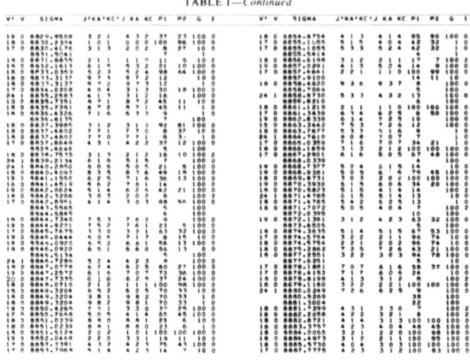
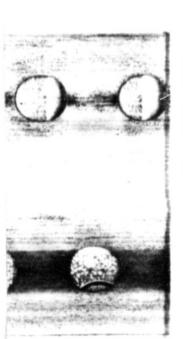


TABLE 1—Continued





FLAUD ET AL.

TABLE I-Continued

٧.	٧	SIGNA	3 * K		611		K.C	-	02	6	-	v ·	-	SIGMA	4*1	KA.	KC.		KA	M.C	P1	PZ	6	-1
-	Ė				-				-		_				-				-					_
#5	1	9999.2768				: :				100	0	1.7	٠,	6916-4872	11	01		10	١		12		1	0
19	0	9998.0594		11			- 2	5.3	1.5	100	0			8919-6396							2		100	
12	0	4649.5197 5559.5522		0 8		: :	2	*2	30	100	0			8920.4450							22	11	100	
÷	ő	8590.2586		ĭ		7 0		63	32	100	ŏ			8921.0518							-8	• •	105	
Á	ŏ	8890.4860		4 7		Á ž		~4			ŏ	1.7			3	3	0	3		3	- 8		100	
*	ŏ	4992.0688	10	4 7		0 3				ī	ō	14		8922.1656	- 2	2	8	ī	8	1	77	53	100	
	2	8892.5950		2 4		5 0		100	77	100	۰			8922.5740							. 7		100	
	0	4463.5122	•	١.		9 9	5	67	27	100	9	17	, (8924 - 1741		2	7	,		۰	36	19	10	
•	0	8993.6121		? \$		6 2	á	.7	27	100	0	16		5924.3566							- Z		100	
٠	0	4896.1932		, ,			•	•1	70	100	۰	17		6925.2217	3	3	÷	á	٠	:	100	49	100	
	0	8896.3433	6	2 5		6 3		- 1			0	1 6	1	4925.4974	- 6	i	٠	- ï	ï	3	100	97	10	
		8896.5508	-					10		100		1.4		4925.4974	. 7	ò	7	- 6	0	6	100	97	1.0	
*	•	8897.0636	3	8 1		2 1	2	10		100	0	11	١ (•	7	9		٠	17	12	100	
		8897.4781	_					. 10		. 1		1.9		8924.4714	- 2	9	ş	- 3		3	35	23	100	
3	8	8898-1917		1 3		: :	:	100	1.9	100		1.0	, ,	8927.2816		١	,			•	-7	23	100	٠,
ä	ő	5599.1256	:	2 3		3 2		100	97	100	ŏ			8927.7390							- 7		100	
ä	ŏ	4499,4917		őś		4 ö		100	97	100		1.6				3	3		3	\$	100	74	100	
ř.	ő	4900.1300	- 6	ă				40	20	100		1.7	, ,	8929.1704	. 9	ï				7	17	-	1	
*	0	4900.7505		1 9		8 1 8 0 5 2		17		10		11	1	8930.1499		. 5	7		0		35		100	
9	0	4900.4420		1 5						0	9	14	, ,		3	2	2		1	3	15		100	
9	0	9901 - 2243		; ;		2 2		10	1.0	100	0	14		8930.8263	10	2		10			.7		100	
3	0	9904.2551	•	5 7		6 O		' 7	14	100	ő	13				ž	ž	٠,	3	ī	'4		100	
ă.	ŏ	9905.7430		7 7		, ,		- 1			ŏ	11			- i	- 3	ī	Ä	ä	٠	12			
6	ě	8905.7430		* 3		7 7	ī			ò	ō	16	١ (8932.2207		,	,				27	21	10	
٠	0	9906.3495		1 3		3 1	- 8	100	1 00	1 00		11					5		3	1	100		100	
*	0	9906.5615		2 2		9 1			46	100		13			2	5	2	:	-8	2	100	95	100	
3	0	4907.9023		3 2		2		100	90	10	ő	13			7	ž	ě.	•	3	•	28	90	100	
ě	ŏ	1908.7403		0 3		ii		45	40	100			٠,	8934.1286		•				-			100	
ä	ő	4909-1093	- 4	11		i i	ı ö	100	97	1 00		10		8934.7396	6	2			2		100	92	100	
ń	0	8909.3514		2 2		1 2		100	1 00	100		11			3		3	3		5			100	
٩	0	4909.4227		3 4		6 1	•	27	25	100		11					5				100	??	100	
,		9910.5026				9 0		2	1.2	1 00	0	11	1		2	1	À	•	•	9	92	71	100	
á	0	4910.6733	10	110		9 0	2	72	52	10	ä	17				ò	ä	,	ò	÷	100	98	100	
7	ŏ	9911.9320		1 6		, i	7	68	52	· ŏ	ŏ	11			- 6	- 6	ä		3	i	14	- 6	10	
ä	ö	A911.7342	*	3 5	1 1	* i		36	35	1 00		1.7		8940.4505		3	ž	3	ż	i	26	18	100	
9	0	9912.2534		1 6		9 1	- 5	100	94	100		11				0	•		- 1	3	9		100	
*	0	9912.9408		0 6	1 1	9 0		100	1 90	100		11		8941-5157		!	2		1	•	100	12	100	
7	0	5915-0014	9	2 6		5 1	3	31	11	10		13			2	ò	ť	2	- 2	à	10	42	100	
á	0	9915.6913		2 3		: ;		-7	• •	1 00		17			÷	ž	À	ē	à	ā	34	77	100	
•		8915.9478	•							100		13			10	- 2	ë	ě	ĭ	ě	15			
		9916.2147						14		100		ie		8946.5436	- 11	3		11	í	10			100	
		4916.4356				-		7	122	100				8946.9815									100	١.
9	0	9917.0077	7	2 6	1 1	7 0		64	56	10		1.2			*	3	1		9	2	-?		!	. !
9	ò	9917.0772	,	1 3		. 0		64	16	10		12			š		3		î	8	100	85	100	
٩	0	9917.5766	5	2 4				100	96	100		12			ž	3	ï	3	3	3	100	84	100	
3	ŏ	9918.4172	- 1	1 4	. 7	: 2	;	12	-	10				5949.7514			-		-			3.4	100	

TABLE I—Continued

•	٧	51544	1.K	۸,	KC .	,	K A	ĸc	01	65	6	1	٧٠	٧	SIGMA	1.4	W. KC	. 7 K	KC	PI	P.S	6	I
	0	9950.0850	3	2		2	0	2	47	38	100	0			6980.9293					. 7		100	
	0	9950.3359	7	s					140	98	100	0	19	0	8981.1218		0 1	. 1		50	36	100	
A	0	0951.0859	*	٠		7	2	5	50		100	0	12	0	8981.9404		013		112	51	20	10	0
6	0	8952.1061	10	2		9	0	9	02	**	100	0	12	ő	8982-1302		2 5		112	21	74	10	ä
	ŏ	9953.4782		2	ž.	- 5		ï	100	96	10	ő	17	ŏ	8982.6011		1 7		9 0	32	26	100	
	ŏ	4954.1285	ő	â	5	- 6		á	100	20	100	ŏ	10	ŏ	8982.7578	7	3 4	6	3 3	83	67	100	
	ě	4955.0518	7	1	6	- 6	ī	5	97	63	1 00	ō	-		8983.1740							- 1	
*	ö	4955.5135		3	3		5	2	23	23	100	0	1.0	0	8983.7047		3 6		3 5	94	56	100	
	0	9955.7196		2	8			3	97	90	100	0	1.0	0	1984.6452		5 3		3 8	97	65		-
	0	9956.2926		2	2	- 2		s	1 00	94	100	0	1.0	0	8984-6452		5 2	3 3		97	7.0	. 00	1
	0	9957.2016		:	3	1	;	2	93	66	100	8	16	0	8984.7548		2 2	3 5		29	11	100	
	0	9957.4585		i.		•		ő	31	40	100	ő	1.6	ő	8987.6940		2 9	- 6		67	30	100	
	ŏ	8957.8687		÷.	ž	- 6		ĭ	100		100	ŏ	1.0	ŏ	8988.2282		014		113	**		10	
	ŏ	8948.6323		01		- 9		à	97	77	100	ö	10	ŏ	6986.2282		114	13 1	113	44		. 1	- 7
	ŏ	8958.9975	. 5	ī.	ī	- 2		0	1.4	10	1 00	ō	1.0	ō	6989.7959	10	1 9	9		57	61	100	
ė.	ō	4960.2200	3	0	3	- 3	1	2	44	26	1 00	0	16	0	8991.0780		5 3		3 4	57	22	100	- 1
	0	9960.3487		٠	2	- 3		1	46	24	100	0			8991.4088	_				- 9	_	100	
6	0	8961.5449		5	1	5	3	2	94	6.2	100	0	16	0	8991.5440		5 3	8	•	72	. 2	100	
_	_	4962.2509		2					. 2	10	100	0	10	0	8992.0657		6 1		1 1	72	25	10	1
	0	9962.4769		-	1	3 6		2	19	10	10	ő		•	9993.4345	,	0 1			12	4.0	100	
	ő	8963.0033	. 0	õ	ò	î	ı,	7	24	1.1	1 00	ŏ	1.6	0	8993.5459	15	115	14	114	1.0		10	
	ŏ	9964.0066	10	š	ă	ė		ò	62		100	ŏ	10		8993.5459		015		114	18		.1	- 7
	ŏ	9954.5295			ž	- 6		í.	- 6	73		ŏ		-	8994.7906							100	
	ŏ	5964.5296	6		ï	- 5	5	ò	100	73	ī	o			8995.4047							1	
	ö	5964.6150	•	2	7	7	5	6		75	10	ō			8995.7322			-		. 5		100	
	0	9964.3959	7	3	5	. 6	2	٠	46	33	100	0	1.0		8996.4363			7 4		64	36	100	
•	0	8965.2325	11	51	0	11	01	11	7		10	0	18	0	8996.8064		2 6	7 ;	2 5	100	84	100	
		5965.9431							. 2		1 00		1.0		8997.5090		016	15 (115	12		100	
	0	9966.2506	11	5	ř	10	3	3	13	63	10	8	18	0	8997.9827		116		113	- 2			1
	ŏ	9966.8676	;;	ĭ		10			100	63	- ;	ŏ	1.6	ŏ	8996.7171		210	10	2 9	65	38	100	
	ŏ	9966.9650		à'	÷			Ä	100		100	ŏ	10		6998.9692		3 7		3 6	95	66	100	
	ŏ	4967.1406	i	ĩ	ĭ			2		49		o		-	8999.2684			_	-	- 6		100	
	ŏ	8967.2305		i	7	7	ī	ň	100	96	1 00	ō	1.0	0	9001.5656	11	110	10			12	10	
	0	4967.7431		3		- 5		3	100		100	0	10	0	9001.6588		4 4			100	69	10	
	0	9969.4898		3	1			•	5.5		1 00	0	1.0	0	9003.4341	3	3 0		!!	. 10	10	100	
	0	9970-1740		9	2	•	2	*	61	46	1 00	0	1.6	0	9004-1748		3 5	7	1 2	57	43	100	
*	0	4971.5187	•	,	6	,	2	5	60	24	100	0	1.0	0	9004.4043	•	9 3	,		97	•3	100	
	0	6973.4307	2	0	2	2			31	1.0	100		1.0	٥	9005-1416		4 0		2 3	7		100	
	ő	9974.9351		ŭ		11			34	::	100	ŏ	19	ŏ	9006.3407		2 8		6 6	ı í		. 1	1
	ŏ	4974.6269	12			ii			96	33	100	ŏ	ié	ŏ	9007.4059		îıï		110	68	1.2	100	i
	ō	9974.7219			2		2	•		13	100	ō	18	0	9008.3140		4 1	5 :	2 4	. 5		1	-
		8976.1229							. 9		1 00		1.0	0	9008.5642	12		11	210	26		100	
	0	4976.9088	7	5	2			3	79	36	100	0	19	0	9009.2767		6 3	7	7 0	11		10	
•	0	8977.3624		٠	•	2	*	3	100	~1	10	0			9009.5620	_	2 7			-2		100	
	0	8977.8717		*	3	2		?	1 6	56	100	9	10	0	9010.0650		2 7	,	6 1	65	39	100	
	0	9978.7432		2	2	7	1 2	÷	100		100	0	16	ő	9010.9459		6 5		: :	21	1.0	100	
9		9979.5263		-	3	- 6		ź	86	52	1 00		10		9011.4359			, ,		16		100	
		4980.4527		-		-	-	-	1.6	9.6	100	~	1.6	0	9012.7908	10	3 8	9	3 7	45	1.0	100	
														-			_						

1.1- μ m BANDS OF $H_1^{16}O$

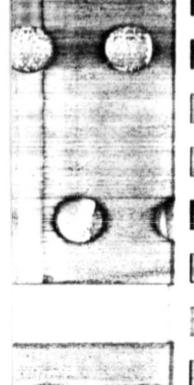
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TABLE 1-Continued

٧,	٧	51544	1.4 W. KC	. 3 KW KC	-1	6.5	6	1	۸.	٧	SIGMA	1.KW.KC.1	KA KC	PI	P 2	6	1
::	0	9914.1294	13 212	.2 2.1	100	**	100	0	12	0	9051-1834	: 1 3	: : :	**		100	
: 7	:	9015.2870	10 3 8	4 1 1	* *		10	0	1.0	0	9053.2108	0 7 2 1	1 7 1	1.0			ē
á	ĕ	9016.0421	7 5 2	2 2 1	**	17	100	0	12	2	9054.0833	202	111	**		100	8
1 2	0	2016.7929	13 112	12 111	15		100	ō	16	ĕ	9054.8002	14 212 1	211	~7		100	ĕ
12	8	9018-9025	9 3 3	2 1 2	22	17	!	8	17	0	9055.2917	10 4 7		**		100	
1 6	ĕ	901 9- 5496	110	101	65	45	100	ö	19	۰	9054.0575					100	
		9020.5920			11		1 00			_	9056.4869			-6			
12	0	9021.1638	10 2 0	2 1 5	80	34	100	0	1.0	0	9057.1753	11 5 7 10		.2		100	
	ė	9023.2070		4 5 4	70		100	ō.			9060.4168			14		10	•
:	0	9023.9154	:::	2 2 3	82		9	0	18	8	9061-7193	111	111	44		100	
	۰	9024.5269	4 3 6	6 3 4	56		100	ö		۰	9062.2535	* * * *		ı i		100	
:	8	9074.7579	14 113	13 115	5.0		100	ė.		_	9062.8725					100	
4	ö	9025.2490	11 3 9	10 3 8	73		1 00	0	19		9063.6982	3 2 2	3 1 3	77		100	
٠	ō	9025.9438		7 3 4	11		10	ŏ	12		9065.6498	6 3 3 1		**		100	
	e	9025.6309			50		100	0	1.0	0	9065.7945	11 6 6 1					ō
6	ō	9029.1505	3 6 3		10		100	ő			9066.3447			11		100	
:	0	9029.5659	9 6 4	4 6 3	**		100	ō	19	۰	9068.8790	4 3 1		1.4		10	
ï	0	9079.6215	10 4 7	10 7 6	.2		100	0	12	0	9068.9773	2 ! 2 !		::		100	
ġ.	0	9031.6112	8 0 8	1 1 1	31		1.0	ŏ		۰	9070.0906			-4		100	۰
2	0	9031.5112	2 ; 1	7 7 0	21		10	0	10	9	9071.2007	9 9 1 1	1 1 2	17		100	
1	v	9034.9411	9 7 2	7 7 1	31		100	0	13	2	9072.0274	111	1 ! 2	27		100	9
	0	20 35. 74 35	3 1 2	3 0 3	72		100	0	19	õ	9073.8698		5 i 5	16		100	ŏ
:	0	9036.4444	15 310	0 0 0	17		100	0	12	9	9073.9602	10 7 4	7 7 3	3			0
٨	ė.	9037.7650	10 3 7	9 7 6	64		10	ĕ	ië	ŏ	9074.3963	10 7 3 1	5.5	13		100	8
:	0	9034.7213	1 : :	7 1 1	1.3		100	0	10	ō	9074.6427	6 2 4 1	0 5	5.5		100	ē
ä	ě	9079.1016	3 2 1	2 1 5	41		100	0	10	2	9075.2815	3 3 0	0 5	32		10	9
	ō	9040.4125	15 510	11 2 9	19		1 00	ö	1.5	ŏ	9076.8832	6 5 2 5	9 9	20		100	ŏ
*	0	9040.9707	10 5 6	: 1 2	10		100	0	12	9	9080.9068			**		100	0
	ĕ	9042.0752	2 2 0	2 1 1	27		100	8	10	ö	9082.0993	.2 1 2 1	9 9	*1		100	8
	ĕ	9043.4185	10 5 5	9 4 4	54		10	ö	19	ő	9082.7545	4 3 2 '	2 3	45		100	ŏ
:	0	9044.4395	11 4 8	10 4 7	12		1 00	8	10	9	9085.0069	111	1 1 2	30		100	
	ō	9045.0516	6 5 2	6 3 3	27		100	ö	19	ĕ	9086.5080	5 3 3 1	1 2 2	18		100	ě
9	0	9045-2354	4 5 3		64		100	0	19	9	9056.6556			11		10	
•	e	9044.1556	13 311	12 310	17		100	0	19	0	9054.7296	6 3 4	0 6	12		100	:
	-	9046.8397			9		100		-		9090.2520			- 6		10	
۰	0	9047.0444	10 6 4	963	23		100	0	12	0	9091 - 3541	10 8 2	1 1 1	97		100	9
		7049.9989			10		100		18	0	9091 6237	10 8 2	: : :	:		- 1	:
		9052.3702			12		100		19	ō	9092.3335	6 3 4 6	2.5	33		100	ō
1.8	0	9050.4434	4 3 2	3 1 3	17		1 00	0	10	۰	9093.3976		2 1				

TABLE 1-Continued

۷,	٧	51644	3. K	۸.,	c.	,	KA	K.C	01	0.2	G	1	۸.	٧	SIGMA	***		KC.1	*	4 60	PI	P 2	6	1
19	0	9093.4728	. ?	2 1			1	0	69		100	0	19	0		•	2 3	,	,		,		100	
1 9	0	9094.4507	11	? !		10	?	;	. 2		10	0	10	0			3	5	3	1 6	21		100	0
š	ě	9095.0372	5	1 3		:	5	2	12		100	0	19			•	ž	2	3	2 1			100	
6	ö	9095.5511	7	5 1	,	6	ĭ	5	15		100	ŏ	16	ŏ		;	į.	á i			10		- :	8
9	0	9096.0107		4 1	2	- 6	- 3	3	15		100	n	19	0	9178.3507	,	3	i i	3	2 6 2 2 5 3 2	20		100	ě
9	0	9099.1139		2 1		- 6	1	6	41		100	0	10						6 :	2 5	29		100	
3	00	9099.2098				- 7	2	?	37		10	0	18	0		5	5	1	•	3 5	7		100	0
ö	ě	9100-1550	2	3 4	í	÷	÷	è			.,	ö	19	0	9184.1746	5		3			18		100	
		9101.2558					-				100		1.6		9192.9583			6	;	2 2	11		100	ĕ
	0	9101.3090	7	3 1			1	5	24		100	0			9193.6964			-			- 8		100	
9	0	9102.9965	2	4 1		3	3	5	85		100	0	10	0					5	3 2 3 3 6	10		100	
ě	ě	9105-0279		1 6		7	ò	7	28		100	0	19			:		: 1	:		25		100	
ġ	ŏ	9107.1465		4 6			3	1	12		100	ŏ	iš					ė i	3 3	3 0	29		100	8
9	0	91 37 . 3299		4 3		-	- 3	٠	28		100	Ö	19		9206.9944	7		5 6	6	ž 4	- 6			ě
3	0	9107.5686		• 3		5	- 2	2	10		10	9	19			2	3	2 (2 3	34		100	ō
ï	ő	9108.1615		1 :		- 3	9	2	23		100	0	18	0			5	2 (3 3	. 9		10	0
ä	ŏ	9108.4360		3 7			3		23		100	ö	16	0		8 7	8	3 1	,	; :	10		.00	9
	ö	9108.7645		ā i		Ä	ź	2	îŕ		100	ŏ	10	ŏ		7				; ;	13		100	8
		9109.5515							5		100		19	ō	9227.6997					íí	20		100	ŏ
3	0	9109.7777	3.	3 6			5	7	15		100	0	19	0		- 5		1 (•	3 2	46		100	ō
ŏ	0	9110.7919		2 2		2 8	١	1	35		100	0	16	0		10	5	2 1	9	3 6	7		100	
ő	ř	9113.9535	7	2 6		7	ű	7	17		10	ö	10	ő	9241.5674	10	3	;	5 1		:		à	0
9	93	9119.639R	7	0 7	•	6	i	6	9		100	ŏ			9242.1965			,	-		- 7		- 7	۰
9	0	9119.9052		1 7		5	0					0	10	0	9247.1972				7 :	3 5	- 6		i	0
:	00	9121.7005		4 3		3	2	;	**		100	0	19	0		6		3 1	5	3 5 2 7	34		100	
~	ŏ	9124.1230	Ä	2 7		à	ï	ě	-;		1 00	0	10	0		2		6 !	9 8	3 3	10		100	8
ġ.	ō	9124.5440		2 1		3	i	2	44		100	ő	1.4	۰	9253.8425		•		,	, ,			100	
٠	0	9126.0991	6	3 4	١.	- 5	- 1	5	16		100	0	19	0	9257.0865	5	5	0 (61		10	2
6	0	9129.2208	2	5 3		7	1	5	. 9		1 00	0	19	0		5	3	1 (. 0	61			ö
ö	ŏ	9129.5885 9132.2972		5 2		ŕ	ô	3	10		100	6	18	0		9					10		100	0
ŏ	ŏ	9132.9710		2 1		ż	ĭ	2	44		100	ě	10	ő	9263.4294	é	:	8 8	2	3 6	10		100	8
9	0	9133.4062	6	5 1		6	ă.	ž	7		100	ŏ	19	ŏ	9274.6446	9	Ä		6	; ;	17		100	ě
9	0	9134.4359		5 2		6		3	15		10	0	19	0	9286.0295		٠		5	3 4	19		100	ŏ
2	0	9134.5294		5 0		2	ż	•	16		10	0	19	0	9280.8090	6		2 1	5 4	٠.	5.6		100	0
ä	ő	9135.1019		5 0	,	5	:	;	13		100	0	19	6	9281.0820	6			5 5	2	11		100	0
ŏ	ŏ	91 36 . 1408		2 4		ě	7	í	12		100	ŏ	19	ő	9263.5903	7					10		100	0
9	0	9136.5526	9	1 6		9	ò	9	7		100	ö	19	ŏ	9304.4030	÷					21		100	ö
3	0	9138.1985		2 1		6	0	6	10		100	0	19	Ö	9305.4474	6		i i	, ,	5 0	34		10	ŏ
:	0	91 19. 1557		3 5		9	?	2	. 6		1 00	0	19	0	9305.4474	6		0 1	5 1	5 1	34		1	ō
3	ö	9142.3167		3 5	í	é	1	•	33		1 00	0	19	0	9323.1785	9			! !		. 9		100	0
Ä	ě	9150.4829				ř	ż	š	19		1 00	ő	10	ő	9325.1147	7			6 1		1.0		100	0
9	ö	9151.0601	3	3 1			2	0	40		10	ö	ië	ŏ	9343.6845	ŕ	,		8	5 2	10		10	0
2	0	9151.0631		• 3		5	2		40		ī	Ö	19	ö	9350.3939	7	7				15		10	ŏ
9	0	9152.5360	3	3 0		2	2	1	72		100	0	19	0		7			6 6		15		1	0
											10		1.6	0	9366.6107		7		, ,	5 2	10		100	0



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TABLE II

Energies (cm⁻¹) of the Rotational Levels for the Vibrational States (130), (031), (210), (111), (012), of H₂¹⁶O^a

		1.10		0.31		210		111		012	
		0273.0773		4373.0443	*.*	4761.3792	*.*	8807.0010	1.9	9000.1402	*.
-	1 0	0277.1790 0121.2920 0129.3009	1:1	0107.4930 0421.1044 0427.3637	1:1	0777.0337	<u>}:</u> }	0830.2312 0844.5359 8850.1080	5:2	9023.4919 9037.1943 9042.8439	1:
9	1	******	112	*******	8 - 9	*******	3.0	0075.3083	1.0	7008.7813	3.
ì	. 1	0.304.000 0.304.000 0.302.2304 0.400.9480 0.407.7039	2 . 2 2 . 1 2 . 4 2 . 5	0402.4079 4402.4079 4400.0379 5049.4649 4001.0218	2.0	8829.5298 8840.2777 8856.7724 5949.8358 8902.6982	3:1	8875.3083 8885.2066 8903.4947 8944.6362 8945.9637	1 . 6	900 n . 7513 907 n . 2809 9095 1601 9135 8415 9137 2554	200
1	ò		3:3		3.0		5:1		1:4	9135.8415	\$:
1	1	0400.0012 0400.7036 0400.7636 0527.3272	2.9	0010.3702 6023.1114 0000.1432 0020.4624 0020.4924 0740.0252	2:1	8494.0325 8400.4782 8433.2035 8404.7544 8470.5522 9657.7643	3 - 1 4 - 5 2 - 2 6 - 7	0939-0627 8985-6596 8979-6738 9014-0205 9020-1717 9098-3961 9098-6088	4.9	9113.5830	1:
-		4527.1272	3:3	0020.9529	117	1909.7544	717	9014.0205	1:3	9205.9698	1:
)		8660.7358	1::	0740.5252	2 · 0	9097.7693	2::	9098.3961	1:3	91 19 0000 91 72 5037 9205 9696 92 2 6670 9207 6 100	and the factor of the
í	:	0504.0711	1:3	*********	1::	3976.2935 4979.6565 9033.0184 9061.1514 9077.3893 9152.6041 9153.6663	7 - 7 7 - 1 7 - 1 7 - 2 7 - 4 4 - 4	9022.3352	1:2	9210.1454	1:
į	1	07:0.2023	7:7	4714.0089	1:3	9001-1010	1:1	9109.4276	1:0	9273.2371	1:
į	- 1	0750.5320	117	1043.2414	1:5	9102.6041	7:5	0103.1018	1:7	9315.1835	1::
:		00-00-07-1-00-00-07-00-07-00-07-00-07-00-07-00-07-00-07-00-07-00-07-07	2 - 5 4 - 8 2 - 5 2 - 6 3 - 6 4 - 4	00 90 0 00 53 0003 0000 0004 0000 0710 0797 0043 0797 0043 0797 0000 0000 0000 0000	2 - 0 2 - 0 2 - 0 3 - 0 4 - 0 5 - 0	9268.0198	1:1	9022-3352 9025-1014 9079-7159 9121-5168 9121-5168 9198-5304 9305-4074 9305-5177	1 - 9	9210.1454 9213.2371 9273.2371 9274.2371 9315.1135 9315.1149 9316.0525 9490.9535	0.000
0	- 5		3 - 4 3 - 1 2 - 5 3 - 1 3 - 5	8502.4849 8791.0025 8830.1708	1:1	90 F5 . 0 00 1 90 FF . 1 0 93 91 F3 . 5 F32 92 F4 . 0 0 25 92 F5 . 0 0 5	3 - () 0 - 7 2 - 7 6 - 7 6 - 8 8 - 0 8 - 1		1:1	9316.1913 9317.0897 9317.0897 9411.0719 9488.07.08 9503.7052 9508.2308	1:
1	;	1002.7452 8040.1044 875.0085 8761.0344 8876.3582 8878.5771	5:1	0030.1708	1::	9173.5732	2 - 7	9210.0414	2.5	9411.0319	300000
1		3070.3500	5:1	4904.2525	5:3	9271.2950	3.0	9310.9930	1:3	9502.7052	5:
i		8878.5771	3.5	2127.2563	2:5	V275.0853	3.1	9424.3703	1:0	9508.2306	5:
* 5 5	i	4257.2230	1.1	0830 - 1708 8830 - 1708 8408 - 3508 8407 - 2471 - 127 - 3503 - 28 - 3840 - 4324 - 0730 - 4324 - 1049				0121.7650 0121.03577 0210.0416 0210.26130 0210.26330 0316.3663 0416.3703 0426.5465 0503.0050	2 - 5 - 5 - 5 - 5 - 5 - 5 - 5 - 5 - 5 -	9508.2308 9611.8095 9611.8095 9745.1862	
0	0	0715.0739 0715.0739 0017.4059 0017.4069 0917.0062	4 · 1 2 · 9 3 · 1 2 · 9 1 3 · 7	0017-2438 0010-0020 0730-0030 0700-7803 0710-7803 0710-0706 0777-1471 0777-1471 0777-1471	3:1	9192.1931	13:3		1:7	0432,5107 0433,5046 0535,4244 0545,3716 0545,2460 0657,762 0756,3004 0757,5546	1:
ì	3	8870.9849	2:4	0910.0039	3:1	9300.1247	2.9	9340.9730 9350.9474	1.9	9535.4244):
í	:	9021.0007	13:7	#100.0419	1:1	9414./241	15:1	9399.9903	1:5	9045.2480	1
:	;	91 90 . 100 3	13.7	4273.0251	4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4	2023.0024 2024.0748	1 3	9567.3079	1:1	9756-3094	13:
9 9	- 1	9402.2322	2.4	9470.2024	317			7700.0003	312	0801.1512	2
0	0			9092.4899	3.0			9230.3301 9230.8775 9350.975 9350.9474 9350.7300 9560.3000 9560.3000 9700.600 9700.600 9700.600 9700.600	2 - 0 - 0 - 0 - 0 - 0 - 0 - 0 - 0 - 0 -	9891.1512 9891.1921 10047.5235 10047.5206	
0	,	0851.0339 0853.9115 9003.7539	3:3	0 0 5 3 . 4 2 4 6 0 0 5 4 . 7 1 2 7 0 1 0 4 . 0 2 4 0 0 1 2 3 . 5 9 3 7 0 1 2 9 . 0 7 1 7 9 2 7 0 . 0 5 0 0	2 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 -	9323.4401 9323.1122 9451.2791 9437.967	;;;	9372.2070 9472.4751 9907.9618 9903.2441 9615.4659 9734.099 9734.099 9737.2298 1037.2298 1037.2351	1:3	9866.8983 9766.4910 9700.3289 9700.3289 9009.7808 9038.308 9038.308 9038.3185 10080.4195 1079.7175 1079.7175 1079.485 1079.4	2::
è	:	9097.4148	1.0	0123.5937	1:1		13:5	9503.2441	1:3	9700.3269	1
ì	- 2			9270.0500	1:)	1307.7702	3.0	9615.9454	1::	9809.7808	1:1
:	1	9368.3000 9372.9513	12:5	9270.0500 9294.3647 9445.1636 9640.5192 9653.4560 9653.1249 9653.1593 10105.2904	7:0			9737.3063	7	9924.9792	0 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1
2	- 1	4572.4515	2.5	9638.4560	1:1	***2.838*	4.7	9873.2015	1:3	10060.9193	1:1
:	ī			10103.2904	13.7			10037.1240	1:3	10713.7133	1
,	ò				13.7			10232.2935	1::	10395.4517	17.7
0 1 1	÷	9004.7208 9006.0325 9188.7016 9203.1071	::	9106.F0F8 9107.2190 9288.1887 9299.7728 9405.6507 9409.0454	3 - 0 3 - 0 3 - 1 3 - 1 4 - 1 2 - 5	9476.3614 9476.4991	3:1	V823.4770	1 - 7 - 7 - 7 - 7 - 7 - 7 - 7 - 7 - 7 -	9718-1984	:::
-		9203.1071	3.4	9494.0098	3:1	9628, 1861	7.7	9674.2183	3.0	vec m. 3053	2.4
i	1			V49V.0454	3:3	W751. W26W	*. "	9846.5303	1:3	9495.3733	** 1
i	:			*******	13.7			9932-8911	1:3	10116.9933	3.1
5	1			*437.4377	4.0			10064.2417	1:1	10404.0925	7.7
0,7	1			10300.3350	13:1			10277.0376	1 . 7		.,,
:	0			10300.5398	6.7			0823.4770 V023.0108 0071.4883 0074.2183 0074.2183 0000.4003 0000.4003 0004.1818 1006.2041 1006.2041 1006.2041 1006.2041 1006.2041 1006.2041 1006.2041 1006.2041	2.0		
°	;	0170.0747 0170.0240	:::	9277-4107 9277-4787 9487-8402 9494-8834	3:1	9844.817 9814.7708	13:7	9042.2202	1.3	9887.8602	:::
١				****.0402	1:1	9814.770m	13.7	V801.0325	1:7	9887.8602 9887.7519 10056.7561 10056.4787	::
•	:			9676.3777				10005.0809	N - 4 - 4 - 4 - 4		
;								10136.0300	4.7		
******	2							10278.3741	1		
*********	•							10279.8545	3.0	10473.8535	3.8
*********	55++3594-							98 92 . 2202 98 92 . 80 88 98 91 . 8325 98 96 8 . 85 9 99 92 . 97 95 100 74 . 68 95 101 36 . 63 80 101 36 . 63 80 102 79 . 35 85 102 79 . 35	3.0	10473.8535	3.6

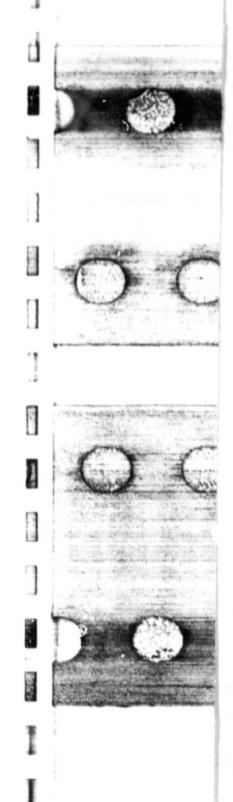


TABLE II - Continued

		130		031		210		111		615	
10	9 10			9465.0067	3.1	9830.8424	13.7	9876.8033 9877.8116 10068.8783	2.0	10074.3509	7:2
10	2 9					10024.6995	13.7	10068-8783	1.7		
10	2 8			9872.5978	3 . 1			10224.7893	3.2		
10	3 7	9084.1735	3.1			10217.0216	3.1	10229.0228	2.0		
10	: :					10336-2134	3.0	10371.1895	2.2		
10	: :							10516.2311	3.0		
10	6 5								2.2		
10	, :							10678.4300	6.2		
10	: :							11101.4461	1.0		
10	: 3							11101.4461	2.1		
11	0 11			9670.0490	6.7	10033.0368	13.7	10081.4128	5.7		
11	1 10			40.010440	6.7			10294.5809	2.3		
11	2 10							10292.3490	2.6		
!!	3 :							10471.3833	1.0		
11	: ?							10626.2573	2.0		
H	5 6							10940.8063	1.6		
11	0 5							11148.8318	1.5		
11	7 :							11366.7813	2.4		
ii	8 3							11306.7858	3.3		
3.1	0 12			9092.0035	7.7			10301.7397	1.0		
12	1 12							10301.4562	1.0		
51	2 11							10533.7093	2.3		
12	3 10							10731.0768	2.1		
12	3 9							10887.6173	2.3		
12								10981.4963	1 . 0		
12	6 7							11078.8781	2.0		
15	6 6 7 6 7 5 8 5							11224.9451	1.6		
1.2	8 5							11054.8053	5.0		
13	0 13							10539.7870	3 - 1		
1.3	1 13							10539.7897	1.7		
13	2 12							10791.4120	5.4		
13	2 12							11004.1858	2.1		
1.3	3 11							11009.0592	1.0		
13	4 10							11194.8327	5.0		
13	5 9										
1.3	6 8							11532.6720	1.7		

^a To the right of each energy level (in cm⁻¹) the statistical unce tainty (68% confidence interval in 10^{-a} cm⁻¹) is given. Additional level appearing through perturbation: (060) [616] 9400.6409 2.4.

Particular attention was paid to lines originating from resonant levels which, as already mentioned, are quite numerous. In general, perturbed lines were located by considering strong lines which cannot be attributed to normal lines of the strongest band, $\nu_1 + \nu_2 + \nu_3$. Among these perturbations are two interesting examples:

— The perturbation between the levels (210)[221] and (111)[211] leads to doublets of lines of nearly equal intensity, instead of the usual single lines belonging to the strongest A-type band $\nu_1 + \nu_2 + \nu_3$ (10).

—An extreme example of resonance is the quadruple resonance between the levels (060)[616], (130)[652], (210)[634], and (111)[624]. In this case, the labeling of the levels will be definitive only when a theoretical calculation provides the mixing coefficients. It may be emphasized that the $6\nu_2$ band appears only through lines originating in the perturbed level (060)[616]; this behavior is similar to that of the

¹ These resonances can be classified for the hexad of the vibrational states $\{(050), (130), (031), (210), (111), (012)\}$ under study as follows: Coriolis-type interactions between $(v_1v_2v_3)$ and $(v_1-1v_2v_3)$ and $(v_1-1v_2+2v_3)$; and Darling-Dennison-type interaction between $(v_1v_2v_3)$ and $(v_1-1v_2+2v_3)$; and Darling-Dennison-type interaction between $(v_1v_2v_3)$ and $(v_1-2v_2v_3+2)$.

 $4\nu_2$ band, in the 1.4- μ m region, which appears only through lines originating in the perturbed level (040)[945], (14).

Because of the large value of the $P \times L$ product for the heated spectrum, we have been able to detect lines belonging to the "hot" band $\nu_1 + 2\nu_2 + \nu_3 - \nu_2$ (relative intensity compared to $\nu_1 + \nu_2 + \nu_3 = 0.2\%$) and lines belonging to the $\nu_1 + \nu_2 + \nu_3$ band of H₂¹⁸O (relative isotopic abundance \approx 0.2%). The assignment of this band of H₂18O has been facilitated by knowledge of the ground state of $H_2^{18}O(15,16)$. The complete list of the transitions with their assignments and the percent absorption at the center of the lines is given in Table I. From these lines and from the ground-state energy levels, we deduced the vibration-rotation energy levels of the upper states, which are given in Table II.

Since the rotational levels of the (111) vibrational state were obtained previously from the study of a flame spectrum (13), we combined the data originating from the latter work and the present study to obtain what we consider to be the most reliable set of rotational energy levels for this state. In the present work, we did not observe lines involving high rotational quantum numbers, and, therefore, we limited the levels listed in Table II to $J \leq 13$. Reference (13) gives the information for J > 13 of the (111) state.

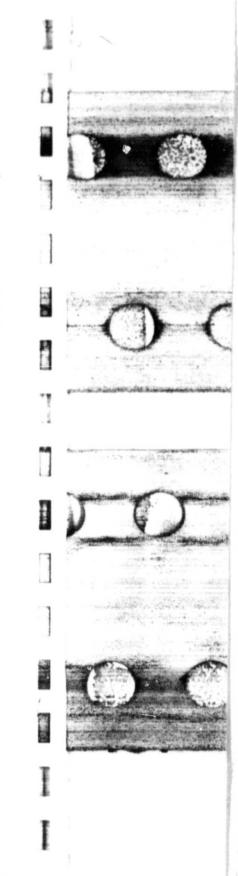
ACKNOWLEDGMENTS

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JOURNAL OF MOLECULAR SPECTROSCOPY 77, 156-159 (1979)

Interpretation of Diode Laser Spectra of Ozone at 14 μ m: (010) and (020) States of Ozone

Introduction. In the present investigation, ozone spectra at different wavenumber intervals have been recorded in the 14- μ m region with a resolution of about 0.002 cm⁻¹ using a tunable diode laser. As may be noticed from Fig. 1, which gives a reproduction of a small portion of the spectrum, the structure is well resolved. Almost all the observed lines have now been assigned to the ν_2 fundamental and to the associated "hot" band $2\nu_2 - \nu_2$ of ozone.

 v_2 band lines of ozone. In interpreting the diode laser spectra, use has been made of the molecular constants reported recently by Monnanteuil et al. (1) for the (010) state of ozone. A synthetic spectrum of the ν_2 band of ozone has been calculated according to the method suggested in Ref. (2). Seventy-eight transitions belonging to the ν_2 band have thus been assigned, and their wavenumbers together with these assignments are given in Table I. This study has given a value for the ν_2 band center as $\nu_0 = 700.9316 \pm 0.0030$ cm⁻¹, which is in good agreement with the value of $700.933 \pm 0.005 \text{ cm}^{-1} \text{ suggested in Ref. (1)}.$

 $2\nu_2-\nu_2$ band lines. Using the rotational constants of Ref. (1) and an estimated band center, we have calculated a synthetic spectrum of the $2\nu_2-\nu_3$ band. As the uncertainty in the estimated band center is almost as large as the spectral range of the diode laser spectra (the extension of which is typically 1 cm⁻¹), it was not readily possible to match the calculated spectrum with the weak unassigned lines of the diode laser spectra. By trial and error, we finally succeeded in assigning 34 lines in the regions studied to the $2\nu_2-\nu_2$ band, and these data are also presented in Table I. Combining these data with the 10 microwave transitions observed in the (020) state (1, 3), a least-squares fit was performed, and the rotational constants thus obtained are given in Table II. For this calculation the sextic centrifugal coefficients have been extrapolated from the values relative to the (000) and (010) states. This procedure seems to be better than fixing the values of the sextic coefficients for the (020) state to those of the ground state. The egreement between the observed microwave transitions as well as that between the observed and calculated energy levels for the (020) state is close, as can be seen from Table III.

Discussion. As a result of the above analysis, it has become possible to locate for the first time

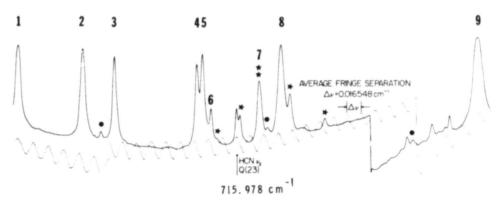


Fig. 1. A sample spectrum of ozone at 14 μ m obtained with a tunable diode laser spectrometer. Lines with asterisks are due to the $2\nu_2-\nu_2$ band and the numerals 1 to 9 are used to locate conveniently the corresponding lines in Table I. Filled circles indicate unassigned lines.

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TABLE I

Wavenumbers (cm $^{-1}$ -vac) and Assignments for Some of the Ozone Lines at 14 μm

OBS.	CALC.	1,	K.	1.	J*	K.,	K"			OBS.	CALC.	J,	K,	$\kappa_{\rm c}^{\prime}$	J"	K",	K,		
83.0760	.076	26	6	2.0	25	,	19	v ₂		704.9286	. 931	1	1	1	0	0	0	v ₂	
.1386	. 144	16	0	16	17	1	17		2-22	.9318	. 934	5	2	4	6	1	5	v 2	
.1916	.197	41		34	40	9	31	v 2		.9747	.977	16	1	15	1.5	2	14		2-2-2
21.00	. 189	23	5	19	22	6	16		2 2 2 2	. 9814	. 98 2	8	0	8	7	1	7	¥2	
. 2405	. 240	30	1	23	30	2 2	22	* 2		705.0143	.315	8	1	7		0			
. 2772	. 308	19	5	15	18	6	12		2 2 2 2	0412	.047	41	- 2	36	40	2	35	·2	
. 3261	. 325	5	í	-5	6	2	4	v 2	2 - 2 - 2	.0645	1001	**	-	,,,,	40	,	33	v 2	
.3419		-			-				2 2	.0880									
. 3544										. 1015	.095	41	7	35	42	6	36 1		
. 36 28										. 2021	. 207	34	6	28	35	5	21	v2	
. 3685	. 370	21	2	20	21	3	19	v 2		. 2055									
.4563	.458	6	1	5	7	2	6		2-2-2	. 2197									
.4772	.479	26	0	26	26	1	25	v2	-	. 2483	. 250	20	4	16	21	3	19	v 2	
.5070	.501	23 12	2	22	11	5	23	*7		.3242	. 326	5	1	5	4	0	4		2 2 2 2
.5377	.541	34	7	27	33	8	26	v2		. 35 79	. 358	6	2	4	7	1	7		
.5442	.545	21	2	20	22	ĭ	21	v 2	2 2 2 2	.4934	.493	14	î	13	13	2	1.2	, 2	
.5581	.558	10	1	9	11	2	10	v 2	2 2	.5021	.501	10	3	7	11	2	10	v 2	2 v 2 - v 2
.5875	.588	16	4	12	15	5	11	- 2	2 2 2 2		.510	40	5	41	45	6	40	v 2	2 -2
.6237	.624	1.8	2	16	18	3	15		2 1 2 - 1 2	.5598	.560	20	1	9	10	0	10	v 2	
.6430										.5847	.585	1.7	4	14	1.8	3	15	ž.	2 2 2 2
.6725										.6091									
.6786	. 6 78	5	3	3	4	4	0	× 2		.6196									
.8370	.836	27	6	22	26	7	19	v 2	2	.6233	.702	10	4		20				
. 90 78	.840	34 19	2	32 18	19	3	17		2 2 2 2	.8356	.833	19	1	16	20 16	0	17	v 2	2
.9078	.906	7	1	7	8	2	6	× 2		.8336	.033	10	1	1.5	10	0	16		2 - 2 - 2
. 72.24	. 920	*		*		*		v 2		715.1221	.120	9	4	6	10	3	7	V -	
96.1343	. 131	20	3	17	19	4	16	$^{\nu}_{2}$.1832		*	-				,	v 2	
. 1615	.160	5	í	5	6	0	6	2	20,000	. 2353	. 233	31	7	25	32	6	26	v 2	
. 2456	. 245	25	3	23	24	4	20		2 v 2 - v 2 2 v 2 - v 2	. 2412	. 239	28	1	27	28	0	28	- 2	2 v 2 - v 2
. 25 39	. 25 3	25	1	25	24	2	22	v 2		. 2471									2 2
. 2856	. 283	13	1	13	12	2	10		202-02	. 2666	. 264	21	3	19	21	2	20		2-22
. 3247	. 321	8	1	7	7	2	6		2 2 2 2 2	. 2920	. 298	50	4	46	50	3	47	V 2	
. 3655	. 366	27	4	24	28	3	25	v 2		. 3107									
.4113	.412	28	3	19	27	5	23 16	* 2		. 4459	.464	42	3	39	42	2	40		
.4567	.427	21	3	19	20	*	10	× 2		.4747	.400	42	3	39	42	2	40	v2	
.5918	.592	7	1	7	6	2	4			.5092	.508	25	2	24	25	1	25		2
.6066	.610	36	5	31	35	6	30	, v 2		.5437	.542	16	3	13	16	2	14		2 2 2 2
.6619	.661	8	ú	8	8	ñ	7	v 2		.5637			-					v2	
.6898	.689	7	1	7	8	ō	8	v 2		.5924	.591	16	5	11	1.7	4	14	v 2	
.6976	,700	19	2	18	18	3	15		2 1 2 - 1 2		.591	23	3	21	23	2	22	2	2 - 2 - 2
.7068	. 715	44	6	38	43	7	37	v 2	2 2	.6407									2 2
										.6877	.686	11	2	10	1.0	1	9		2 2 2 2
01.4894	.490	38	3	35	39	2	38	v 2		.7018	. 703	36	4	34	38	3	35	V 2	
.5094	.499	50	6	44	49	7	43 ?	V		.7107	. 709	20	0	20	19	1	19	v ₂ v ₂ v ₂ v ₂	S.
.5566	.554	16	2	14	17	1	17			.7309	.729	21	2 2	20	21	1	21	V 2	
.6038	.624	18	2	16	19	1 2	19			.8008	. /99	7	4	6	0	1	5	× 2	
.6548	.657	2	1	15	19	0	2	v22	2	.8358	.836	26	1	25	26	0	26		
.6939	.701	42	5	37	41	6	36		2 2 2 2	.9314	.932	34	2	32	34	1	33	V 2	
	.699	43	5	39	42	6	36	v2		.9384	.937	6	2	4	5	î	5	v 2	
.7203	.720	24	4	20	25	3	23	V.		.9479	.947	23	6	18	24	5	19	v 2	
.7703	.771	29	2	28	28	3	25	V		.9540	. 955	21	6	16	22	5	17	2	20,00
. 796 3	. 797	25	2	24	24	3	21	v 2		. 98 21	. 98 2	26	1	25	25	2	24		2 2 2 2 2
.8038	.804	14	2	12	15	1	1.5	v 2				944							
.8212	.823	4	1	3	4	0	4		2 - 2 - 2	16 0060	. 005	8	4	4	9	3	7	¥2	
. 8 25 2	.827	29	3	27	28	4	24	v 2			. 004	23	1	23	22	0	22		2 2 2 2 2
.8488	.850	18	2	16	17	3	15			0163	.006	25	3	23	25	2	24		2+2-+2
. 86 29	.866	34	4	32	33	5	33			.0163	.029	14	3	11	14	2	1.2		
.9499	.951	27	6 2	32 26	26	3	23			.0319	.029	24	0	24	23	1	23	v 2	2
. 96 08	.963	20	2	18	21	1	21	* 9		.0841	.084	32	4	28	32	3	29		2 2 2 2 2
.9673	. 968	26	3	23	25	4	22			.1773	.190	52	4	48	52	3	49	ν.	2 2
. 2013	. 200	840	-	- 3	-	-		v 2		. 2054	. 210	30	7	23	31	6	26	v 2	
02.0035	. 004	15	3	13	10	2	14			. 2200				-				v2	
.0983	.099	6	1	5	4	ć	6	v 2	2,	. 2550	. 256	19	1	19	18	0	18	v_2	
. 2549				-		4			2,2,2	. 3314	.332	32	2	30	31	3	29	2	2 - 2 - 2
										. 3689	. 371	37	8	30	38	7	31	v 2	
										1.005	. 376	46	4	42	45	5	41	V 2	
										.4085	.408	1.2	3	9	1.2	2	10	v 2	

These serial numbers correspond to the ones in Fig. 1.

the vibrational energy $E_{(020)}$ for the (020) state of ozone. The value obtained is $E_{(020)} = 1399.2744 \pm 0.0035$ cm⁻¹.

From the two vibrational energy values $E_{(010)}$ and $E_{(020)}$ obtained here, the value of x_{22} is obtained using the relation $2x_{22} = E_{(020)} - 2E_{(010)}$ as $x_{22} = -1.294 \pm 0.006$ cm⁻¹, which compares reasonably well with the value $x_{22} = -1.0$ cm⁻¹ obtained in Ref. (4).



NOTES

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 $TABLE \ II$ Rotational Constants for the (020) Vibrational State of O_3

Ev	1399.27436	± 0.0035	6 °	(0.68458 ₉ ± 0.00097) × 10 ⁻⁷
A^{\vee}	3.662409092	± 0.0000057	H _K [∨]	0.52703 × 10 ⁻⁷
BV	0.442744919	# 0.00000076	H _{K.T}	-0.2098 × 10 ⁻⁸
CV	0.39007248	± 0.00000026	H_{JK}^{V}	-0.6004 × 10 ⁻¹¹
Δ_K^{\vee}	(0.25660	± 0.00069) x 10 ⁻³	H,	0.3669 x 10 ⁻¹²
$\Delta_{J,K}$	(-0.17920	± 0.0034) x 10 ⁻⁵	h.V	0.2902 × 10 ⁻⁸
ΔY	(0.45742	± 0.0012) × 10 ⁻⁶	h_{JK}^{V}	-0.6004 × 10 ⁻¹¹
δ×	(0.4583	± 0.0013) x 10 ⁻⁵	h,	0.2035 × 10 ⁻¹²

The quoted errors are one standard deviation.

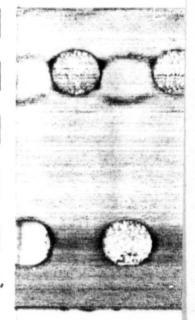
The sextic coefficients have been extrapolated from the values relative to the (000) and (010) states of Ref. [1].

TABLE III

Agreement between Observation and Calculation for the (020) State of Ozone

					M	icrowave	Transitions	
, K	i K		J	Ka	KC	OBS	(in MHz)	OBS-CALC (in MHz)
8	3 1	5	19	2	18	46	326.84	0.00
1	1	1	2	0	2	46	600.95	-0.01
6	2 14	1	17	1	17	46	977.39	0.00
5	3 1	3	16	2	14	48	532.34	0.00
4	2 12	2	15	1	15	49	041.20	-0.03
8	2 16	5	19	1	19	54	725.88	0.01
2	2 10)	13	1	13	60	443.46	0.04
6	0 (5	5	1	5	63	115.29	0.01
7	2 (5	8	1	7	64	381.84	0.00
2	1 11		11	2	10	66	210.13	0.00
						Infrare	ed levels	
	J	Ka	Kc			OBS	$(in cm^{-1})$	OBS-CALC (in cm ⁻¹)
	:	1	1			14	405.0951	-0.0024
		1	3			14	411.1079	-0.0018
		1	5			14	414.6135	0.0007
	(1	5			14	420.5506	-0.0009
	8	1	7			14	433.4184	0.0036
	11	2	10			14	467.1482	0.0018
	1:	3 1	13			1.4	475 • 7092	0.0029
	16	0	16			15	510.7506	-0.0050
	16	1	15			15	18.7821	0.0003
	16	4	12			15	564.5287	-0.0005
	17	4	14			15	578.7103	-0.0008
	18	2	16			15	556.7493	-0.0002
	15	2	18			15	69.8232	-0.0023
	21	2	20			16	603.6494	-0.0007
	21	3	19			16	621.2083	0.0026
	24	0	24			16	642.3163	0.0019
	25	2	24			16	580.9773	0.0015
	26	1	25			16	699.6124	0.0001
	28	1	27			17	745.2668	0.0019
	30	1	29			17	793.9980	-0.0019
	32	2	30			18	361.3107	-0.0006
	32	4	28			18	392.8885	-0.0001





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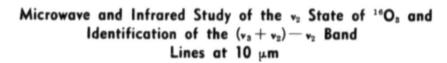
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A study of the ν_2 state of ozone is presented both in the microwave and infrared regions. Forty observed microwave lines led to the determination of molecular parameters which allowed the calculation of the ν_2 state energy levels. One-hundred and twenty transitions of the $(\nu_2 + \nu_2) - \nu_2$ band, observed as a residual spectrum of the ν_2 band, are identified leading to the determination of the band center of the ν_2 fundamental. The interpretation of an observed spectrum of the ν_2 band which is consistent with these molecular parameters is presented.

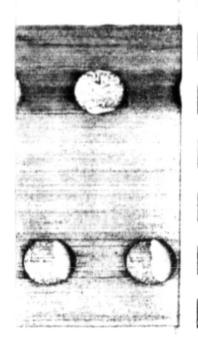
INTRODUCTION

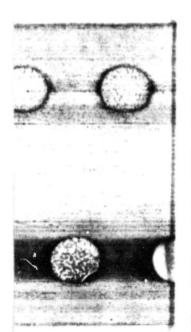
Ozone, a bent triatomic molecule has its three fundamental vibrations located at 701 cm⁻¹ (ν_2), 1042 cm⁻¹ (ν_3), and 1103 cm⁻¹ (ν_1). Microwave study of the ground state (1) and of the two excited states ν_2 and ν_1 which are strongly coupled by a Coriolis interaction have previously allowed an analysis of the ν_3 state both in the microwave and infrared regions (2, 3). In the same way, the microwave study of the ν_2 state (Section I) leads to the possibility of computing the ν_2 state energy levels. These calculated levels along with the values for the energy levels of the ($\nu_3 + \nu_2$) state obtained from the study of the ($\nu_3 + \nu_2$) band (4) have allowed the assignment of 120 transitions of the ($\nu_3 + \nu_2$) — ν_2 band (Section II). In Section III, we present an interpretation of the ν_2 band which is consistent with the information summarized in Sections I and II.

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MONNANTEUIL ET AL.

TABLE I
Microwave Lines in the >2 State (values in MHz)

Lower State			Upper State		Calculated	σ	Obe	erved	ObsCalc	
J	K-1	$^{\kappa}\check{i}$	J	K-1	r _K	frequency		fre	quency	
. 3	1	3	4	0	4	9 077.27	0.01	9	077.22	-0.05
32	3	29	33	2	32	20 050.16	0.05	20	050.14	-0.02
22	3	19	23	2	22	23 551.10	0.02	23	551.12	0.02
24	3	21	23	4	20	27 947.62	0.02	27	947.60	-0.02
17	3	15	18	2	16	28 915.19	0.02	28	915.23	0.04
34	5	29	35	4	32	32 932.60	0.03	32	932.55	-0.05
19	2	18	18	3	15	34 916.39	0.02		916.38	-0.01
17	1	17	16	2	14	36 141.37	0.02		141.37	0.00
25	4	22	26	3	23	39 094.94	0.03		094.99	0.05
16	2	14	15	3	13	39 149.89	0.02	39	149.83	-0.06
15	1	15	14	2	12	39 459.17	0.02		459.16	-0.01
24	3	21	25	2	24	41 922.18	0.02		922.10	-0.08
19	1	19	18	2	16	42 427.71	0.03		427.79	0.08
28	4	24	29	3	27	43 051.91	0.02	43	051.91	0.00
25	3	23	24	4	20	43 190.86	0.02		190.82	-0.04
2	0	2	1	1	1	44 686.12	0.01		686.09	-0.03
33	5	29	34	4	30	46 169.38	0.04		169.36	-0.02
39	5	35	38	6	32	46 861.45	0.03		861.46	0.01
28	3	25	29	2	28	51 274.31	0.03		274.34	0.03
26	3	23	27	2	26	51 462.79	0.03		462.88	0.09
13	1	13	1.2	2	10	51 913.40	0.02		913.37	-0.03
41	6	36	42	5	37	54 981.47	0.05		981.45	-0.02
21	1	21	20	2	18	58 385.94	0.04		385.92	-0.02
8	ī	7	7	2	6	58 950.79	0.02		950.80	0.01
42	6	36	43	5	39	59 189.98	0.05		190.01	0.03
38	5	33	37	6	32	64 861.30	0.03		861.26	-0.04
5	í	5	6	o	6	65 267.06	0.02		266.99	-0.07
31	4	28	30	5	25	67 701.47	0.03		701.46	-0.01
11	2	10	12	1	11	71 449.28	0.02		449.18	-0.10
17	2	16	16	3	13	72 473.72	0.02		473.66	-0.06
11	1	11	10	2	8	72 657.43	0.02		657.44	2.01
30	4	26	29	5	25	78 487.62	0.03		487.63	0.01
30	4	26	31	3	29	80 191.49	0.03		191.46	-0.03
2	0	2	2	í	1	97 930.66	0.02		930.68	0.02
37	5	33	36	6	30	100 209.86	0.04		209.87	0.01
9	1	9	8	2	6	100 582.93	0.02		582.92	-0.01
19	3	17	20	2	18	100 899.89	0.03		899.88	-0.01
14	2	12	13	3	11	103 150.32	0.03		150.39	0.07
4	0	4	4	1	3	103 554.00	0.03		554.07	0.07
29	4	26	28	5	23	119 316.35	0.03		316.39	0.04

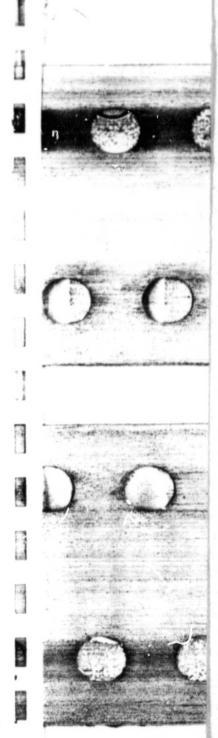
^{*}Lines measured by TANAKA and MORINO $(\underline{5})$.

I. MICROWAVE STUDY OF THE PS STATE

The pure rotational spectrum of ozone in the ν_2 state was first studied by Tanaka and Morino (5). They measured and assigned 17 lines with $J \leq 26$ and $K_{-1} \leq 4$. In determining the rotational constants they assumed τ_{xxxx} , τ_{xxxx} , and τ_{xxxx} to have the same values as given by Pierce (6) for the ground state; τ_{xxx} was chosen to obtain the best fit of the observed with the calculated frequencies of all the 17 lines. The rotational constants thus obtained are:

 $A = 108137.57 \pm 3$ MHz, $B = 13311.20 \pm 0.6$ MHz, $C = 11765.17 \pm 0.6$ MHz, with $\tau_{zzzz} = -27.245$ MHz.

In the present investigation, new measurements were made in the range 18 to 120 GHz, using a classical Stark modulation spectrometer. A modulation frequency of



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 $\label{thm:thm:thm:constant} TABLE~II$ Rotational Constants of the Ground, ν_1 and $2\nu_2$ States (Values in MHz \times 10×)

CONSTANTS		GROUND ST	TATE (†)	V2 STATE		2v2 STATE	
	*	Constants	0	Constants		Constants	0
	0	106 536.236	0.004	106 137.97	0.03	109 796.25	0.09
	0	13 349.2548	0.0007	13 311.443	0.004	13 273.153	0.000
c	0	11 834.3613	0.0006	11 765.034	0.003	11 694.077	0.00
۵,	2	1.3618	0.0002	1.367	0.001	1.36	0.02
Δ_{JK}	2	-5.534	0.004	-5.45	0.03	-5.6	0.6
△ _K	2	634.54	0.03	698.2	0.2	770	4
6,	2	0.20924	0.00002	0.20744	0.06009	0.205	0.00
6 K	2	9.692	0.003	11.647	0.009	13.73	0.01
×,	6	0.011	0.002	0.011*		0.011*	
H _{JK}	6	-0.18	0.07	-0.18*		-0.18*	
R _{KJ}	6	-55.1	0.5	-59	2	-55.1	
14	6	1 178	7	1 379	44	1 176*	
h,	6	0.0053	0.0002	0.0057	0.293	0.0053	
h _{JK}	6	-0.18	0.06	-0.18*		-0.18*	
h _K	6	67	4	77	4	67 *	
8 _K	6	0.021	0.003				

^{*} Ref. (1)

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50 KHz was employed. The cell, cooled with dry ice, is a 3 m long cell in a RG 52 U guide and the source is a klystron locked by a Schommand, FDS unit.

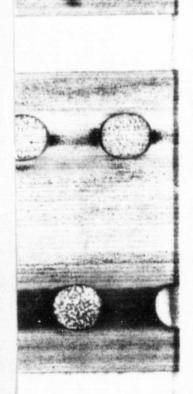
Thirty-three lines, twenty-three of them being new and corresponding to $J \leq 43$ and $K_{-1} \leq 6$ were measured. Using a Watson-type Hamiltonian developed up to P^6 terms in angular momentum, forty transitions (Table I) were fitted by a least squares process. The molecular constants thus obtained are given in Table II. The uncertainties

TABLE III

Microwave Lines in the 2_{P2} State (Values in MHz)

	tat	-		ppe		Ca	lculated		Obi	served	
J	K-1	к1	J	K-1	к1	Fr	equency			equency	ObsCalc
19	2	18	18	3	15	46	326.84	0.04	46	326.84	0.00
* 2	0	2	1	1	1	46	600.96	0.03	46	600.95	-0.01
*17	1	1.7	16	2	14	46	977.39	0.03	46	977.39	0.00
16	2	14	15	3	13	48	532.34	0.04	48	532.34	0.00
15	1	15	14	2	12	49	041.24	0.02	49	041.20	-0.04
*19	1	19	18	2	16	54	725.88	0.04	54	725.88	0.00
13	1	13	12	2	10	60	443.43	0.03	60	443.46	0.03
5	1	5	6	0	6	63	115.28	0.03	63	115.29	0.01
5	1	7	7	2	6	64	381.84	0.04	64	381.84	0.00
11	2	10	12	1	11	66	210.14	0.04	66	210.13	-0.01

^{*}Lines measured by Tanaka and Morino (5).



^{*} Fixed to the ground state value.

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TABLE IV

Evolution of the \(\tau_{aaaa}\)'s with Respect to the Quantum Number \(\tau_2\) (Values in MHz)

	Ground State	v ₂ State	2v2 State
TXXXX	- 0.071211	- 0.07128	- 0.0708
туууу	- 0.037733	- 0.03809	- 0.0381
TEEEE	-25.215	-27.77	-30.6

 σ for the parameters and for the calculated frequencies are also reported in Table II and Table I, respectively.

Molecular parameters are presented for the $2\nu_2$ state in Table II. These were obtained by a least squares process in which some of the constants, as indicated in Table II, were constrained to ground state values. The input data of the analysis consisted of ten lines in the $2\nu_2$ state (Table III), some of which were measured previously by Tanaka and Morino (5), and the remaining ones by the present experiments.

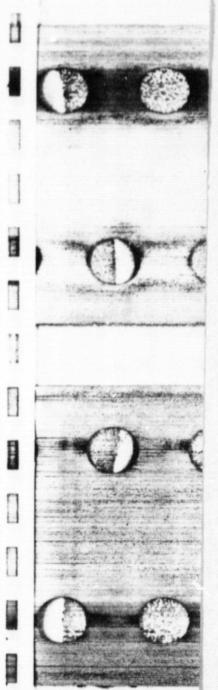
The molecular parameters of the ν_2 and $2\nu_2$ states can be compared to those of the ground state (recalled in Table II), which were obtained by fitting 107 microwave lines with $J \leq 56$ and $K_{-1} \leq 8$ (1). In particular, it may be pointed out that Δ_J , Δ_{JK} , and δ_J have almost the same values in these three states. On the other hand, the corresponding values of Δ_K and δ_K differ significantly. This is to be expected because the bending vibration ν_2 induces an important alteration in the value of τ_{seed} . In Table IV the values of τ_{aaaa} for the ground, ν_2 and $2\nu_2$ states are compared. Interestingly, the τ_{seed} varies almost linearly with respect to the vibrational quantum number (see also Ref. 7).

II. INFRARED STUDY OF THE (P2 + P2) - P2 BAND

Ozone has taken on a new practical significance because of atmospheric considerations. In particular, the ν_3 band appears in the atmospheric window in the 10 μ m region. Therefore, this band has been chosen for laboratory study. It was recorded using a SISAM spectrometer with a resolution of 0.012 cm^{-1} (3). About 1200 lines have been obtained with a calibration precision of $1.10^{-3} \text{ cm}^{-1}$. Most of these lines have been assigned to the $(001) \leftarrow (000)$ band (2, 3). As the average population in the (010) state is about 3.1% of the average population in the ground state at 296 K (experimental temperature), most of the residual spectrum can be attributed to the $(011) \leftarrow (010)$ band.

Calculations

The spectra of the $(\nu_1 + \nu_2)$ and $(\nu_3 + \nu_2)$ bands have also been recorded with resolution of 0.017 cm⁻¹ by means of the Reims-SISAM (8). The (011) and (110) states in Coriolis resonance have been fitted using a Watson-type Hamiltonian and the results will be published separately (4). From this work we have a knowledge of energy levels for the (011) state. The (010) energy levels can be derived from the rotational constants obtained by the microwave study (Section I) and the ν_2 band center, 700.93 cm⁻¹, given





by Tanaka and Morino (9). Thus, we can compare the observed spectrum of the "hot" band $(\nu_3 + \nu_2) - \nu_2$ with a calculated one. Most of the lines of the residual spectrum in the ν_3 region have been assigned by adopting a slightly different value for the ν_2 band center: 700.933 cm⁻¹. The uncertainty in this value is taken as the sum of the uncertainties in the $(\nu_3 + \nu_2)$ and $(\nu_3 + \nu_2) - \nu_2$ band centers (respectively, 1726.5277 ± 0.0027 cm⁻¹ and 1025.594 ± 0.0018 cm⁻¹), that is 0.005 cm⁻¹.

Atlas of the $(\nu_3 + \nu_2) - \nu_2$ Band

Table V gives the listing of this band with calculated wavenumbers and assignments. The absolute intensities have been derived taking into account the Boltzmann factor at 296 K. Only the lines of absolute intensity larger than 0.3 cm·molecule⁻¹,² in the range 985 to 1052 cm^{-1} , are reported. In fact, only 120 lines have been observed because several lines are blended by the ν_3 lines.

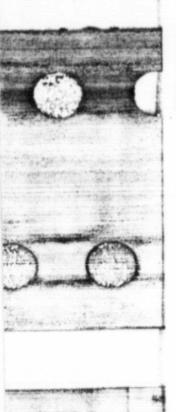
III. INFRARED STUDY OF THE P2 BAND

Experimental Procedure

High resolution infrared studies of the ν_2 band of ozone have been sparse because in the region where it occurs the technology progressed somewhat more slowly than elsewhere. Also, in the same region, the strong v2 fundamental of carbon dioxide is observed and so it is necessary to use a vacuum spectrograph to record its spectrum. In 1955, Nexsen (10) recorded the spectrum using a one-meter focal length Pfund-type vacuum grating spectrograph equipped with a thermocouple as a detector. Although the resolution was not very high, as mentioned in the above section, Tanaka and Morino (9) were able to reinterpret his data and thereby obtain a value for the band center. The present measurements also used a vacuum grating spectrograph. However, it was possible to employ a liquid helium cooled Ge: Cu detector and therefore improved results were obtained. Actually, the resolution achieved was about 0.1 cm-1. The spectral positions were determined relative to the absorption standards of carbon monoxide (11). The ozone sample was prepared by flowing dry and pure oxygen through a silent electric discharge. The gas was condensed in a liquid nitrogen trap and the ozone concentration was increased by pumping off excess oxygen. An estimate of the concentration of ozone was made using chemical methods. Figure 1 illustrates the observed spectrum which corresponds to the data of Table VI. The v2 band which is a B-type band shows an irregular structure with no distinct Q branch present. The lines indicated by an asterisk² in Fig. 1 are due to the CO₂ present in the ozone sample because of reaction with the cell which is made out of stainless steel. Earlier measurements (12) of some of these CO2 lines served as internal standards in the measurement of the ozone spectrum.

⁹ Conversion factors to obtain other intensity units are available in Chapter 4 of Molecular Spectroscopy: Modern Research (K. Narahari Rao, Ed.), Academic Press (1976).

⁸ These asterisks are placed below the lead marks used for indicating the serial numbers of the lines in Fig. 1. They are somewhat difficult to locate in the reduced spectrum. The extent of the CO₂ interference can be inferred from the Q-branch region of the ν_2 band at 667 cm⁻¹.





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TABLE V Wavenumbers and Assignments of the $(\nu_1 + \nu_2) - \nu_2$ Band of Ozone

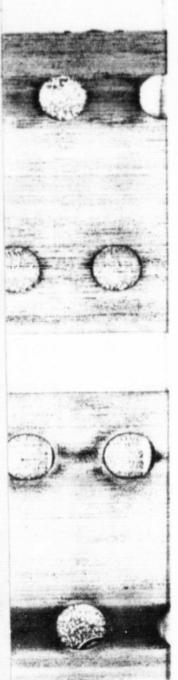
		Waven	umbers a	and Assig	nments of	f the $(\nu_1 +$	ν ₁) — ν	Band o	of Ozone		
ele.	Caper state / E., E.,	1.0007 21.010 2.5.1.5.1	tioner state ener- gr(cm ⁻¹)	100. 101. 10 ⁻²³ cm ms1 ⁻¹	6 · c (10 ⁻³ cm ⁻¹)	Catr.	Opper state / t ₋₁ t _{e1}		state seer. SF(cm ⁻¹)	101 10 ⁻²¹ cm mol ⁻¹	0 - 0 (10 ⁻³ cm ⁻¹)
1867 1991 1888 1888 1888 1888 1888 1888 188		37 1 37 38 38 38 38 38 38 38 38 38 38 38 38 38	1 364 A 568	0 1988 0		987 Juli 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	17	1	1 (86) See 1 (15) See		
997 , s 6, 1 997 ,	27 3 24 14 8 11 22 1 22 23 2 1 22 24 2 22 2	## 3 #5 # 172 # 173 # 173 # 174 # 17	1 077.464 1 078.775 1 078.775 1 08.785 1 0	0.564 0.101 1.021 1.021 0.171 0.879 0.864 0.864 0.864 0.864 0.871 0.714 0.871 0.714 0.714 0.871 0.714 0.871 0.714 0.881 1.02 0.861 1.03 0.861 1.03 0.861 1.03 0.861 1.03 0.861	-2 -4 -2 1 -4	1 001.414 1 009.401	12	35 2 82 34 2 40 40 40 40 40 40 40 40 40 40 40 40 40	10.1 (44) 21.7 (24) 217.7 (25) 217.7 (25) 217.7 (25) 217.7 (25) 217.7 (25) 217.7 (25) 217.7 (25) 217.7 (25) 217.7 (25) 217.7 (25) 217.7 (25) 217.7 (26) 2	0.919 1. 21 1. 21 1. 21 1. 21 1. 21 1. 21 2. 21	-1 -1 -1 -1 -1 -1 -1 -1 -1 -1 -1 -1 -1 -
014 951 014 75	0 6 3 10 6 7 10 10 10 10 10 10 10 10 10 10 10 10 10	10 8 h 11 a 8 11 a 9 11 a 1 a 10 10 a 6 10 a 7 10 a 8 10 a 9 10	861, A48 807, 130 70, 481 818, 538 821, 682 921, 681 921, 681 931, 683 931, 783 931, 78	0.391 0.797 0.797 0.790 0.341 0.407 1.11 0.461 0.461 0.461 0.771 1.02 0.771 1.02 0.771 0.9	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	1 021.370 1 021.170 1 021.170 1 021.170 1 021.170 1 021.170 1 021.170 1 021.170 1 022.187 1 022.187 1 022.187 1 022.187 1 022.187 1 022.187 1 022.187 1 022.187 1 022.187 1 022.187 1 022.187 1 022.187 1 023.180 1 023.	12	12 8 4 11 8 4 11 8 4 11 8 4 11 8 4 11 8 1 1 1 1	(6) 138 (6) 247 (7) 20 (6) 27 (7) 20	0 1864 0 4.17 0 173 0 17	-1 -1 -1 -1 -1 -1 -1 -1 -1 -1 -1 -1 -1 -

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030.136 030.177 030.352 030.454 030.499 030.850 030.877 031.089			718.276 728.227 789.462 603.933 787.188 885.569 731.286 718.451 775.320 722.215	0.71A 0.425 0.467 0.901 0.60A 0.527 0.735 0.860	;	1 036 336 1 036 356 1 036 361 1 036 561 1 036 67 1 036 907 1 036 907 1 036 907 1 037 037	13 3 10 14 6 9 15 7 8 13 7 11 13 6 9 14 5 10 17 8 9 15 6 9 16 7 10	11 3 8 8 13 6 8 14 7 7 11 2 10 13 6 8 13 5 8 14 6 8 15 7 9	780 - 668 891 - 575 864 - A87 768 - 818 817 - 179 834 - 461 1 017 - 849 902 - 797	0.886 0.586 0.610 1.10 0.880 6.753 0.82 0.417 0.487 1.63 1.8	
031.591 031.443 031.554 031.574 031.67	:::	1111	779 - 320 722 - 215 810 - 627 75 3 - 665 73 - 790 803 - 666 737 - 692 782 - 616 616 - 159 794 - 563	0.827 9.487 0.708 0.913 0.403 0.323 0.817		1 027 027 0 0 025 006 1 705 1 106 1 025 366 1 025 365 1 025 365 1 025 379 1 025 666 1 025 720 1 025 726 1 025 726	17 4 9 10 5 10 17 8 9 15 6 9 16 7 10 13 0 13 14 1 18 13 2 11 14 1 18 13 2 11 14 1 18 13 7 10 14 6 11 14 6 11 14 7 10 14 7 10 15 7 10 16 8 11 17 7 10 18 8 11 18 7 10 18 8 11 18 7 10 18 8 11 18 7 10 18 8 11 18 8 11 1	12 1 1	700 - 00.0 No. 34.3 829 - 004 777 - 410 778 - 423 1 033 - 006	1.03 1.34 0.911 1.34 1.35 0.370 0.773	
031.836 032.116 032.129 032.367 032.367 032.362 032.362 032.362 032.362 032.860 032.860 032.860 032.860 032.860 032.860	# 3	# 6 6 # 5 5 # 9 5 # 0 8 # 0 8 # 0 8 # 0 8 # 1 7 # 1 7 # 1 7 # 1 7 # 1 8 # 1 7	911.857 963.900 730.910 789.569 826.528 730.126 967.778	8.46.7 8.56.0 9.70 8.46.7 0.373 0.919 1.02 0.721 0.425 1.00 0.470 1.07 0.317 0.519	4	1 036 - 362 1 936 - 186 1 036 - 260 1 036 - 316 1 036 - 327 1 036 - 333 1 036 - 462	16	13 1 13 12 2 16 13 8 16 14 5 8 14 7 8 15 6 16 15 6 16 15 1 11 15 2 12 15 1 11 15 2 12 16 1 16 17 7 11 16 6 10 16 6 10 16 6 10 17 7 11 16 6 12 17 8 12 18 7 11 18 8 10 18 8 10 18 8 10 18 8 10 18 8 10 18 9 10 18 9 11 19 1 11 10 1	970, 4.29 913, 864 800, 913 771, 199 789, 657 839, 796, 1 067, 155 880, 465 787, 480 929, 238 817, 606, 801, 058	0.497 0.493 1.00 1.22 1.17 0.932 0.374 0.786 0.500 1.20 0.401 1.07 1.07 1.00	
033.036 033.121 033.121 033.533 033.636 033.640 033.640 033.71a 033.71a 033.853 034.233 036.273 036.273 036.273 036.273	10 1 10 10 10 10 10 10 10 10 10 10 10 10	12 7 5 9 2 8 10 A 6 11 5 7 14 8 6 10 5 7 12 6 8 13 7 7 10 0 10 11 4 8 10 2 8 11 1 11 12 5 7 13 6 8	Mic. 350 Mic. 516 870. 553 971. 886 751. 780 797. 973 687. 775 981. 775 981. 775 867. 130 775. 661 137. 777 867. 130 775. 687 130. 481 771. 687 860. 481 771. 687 860. 481 771. 487	0.40 0.992 0.786 0.678 0.301 0.936 0.936 1.13 0.666 1.13 1.06 1.18 0.720 0.358		0 the 452 0 the 452 1 0 the 700 1 0 the 700 1 0 the 840 1 0 the 840 1 0 the 840 1 0 the 840 1 0 the 670 1 0 the 6	20 8 13 17 3 12 18 7 17 15 1 15 16 6 13 16 3 14 16 3 15 15 1 16 17 4 13 21 8 13 17 0 17 18 5 14 20 7 14 18 6 13 18 1 18	10 8 12 10 5 11 18 7 11 10 2 12 17 6 12 15 3 13 15 3 13 15 3 14 16 4 12 16 0 16 17 1 13 10 6 12 10 7 13 10 6 12 10 7 13 10 6 12 11 1 17	##1, 275 ##6 - 644 #	0. 273 0. 786 0. 699 1. 19 0. 662 1. 68 1. 19 1. 26 0. 362 1. 27 0. 791 0. 493 0. 493 1. 28	4
037.802 037.965 036.301 036.157 036.161 036.214 036.354 036.354 036.406 036.452	17 3 14 22 8 15 18 4 15 21 7 14 18 5 14 20 6 15 18 3 18 18 2 17 17 2 15 17 1 18 18 0 19 23 8 15 19 4 15	18 3 13 21 8 16 17 6 16 20 7 13 18 5 13 19 6 16 17 2 16 18 2 16 18 1 15 18 1 15 18 0 18 22 8 16 18 6 16 18 6 16 18 6 16 18 6 16	863.659 1 097.358 880.012 1 032.356 923.667 976.667 976.667 861.362 828.816 820.815 861.363 1 115.765 899.109	1.08 0.950 0.950 0.462 0.762 0.690 1.07 1.17 1.19 1.25 1.23 0.865		1 061.115 1 061.165 1 061.165 1 061.166 2 061.352 1 061.560 1 061.660 1 061.690 1 061.707 1 061.707	23 % 19 25 % 19 21 2 30 24 5 20 25 0 25 27 7 25 28 6 21 28 6 21 25 5 20 25 9 20 22 9 20 23 1 22 25 9 20 27 7 22 28 6 21 27 7 20 27 9 27 9 20 27 9 27 9 27 9 27 9 27 9 27 9 27 9 27 9	22 A 18 26 6 18 20 2 16 23 5 19 26 0 26 25 1 25 25 2 22 25 2 22 26 5 19 22 3 19 22 3 19 22 3 19	963,910 1 066,596 692,176 1 011,613 965,217 1 150,366,982,163 965,467 963,169 1 087,676 1 031,731 962,389	0.808 0.31a 1.09 0.657 0.956 0.891 0.891 0.986 0.48a 0.48a 0.48a 0.48a	1
036,344 036,443 036,749 036,761 036,820 038,820 038,831 039,132 039,143 039,241 039,393 039,393	20 5 16 21 6 15 20 1 20 26 8 17 19 3 16 20 6 17 23 7 16 21 5 16 22 6 17	19 5 15 20 6 14 19 1 19 23 8 16 18 3 15 19 6 16 22 7 15 20 5 15 21 6 16	939.560 991.183 857.336 1 135.009 873.125	0.92s 0.468 0.786 0.811 1.19 0.331 1.05 0.903 0.451 0.745 0.591		1 On 1.89 2 3 On 1.95 2 5 On 2.175 5 On 2.275 5 On 2.250 1 On 2.301 1 On 2.301 1 On 2.419 1 On 2.419 1 On 2.419	25 & 21 25 & 21 25 & 21 28 1 28 29 7 27 28 1 28	23 3 21 21 21 21	86.3 (16.4) 1 087 - 076 1 091 - 731 96.2 (16.4) 96.0 (16.4) 1 177 - 96.3 986 - 06.1 1 105 - 26.4 1 007 - 80.7 1 007 - 80.7 1 196 - 400 983 - 219	1.03 0.867 0.386 0.860 0.453 0.585 0.727 0.996 0.781 0.513	0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0
039.4% 039.403 039.403 039.48 039.49 039.80 039.80 039.80 039.903 039.903 040.439 060.439	21 4 17 24 7 18 22 5 18 23 6 17 23 6 17 25 7 18	19 3 17 18 1 17 26 8 16 18 2 16 21 1 21 20 6 16 23 7 17 21 5 17 22 6 16 22 0 22	1 088 3s.3 998 314 998 314 872,834 871,854 888 A98 850,433 1 155,990 858,753 890,228 977,820 1 087,592 973,907 1 077,186	1.16 1.17 1.02 1.20 0.315 1.15 1.11 0.876 0.566 1.06 0.410		00.1 10.5	25 1 24 29 0 29 26 3 26 25 3 22 30 1 50 29 6 23 28 5 26 27 28 28 2 27	28 6 20 25 5 210 25 5 210 27 1 2 20 27 1 2 20 27 2 2 20 25 2 2 20 26 2 2 20 26 2 2 20 26 3 2 21 26 3 2 21 26 1 2 20 26 2 2 20 26 2 2 20 27 1 2 20 28 2 2 2 2 20 28 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	983.62% 1 33.462 1 024.487 999.670 1 030.024 1 031.563 1 032.250 1 053.277 1 155.316 1 097.111 1 046.270 1 027.306 1 027.405	0. h27 0. h83 0. h83 0. 917 0. 723 0. 748 0. 821 0. 666 0. 991 0. 509 0. 724 0. 889 0. 724	0 3 2 2 3 3 4 3 4 4 5 4 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6
060,003 060,439 060,439 060,450 060,470 060,580 060,833 060,833 060,833 061,018	21 3 18 22 4 19 24 6 19 23 5 18 22 2 21 24 1 26 22 3 20 21 1 20 26 7 20	28 7 17 280 5 17 21 4 18 23 6 18 22 5 17 21 2 20 23 1 23 21 3 19 20 1 19 25 7 19	906 .023 965 .411 1 066 .642 992 .361 905 .260 926 .272 923 .181 883 .763 1 128 .603	0.99a 0.8aa 0.5a2 0.690 1.0a 1.01 0.955 1.13		1 063,719 1 063,749 1 063,838 1 063,838 1 063,968 1 064,051 1 064,067 1 064,346 1 064,373	25 2 23 31 0 31 27 1 26 30 4 25 29 5 20 32 1 32 28 3 26 31 6 25 27 5 38	2h 3 J1 29 1. 19 28 6 27 27 5 28 26 4 27 27 2 26 36 2 27 27 2 36 36 1 25 29 6 36 27 4 36 27 4 36 28 5 13 27 3 25 30 6 2 27 30 6 2 27 4 36 27 4 36 28 5 23 29 6 3 23 30 6 2 27 30 7 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	1 077.105 1 002.401 1 179.408 1 088.774 1 120.413 1 101.883 1 086.372 1 208.761 1 025.588	0.610 0.799 0.360 0.593 0.670 0.556 0.666 0.329 0.720	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1
						1 064 A 20 1 066 A 54 1 066 A 57 1 066 A 57 1 066 A 57 1 066 A 57 1 066 B 66 1 066 B 66 1 065 O 67	33 G 55 30 2 29 30 5 24 29 1 28 29 4 25 36 1 36 27 2 25 32 6 27 30 6 27 31 5 26	\$2 0 32 29 7 288 29 5 25 28 1 27 28 4 26 33 1 33 36 2 24 31 6 28 29 4 26	1 127.303 1 074.180 1 144.921 1 046.380 1 092.548 1 153.617 1 013.146 1 230.712 1 114.633 1 170.115	0.505 0.615 0.433 0.661 0.548 0.456 0.775 0.300 0.503	-6 7 -1 3 -4 +2 3 0
						1 045 .082 1 045 .075 1 045 .252 1 045 .252 1 045 .372 1 045 .464 1 045 .462 2 045 .662 1 045 .755		29 3 27	1 098 ,006 1 180 ,619 1 136 ,260 1 097 ,221 1 308 ,6 hp 1 196 ,096 1 072 ,46 h	0.564 6.410 9.511 9.570 9.367 9.360 0.417 9.327 9.461 0.460	3 3
						1 965 (862) 1 965 (762) 2 965 (762) 2 965 (762) 3 965 (762) 3 965 (762) 3 965 (762) 3 965 (762) 3 965 (762) 3 965 (762) 3 965 (762) 3 965 (762) 3 965 (762) 3 965 (762) 3 965 (762) 3 966 (762) 3 966 (762) 3 966 (762) 3 966 (762) 3 966 (762) 3 966 (762) 3 966 (762) 3 966 (762) 3 966 (762) 3 966 (762) 3 966 (762) 3 966 (762) 3 966 (762) 3 966 (762) 3 966 (762) 3 966 (762) 3 966 (762) 3 967 (762) 3 967 (762) 3 967 (762) 3 967 (762) 3 967 (762) 3 967 (762) 3 967 (762)	30 0 2 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3	Section Sect	1 14 2 303 1 060 264 1 167 267 1 169 269 1 167 269 1 177 27 1 222 999 1 233 396 396 1 233 267 1 110 710 1 122 876	0.418 0.467 0.468 0.416 0.326 0.376 0.335 0.550	
						1 066 - 912 1 066 - 913 1 067 - 666 1 067 - 709 1 067 - 996 1 068 - 066 1 068 - 066 1 069 - 083	33 4 29 34 4 31 36 3 34 33 2 31 33 3 30 35 4 31 30 2 33 35 3 32	36 1 33 35 2 34 30 2 28 30 3 27 33 3 31 32 4 28 33 4 30 35 3 33 37 2 30 32 3 29 34 30 35 3 33 37 2 30 38 4 30 38 3 33 31 2 30 32 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3	1 195 .4 ne 1 222 .4 59 1 254 .7 53 1 164 .4 59 1 176 .403 1 252 .1 78 1 221 .4 57 1 236 .223	0.381 0.379 0.340 0.903 0.469 0.425 0.505 0.358	7



MONNANT EUIL ET AL. OF POOR QUALITY

TABLE VI
Wavenumbers and Assignments of the P2 Band of Ozone

			Wavenuml	bers a	ind Assi	gnments	of the v	Bar	id of	Ozone				
	tearved evenuates og tvar.)	Looke France	Toper trace ty	Line No.	(Sparred Versionles (cm ⁻¹ mm.)	Communitates	Toper Trace	i,	Line No.	Observed Vavenumber (cm ⁻¹ rac.)	tower frate	Oppor State to	ci	
	13-156	18 16 2	15 13 1 0.41	85.7 85.7	676 -276 626 -625 626 -775 627 -119	23 10 14 17 11 7	22 + 13 18 10 + 27 8 20	0.79	***	638.733 139.662	2 1 5	52 5 27 0.59 25 8 20 1.08	1.3	and the second second
	16 (M) 16 (B) 16 (D) 10 (D)	15 16 2 10 13 7 16 16 0 19 13 7 26 12 12	16 13 1 0.43 18 12 8 0.43 13 13 1 0.46 18 12 6 0.48 73 11 13 0.43	80.7 80.9 80.6 80.7 8.74		23 10 16 17 11 7 28 9 18 22 10 12 18 11 5 21 10 12 12 8 28 15 .1 3	18 10 6 27 8 20 21 9 13 15 10 6 20 9 11	0.86	9'96. 9'96. 9'97 1003 1025 1006. 1007	639.683 639.537 639.792	20 8 12 16 8 3 52 8 26 27 11 17	32 3 27 0.54 25 8 00 1.08 19 7 1.3 1.45 1.3 8 6 1.07 31 3 27 0.66 27 10 18 0.41 28 10 18 0.45 27 10 16 0.66 30 18 1.18 30 18 1.18 30 18 1.18 30 18 0.66 31 18 0.66 31 18 0.66 31 18 0.66 32 18 0.66 33 1.88 34 0.66 35 0.66 36 0.66 37 1.88 38 0.66 38 0.66		Southern and John Control
	13 101 14 222				1.28 .099 1.28 .106		66 10 A	0.50 1.06	1-003 1-006	840.156 840.184 840.288	77 11 17 78 11 15 73 14 15 75 14 15 76 11 13 76 1 19	27 10 18 0.41 28 10 18 0.43 23 10 18 0.46 20 10 18 0.46 25 8 18 1.18	-	THE NAME OF THE PARTY.
6.1	10 688	23 17 12 17 13 3 22 12 10	29 13 11 0.47 18 12 4 0.53 21 11 11 7.51	883 887 891 891 895 919 903	6.78.66. 6.79.770 6.79.586 6.79.870	28 * 17 20 10 10 14 11 3 25 * 17 19 10 10	25 # 18 10 # 11 13 10 11 18 8 18	0.00		640.321 640.382	25 7 18 25 7 18 18 8 17	26 10 16 0.46 26 6 18 1.18 18 7 11 1.56		100 M
*1	17 #33 18 11# 18 393	18 19 3 21 12 10 26 11 13 15 13 3	15 12 a 0.56 20 11 9 0.55 25 10 18 0.48 18 12 2 0.60	***	627.366 630.187	20 10 10 18 11 3 25 9 17 19 10 10 13 11 3	12 10 2	1.05	1011	840.435 840.438 840.855	10 11 1	12 8 4 1.76 20 10 10 7.32 17 10 8 0.42 18 10 6 0.43		
8.1	18.717			900	6:50 18" 6:50 26 6:50 688 6:30 688	30 8 22 36 9 13 18 10 8 12 11 1 29 8 22	7 1 1	の の の の の の の の の の の の の の	1018	640.939 641.328	/m / 19 12 12 12 13 14 15 15 15 15 15 15 15 15 15 15 15 15 15	32 5 27 0.38 5 5 27 1.3 1.30 13 7 13 1.35 13 8 1 1.35 27 13 1.35 27 13 1.35 27 13 1.35 27 13 1.35 27 13 1.35 27 13 1.35 27 13 1.35 27 13 18 0.35 28 18 1.35 28 18 1.35 29 19 10 0 0.35 20 10 10 0 0.35 20 10 10 0 0.35 20 10 10 0 0.35 20 10 10 0 0.35 20 10 10 0 0.35 20 10 10 0 0.35 20 10 10 0 0.35 20 10 10 10 0.35 20 10 10 0 0.35 20 10 10 10 10 0.35 20 10 10 10 10 0.35 20 10 10 10 0.35 20 10 10 10 10 0.35 20 10 10 10 10 0.35 20 10 10 10 0.35 20 10 10 10 10 0.35 20 10 10 10 10 0.35 20 10 10 10 10 10 0.35 20 10 10 10 10 10 10 10 10 10 10 10 10 10	100	
*	19.021	25 11 15 16 13 1 30 10 20 15 17 8 56 13 15	2h 10 16 0.52 13 12 7 0.65 29 9 21 0.61 18 11 7 0.63	910 913 915 916 917	531.078 631.702 631.628	23 9 15	11 18 / 28 : 21 22 8 18	0.70 0.40	1017	691-430	28 7 17 13 9 3 30 9 26 17 8 10	17	- 1	Aught and and
	14 425 20 282 20 476	10 10 20 15 12 8 26 13 15 13 13 1	18 11 / 0.83 13 10 16 0.57 12 12 0 0.82 28 9 19 0.68	917 918 920	6.31 696 6.31 656 6.32 126 6.37 501	17 10 8 11 13 1 28 8 20 22 9 13	10 10 0	1.25	1023	647.145 642.701	13 7 47	29 5 25 0.82 16 7 9 1.72 10 8 2 1.88 22 8 16 1.38 28 5 25 0.91 9 8 2 1.86		
6.7	20 633 20 800 21 168 21 163	13 13 1 24 10 20 18 12 8 23 11 13 28 10 18	18 11 7 0.63 13 10 16 0.57 13 12 0 0.69 13 12 0 0.69 13 13 0.69 13 13 0.69 13 13 0.69 13 13 0.67 13 13 0.71 16 11 5 0.71	923 928	632.555 632.986 633.038	18 10 8 13 7 27 27 8 20 13 10 6	15 9 7 12 6 26 26 7 19	1 06 1 25 0 53 0 53	1030	64 2 - 54 2 64 2 - 078	10 0 0	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	1	
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	23.488	76 1.0 16 15 12 6 20 11 9	25 4 17 0.62 16 11 3 0.80 19 10 10 0.77	44.7	435.961	18 9 9	17 8 10	1.37	1065	046 -571 046 -771 044 -886	15 # #	20 8 16 1.58 11 9 23 0.43 30 9 21 0.47 13 7 7 1.96	6.0	and the second second
81	76 561	23 10 18	25 w 17 0.42 16 11 9 0.80 19 10 10 0 77 36 8 22 0.47 18 10 3 0.86 18 10 0.47 18 10 3 0.86 18 10 0.57 19 8 22 0.52 10 17 1 0.89 21 11 0.89 21 11 0.89 22 0.52 23 15 0.73 24 0.85 25 0.85 26 0.85 27 0.85 28 0.85 29 0.85 20 0.85 2	96.7 970 976	6.36.75A 6.37.008 6.37.6.33	28 7 21	11 + 3 22 7 15 10 + 1 77 h 22	1 31 0 99 1 22 1 58 1 50 1 56 0 40 1 37 1 50 1 18 1 50 1 18	1067	643 .090 643 .200	20 7 13 76 6 20 13 3 29 26 10 16 23 10 16	29 9 23 0.50 18 8 16 1.48 25 5 21 1.21 32 6 28 0.62 26 9 17 0.62 25 9 17 0.65	100	The state of the s
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	19 187	11	12 7 1 2.06	1117	632.333	18 8 12	14 # 6	1.60	1187	141.571	N 7 27		-	N. Commission
	3 - 4a 1 3 - 973	19 10 10 18 10 8 23 8 20	10 0 11 0.75 10 0 7 0.85 10 10 1 12 11 12 12 0 7 0.85 10 0 7 0.85 10 0 7 0.85 10 0 7 0.85 11 7 0.85 12 0 7 0.82 11 7 5 7.11 17 6 12 1.86	1119	632.943 633.200 633.016	11 7 3 17 8 13 10 7 3	10 6 6	7.18	1100	680.710	43 2 44 32 7 23	36 6 28 0.76 66 1 63 0.46 32 6 29 0.90 16 6 10 2.49 30 6 26 1.08 29 8 36 1.17 27 6 22 1.36 7 3 3 2.76		
	6.167	15 10 8 14 10 a 13 10 4 12 10 2 12 8 4 18 7 11	15 9 3 0.51	1130 1131 1132	656.712	18 6 10 28 6 28 33 6 30 22 3 17	15 1 11 27 3 25 12 3 28 21 4 18	2.59 2.27 1.02 0.65	1190	660.966 661.126	15 5 11 30 7 23 29 7 23 27 7 21	30 6 36 1.08 29 6 36 1.17 27 6 22 1.36 7 1 3 2.76		And the second s
84	6.810 6.762	12 10 2 12 8 6 18 7 11 51 5 27 26 6 18 10 7 25 17 7 11 27 6 18 18 7 25 18 5 25 6 8 2	12 * 1 0.42 11 7 3 2.11 17 6 12 1.88	1135	654 .425 654 .718 654 .896	9 7 3	13 3 29 21 4 18 8 8 2 33 7 29	0.65 1.77 2.56 0.52	1196	881.364	8 8 2 25 7 19 28 3 21	7 5 3 2.76 25 6 20 1.53 23 2 22 1.08	1	
0.0	7 154	26 6 LB 10 1 25	11 7 5 2 18 17 6 12 1 88 30 0 28 0 79 23 5 19 1.49 19 0 28 0 0.88 14 0 10 1.56 13 0 10 2.86 15 17 1.56 15 0 10 2.86 21 5 17 1.66 21 5 17 1.66 22 5 17 1.66 23 5 17 1.66 23 5 17 1.66 24 5 17 1.66 25 5 17 1.66 27 5 17 1.66 27 5 17 1.66 28 7 1 7 1 7 1	1139	633 - 386	21 3 17	35 / 29 27 2 26 50 7 23 20 4 16 28 7 21	0.55	1197	461.591	26 7 17 21 15 21 8 18	25 6 20 1.55 23 2 22 1.05 26 6 18 1.62 21 6 16 1.62 20 3 17 1.92 13 4 10 2.58 19 8 16 1.90 18 6 12 1.92 43 1 43 0.83	- 1	
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45	1.231	13 7 7 15 8 7 14 8 5	13 8 8 7.30	1177 1177 1178	659.573 659.606 659.663	10 6 6	16 4 12 4 5 5 15 4 12 45 1 45	2.68 2.41 0.62	1251	063 118 063 698 865 789	10 3 3 18 6 12 60 0 60	9 4 8 2.56 15 3 13 2.44		The Property of
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- 7	1991 1996 2000 2001 2002	713.233 15 1 15 38 2 36 713.333 12 3 9 713.652 22 2 20 713.996 17 1 17 714.026 3 1 12 72 2 5 26 5 21		333 717	15 7 14 4 2 7 8 5 7 7.132 17 2 16 7.281 21 1 21 19 2 18 7.427 15 4 12 34 3 31	15 3 13 4.46 4 3 1 1.60 9 2 8 2.62 17 3 15 4.64 22 0 22 6.77 16 3 17 4.26 14 5 9 0.86 34 4 3r 1.69	2084 2085 2086 2089 2090	721.522 721.597 721.765 722.061 722.226	26 0 26 24 3 21 27 1 27 35 2 36 18 1 17 5 2 4 22 3 19 36 0 36	27 1 27 5.54 24 4 20 3.54 28 0 28 5.21 35 3 33 1.12 19 2 18 3.48 6 3 3 2.53 22 4 18 3.65 36 1 33 0.76

TABLE VI-Continued

							TABL	E VI—	Continued	ı						1
Line No.	Observed Varequater (cm ⁻¹ vac.)	Lower State	Uppe J	r State	1,	Line No.	Observed Vave quality (cm ⁻¹ vec.)	Lower State $x' = x'_0 - x'_1$	Upper State $J' = K_{\underline{a}}^{\prime} - K_{\underline{a}}^{\prime}$	1 _p	Line Ro.	Observed Vavenaber (cm ⁻¹ vac.)	Lorer State	Spper State L _p	d	Company of the property of the party of the
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7096	722.626	31 3 29 29 2 28 25 3 23 11 1 11	31 30 25	4 26 3 35 4 28 4 28 1 29 4 22	0.66 3.68 6.89 6.89 2.35 0.86 2.76 1.66 1.61 1.66 1.61 1.66 1.61 1.66 1.61 1.66 1.61 1.66 1.61 1.66 1.61 1.66 1.61 1.66 1.61 1	2128 2129 2130 2130	725 0 726 . 111 726 . 293 726 . 407 726 . 865	36 6 32 36 0 36 30 1 29 35 1 35 36 6 30	36 5 31 35 1 35 51 2 30	1.06 2.95 2.53 2.66	2160	729.836 729.931	17 4 14 16 4 12 40 0 40	18 5 13 3.47 17 5 13 3.49 16 5 11 1.46 41 1 41 1.49 15 5 11 3.44 14 5 9 3.37 13 5 9 3.25	100	-
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TABLE VI_Continued

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2552	771.070	24	10	84	24	1.1	6.3	0.67	265.1	783.563	1.3	1.0	*	14	1.1	3	1-67			95	16	12		3.7	1.3	5	0.85	
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2535	771.507	25	,		24	*	18	1.32	2654 2657	786 - 165 786 - 354	14	10	15	35	11	*	1.55	2784	801.3	6.7	17	11	1.7	28	12	14	0.60	
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2567	772.360	24	,		27		20	1 - 21	286.7	781 - 908	24	,	15	25	10	18	1.01	2791			78	8.1	13	29	12	18	0.46	
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20-18	779.384	26		1.0	27		19	1.09	2731	793.622	4.7	4.4	7	18	4.2		1.06	1861	701.818	26			*** **	in th			ttet g	alme
26.70	779.592	4.7			18	10		1.63	2735	796 263	27	10	18	78	11	1.7	0.62			1.7		15	18	2 14		1.50		
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Ze 28	780 . 790	28		20	24		21	0.86	7.94-3	195.367	19	1.1		20	1.2		0.97	5509	730.214	4.5			44	1 45		0.51		
29.31	781.157	19		11	20	10	10	1.45	2.76a	795 499	29	10	200	30	1.1	19	0.51			36				to th		***	1585	
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26 30 26 37	781.511	29	٠	5.5	30		21	0.77	2750	790 .878	21	11	11	2.7	1.7	4.0	0.80			1.6	7	٠	1.7	# 10		2.35		
26.37	781.897	20		8.3	2.4	10	1.7	1.30	2767	797.830	2.2	8.1	1.1	2.3	4.7	1.7	0.74											
26.38	781.976	6.6	10	7	1.7	8.5		1.77	2784	797.865	13	1.7	. 2	36	1.3	. 1	1.00											
7664	PB2.452	2.4		1.3	2.2	10	1.7	1 27	276.6	798 189	2.3	1.1	1.3	24	1.1	1.2	0.69											
284.5	782.777	1.7	10	*;	1.3		1	1.70	2770	798.653	44	17	1		13		0.50											

Interpretation of the v2 Structure of Ozone

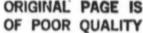
Table VI gives the measured positions and assignments of some of the structure observed for the ν_2 band. The basis of these assignments was the computed spectrum for the ν_2 band. For these computations the energy levels for the ground state were obtained by using 107 microwave lines with $J \leq 56$ and $K_{-1} \leq 8$. The ν_2 state levels were also calculated by making use of the microwave data discussed in Section I and of the band center already described in Section II. The relative intensities of the lines are also important parameters and they were calculated according to the formula

$$I_r = S[\exp(-W/kT) - \exp(-W'/kT)],$$

where S is the calculated line strength of the transitions, W and W' are the energies of the ground state and z_2 state levels, respectively, and T is the temperature, set to 300 K. Table VI includes I_r values also. About 950 lines in the observed spectrum have been identified in this manner with J values up to 46 and K_{-1} values up to 14 with $I_r \geq 0.4$ which corresponds to 1/18 of the intensity of the strongest line. The spectrum itself showed several other weaker features which were all documented by Hoh (13). In fact, the serial numbers given in the first column of Table VI were taken from Hoh's compilation.

⁵ Details of these calculations are available in Ref. (2).

⁴ In Table VI the ozone assignments were omitted whenever there was interference due to impurity lines of CO_2 . Also, the assignments are given in terms of $JK_{\alpha}K_{\epsilon}$ instead of $JK_{-1}K_{+1}$ used in the rest of the paper. Both these equivalent notations appear in the literature (see, for instance, K. Narahari Rao in Physical Chemistry (D. A. Ramsay, Ed.), Series 2 Vol. 3, p. 348, Butterworths (1976)).



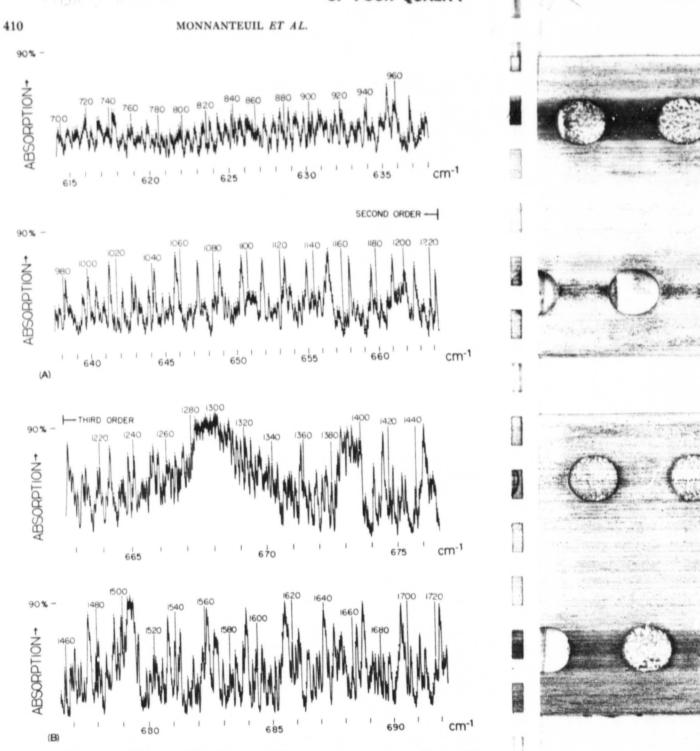
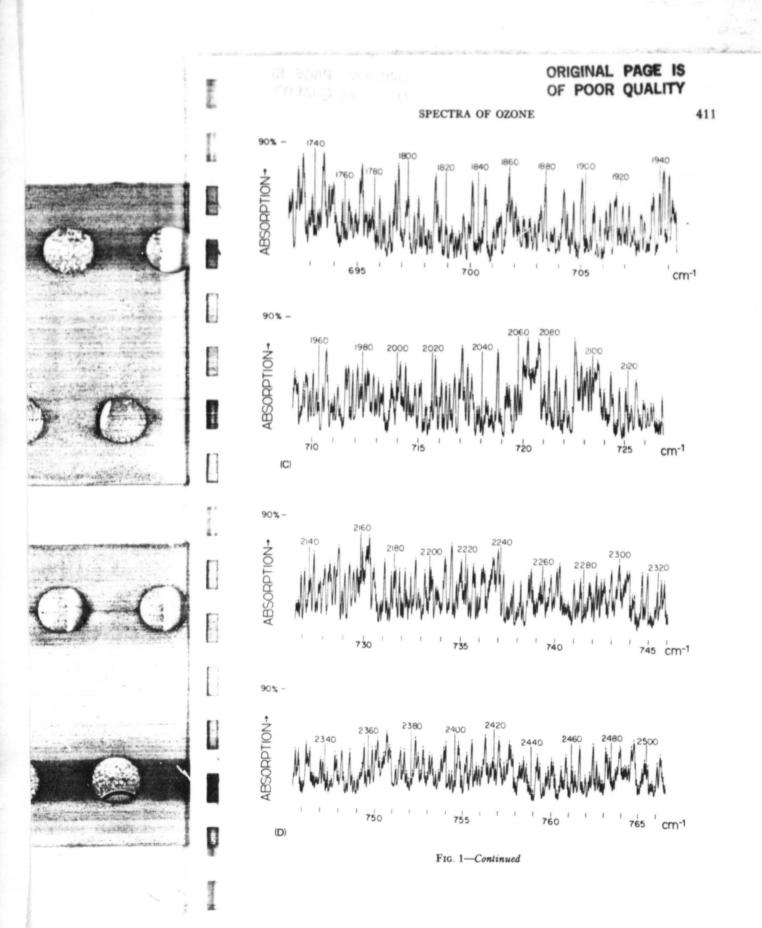


Fig. 1. ν_2 band region of ¹⁸O₃ from 615 to 806 cm⁻¹. Path length: 4 meters. Pressure: 25 mm of Hg. Ozone concentration: 50 to 35% Temperature of cell: 300 K.



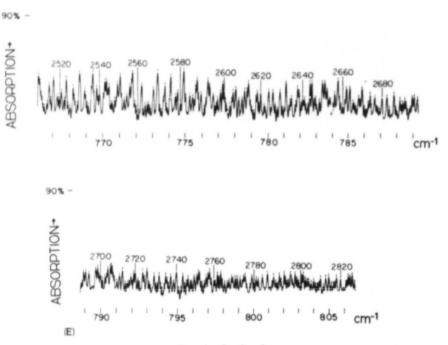


Fig. 1-Continued

Since the tunable diode laser spectroscopy in this region is progressing rapidly (14), before long it should be possible to record the ν_2 band of ozone with a resolution of 0.002 cm⁻¹ and improve on the present analysis.

ACKNOWLEDGMENTS

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We also wish to express our thanks to Dr. C. Camy-Peyret and Dr. J. M. Flaud for participating in the calculations of the $(\nu_2 + \nu_2) - \nu_2$ band.

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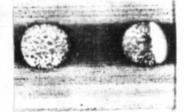
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NOTE

Diode Laser Spectra of Acetylene: ν₅ Region at 15 μm¹

In 1979, during the early studies of diode laser spectra in this laboratory (1) it became evident that useful spectroscopic information can be derived for the acetylene molecule and its isotopic species even though only limited regions can be scanned by this technique. In the present study, the Q branches of the ν_5 of $^{12}\text{C}_2\text{H}_2$ and $2\nu_5^6 - \nu_5^4$ "hot" bands of $^{12}\text{C}_2\text{H}_2$, $^{12}\text{C}^{13}\text{CH}_2$, and $^{13}\text{C}_2\text{H}_2$, all occurring in a natural sample of acetylene, were recorded and measured by using a diode laser spectrometer. For wavenumber calibration, lines from the ν_2 bands of HCN (2) and CO₂ (3) were used. The data obtained are estimated to be internally consistent to within ± 0.0005 cm⁻¹ (4). Figure 1 gives an illustration of the spectra obtained along with a small portion of the grating spectrum (5).

In obtaining molecular constants from the observational data, for the ν_3^a and $2\nu_5^a$ states of ${}^{12}C_2H_3$ we combined our data with those of Ref. (5). This resulted in improved values for the band centers and some of the rotational constants. Tables 1–III present experimental data and the molecular constants determined.

While treating the data for $\Pi - \Sigma$ bands, we found it convenient to use the following expressions: For the P and R transitions of ν_5 ($\Pi \leftarrow \Sigma$)

$$\nu(m) = (\nu_0 - B'_e - D'_e - H'_e) + (B'_e + B''_e + 2D'_e + 3H'_e)m + (B'_e - B''_e + D'_e + D''_e)m^2$$

$$+ (-2D'_e - 2D''_e - 5H'_e + H''_e)m^3 + (-D'_e + D''_e - 3H''_e)m^4$$

$$+ 3(H'_e + H''_e)m^5 + (H'_e - H''_e)m^6 \cdots$$

For above Eq. and what follows below in Eq. 2

$$m = -J$$
 for a P line

$$= J + 1 \text{ for a } R \text{ line.}$$

and

$$B_r = B_r - (1/2)(q_r - 2q_r^J + 3q_r^{JJ}),$$

$$B_f = B_r + (1/2)(q_r - 2q_r^J + 3q_r^{JJ}),$$

$$D_r = D_r - (1/2)(q_r^J - 3q_r^{JJ}),$$

$$D_f = D_r + (1/2)(q_r^J - 3q_r^{JJ}),$$

$$H_r = H_r - (1/2)q_r^{JJ},$$

$$H_f = H_r + (1/2)q_r^{JJ}.$$
(1)

The meaning of the q constants has been given by Maki and Lide (6). For the Q transitions of the ν_5 band

$$Q(J) = \{ \nu_0 + (q_T - q_T^J + q_T^{JJ}) - B_T' - D_T' - H_T' \} + (B_T' - B_T'' + 2D_T' + 3H_T') \{ J(J+1) \}$$

$$+ (-D_T' + D_T'' - 3H_T') \{ J^2(J+1)^2 \} + (H_T'' - H_T'') \{ J^3(J+1)^3 \}.$$
 (2)

From these equations we note that the band origin ν_0 has been identified with a P or R branch transition. As a result, a Q(J) transition for a $\Pi \leftarrow \Sigma$ band is essentially shifted by $(q_v - q_v^I + q_v^{IJ})$.

Support extended this research by the National Aeronautics and Space Administration is gratefully acknowledged.

NOTE

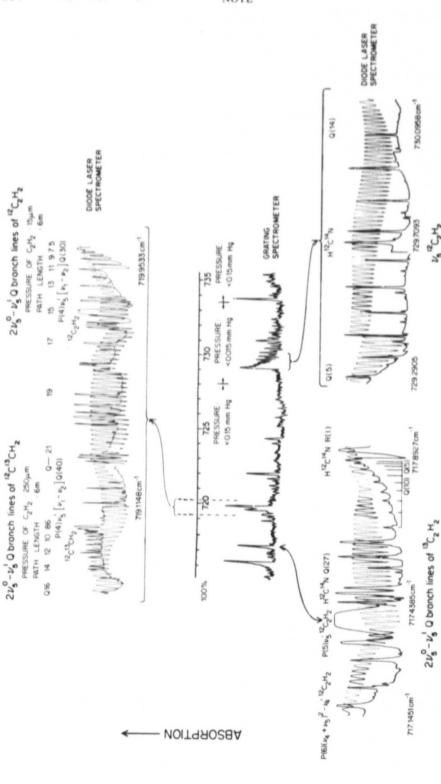


Fig. 1. Acetylene spectrum at 15 μm.

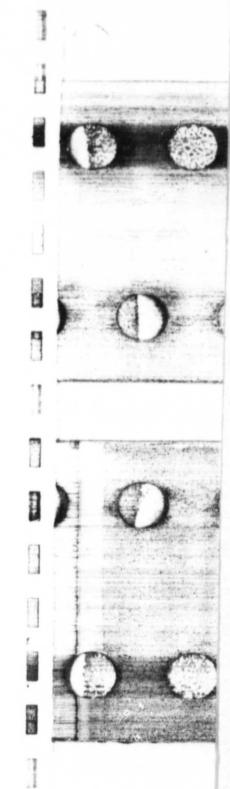


TABLE I

Wavenumbers (vac. cm⁻¹) and Molecular Constants (in cm⁻¹) for the Q Branch of the ν_5^4 Band of $^{12}\mathrm{C}_2\mathrm{H}_2$

Q(J)	*vcalculated	(0-C)×10 ⁴
Q(2)	729.1832	7
Q(3)	729.2101	- 6
Q(4)	729.2459	2
Q(5)	729, 2907	- 2
Q(6)	729.3444	0
0(7)	729.4070	0
Q(8)	729.4787	- 5
Q(9)	729.5592	1
Q(10)	729.6487	- 2
Q(11)	729.7472	* 2 2
0(12)	729.8546	6
Q(13)	729.9710	6 3 - 5
Q(14)	730.0963	- 5

$$v_0 - B_f' - D_f' + q = 729.15639 \pm 0.00019$$

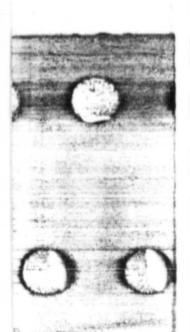
$$B_{\epsilon}' - B_{\mu}'' + 2D_{\epsilon}' = 0.0044759 \pm 0.0000018$$

TABLE II

Wavenumbers (vac. cm⁻¹) and Molecular Constants (in cm⁻¹) for the Q Branch of the $2\nu_0^6 - \nu_0^1$ Band of $^{12}\text{C}_2\text{H}_2$. $^{12}\text{C}^{13}\text{CH}_2$, and $^{13}\text{C}_2\text{H}_2$. The ν 's in columns 2, 4 and 6 are calculated values

	12	C2H2	12 _C 1	¹³ сн ₂	13 _{C2} H2				
Q(J)	v	(0-C)×10 ⁴	V	(0-C)×10 ⁴	ν	* (0-C)×10 ⁴			
Q(5)	719.9533	2			717.8709	1			
Q(6)	719.9491	1	719.0239	- 2	717.8652	- 2			
Q(7)	719.9428	1	719.0190	2	717.8570	- 3			
Q(8)	719.9341	- 2	719.0123*		717.8457	2			
Q(9)	719.9224	- 2	719.0034	0	717.8304	4			
Q(10)	719.9069	- 3	718,9918	0	717.8105	1			
Q(11)	719.8869	0	718.9768	0	717.7851	- 1			
Q(12)	719.8618	- 4	718.9577	- 1	717.7533	- 3			
Q(13)	719.8308	6	718.9338	- 1	717.7144	- 2			
Q(14)	719.7932	6	718.9641	2	717.6675	3			
Q(15)	719.7481	- 3	718.8678	0	717.6119	2			
Q(16)	719.6948	0	718.8237	0	717.5468	- 2			
Q(17)	719.6327	- 4							
0(18)	719.5609	2							
Q(19)	719.4789	4							
Q(20)	719.3862	- 2							
Q(21)	719.2822	- 5							
Q(22)	719.1667	3							

	12 _{C2} H2	¹² C ¹³ CH ₂	13 _{C2} H2
v _o + B _f " + D _f "	719.95970 ± 0.00037	719.03330 ± 0.00038	717.87953 ± 0.00055
B _e ' - B _f " - 2D _f "	(- 1.072 ± -4 0.064)x10-4	(-1.345 ± -4	(- 1.54 ± -4 0.15)×10
De' - Df"	(3.535 ± 0.029)x10 ⁻⁶	(2.119 ± 0.070)×10 ⁻⁶	(4.55 ± 0.12)×10-6
H _e ' - H _f "	(1.283 ± 0.036)x10 ⁻⁹	0.15)×10 ⁻⁹	(2.29 ± 0.25)×10 ⁻⁹



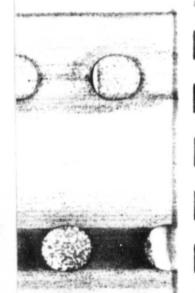


TABLE III

Molecular Constants of ¹²C₂H₂ (in cm⁻¹)

$$v_o(0000^01^1 - 0000^00^0] = 730.33281 \pm 0.00019$$
 $v_o(0000^02^0 - 0000^01^1) = 718.78334 \pm 0.00037$
 0000^00^0 0000^01^1 0000^02^0
 $B_o = 1.176608 \pm 0.000014^*$ $B = 1.178768 \pm 0.000023$ $B = 1.181008 \pm 0.000014$
 $D_o = (1.610 \pm 0.007) \times 10^{-6^*}$ $D = (1.633 \pm 0.013) \times 10^{-6^*}$ $D = (5.190 \pm 0.031) \times 10^{-6^{**}}$
 $Q_v = 0.004711 \pm 0.000023^*$ $H = (1.283 \pm 0.036) \times 10^{-9^{**}}$

In the case of the $2\nu_5^0 - \nu_5^1$ band ($\Sigma \leftarrow \Pi$), P and R transitions have been represented by

$$\nu(m) = (\nu_0 + B_e^n + D_e^n + H_e^n) + (B_e^n + B_e^n + 2D_e^n + 3H_e^n)m + (B_e^n - B_e^n - D_e^n - D_e^n)m^2 + (-2D_e^n - 2D_e^n + H_e^n - 5H_e^n)m^3 + (-D_e^n + 3H_e^n)m^4 + 3(H_e^n + H_e^n)m^5 + (H_e^n - H_e^n)m^6 \cdots (3)$$

In this case, the Q branch transitions would be given by

$$Q(J) = [\nu_0 - (q_r'' - q_r''^J + q_r''^J) + B_f'' + D_f'' + H_f''] + (B_r' - B_f'' - 2D_f'' - 3H_f'')J(J+1)$$

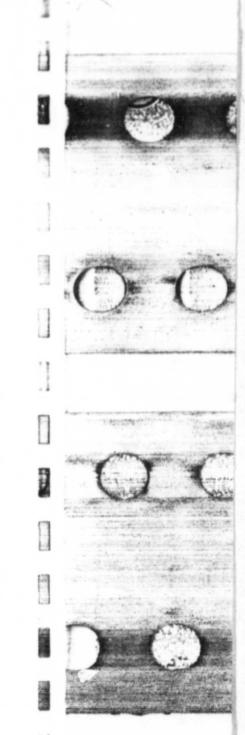
$$+ \{-(D_r' - D_f'') + 3H_f''\}J^2(J+1)^2 + (H_r' - H_f'')J_3(J+1)^3. \quad (4)$$

The reliability of the molecular constants given in Tables I-III and the validity of the equations (1)-(4) were checked by calculating the wavenumbers of some P(J) and R(J) lines which were also recorded in this work. The observed wavenumbers for P(4) and P(5) of the ν_5 band of $^{12}C_2H_2$ are 719.7408 and 717.3875 cm⁻¹, which agree well with our calculated values of 719.7416 and 717.3873 cm⁻¹, respectively. Similarly, for the $2\nu_5^0 - \nu_5^1$ band, the observed wavenumbers of P(1) and P(3) lines are 717.6070 and 729.4612 cm⁻¹ and the corresponding calculated wavenumbers are 717.6069 and 729.4611 cm⁻¹.

In conclusion, it may be mentioned that the astrophysical interest in these bands of C_2H_2 was pointed out by Ridgway (7), who identified them in the infrared spectrum of Jupiter. Measurements of intensities of these 15- μ m lines of acetylene have been deferred until we have access to another diode laser operating at these wavelengths because the present one ceased to function after the recording of the above spectra.

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^{*}Values taken from Ref. (5).

^{**} Represent only effective constants

NOTE

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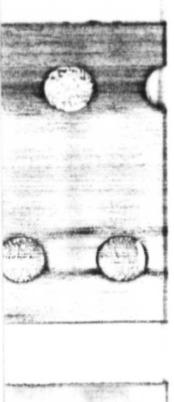
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Acetylene Spectra with a Tunable Diode Laser: $(\nu_4 + \nu_5)^{0+} - \nu_4^{1f} Q$ Branches of $^{12}\text{C}_2\text{H}_2$ and $^{12}\text{C}^{13}\text{CH}_2$

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The rotational structure of the Q branches of the $(\nu_4 + \nu_5)^{0+} - \nu_4^{1/}$ bands of $^{12}\text{C}_2\text{H}_2$ and $^{12}\text{C}^{13}\text{CH}_2$ at 13.7 μm has been observed in a natural sample of acetylene by using a tunable diode laser as a source in a high-resolution infrared grating spectrometer equipped with a precision grating drive. Altogether 23 lines from J=6 to 28 for $^{12}\text{C}_2\text{H}_2$ and 15 lines from J=6 to 20 for $^{12}\text{C}^{13}\text{CH}_2$ have been identified. The observed full width at half maximum of the resolved lines of these Q branches is very close to the calculated Doppler width. Molecular constants ν_0+B'' , B'-B''-2D'', D'-D'', and H'-H'' have been derived from the measured line positions of the rotational structure.

I. INTRODUCTION

The objective of this paper is to report the observation and interpretation of tunable diode laser data for the rotational structure of the Q branches of the $(\nu_4 + \nu_5)^{0+}(\Sigma^+) - \nu_4^{10}(\Pi)$ band of $^{12}C_2H_2$ and $^{12}C^{13}CH_2$ at 13.7 μ m. In recent years diode lasers have been in use for the studies of molecular spectra at Doppler-limited spectral resolution. The present paper not only confirms these findings but also provides an example to demonstrate the sensitivity of the technique by observing the "hot" band of the carbon-13 variety of acetylene occurring in the natural sample. The previous grating data obtained with a conventional source did not have even an indication of this $^{12}C^{13}CH_2$ Q branch.

A large amount of experimental data on the high-resolution infrared absorption spectra of acetylene ($^{12}C_2H_2$) and its isotopic species using conventional thermal radiation sources such as a carbon rod furnace have been obtained over the past several years (see, e.g., Ref. (I)). In particular, in the spectral region near 13.7 μ m the infrared active $\nu_5^4(\Pi)$ band of $^{12}C_2H_2$ and its five "hot" bands, $2\nu_5^{0.2}-\nu_5^1$ and $(\nu_4+\nu_5)^{0+,0-,2}-\nu_4^1$, have been studied in considerable detail by Palmer et al. (2) at a spectral resolution of ~ 0.04 cm⁻¹. In the present study the high intensity and the narrow width of the modes of the PbSnTe tunable diode laser have made it possible to resolve and analyze the structure of the two

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Q branches of the acetylenes ${}^{12}C_2H_2$ and ${}^{12}C^{13}CH_2$ occurring in the spectral region 713.5–716.5 cm $^{-1}$, both in a commercial sample of acetylene.

II. EXPERIMENTAL DETAILS

The experimental technique employed in the present investigation has been described in detail elsewhere by Reddy et al. (3). The tunable diode laser assembly used was supplied by Laser Analytics, Inc. A closed-cycle cryogenic helium compressor unit cooled the crystal diode, a cryogenic temperature stabilizer maintained the required temperature in the range 10–100 K in the diode, and a laser control module unit enabled precise current tuning in the laser thereby producing different laser modes or sweeping a given laser mode continuously through its lasing region.

A 2-m focal length Czerny-Turner vacuum spectrometer equipped with a 31grooves/mm echelle grating having a ruled area of 25 × 12.5 cm² was used for the laser mode separation. With the help of a precision drive, the grating position could be set to match one of the laser modes and the speed of rotation of the grating conveniently adjusted so that it would efficiently synchronize with the sweep rate of a particular laser mode. A wedge-shaped KBr window was used as a beam-splitter after the exit slit of the spectrometer. The transmitted beam from the beam-splitter reached a liquid-nitrogen-cooled HgCdTe detector after passing through the absorption cells. The reflected beam from the front surface of the beam-splitter reached a second liquid-nitrogen-cooled HgCdTe detector after passing through an air-spaced etalon 30.09 cm long in order to produce a Fabry-Perot fringe pattern. The air-spaced etalon consisted of two CdTe plates mounted on the parallel ends of a quartz tube spacer. The absorption spectrum and the Fabry-Perot fringes were simultaneously recorded on a chart paper with a two-pen recorder. The fringe spacing of the etalon in the spectral region under study was 0.016548 ± 0.000030 cm⁻¹. The line positions of the acetylene spectra relative to the absorption lines of the reference spectra were measured as a function of fractional fringe position. Spectra of acetylene in the region 715.0-716.5 cm⁻¹ were recorded with an absorption cell 6 m long using gas pressures in the range $50-170 \mu m$ of Hg. In the spectral region investigated, four lines Q(21)-Q(24) of ν_2^1 band of H¹²C¹⁴N with a sample path length of 20 cm and a pressure of about 1 mm of Hg and two lines $(\nu_1 - \nu_2^{i})P(7)$ and ν_2^1 R(60) of ${}^{12}C^{16}O_2$ with a sample path length of 1.4 m and a pressure of about 2 mm of Hg were also recorded. The wavenumbers of the acetylene lines were measured relative to the four Q lines of H¹²C¹⁴N, the positions of which were calculated from the rotational constants given by Maki (4, 5) and the vibrational band origin given by Yin and Rao (6). The calculated wavenumbers of Q(21), Q(22), Q(23), and Q(24) of the ν_2^1 band of $H^{12}C^{14}N$ are 715.3307, 715.6474, 715.9779, and 716.3224 cm⁻¹, respectively. The separations $\Delta \nu = Q(J+1) - Q(J)$ thus calculated agree within a few ten-thousandths of 1 cm-1 with those obtained experimentally in Ref. (6) and by Wang and Overend (7). It is to be noted that the band origin of the ν_2 band of H¹²C¹⁴N determined by Yin and Rao is 0.002 cm⁻¹ less than that determined by Wang and

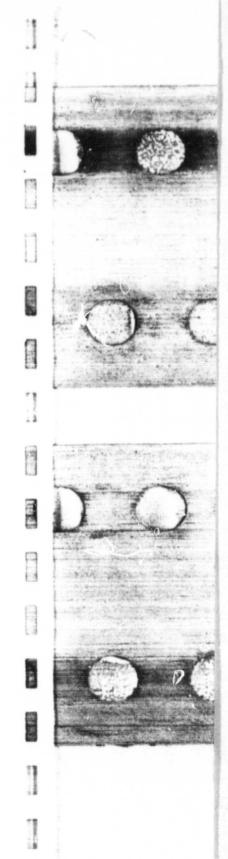


TABLE I

Wavenumbers (in cm⁻¹ vac.) of the Q branches in the $(\nu_4 + \nu_5)^{0+} - \nu_4^{1f}$ bands of $^{12}C_2H_2$ and $^{12}C^{13}CH_2$

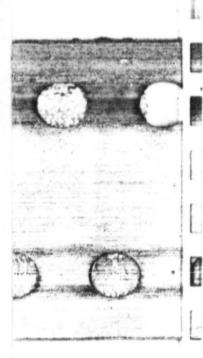
		12 _{C2}	42	12 _C 13 _{CH2}						
J	Observed Q(J)	Observed &v	(Avobs Avcalc.)	Observed Q(J)	Observed &v	(Avobs. 4 Avcalc.)				
6	716.3752	-0.0035	-0.6	/15.4805	-0.0033	-3.3				
7	. 3717	-0.0048	0.0	.4772	-0.0039	1.5				
8	. 3669	-0.0066	-1.1	.4733	-0.0052	2.1				
9	. 3603	-0.0084	1.6	.4681	-0.0070	0.7				
10	. 3519	-0.0109	1.5	.4611	-0.0089	1.7				
11	. 3410	-0.0141	-1.2	.4522	-0.0111	3.2				
12	. 3269	-0.0172	2.1	.4411	-0.0144	-2.1				
13	. 3097	-0.0215	-1.4	.4267	-0.0177	-3.1				
14	. 2882	-0.0262	-3.5	.4090	-0.0215	-4.5				
15	.2620	-0.0307	2.2	. 3875	-0.0251	1.1				
16	. 2313	-0.0363	3.1	. 3624	-0.0293	6.1				
17	. 1950	-0.0426	3.2	. 3331	-0.0362	-10.2				
18	. 1524	-0.0500	-1.0	. 2969	-0.0402	8.4				
19	.1024	-0.0578	-2.5	. 2567	-0.0477	-1.6				
20	.0446	-0.0655	-5.7	.2090						
21	715.9781	-0.0752	-2.5							
22	. 9029	-0.0844	3.1							
23	.8185	-0.0952	0.2							
24	.7233	-0.1069	-4.3							
25	.6164	-0.1172	12.4							
26	. 4992	-0.1311	0.6							
27	. 3681	-0.1452	-5.7							
28	. 2229									

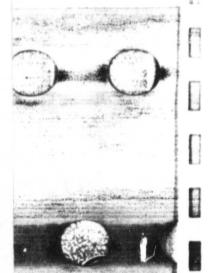
Overend and these two studies agree with one another within their respective uncertainties. Considering all aspects, the absolute wavenumbers of the acetylene lines given in Table I are believed to have an uncertainty of about $\pm 0.003~\rm cm^{-1}$ at the most. The wavenumbers of $(\nu_1-\nu_2^1)~P(7)$ and $\nu_2^1~R(60)$ of $^{12}\rm C^{16}\rm O_2$ calculated in a similar way are 715.3195 and 716.4285 cm $^{-1}$, respectively, whereas the values determined by Gordon and McCubbin (8) are 715.321 and 716.426 cm $^{-1}$, respectively. In the spectral region 715.0–716.5 cm $^{-1}$, in addition to the two Q branch lines presented in this paper, five absorption lines of acetylene were also observed. One of these lines was unambiguously assigned as the P(2) line of $2\nu_5^0-\nu_5^1$ band of $^{12}\rm C_2H_2$. The estimated wavenumber of this line in the present work is 715.2645 cm $^{-1}$ as compared to the value of 715.262 cm $^{-1}$ reported in the previous work (2).

III. RESULTS

Observed Spectra

A major portion of the spectrum observed for the Q branch of the $(\nu_4 + \nu_5)^{0+}(\Sigma_u^+) - \nu_4^{1/}(\Pi g)$ band of $^{12}C_2H_2$ is reproduced in Fig. 1. The vibrational assignment of this band was made with the help of the previous work (2). As $^{12}C_2H_2$ has a center of symmetry (point group $D_{\infty h}$) and the nuclear spins of





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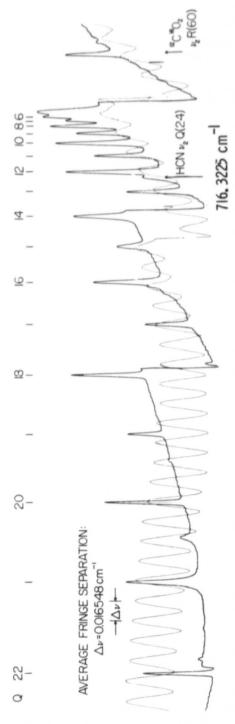
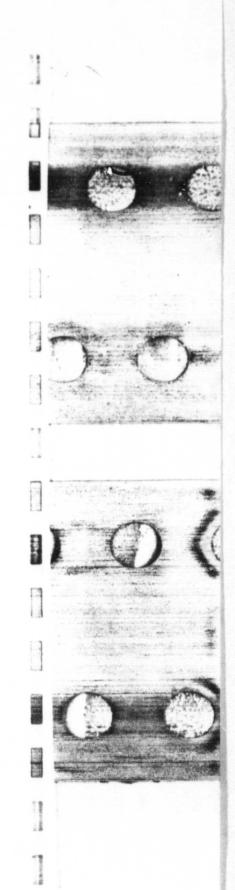


Fig. 1. A major portion of the observed Q branch of the $(\nu_4 + \nu_5)^{\nu_1}(\Sigma_u^*) - \nu_4^{\nu_1}(\Pi_g)$ band of ${}^{12}C_2H_2$. Also shown are the Q(24) line of the ν_2^4 band of ${}^{12}C^{16}O_2$. (Sample path length: 6 m; pressure of gas: 70μ m of Hg).



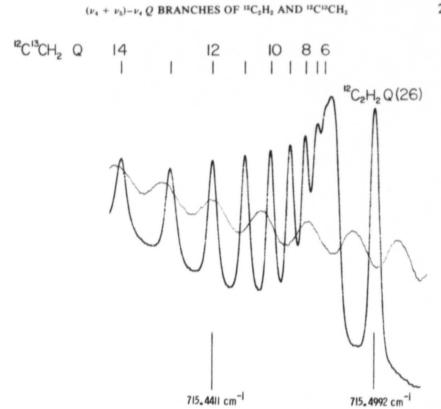


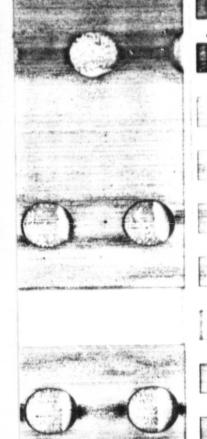
FIG. 2. A portion of the observed Q branch of the $(\nu_4 + \nu_5)^{0+}(\Sigma^+) - \nu_4^{1/4}(\Pi)$ band of $^{12}\text{C}^{13}\text{CH}_2$. Also shown is the Q(26) line of the corresponding band of $^{12}\text{C}_2\text{H}_2$. (Sample path length: 6 m; pressure of gas: 160 μ m of Hg).

 12 C and H atoms are 0 and $^{1/2}$, respectively, the rotational levels with even and odd J of the 0001^{10} 0 state have statistical weights 3 and 1, respectively (9). In fact, the observed spectrum shows such an alternation of intensity in the rotational structure of the Q branch (Fig. 1). In assigning the J numbers to the rotational structure the calculated line positions and the intensity alternation within

TABLE II $\label{eq:Molecular constants*} \mbox{ (in cm$^{-1}$ vac.) for the } (\nu_4 + \nu_5)^{0+} - \nu_4^{1/} \\ \mbox{ bands of } {}^{12}C_2H_2 \mbox{ and } {}^{12}C^{13}CH_2 \\$

Molecule	v ₀ + B"	B'-B"-2D"	(D'-D")×10 ⁶	(H'-H")×10 ¹⁰	Std. dev. of fit
² C ₂ H ₂	716.3809 ±0.0030	-0.000068 ±0.000006	-1.828 ±0.019	1.91 ±0.02	1.9 x 10 ⁻⁵
12 _C 13 _{CH2}	715.4852 ±0.0051	-0.000076 ±0.000009	-1.39 ±0.02		3.2×10^{-5}

^{*}The values of v_0 + B" are obtained from a least-squares fit of the observed data to Eq. (1) and those of the remaining parameters are obtained from a similar fit of Δv to Eq. (2). The errors quoted are standard deviations.



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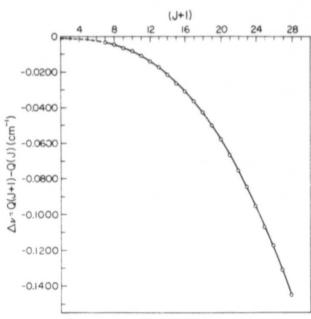


Fig. 3. A plot of $\Delta\nu$ [=Q(J+1) - Q(J)] versus (J+1) for the $(\nu_4 + \nu_5)^{0+}(\Sigma^+) - \nu_4^{1f}(\Pi_g)$ band of $^{12}C_2H_2$.

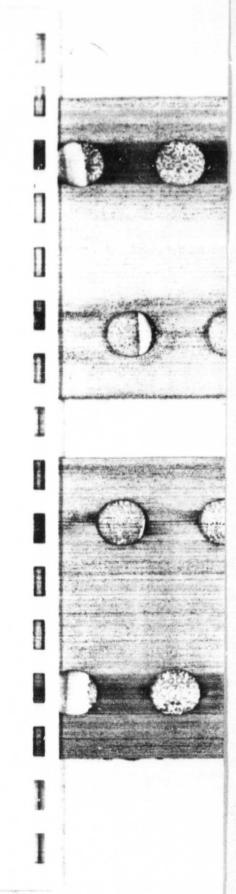
the Q branch were taken into account. Altogether 23 lines from J=6 to 28 were identified for the Q branch of this band of $^{12}\mathrm{C}_2\mathrm{H}_2$. For the resolved lines of this branch the full width at half maximum absorption is found to be 0.0025 cm⁻¹ whereas the calculated Doppler width of these lines of $^{12}\mathrm{C}_2\mathrm{H}_2$ at room temperature is 0.0017 cm⁻¹.

A portion of the spectrum of the Q branch of the $(\nu_4 + \nu_5)^{0+}(\Sigma^+) - \nu_4^{1/}(\Pi)$ band of $^{12}C^{13}CH_2$ is reproduced in Fig. 2. Although lines from J=6 to 20 are identified, only the ones up to J=14 are shown in this figure. As $^{12}C^{13}CH_2$ does not have a center of symmetry (point group $C_{\infty v}$) no alternation of intensity is expected in any given branch of a band of this molecule, and this is what in fact has been observed experimentally (Fig. 2) and was responsible for its identification. The J assignments of the rotational structure of the Q branch of $^{12}C^{13}CH_2$ were confirmed by the rotational constants of this molecule obtained in the previous studies (10, 11).

The wavenumbers (in $cm^{-1}vac$.) and the J assignments for the lines measured in this work are listed in Table I.

Evaluation of Molecular Constants

The wavenumber data were fitted by least-squares techniques to not only the relation representing the Q(J) lines but also the $\Delta\nu$ formula which can be readily represented, respectively, by the following equations:



$$Q(J) = \nu_0 + B'[J(J+1) - l'^2] - D'[J(J+1) - l'^2]^2 + H'[J(J+1) - l'^2]^3 - B''[J(J+1) - l'^2] + D''[J(J+1) - l''^2]^2 - H''[J(J+1) - l''^2]^3.$$
 (1)

Here $\nu_0 = G_0(v',l') - G_0(v'',l'')$ is the vibrational band origin, l' = 0 for the upper Σ state, l'' = 1 for the lower Π state, and the constants of the lower state refer to the f-sublevels.

$$\Delta \nu = 2[B' - B'' + 2(D'l'^2 - D''l''^2) + 3(H'l'^4 - H''l''^4)](J+1) - 2[2(D' - D'') + 6(H'l'^2 - H''l''^2) - (H' - H'')](J+1)^3 + 6(H' - H'')(J+1)^5. (2)$$

The molecular constants so derived are presented in Table II. That the relative accuracy of the data is high is clearly indicated by the fact that the observed $\Delta\nu$ values could be reproduced to within a few ten-thousandths of a cm⁻¹ by the molecular constants (see also Fig. 3). Although the J values observed in the present work are not very high, there is definitely a need to include the H terms in Eqs. (1) and (2).

ACKNOWLEDGMENTS

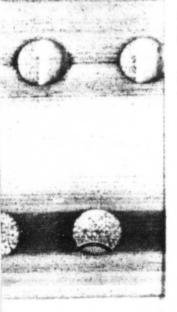
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Interpretation of the Acetylene Spectrum at 1.5 µm

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Bands of acetylene in the region 6680–6460 cm⁻¹ recorded with a high resolution infrared spectrograph are assigned to transitions and molecular constants of the levels involved have been evaluated.

INTRODUCTION

This article presents an analysis of the band systems observed in the absorption spectrum of acetylene at 1.5 μ m (between 6680 and 6460 cm⁻¹). Numerous overlapping bands of both $^{12}\text{C}_2\text{H}_2$ and $^{12}\text{C}^{13}\text{CH}_2$ occur here and the interpretation of the spectrum has been done by adopting procedures described in previous publications. These procedures involve the use of graphical plots, examination of the patterns of intensity alternation, and evaluation of approximate vibrational term values.

Figure 1 shows a reproduction of the acetylene absorption spectrum observed in the present study. The conditions of the experiment along with the identifications of the different bands are given in the caption to this figure. The effective spectral resolution of the data achieved was about $0.03~\rm cm^{-1}$. About 80% of the nearly 1000 lines recorded have been given assignments. The next section summarizes the results along with a discussion of some of the more important aspects of the spectrum.

RESULTS AND DISCUSSION

Main Features (Combination Bands and Some "Hot" Bands)

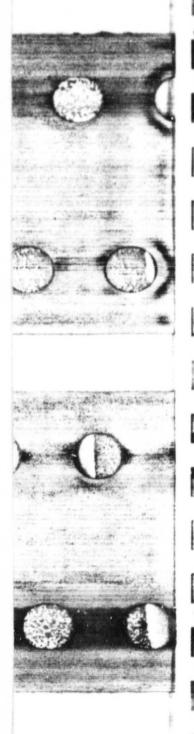
The $\nu_1 + \nu_3$ and $\nu_1 + \nu_2 + (\nu_4 + \nu_5)^0$ bands of $^{12}\text{C}_2\text{H}_2$ originating from the ground state of this molecule represent the stronger absorptions. They are the bands labeled A and Q in Fig. 1. The two bands centered at about 6529.8 and at 6534.7 cm⁻¹ are, respectively, the "hot" bands $\nu_1 + \nu_3 + \nu_4{}^1 - \nu_4{}^1$ and $\nu_1 + \nu_3 + \nu_5{}^1 - \nu_5{}^1$ of this molecular species.

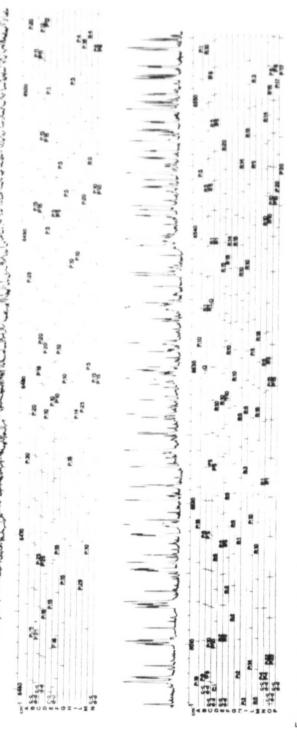
For all these four bands, the present study gives improved values for the molecular constants of the levels involved in these transitions as compared to previous work in this region (1). The improvements reflect the better technology available at this time.

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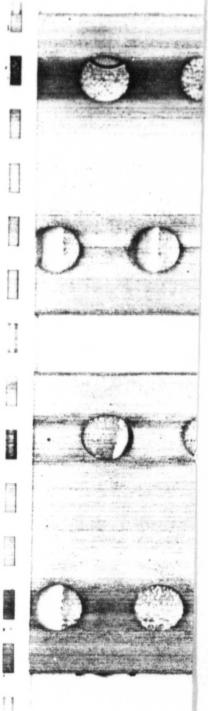




graph at The Ohio State University equipped with a 79-grooves/mm 40 × 20-cm² (16 × 8-in.²) echelle. Detector: PbS cooled to dry ice temperature. Path Fig. 1. Reproduction of the rotational structure for the acetylene bands at 1.5 μm. Spectrograph used: 10-m-focal-length Czerny-Turner vacuum spectrolength and pressure: An absorption path of 24 m with pressure of acetylene ranging between 0.2 and 10 Torr was used. The identification of the bands! marked A, B, C, D, etc., is given below (L, M, N refer to $^{12}\text{C}^{13}\text{CH}_2$ and the rest to $^{12}\text{C}_2\text{H}_2)$:

0. $2p_1 + p_4^1 - p_5^1$ P. $p_1 + p_2 + (2p_4 + p_5)_{11}^1 - p_4$ Q. $p_1 + p_2 + (p_4 + p_5)_{12}^2$	R. $p_1 + p_2 + (2p_4 + p_2)_1^4 - p_4^4$ S. $p_1 + p_2 + (p_4 + 2p_2)_1^4 - p_2$ T. $2p_3 + p_1^4 - p_2^4$	
G. $p_1 + p_3 + (p_4 + p_5)^2 - (p_4 + p_5)^2$ H. $p_1 + p_3 + 2p_5^2 - 2p_5^3$ I. $p_1 + p_3 + 2p_5^2 - 2p_5^2$	L. $v_1 + v_2$ M. $2v_3$ N. $v_1 + v_2 + v_4 - v_4$	
A. $v_1 + v_3$ B. $v_1 + v_2 + v_4^1 - v_4^1$ C. $v_1 + v_2 + v_3^1 - v_5^1$	D. $p_1 + p_2 + 2p_4^0 - 2p_4^0$ E. $p_1 + p_3 + 2p_4^2 - 2p_4^2$ F. $p_1 + p_3 + (p_4 + p_5)_+^0 - (p_4 + p_5)_+^0$	

¹ The levels $(\nu_4 + \nu_5)^{-1}$, $(\nu_4 + \nu_5)^{-2}$, $(\nu_4 + \nu_5)^2$ have, respectively, the symmetries $\Sigma_u^+, \Sigma_u^-, \Delta_u$. The subscript I refers to the upper and II to the lower level of the doublet.

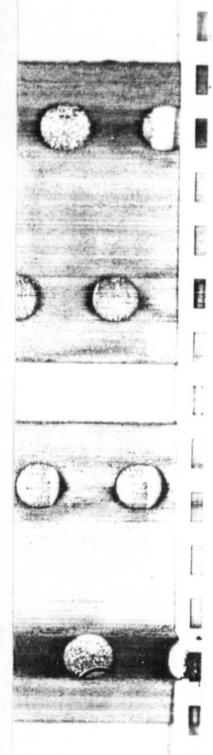


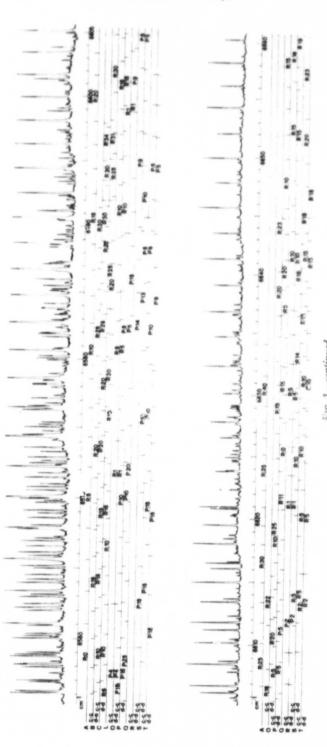
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 C_2H_2 BANDS AT 1.5 μm

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BALDACCI, GHERSETTI, AND RAO

TABLE I

Vibrational Term Values (cm⁻¹) of ¹²C₂H₂ Obtained from the Bands at 1.5 μm

LEVEL	SPECIES	G ₀ (v)(cm ⁻¹)
010000	ε.*	6556.461:2
10(11)0	ε.	6623.140+2
011100	",	7142.657±6
10(21)1 11	t _u	7207.113:9
0020011	n _u	7219.370±8
10(21)1 1	n _u	7229.448:10
010011	1 ₆	7265.060±6
001100	т,	7297.547:8
10(.2)1 11	1,	7330.531:12
U12000	, s	7732.78 ±6
012200		7737.03 ±6
01(11)0	r.*	7835.014:12
01(11)2	Δ.	7852.417:9
010020	Δ. Ε.	7961.837±8
10022	Δ _u	7976.274±9

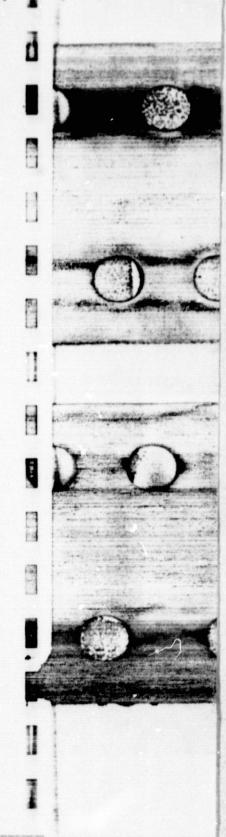
Other "Hot" Bands from the 0001'00 and 0000011 Levels

The medium intensity band at 6567.2 cm⁻¹ belongs to a transition arising from a u state as shown by the type of the intensity alternation in the R branch (no splitting was observed in the P branch): on this basis and in agreement with a vibrational calculation it was assigned to the $2001^{100} \leftarrow 0000^{0}1^{1}$ transition.

The medium-weak band at 6606.5 cm⁻¹ presents the *l*-type doublets resolved in both P and R branches with the intensity alternation for a g vibrational lower state and the $B''_{e/d}$ constants are close to those measured for the 0001 1 0 0 state [see Refs. (2,3)]. Among the possible transitions compatible by symmetry and expected to occur in this wavenumber range, the most probable one is the transition 0020° 1 $^{\circ}$ 1 \leftarrow 0001 $^{\circ}$ 0 $^{\circ}$ 1 in fact from Keller's results for $2\nu_3 + \nu_5$ 1 (4) and from ν_4 1 (2, 3) a band origin at 6605.8 cm⁻¹ is calculated.

Three additional weak bands were assigned as "hot" bands accompanying the main transition $110(11)_{+}^{0} \leftarrow 0000^{0}0^{0}$ which absorbs at 6623.1 cm^{-1} . The band at 6594.2 cm^{-1} belongs to a transition arising from the $0001^{1}0^{0}$ state and was identified as $\nu_{1} + \nu_{2} + (2\nu_{4} + \nu_{5})_{11}^{1} - \nu_{4}^{1}$. By using the results at $5 \mu \text{m}$ (2), a second band ($\nu_{1} + \nu_{2} + (2\nu_{4} + \nu_{5})_{1}^{1} - \nu_{4}^{1}$) should be located at about 20 cm⁻¹ higher, that is at about 6614.2 cm^{-1} . The only absorption observed in this range is a weak badly overlapped band at 6616.6 cm^{-1} which was tentatively assigned to this transition. In passing, it may be noted that the c and d rotational sublevels² of the upper state $[(11021)_{1}^{1}]$ appear here much closer to each other than in the Π states in general.

² It may be noted that the recommendation of J. M. Brown, J. T. Hougen, K.-P. Huber, J. W. C. Johns, I. Kopp, H. Lefebvre-Brion, A. S. Merer, D. A. Ramsay, J. Rostas, and R. N. Zare [J. Mol. Spectrosc. 55, 500 (1975)] designates the c and d labels as e and f.



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The band at $6600.2~\rm cm^{-1}$ was attributed to the $(11012)_{\rm H}{}^{1} \leftarrow 0000^{\rm o}1^{\rm o}$ transition. A value of about $6601~\rm cm^{-1}$ was estimated for the band origin on the basis of vibrational calculations. The rotational constants, as well as the type of the alternation of intensity are in agreement with such an assignment for the lower state. No band of a transition from the same lower state to the $(11012)_{\rm I}{}^{\rm I}$ level could be identified.

"Hot" Bands Arising from the 00020.200, 0000020.2 and 000(11)0.2 States

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The band at 6502.4 cm⁻¹ was assigned to the $\Sigma_u^+ \leftarrow \Sigma_g^+ 1012^{\circ}0^{\circ} \leftarrow 0002^{\circ}0^{\circ}$ transition $(\nu_1 + \nu_3 + 2\nu_4^{\circ} - 2\nu_4^{\circ})$ on the basis of the appropriate term values known from the literature (1012°0° given by Keller (4), and 0002°0° quoted by Plíva (5)). It may be recalled that although the type of alternation of intensity agrees with a g vibrational

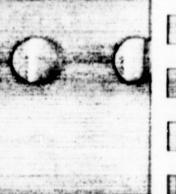
TABLE II

Spectroscopic Constants (cm⁻¹) of ¹²C₂H₂ Obtained from the Bands at 1.5 µm

LEVEL	.5	Species of			
UPPER	LOWER	States	A0-B,T,:+B,T,:	(B'-B")×103	(D'-D")x10
1010000	0000000	1.+1.	6556.461±0.002	-12.865±0.008	-0.029:0.006
(1010 ⁰ 1 ¹) _e	(0000°11)c	1,-1,	6534.742±0.004	-12.90 ±0.02	-0.01 ±0.02
(1010 ⁰ 1 ¹) _d	(0000°11) _d			-12.68 ±0.02	0.03 ±0.02
(1011 ¹ 0 ⁰) _e	(0001 ¹ 0 ⁰) _c	10-16	6529.800±0.004	-12.60 10.02	0.02 10.02
(1011 ¹ 0°) _d	(000110 ⁰) _d		.,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	-12.39 +0.02	0.02 ±0.02
1010022	2000022	4.*4g	6513.265:0.005	-12.65 ±0.09	0.5 ±0.4
1010020	0000020	1	6512.716:0.004	-12.69 to.09	-1.1 ±0.4
(10111)0	(00011)0	1	6506.927:0.007	-12.65 ±0.11	-0.6 ±0.4
(10111)2	(00011)	Ag**Au	6504.949±0.005	-12.46 ±0.09	-0.5 ±0.3
(1012°0°)	(0002-0°)			-14.67 ±0.08 ⁽⁺⁾	-3.3 ±0.2
(1012 0°) d	(0002-02)d	^u-Ag	6503.563±0.008	-12.06 ±0.08 ⁽⁺⁾	0.6 ±0.3
1012000	0002 000	Σ Σ. g.	6502.389:0.006	-9.7 to.2	2.1 ±0.9
(11011)0	0000000	: -: -:	6623.140±0.002	-8.77 ±0.02	1.98 ±0.03
(11021)1 _c (1)	(0001100) _c		4414 444-0 004	-7.03 ±0.10	-2.7 ±0.4
11021)1 _d (1)	(0001 ¹ 0 ⁰) _d	"-"g	6616.586:0.006	-8.60 ±0.05	0.8 ±0.2
11012)1 (11)	(0000°11)			-11.36 ±0.10	0.6 ±0.3
11012)1 _d (11)	(0000°11)d	ng-nu	6600.205±0.010	-5.77 ±0.10	0.6 ±0.3
11021)1 (11)	(0001100)			-10.48 ±0.05	1.07 ±0.10
11021) ¹ d(11)	(0001 ¹ 0 ⁰) _d	1,-1,	6594.251+0.007	-5.51 ±0.06	1.94 ±0.13
0020 ⁶ 1 ¹) _c	(0001 ¹ 0 ⁰) _c		4404 \$1040 OC1	-8.98 ±0.05	0.3 ±0.1
0020011) _d	(0001 ¹ 0 ⁰) _d	"u-II g	6606.510±0.006	-11.65 ±0.06	0.5 ±0.2
2001100)	(00000°11)			-14.56 ±0.04	-0.41 ±0.06
2001 ¹ 0 ⁰) _d	(0000°11) _d	Πg+Πu	6567.231±0.006	-13.96 ±0.04	-0.09 ±0.07

⁽⁺⁾ Effective values

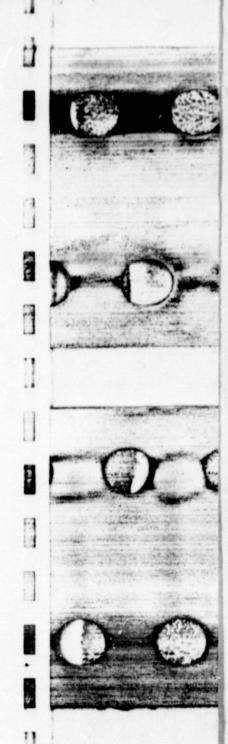




Wavenumbers (vac. cm⁻¹) of the Rotational Structure of Σ_{u(g)} + ← Σ_{g(u)} + Bands of ¹³C;H₂ (Estimated Accuracy of Data: ±0.005 cm⁻¹) TABLE III

Colorado	# # # # # # # # # #	F(J) 038. F(J)		0.000	110(11)0	110(11)0 - 0000000	101200	1012600 - 0002600	101(11)0	101(11)0 - 000(11)0	1010020	1010520 - 0000020	
6551-351 655	6551-731 6552-409 655	6551-781 6552-409 655	J R(J)08S.	P(J)08S.	R(J)OBS.	P(J)OBS.	R(J)08S.	P(J)085.	R(J)08S.	P(J)085.	R(J)OBS.	P(J)085.	7
6562-149 656	6563-714 6502-718 6502-718 6502-718 6502-718 6502-714 650	685-821 8058-119 6018-129 6010-139 6010	6558.798		6625.489								9
Color	Column C	665.242 (1972) (1 6561.087	6117-5559	6627.789	9620.800	6507.056	6500.044			6517.363	6510.342	
655, 241 655, 241 655, 243 655, 244 655, 243 655	Color Colo	Color	6363.374	*17. 98.00	111 0000	6518.423		1/9./679	6513.846	***********	9519.655	194.700	N# 1
10,000 1	6570.000 6544.442 6546.85 4.642.1185 6542.35 6550.377 655	657, 205 654, 442 645, 85 4, 641, 118 651, 118 6	4 6567.842	8545 878	6634. 648	6619 634	6511.620	6495.243	6516-136	197.6640	6521.911	6305.336	
6575.205 6541.207 6665.207 6665.207 6665.207 6665.207 656	6572.205 6561.205 6562.305	657,2705 6641,842 6661,842 6661,327 6662,324 6516,337 6662,324 6516,324 6516,324 651	5 6570.050	6544.442	6636.874	6611.185	4516.136	64.00 175	6530.5727	08 7077	6536 353	*****	• •
6578, 213 6578, 213 6570, 213 6570, 213 6570, 213 6570, 213 6570, 214 6570, 203 <t< td=""><td>6578-378 6551-378 6551-321 6550-377 6551-324 655</td><td>6578-373 6551-323 6661-323 6661-324 6552-545 6552-575 655</td><td>6 6572.205</td><td>6541.942</td><td>6639.107</td><td>5608. 756</td><td>6518.336</td><td>6487.9257</td><td></td><td></td><td>6538 5017</td><td>091 80.79</td><td></td></t<>	6578-378 6551-378 6551-321 6550-377 6551-324 655	6578-373 6551-323 6661-323 6661-324 6552-545 6552-575 655	6 6572.205	6541.942	6639.107	5608. 756	6518.336	6487.9257			6538 5017	091 80.79	
6578-581 6518-785 660-1-302 6520-999 6600-1-302 6520-999 6600-1-302 6520-999 6530-99	6578-581 6518-785 660-1-302 6522-399 6600-1-302 6532-399 6532-399 6532-399 6532-399 65	6578-31 651-37 666-32 6601-105 6521-99 6481-344 6527-01 6481-344 6527-99 6532-399 65	7 6574.378	6539.455	6641.321	6606.284	6520.577	6485.424	6524.899	816.6879	6530.693	6495.664	
6550-540 6531-745 9645-749 6460-105 6524-99 6460-114 6446-799 6534-749 6460-114 6446-799 6534-749 6460-114 6546-749 6462-749 6534	6550-540 6531-745 98-5-548 6601-305 6524-399 6460-414 6448-779 9584-719 6550-540 6531-740 6548-719 6550-540 6531-740 6548-719 6552-400 6531-740 6548-719 6552-400 6531-740 6553-740 6531-740 653	6550-540 6531-745 684-7745 684-7745 684-7745 684-7745 684-7745 684-7745 685-874 6550-540 6531-745 685-874-745 685-874-745 685-874-745 685-745	8 6576.471	6536.905	667.5799	6603.822		6482.944	6527.017	6487.347	6532.799	6493.108	
6550-540 6531-758 6667-868 6558-613 6527-052 6477-831 6511-221 6468-235 6458-344 6531-758 6667-868 652-022 6559-646 6531-758 6667-231 6525-318 6527-322 6531-232 6457-323 6531-232 6457-323 6532-323 6531-232 6457-323 6532-323 6531-232 6457-323 6532-323 6531-232 6457-323 6532-323 6531-232 6457-323 6532-323 6531	6552.480 6513.784 6667.88 658-413 6517.052 6473.311 6469.494 6517.251 6443.205 6459.494 6517.251 6459.494 6517.251 6459.494 6517.251 6459.494 6517.251 6459.495 6459.405 6525.405 6552.	6582-660 6531.768 6642.789 6542.789 6531.215 6438.235 643	9 6578.581	6534.367	8645.685	6601.305	6524.939	6480.414		6484.789	6534.933	6490.567	
CASE-186 6521-173 6647-173	6554-113 6512-255 6417-802 6552-113 6586-315 6417-255 6417-802 6552-315 6585-315 658	6554-113 6512-255 6419-402 6552-113 6512-255 6419-402 6552-115 652-205 6419-402 6521-115 6521-255 6419-402 6521-115 6521-255 6419-402 6521-115 6521-255 6419-402 6521-115 6521-255 6419-402 6521-115 6521-255 6419-402 6521-115 6521-255 6419-402 6521-115 6419	0 6580.540	6531.768	16647.808	6598.813	6527.062	6477.893	6531.221	6482.205	6536.994	6488.024	10
CSB-173 CSB-173 <t< td=""><td>COSE-1731 COSE-1732 COSE-1733 COSE-1734 <t< td=""><td>6556-573 6522-588 6552-020 6559-1225 6547-120 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-305 65</td><td>1 6582.680</td><td>6529.173</td><td>876 6799</td><td></td><td></td><td>6475.311</td><td>6533.255</td><td>6479.602</td><td>6539.065</td><td>6485.375</td><td>=</td></t<></td></t<>	COSE-1731 COSE-1732 COSE-1733 COSE-1734 COSE-1734 <t< td=""><td>6556-573 6522-588 6552-020 6559-1225 6547-120 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-305 65</td><td>1 6582.680</td><td>6529.173</td><td>876 6799</td><td></td><td></td><td>6475.311</td><td>6533.255</td><td>6479.602</td><td>6539.065</td><td>6485.375</td><td>=</td></t<>	6556-573 6522-588 6552-020 6559-1225 6547-120 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-100 6555-305 6542-305 65	1 6582.680	6529.173	876 6799			6475.311	6533.255	6479.602	6539.065	6485.375	=
659-59 651-203 666-218 658-59 651-304 656-218 658-59 640-112 651-33 640-116 651-33 656-218 656-218 651-33 6	C58-539 6521-329 6521-324 6470-312 6437-327 6437-327 6437-327 6442-348 6548-310 6447-324 6548-310 6447-324 6548-310 6447-324 6548-310 6447-324 6548-310 6447-324 6548-310 6447-324 6548-310 6447-324 6548-310 6447-324 6548-310 6447-324 6548-310 6447-324 6548-310 6447-324 6548-310 6447-324 6548-310 6447-324 6548-310 6447-324 6548-310 6447-324 6548-310 6447-324 6468-310 6447-324 6468-310 6447-324 6468-310 6447-324 6468-310 6447-324 6468-310 6447-324 6468-310 6447-324 6468-310 6447-324 6468-310 6447-324 6468-310 6447-324 6468-310 6447-324 6468-310 6447-324 6468-310 6447-324 6468-310 6448-320 6448-321 6448-321 6448-321 6448-321 6448-321 6448-321 6448-321 6448-321 6448-321 6448-321 6448-321 6448-321	C596-239 6521-239 6521-239 6540-112 6531-137 6440-112 6531-137 6440-112 6531-137 6440-113 6531-137 6440-136 6440-137 6540-139 6541-137	2 6584.713	6526.538	6652.052	6593.726		6472.710	6535.305		6541.103	6482.761	12
6590.593 6518.468 6656.218 6588.557 6462.423 6519.232 6417.423 6557.423 6519.232 6517.423 6557.423 6557.423 6518.232 6517.423 6517.424 6517.423 6517.423 6517.423 6517.423 6517.423 6517.423 6517.424 6517.423 6517.424 6517.423 6517.424 651	6590.593 6518.468 6618.213 6588.551 6588.552 6541.633 6555.218 6588.552 6545.233 655	6590.499 6511.201 6566.218 6588.557 6462.423 6519.232 6417.423 6567.423 6568.340 6568.240 6567.423 6568.240 656		6523.868	6654.142	6591.137		6470.112	6537.327	6474.294	6543.105	6480.106	13
6.9999 6.81868 6.6632 6.64578 6.64539 6.45239 6.9998 6.81868 6.81827 6.86237 6.46237 6.64539 6.9988 6.81878 6.66237 6.66237 6.66237 6.66331 6.9988 6.50234 6.66234 6.66331 6.66331 6.66331 6.66331 6.9088 6.00334 6.66335 6.66331 6.66332 6.66332 6.66332 6.66331 6.66332 6.66332 6.66332 6.66332 6.66332 6.66332 6.66332 6.66332 6.66332 6.66332 6.66332 6.66332 6.66332 6.66332	6.9828 6.68273 6.6623	6592.39 6518.49 6518.49 6660.232 6581.29 6462.07 6464.77 6466.19 6466.19 6582.39 6518.40 6518.40 6518.		6521.203	6656.218	6588.557		5467.423	6539.282	6471.633	6545.110	6477.427	1
6596-238 6132-761 6660-255 6583-258 6462-073 6466-195 6462-195 6596-238 6132-394 6662-239 6666-1395 6462-317 6469-239 6666-1395 6596-238 6510-214 6669-288 6559-243 6552-243 6	6594.28 6513.781 6660.213 6583.239 6462.073 6663.435 6693.237 6462.073 6663.431 6669.233 6594.284 6512.344 6560.213 6650.214 6660.213 6552.243 6660.671 6660.671 6660.671 6660.671 6660.671 6660.671 6660.671 6660.671 6660.671 6660.671 6660.671 6660.671 6660.414 6660.414 6660.414 6660.414 6660.414 6660.414 6671.244 6671.244 6671.244 6671.245 6671	6594.384 6513.781 6660.233 6583.239 6462.073 6660.435 6692.237 6660.435 6593.248 6512.991 6660.431 6660.435 6593.248 6512.991 6660.431 6650.237 6660.431 6650.237 6660.431 6650.237 6660.431 6650.444 6692.237 6650.444 6650.442 6650.444 6650.442 6650.444 665		6518.468	665/8.227	6565.928		6464.778		6468.940		6474.729	115
6596.238 6510.249 6660.231 6555.243 6460.431 6469.233 6556.238 6510.244 6698.359 6656.213 6555.243 6556.214 6556.213 6556.214 6556.213 655	6596.234 6312.395 6662.237 666	659-334 6512-89 6512-89 6662-317 6660-237 6660-2		6515.761	6660.255	6583.298		6462.073		6466.195			*
6596.238 6510.214 6669.33 6572.243 6460.471 6558.243 6569.344 6569.38 6572.243 6569.344 6569.38 6572.243 6569.384 6569.384 6569.384 6569.384 6569.384 6569.384 6569.384 6572.508 6669.881 6572.508 6569.384 6572.508 6569.384 6572.508 6572.394 6572.394 6572.394 6572.394 6572.394 6572.394 6572.394 6569.384 6572.395 6660.408 6573.394 6569.384 6572.395 6660.408 6552.331 6462.395 6462.3	6596.238 6510.214 6666.018 6572.243 6460.671 6598.882 6507.35 6666.018 6572.243 6572.243 6572.243 6572.243 6572.243 6566.018 6572.243 6566.018 6572.243 6566.018 6572.243 6566.018 6572.243 6566.018 6572.244 6571.744 6771.744 6771	6596.28 6510.214 6566.018 6572.243 6460.671 6598.099 6507.397 6666.018 6572.243 6572.243 6572.243 6569.885 6569.886 6599.754 6507.344 6569.886 6592.543 6569.886 6572.548 6572		6512.983	66652.233					6463.431		6469.257	13
6598.082 6506.339 6666.133 6575.243 6595.8243 6505.344 6598.8815 6595.734 6659.088 6572.538 6503.144 6598.8815 6569.734 6659.088 6573.548 6569.734 6659.734	6598.082 6504.387 6666.138 675.243 6656.138 675.243 6656.138 675.243 6656.138 675.243 6656.138 675.243 6659.244 6659.888 6561.734 6659.888 6517.34 6559.744 6659.888 6517.34 6559.734 6657.045 6517.34 6559.734 6657.045 6517.34 6577.045 6517.34 6577.045 6517.34 6577.045 6517.34 6577.048 6517.34 6577.048 6517.34 6577.048 6517.34 6577.048 6517.34 6577.048 6517.34 6577.048 6517.34 6577	6503.44 6509.88 15 6656.134 6555.243 6555.243 6550.134 6559.88 15 6659.88 15 6659.88 15 6659.88 15 6659.88 15 6659.88 15 6659.88 15 6659.88 15 6659.88 15 6659.88 15 6659.88 15 6659.89 15		6510.214	\$885a. 179	6577.948				6460.671			18
6603.414 6698.882 6501.213 6668.018 6572.508 6603.414 6669.881 6572.508 6571.213 6669.812 6571.213 6669.812 6571.213 6671.213 6571.214 6771.214 677	6603.414 6698.882 6501.213 6668.018 672.508 6503.414 6698.883 6571.254 6569.734 6569.734 6569.734 6571.254 6571.254 6571.254 6560.414 6699.893 6571.034 6571	6603.414 6698.882 6504.233 6668.018 672.508 6505.414 6698.883 6571.754 6568.013 6569.754 6571.754 6571.754 6571.754 6571.754 6571.754 6571.754 6571.754 6571.754 6571.754 6571.754 6571.754 6589.017 6578.799 6575.799 6775		6507.397	8866.133	6575.243							61
6605.142 6698.815 6671.735 6671.736 6671.736 6671.736 6671.736 6671.736 6671.736 6671.736 6671.736 6671.736 6671.736 6671.736 6671.737 6711.737 671	6603.414 6698.815 6569.754 6603.414 6609.818 6569.754 6605.424 6698.818 6569.754 6605.424 6605.424 6671.724 6671.724 6671.724 6571.724 6671.724 657	6603.44 6698.81 6659.754 6669.88 6569.754 6669.88 6569.754 6669.88 6569.754 6669.88 6569.754 6669.81 6671.754 6671.757 6660.486 6671.754 6671.757 6660.486 6671.754 6671.757 6660.486 6671.757 6660.486 6671.757 6660.486 6671.757 6660.486 6671.757 6660.486 6671.757 6660.486 6671.757 6660.486 6671.757 6660.486 6671.757 6660.486 6671.757 6660.486 6671.757 6660.486 6671.757 6660.486 6671.757 6660.486 6671.757 6660.486 6671.757 6660.486 6671.757 6660.486 6671.757 6660.486 6671.757 6660.486 6671.757 6660.486 6671.757 6		6504.550	810.8999	6572.508							2
6603.424 6698.815 6671.754 667	6603.414 6699.815 6671.754 6564.131 6606.821 6673.526 6554.131 6606.821 6673.526 6573.526 6573.526 6573.526 6573.526 6573.526 6577.034 657	6603.424 6499.815 6471.754 6554.131 6606.821 6499.90 6473.546 6554.131 6606.821 6499.90 6473.549 6557.094 6577.094 6610.181 6689.047 6677.094 6558.431 6610.181 6689.047 6677.999 6611.995 6471.999 6612.937 6462.299 6622.331 6462.299 6633.895 6633.896 6633.896 6633.896		6501.713	6669.888	6569.754							21
6605.812 6495.900 6551.346 6551.346 6551.326 6551.326 6551.326 6551.326 6551.326 6551.326 6551.326 6551.326 6551.326 6551.326 6551.326 6551.326 6551.326 6551.326 6551.326 6551.326 6551.326 6551.327 655	6605.14.2 6492.890 6513.546 6513.546 6513.546 6513.546 6513.546 6513.546 6513.546 6513.546 6513.546 6513.546 6513.546 6513.546 6513.546 6513.546 6513.546 6513.546 6513.547 65	6605.312 6492.90 6551.346 6551		64.98.815	6671.754								22
6606.512 6492.329 6561.292 6610.181 6637.024 6558.431 6610.181 6637.024 6577.032 6611.374 6680.426 6555.531 6611.374 6680.426 6555.531 6611.375 6477.890 6611.375 6477.897 6621.373 6462.285 6627.373 6462.285 6627.835 663.424 6637.835 6637.896 6637.836 663.424 6637.836 663.434 6637.836 663.434	6606.512 6492.329 6506.232 6517.034 6556.313 6610.181 6637.034 6556.313 6610.181 6637.034 6575.331 6610.184 6638.037 6637.034 6555.531 6637.035 663	6606.512 6497.329 6501.292 6610.181 6637.029 6577.032 6610.181 6637.039 6577.031 6611.374 6680.496 6555.531 6611.374 6680.496 6555.531 6611.395 6477.890 6611.395 6477.890 6611.395 6477.890 6621.391 6462.295 6621.793 6462.295 6621.793 6462.295 6621.793 6621.896 6631.650		64.85.900	6673.546	6564.131							23
6600.181 6687.094 6585.531 6610.181 6687.094 6585.531 6611.374 6689.007 6680.468 6611.374 6687.699 6612.375 6680.468 6612.375 6680.468 6612.375 6462.295 6612.375 6462.295 6612.386 6612.386 6612.386 6612.386	6600.181 6687.094 6585.531 6610.181 6689.007 6689.007 6611.784 6689.007 6680.468 6611.784 6678.799 6680.468 6611.784 6471.887 6611.495 6640.468 6620.395 6640.666 6620.391 6462.295 6621.313 6462.295 6621.314 6462.295 6621.318 6462.295 6621.318 6462.295 6621.318 6462.295	6610.181 6687.094 6588.431 6611.184 6689.007 6678.794 6558.431 6611.371 6680.469 6555.531 6611.371 6680.469 6555.531 6617.995 668.466 6620.948 665.494 6621.371 6462.295 6621.371 6462.295 6631.600 6631.600		8492.988	6675.329	6561.292							2
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6626-512 6629-148 6630-110 6631-850 6631-866 6631-866	6625-312 6627-1835 6627-183 6633-6410 6631-640 6632-866 6632-866	6626-512 6627-188 6630-110 6630-410 6631-650 6631-866											1
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	0 - 0	v	*	p - p	3 - 3		9	9 - 9			**	9-9	
2	R(J)08S.	F(J)08S.	R(J)08S.	P(J)085.	R(J)085.	P(J)08S.	R(J)08S.	P(J)OBS.	R(J)0BS.	P(J)08S.	R(2)08S.	P(J)08S.	7
6534.	14.428		6534.428		6539.406		6539.406.						-
653	6536.705	6525.073.	6536.718	6525.073.	6541.645	6530.016.	6541.645	6530.016	6613.446	944.1099	6613.446	6601.746	~
653	5538.955	6522.664	6538.994	6522.664	6543.903	6527.577	6543.935	6527.377	6615.733	6599.396	6615.733	6599.354	
654	6541.173	6520.246	6541.230	6520.209	6546.116	6525.184	6546.163	652: 219.		6596.990		6596.900	
654	3.366	6517.788	6543.438	6517.736	6548.311	6522.710.	6548.376	6522.834,	6520.258.	6594.580	6620.200	6594.483.	*
654	6545.552	6515.329	6545.633	6515.275	167.0559	6520.246	6550.564	6520.208	6622.461	6592.125	•	6591.987	•
654	6547.682	6512.816	6547.783	6512.754	6552.621	6517.736	6552.709	6517.657	5624.691	6589.683	6624.583	6589.488	-
65	5549.825	6510.294	6549.938		6554.763	6515.210	6554.865	5515.145	6626.854		6626.696	6586.953	•
65	6551.895	6507.746	6552.026	6507.667	6556.833	6512.641	6556.949	6512.571	6629.035		6628.828	6584.429	
65	6553.993	6505.160		6505.076	6558.922	6510.057	6559.054	6509.978	6631.175	6582.186	6630.929		10
65	6556.023	6502.573	6556.174	6502.484	6560.951	6507.443		6507.382	6633.296	6579.680	6632.988	6579.262	=
65	5558.051	6499.927	6558.230	6499.831	6562.966	6504.804	6563.127	6204.719	6635.418		6635.063	6576.636	12
6	5560.052	6497.304	6560.242	6497.202	6554.964	6502.168	6565.136	6502.076	6637.483	6574.565	6637.055	6274.009	13
65	2561.997	6494.603	6562.206	6494.498	6566.910	6499.460	6567.096	6499.365	6639.578	6571.974	6639.057	6571.322,	*
5	6563.960	6491.923	6554.187	808.1679	6568.865	6496.771	6569.069	6496.672	6641.614	6569.384	6641.052	6568.610	15
65	5565.862	161.6859	6566.104	6489.076	6570.756	6494.021	6570.973	6493.922	629.659	6566.778	6642.986	6565.927	100
65	6567.756	6485.441	6568.018	6486.321	6572.635	6491.264	6572.876	6491.156			6644.926	6563.177	17
65	8569.638	6483.685	6569.911	6483.566	6574.506.	6488.493	6574.758	6488.386	6647.620		6646.823	6560,435	*
65	5571.441	998.0879	6571.744	6480.745	6576.303	6485.661	6576.582	6485.547	5649.582				18
65	6573.270	6478.071	6573.596		6578.128	6482.851	6578.421	6482.739	6651.548				20
65	6575.059	6475.207	6575.397	6475.090	6579.915	6479.983	6580.205	6479.872	6653.439				21
65	5576.799	6472.330	6577.153	6472.203	6581.638	6477.092	6581.961	6476.972					22
		195.6959	6578.932	6469.337	6583.381	6474.204	6583.719.	6474.084	6657.226				23
8	580.246	6466.510		6466.379	6533.071	6471.252	6585.444	4471.136					24
65	581.916	6463.565	6582,337	6463.431	•	6468.298	6587.100	6438.173					25
5	109.185	6460.611	6584.054	187.0979	6586.386	6465.312,	6588.776	6465.187					58
3	5585.218		6585.676		6590.003	8462.295	6590.407	6462.162					27
55	5586.789		6587.294		6591.565		6591.987						2
65	6588.386		6588.906		6593.141		6593.599.						29
55	5589.941		6590.478		6594.688		6595.177						8
65	5591.451		6591.987				6596.671						31
			6593.541		6597.663		6598.169						32
			6595.035										33
			6596.491										*

lines not used in calculations.

TABLE IV-continued

	- 3	u		p - p	3 - 3		7	P - P	3 - 3			9 - 9	
~	R(J)08S.	P(J)08S.	R(1)08S.	P(1)08S.	R(J)085.	P(J)08S.	R(J)oBS.	P(J)08S.	ki,J)085.	P(1)085.	R(J)085.	F(J)08S.	*1
-	616.8659		6598.919		6621.261	,	6621.261						
~	6601.197	6589.532		6589.532	6623.579	6611.838	6623.579	6611.838					74
			6603.597		6625.857	6609.482	6625.857				6609.519		_
	6605.697		6605.884		6628.131	6607.077	6628.131	6607.038	5611.612		6611.838	6593.700	
	6607.908	6582.284	6608.179	6582.337	6630.410	6697.099	6630.410	6604.629	6613.840	6588.206	041.140	6588.265	
		6579.845	6610.465		6632.632	6602.250	6632.632	6602.140		6585.734			•
	6612.289	6577.358	6612.718	6577.486	6634.898			6599.690	6618.201	,	6618.7087	6583.443	-
00	6614.474	6574.876	6614.992		6637.126		6637.055	6597.198	6620.363	6580.723		6580.975	
•	5616.579	6572.336	6617.210	6572.587	6639.349	6594.944	6639.251	6594.688	6622.461		6623.188		•
10	6618.708	6569.813		6570.129	6641.557		6641.412	6592.204	6624.583	6575.648	8625.426	6576.069	ä
=		6567.227	6621.668	6567.655	6643.757	6590.003	845.548	6589.693	6626.633	6573.06%	6627.625		22
12	6622.858	6564.675	6623.875	6565.186	6645.965	6587.523			6628.693	6570.467	6629.847	6571.096	7
13	6624.921	6562.043	6626.051	6562.659	6648.105	6585.018	5647.761	6584.543	6630.725		6632.012	6568.610	2
7	6626.914	6559.435	6628.207	6560.147		6582.511	5549.851		6632.695	6565.186	6634.209		ž
15	6628.917	6556.753	6630.361	6557.594	6652.425		6651.913			6562.495	6636.354	6563.558	12
91	6630.882		6632.474	6555.069			6653.963		6636.617	6559.827	887.8899		*
11	6632.807	6551.380		6552.475	6656.713		6655.983		9638.539		6640.634	6558.483	7
18	6634.742	6548.669	6636.684		6658.830		6657.969		6640.453		6642.741.		*
19	6636.617	6545.918	6638.765	6547.316							6981.199		=
20	6638.488		6640.826	6544.712			5661.90						2
21	6640.335	6540.367		6542.047									77
22			6584.9599										55
3.3	4447 970												3.5

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~	R(J)08S.	R(J)08S.	bp(1)08S.	"	R(J)08S.	R(J)oBS.	^b P(J)08S.
-	6571.848	6571.848		31	6599.055	6599.319	6531.67
2	6574.133	6574.133	6562.495	15	6600.913	6601.197	6528.891
3	6576.333	6576.393		10	6602.778	6603.090	6526.15
			6557.638	17	6604.829	5604.962	6523.31
'n	6580.723	6580.808	6555.152	18	6606, 400	6606.771	6520.51
	6582.886	6582.980	6552.647	19	6606 . 179	6608.572	6517.63
	6585.018	6585.123	6550.116	30	6609.931	6610.352	6514.74
	6587.100		6547.582	21	6611.612	6612.084	6511.84
•	6589.174	6.89.326	6544.976	22	6613.319	5613.840	
10	6591.180	6591.352	6542.389	23		6615.501	6505.95
11	6593.202	6593.392	6539.730	24			
12	6595.177	6595.391	6537.086	25		6618.796	
13	6597.108	6597.347					

b No aplitting is observed in the P branch.

times not used in calculations

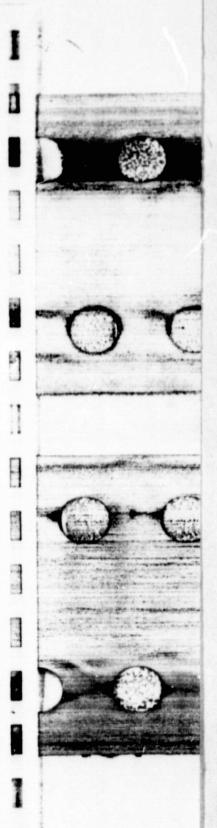


TABLE V

Wavenumbers (vac. cm⁻¹) of the Rotational Structure of $\Delta_{w(g)} \leftarrow \Delta_{g(w)}$ Bands of ${}^{12}C_2H_2$ (Estimated Accuracy of Data: ± 0.005 cm⁻¹)

		1012-00 -	0002-00		101(11)2 -	000(11)2	1010022 -	0000022	
	c -	e	4	- d					
J	R(J)OBS.	P(J)OBS.	R(J)OBS.	P(J)OBS.	R(J)OBS.	P(J)OBS.	R(J)OBS.	P(J)OBS.	
2	6510.481		6510.481		6511.880				
3	6512.717	6496.406	6512.717	6496.406	6514.123	6497.803	6522.459	6506.106	
•	6514.919	6493.952	6514.968	6493.962	6516.380	6495.352	6524.694	6503.660	
•	6517.108	6491.510	6517.177	6491.510	6518.555	6492.919	6526.897	6501.211	
•	6519.242	6489.032	6519.361	6489.032	6520.776	6490.410	6529.061	6498.715	
,	****	6486.486	6521.535	6486.539	6522.913	6487.925.	6531.221	6496.205	
	6523.431	6483.956	6523.651	6484.009		6485.375	6533.345	6493.671	
•	6525.497	6481.353	6525.788	6481.449	6527.164		6535.476	6491.094	
	6527.512	6478.770	6527.839	6478.896	6529.258	6480.241	6537.534	6488.517	
ŀ	6529.530	6476.112	6529.926	6476.278	6531.306	6477.650	6539.615	6485.885	
1	6531.520	6473.486	6531.946	6473.675	6533.345	6475.022		6483.277	
1	6533.474	6470.811	6533.967	6471.017	6535.366	6472.355	6543.643	6480.579	
٠	6535.430	6468.111	6535.952	6468.345	6537.372	6469.701		6477.893	
	6537.327	6465.402	6537.904			6466.972	6547.539		
	6539.220	6462.632	6539.858	6462.901	6541.283	6464.251	6549.458		
	6541.103	6459.898	6541.768	6460.164		6461.485	6551.380		
1	6542.932						6553.219		
•	6544.761								
•	6546.551								

Lines not used in calculations.

lower state, a check by the B'' and D'' constants could not be made since they are not available at this time. The corresponding $\Delta_u \leftarrow \Delta_g$ transition $(\nu_1 + \nu_3 + 2\nu_4^2 - 2\nu_4^2)$ has been tentatively assigned to the nearby weak band at 6503.6 cm⁻¹ because only higher order terms are responsible for different spectral positions for the two vibrational transitions and they are known to be rather small.

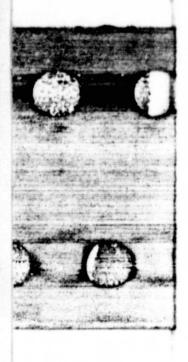
From a rotational analysis of this band a set of molecular parameters is obtained and they give rise to an unusually large splitting between the c and d components both in the upper and in the lower states. It may be noted that Innes (6) also obtains from up data similar results for the 0002^{200} state.

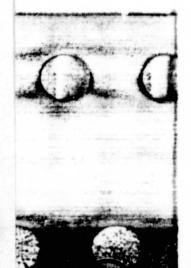
By comparing the above mentioned $\Sigma^+ \leftarrow \Sigma^+$ and $\Delta \leftarrow \Delta$ transitions, it may be noted that the $\Delta^-\Sigma^+$ separation in the upper states is 4.2 cm^{-1} , which is of the same order of magnitude as that quoted for the lower states (3.1 cm⁻¹) (5).

The two bands at 6512.7 and 6513.3 cm⁻¹ are, respectively, assigned to the $1010^{0}2^{0} \leftarrow 0000^{0}2^{0}$ and $1010^{0}2^{2} \leftarrow 0000^{0}2^{2}$ transitions. Such an assignment can also be supported by an approximate vibrational calculation.³ These two bands are distinguished from one another by consideration of the intensity alternations and the molecular constants of the lower levels.

It is of interest to note that in this case the separation between the Σ^+ and Δ levels is 14.4 cm⁻¹ for the upper states $1010^{\circ}2^{\circ}$. This may be compared to the value 13.9 cm⁻¹ found for the $0000^{\circ}2^{\circ}$ levels (3, 5). It may be recalled that in the previous section the $\Delta^-\Sigma^+$ separation was found to be 4.2 cm⁻¹ which is an order of magnitude lower. That means $g_{55} > g_{44}$ which agrees with Ref. (5)

"Double hot" bands can also arise from the $(\nu_4 + \nu_5)_{+/-}^{0,2}$ levels. Again, by an





 $^{^{2} \}nu_{0}$ of $\nu_{1} + \nu_{2} = 6556.5$; ν_{0} of $\nu_{1} + \nu_{2} + \nu_{5}^{1} - \nu_{5}^{1} = 6534.7$. The difference of 21.8 cm⁻¹ $\simeq -(x_{15} + x_{25})$. Therefore, ν_{0} of $(\nu_{1} + \nu_{2} + 2\nu_{5}^{0.2} - 2\nu_{5}^{0.2}) \simeq \nu_{0}$ of $(\nu_{1} + \nu_{2}) + 2(x_{15} + x_{25}) \simeq 6512.9$.

TABLE VI

Wavenumbers (vac. cm⁻¹) and Spectroscopic Constants (cm⁻¹) of the Bands Pertaining to the ¹³C¹³CH₂ Species in Natural Abundance (Estimated Accuracy of Data: ±0.005 cm⁻¹)

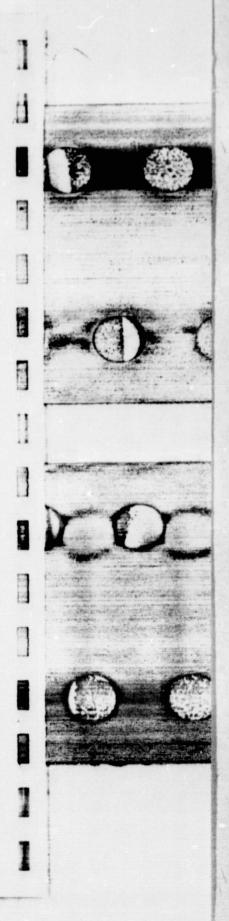
	1010202 -	0000707	0020702	- 005do 019		1011100 +	0001102		
						e	4 -	4	
J	R(J)OBS.	P(.!)OBS.	8(J)085,	P(!)088.	R(J) BS.	P(J)085.	R(J)OBS.	P(J)OBS.	
0	6545.155		6495,151						
1	6547.380	6540.581	6497.392	6490.567	6521,644		6521.699		
2	6549,630	6538.246	6499,600	6488.201		6512.534"		6512.534	
3	6551.801	6535.916	6501.822	6485.885	6526,055	20000000	6526.153"		
		6533,529	6503.996		6528.261	6507,842	6528.327	6507.842	
5	6556.141	6531.149	6506,175	6481.143	6530,404	6505.444	6530,514	6505.444	
6	6558.279	6528.722	6508,303	6478,770	6332.535	6503.018	6532.625*	6503.018	
7	6560.381	6526.295	6510.429	6476.321		6500,600	6534.764		
4	6562.445	6523.830	6512.534	6473.890		6498.160		6498.101	
9	6564.519	6521.348	6514.596	6471.509	6538.782	6495,664		6495.607	
)	6566,526	6518.810	6516,660	6468,940	6540,819	6493,159	6540.988	6493.108*	
1	6568.546	6516.295	6518,670		6542.796*	6+90.615	6543.025	6490,567	
2	6570,516	6513.710	4520,707	6463,856	6544.823	6488.093	5545,024	6488.024	
3	6572.464	6511,145			6546.785	6485.502	6547,016	*******	
		6508,516	6524.643		6548,756	6482.944	6548.965		
5	6576,303	6505,901	6526.592		22401120	6480,307	6550.873	6480.241	
6	6578,180	6503,235	6528,501			6477.697	6552,798		
7	6580,041	6500, 554	6530,504				6554,702	6474.941	
8	6581.849	6497.864	6532,244				6556.532	6472,237	
9	6583,670	6495,151							
)	6585.444	6492,391							
1	6587,193	6489,601							
2	6588,940	6486,811							
1		6181,309							
5	6592,305	6481.143							
	6593.978	6478.293							
4	6595,596	6475,400							
6	6597,198	6472,463							
,		6469.526							
9	6600,335	0400, 184							
)	5501.848								
l.	3411.340								
2	6604.838								

* Lines not used in calculations.

LEV	ELS				
CPPER	LOWER	Species of States		(8'-8")×10 ¹	(D'-D")×106
1010 ⁶ 0 ²	(w)entity?		6542.87310 002	+12.11:0.02	(.02±0.02
3020207	0000 02	•. •	6492,860101,005	-10.8810.08	1.0 :0.4
(1011/05)	0.0011-073		6117, 19710, 1938	-11.96:0.10	-1.5 ±0.4
(101110-)	(0001 0),		W-1771377070	-11.31 -11	0.4 ±0.3

approximate vibrational calculation⁴ they can be predicted to be located at about 6508 cm⁻¹. Out of the three bands $\Delta_g \leftarrow \Delta_u$, $\Sigma_g^+ \leftarrow \Sigma_u^+$, $\Sigma_g^- \leftarrow \Sigma_u^-$ allowed by symmetry, two could be located both at a wavenumber value slightly lower than that expected. The first of these ($\nu_0 = 6506.9 \text{ cm}^{-1}$) arises from the $000(11)_+^0$ level according to the alternation of intensity of the rotational lines and to the value of B'', and therefore it is assigned to the $(10111)_+^0 \leftarrow (00011)_+^0$ transition. The second one ($\nu_0 = 6504.9 \text{ cm}^{-1}$), which arises from the $(00011)^2$ level has been assigned to the $(10111)^2 \leftarrow (00011)^2$ transition. The separation between the Σ^+ and Δ states is 17.7 cm⁻¹ in the upper state as compared to the value of 19.4 cm⁻¹ found for the lower state (5).

⁴ ν_0 of $\nu_1 + \nu_3 = 6556.5$; ν_0 of $\nu_1 + \nu_2 + \nu_4^1 - \nu_4^1 = 6529.8$. The difference of 26.7 cm⁻¹ $\simeq -(x_{14} + x_{34})$. $(x_{15} + x_{35}) = -21.8$ (see footnote 2). Therefore, ν_0 of $(\nu_1 + \nu_3 + (\nu_4 + \nu_5)_{+/-}^{0.2} - (\nu_4 + \nu_5)_{+/-}^{0.2}) \simeq \nu_0$ of $(\nu_1 + \nu_3) + (x_{14} + x_{34} + x_{15} + x_{35}) = 6508.0$.



VIBRATIONAL LEVELS OF ACETYLENE

Table I lists 15 vibrational levels of the ¹²C₂H₂ molecule referred to the ground vibrational state which were obtained by the present analysis. Although a few of these were previously measured, all were reported here because of the better resolution achieved in the present work and the internal consistency of the data. The molecular constants for these levels are presented in Table II and the observational data along with the assignments are given in Tables III–V. The data for the higher vibrational levels of acetylene like the ones obtained in the present study should be useful for the force field calculations of the type discussed by Strey and Mills (8).

BANDS PERTAINING TO 12C12CH2

Although the concentration of the isotopic species $^{12}C^{13}CH_2$ in the natural sample is only around 1%, there were observed, measured, and assigned a few sequences belonging to such an isotopic variety. Of course, no alternation of intensity was either expected or observed in the rotational structure. The $\nu_1 + \nu_3$ and $2\nu_3$ bands were already measured by using an enriched sample (7) although the basic observational data for the bands were not given in this publication. As far as we are aware the $\nu_1 + \nu_2 + \nu_4^1 - \nu_4^1$ band was measured here for the first time. The wavenumbers of the individual lines of all the three bands and the spectroscopic constants determined from these data are summarized in Table VI.

ASTROPHYSICAL INTEREST OF ACETYLENE

In recent years the acetylene spectrum in the infrared has assumed much astrophysical importance. In 1974, from observations made in the 10- μ m window acetylene has been identified as a constituent of the atmosphere of Jupiter (9). About the same time, it has also been identified in the spectra of stellar photospheres; from observations made in the photographic infrared, bands of ${}^{12}C_2H_2$ have been identified in carbon stars by Hirai (10). In addition, the strong absorption features in the spectra of carbon stars at longer wavelengths as, for instance, at 3 μ m may be, at least in part, due to acetylene (11). To maintain consistency in the notation as compared to current laboratory studies (12), the ${}^{12}C_2H_2$ bands at 9802, 9668, and 9603 cm⁻¹ would, respectively, be due to the transitions $2011{}^{10}$ 0- $0001{}^{10}$ 0, $1112{}^{00}$ 0- $0000{}^{00}$ 0, and $0031{}^{10}$ 0- $0001{}^{10}$ 0 instead of as indicated in Ref. (10).

ACKNOWLEDGMENTS

One of us (K.N.R.) expresses thanks to the Atmospheric Research Section of the National Science Foundation and the National Aeronautic and Space Administration for support of this research.

RECEIVED: January 10, 1977

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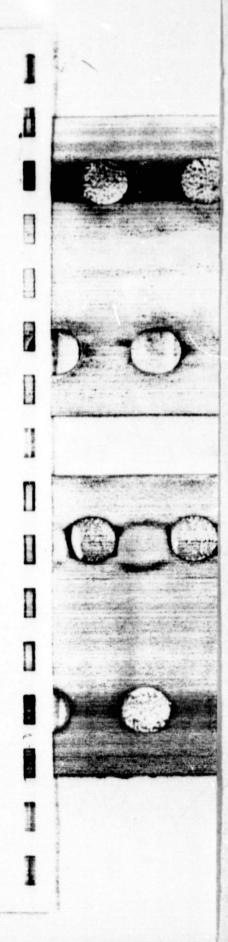
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Analysis of the ν_3 Band of $^{12}CH_3D$ at 7.6 μ m

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The previously reported (J. Mol. Spectrosc. 68, 195-222 (1977)) study of the CH₃D spectrum occurring at $1033-1270~\rm cm^{-1}$ which was mainly concerned with the ν_6 fundamental has now been extended to cover the region $1270-1420~\rm cm^{-1}$. In all, 342 transitions belonging to the ν_3 band are now assigned. Both the ν_3 and ν_6 bands are processed simultaneously taking into account of the Coriolis interaction between them, and the fitting of all the experimental data led to 21 significant spectroscopic constants for the states $\nu_6 = 1$ and $\nu_3 = 1$ of CH₃D.

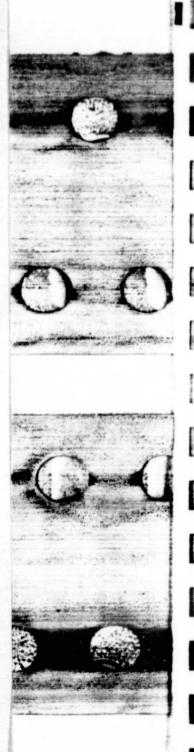
INTRODUCTION AND EXPERIMENTAL RESULTS

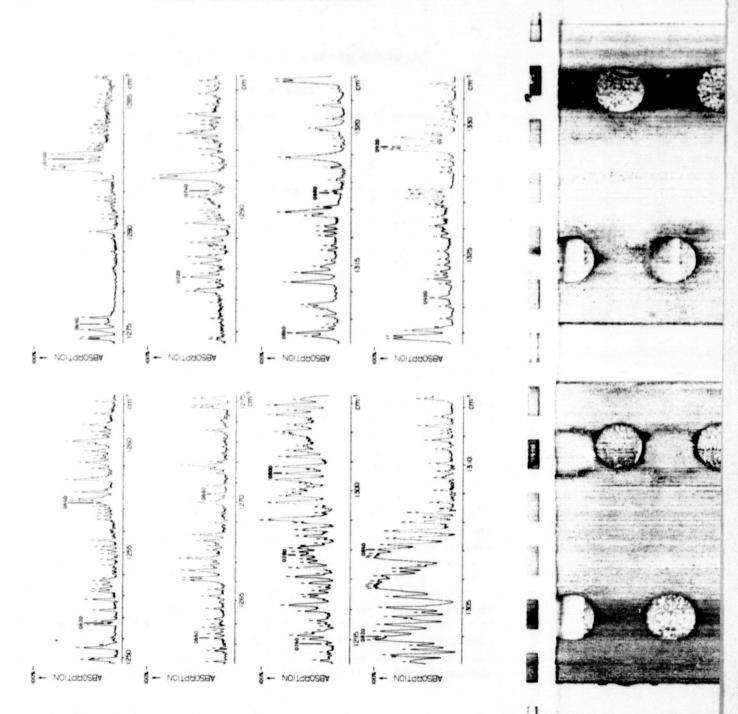
In a previous paper (1), the absorption spectrum of $^{12}\text{CH}_3\text{D}$ was investigated in the range $1033-1270~\text{cm}^{-1}$. The perpendicular-type $\nu_6(E)$ band, responsible for the absorption in this region, was analyzed up to J'=17. A second-order Coriolis interaction with the parallel-type $\nu_3(A_1)$ band was expected to perturb transitions PP(J,|K|), PQ(J,|K|), and PR(J,|K|) near |K|=13, 14. Unfortunately, such transitions were generally too weak in our spectra of ν_6 to be observed or unambiguously assigned. On the other hand, the same perturbation was predicted to be observed in the ν_3 band for lower |K| values, the subbands |K|=10, 11 being expected to be the most perturbed.

In the present paper, the investigation of the spectrum of $^{12}CH_3D$ is extended to the range 1270-1420 cm⁻¹, where the absorption is mostly due to the ν_3 band. The investigated spectra are recorded with a resolution of about 0.04 cm⁻¹ (2), which is better than previous recordings in the same region (3, 4).

A reproduction of this spectrum is given in Fig. 1, which can be added to Fig. 1 of Ref. (1) to get the overall absorption of ¹²CH₃D between 1033 and 1420 cm⁻¹.

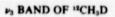
The identified transitions of ν_3 run from 1155 to 1420 cm⁻¹ with three very strong absorption peaks in the Q branch at 1303.898, 1305.709, and 1306.771 cm⁻¹. The high frequency side of the band is progressively overlapped by the very strong ν_2 band of H_2O , and also by transitions assigned to the ν_5 band of $^{12}CH_3D$. In addition, lines belonging to the ν_4 band of $^{12}CH_4$ are also identified throughout the spectrum showing the presence of this molecule as impurity in the gas sample.





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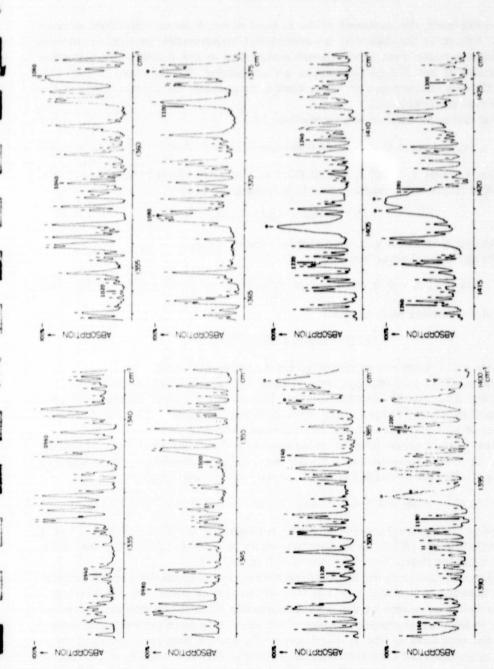


Fig. 1. The spectrum of ¹²CH₃D from 1270-1420 cm⁻¹. The reproduced spectra were recorded with the gas sample contained within a 1-m-length absorption cell at a pressure of 30 Torr.

VIBROTATIONAL ENERGIES IN $v_3 = 1$ STATE

As expected, the treatment of the ν_3 band alone, without vibrational interaction, fails to fit the data within experimental uncertainties; in fact, anomalous deviations occur even if vibrational corrections to the fourth-order distortion constants H and K-type resonances are considered. As discussed in Ref. (1), the Ceriolis interaction with the ν_6 band is responsible for at least a part of the observed discrepancies.

The first-order term of this interaction, i.e.,

$$(v_3 = 1, v_6 = 0, J, K, l_6 = 0 | H | v_3 = 0, v_6 = 1, J, K \pm 1, l_6 = \pm 1)$$
 (1)

connects levels J|K| of $v_3 = 1$ to the two Coriolis components of $v_6 = 1$, the coupled levels being roughly spaced, in terms of |K|, by

$$\Delta E_1^{\pm} = 151 \pm 3.5 |K| \text{ (in cm}^{-1)}$$

for the positive and negative component, respectively.

For the second-order term, i.e.,

$$(v_3 = 1, v_6 = 0, J, K, l_6 = 0 | H | v_3 = 0, v_6 = 1, J, K \pm 2, l_6 = -1)$$
 (2)

the corresponding spacings are

$$\Delta E_2^{\pm} = 128 \pm 11.7 |K| \text{ (in cm}^{-1}\text{)}.$$

Higher order terms can be neglected for the present purpose.

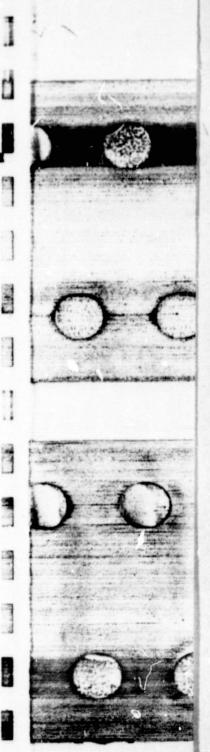
The spacings ΔE_1^{\pm} and ΔE_2^{\pm} remain large for any observed value of |K|, the spacing ΔE_2^{\pm} rapidly decreases when |K| increases, thus making clear the possibility of a "second-order level crossing" between the two rotational series of $v_3 = 1$ and $v_6 = 1$, for |K| between 10 and 11.

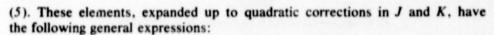
Now, if the Coriolis interaction between $v_3 = 1$ and the other near degenerate state $v_5 = 1$ is considered in a similar way, using a preliminary analysis of the v_5 band, the related quantities ΔE_1^{\pm} and ΔE_2^{\pm} are found to be

$$\Delta E_1^{\pm \prime} = -169 \mp 5.5 |K|$$
 and $\Delta E_2^{\pm \prime} = -165 \pm 2.7 |K|$ (in cm⁻¹)

showing that no level crossing between rotational series of $v_3 = 1$ and $v_5 = 1$ is expected before $|K| \approx 30$ (for interactions up to second order). The same situation holds for interactions between $v_6 = 1$ and $v_5 = 1$.

So, a convenient model for available data is one in which the Coriolis interaction between $v_3 = 1$ and $v_6 = 1$ is treated exactly, all other vibrational interactions being calculated as usual by perturbation. The computational procedure used in the present analysis consists of the same three steps previously described for the treatment of the v_6 band alone (1), except that now the data on v_3 and v_6 are analyzed simultaneously. The energies for $v_3 = 1$ and $v_6 = 1$ states are obtained as eigenvalues of an effective twice transformed Hamiltonian $\binom{C}{v_1,v_2}H^+$, diagonal in v_8 , except for the two elements (1) and (2) mentioned above





$$(v_n, v_t, J, K, l_t|_{(v_n, v_t)}^{(C)} H^+/hc | v_n + 1, v_t - 1, J, K \pm 1, l_t \pm 1)$$

$$= ((v_n + 1)(v_t \mp l_t))^{1/2} \times \sqrt{J(J + 1)} - K(K \pm 1)[C_{11}^{(1)} \pm C_{11}^{(2)}(2K \pm 1) + C_{11}^{(3a)} J(J + 1) + C_{11}^{(3b)}(2K \pm 1)^2 + C_{11}^{(3c)}]$$

$$= (v_n + 1, v_t - 1, J, K \pm 1, l_t \pm 1|_{(v_n, v_t)}^{(C)} H^+/hc | v_n, v_t, J, K, l_t). \quad (3)$$

Coefficients $C_{11}^{(1)}$, $C_{11}^{(2)}$, and $C_{11}^{(30)}$ (i = a, b, or c) originate with first, second, and third order terms of the Hamiltonian, respectively. $C_{11}^{(1)}$ has a well-known expression in terms of molecular parameters, i.e.,

$$C_{11}^{(1)} = \frac{B_e \zeta_{nt}}{2} \times \frac{\lambda_n^{1/2} + \lambda_t^{1/2}}{\lambda_n^{1/4} \lambda_t^{1/4}}$$

where $(\zeta_{nt} = \zeta_{nt_h}^x = -\zeta_{nt_n}^y)$

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$$(v_n, v_t, J, K, l_t|_{(v_n, v_t)}^{(C)} H^+/hc | v_n + 1, v_t - 1, J, K \pm 2, l_t \mp 1)$$

$$= \mp ((v_n + 1)(v_t \pm l_t))^{1/2} \times (J(J+1) - K(K \pm 1))^{1/2} (J(J+1) - (K \pm 2))^{1/2}$$

$$\times [C_{21}^{(2)} \mp C_{21}^{(3)} (2K \pm 2) + C_{21}^{(4a)} J(J+1) + C_{21}^{(4b)} (2K \pm 2)^2 + C_{21}^{(4c)}]$$

$$= (v_n + 1, v_t - 1, J, K \pm 2, l_t \mp 1|_{(v_n, v_t)}^{(C)} H^+/hc | v_n, v_t, J, K, l_t). \quad (4)$$

Coefficients $C_{21}^{(2)}$, $C_{21}^{(3)}$, and $C_{21}^{(4)}$ (i=a,b, or c) originate from second, third, and fourth order terms of the Hamiltonian, respectively. Of course, any analysis does not allow us to separate the constant $C_{11}^{(3c)}$ from $C_{11}^{(1)}$ in Eq. (3), and $C_{21}^{(4c)}$ from $C_{21}^{(3c)}$ in Eq. (4).

Expressions (3) and (4) assume for basis eigenvectors, the phase conventions defined in Ref. (6).

Besides the vibrational interaction terms given by Eqs. (3) and (4), the energy matrix $\left| {\binom{C}{(v_0,v_0)}} H^+ \right|$ includes of course the usual essential resonances within $v_3 = 1$ and $v_6 = 1$. The related terms, i.e., "2, 2," "1, -2," "3, 0," and "4, -2" are expressed elsewhere (1, 6).

RELATIVE INTENSITIES OF LINES

The method described in Ref. (1) is extended to take into account transitions going up to levels with $v_3 = 1$ besides $v_6 = 1$. The leading operator of M_Z responsible for all investigated transitions may be written as

$$(M_Z)_{3,6} = d_6[\cos(Z,x)q_{6a} + \cos(Z,y)q_{6b} + \rho_{3,6}\cos(Z,z)q_3]$$
 (5)

where

$$d_6 = \frac{\partial M_x}{\partial q_{6a}} = \frac{\partial M_y}{\partial q_{6b}}; \qquad \rho_{3,6} = \frac{d_3}{d_6} \qquad \text{with} \qquad d_3 = \frac{\partial M_z}{\partial q_3}.$$

The coefficient $\rho_{3.6}$, as well as the eigenvectors arising from the diagonalization of the matrix $\left| {{}^{(C)}_{\nu_3,\nu_6}H^+} \right|$, are involved in the calculation of the relative intensities. The values of these intensities are obtained in the same arbitrary scale as in Ref. (1).





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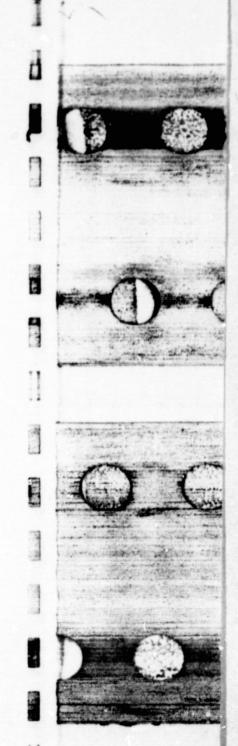
THALL SECTION

TABLE I

Observed and Calculated Wavenumbers of ν_a of $^{12}\text{CH}_aD^a$

775	(1)	(11)	111	1)	(17)	(V)	(VI)	(1117)	(1111)
OPIIR	*****	***************************************	******	*****		**********	*****	*********	
OPIIR	V3 TR	ANSITIONS							
9P117		OP(14. 1.A1)							0.0
OP(17, 2,E)									0.0
OP(17, 3,42) 16 2 2 E									0.0
OP(17, 3,42) 16 3 3 A1 1155,4510 0,316E+03 OP(17, 4,47) 16 4 4 F 1156,1918 0,294E+03 OP(17, 4,47) 16 6 6 A 1158,1918 0,476E+03 OP(17, 9,4) 16 9 9 A 1164,2589 1163,3836 0,476E+03 OP(16, 0,41) 15 0 0 A2 1164,2589 1164,2640 -27 0,682E+03 320 OP(16, 3,41) 15 3 A2 1165,1381 1165,1383 10,597E+03 320 OP(16, 3,41) 15 3 A2 1165,1381 1165,1381 1165,1383 10,597E+03 322 OP(16, 3,41) 15 3 A2 1165,1381 1165,1385 1165,1385 0,597E+03 322 OP(16, 3,42) 15 3 A1 1165,1381 1165,1385 1165,1385 10 0,597E+03 320 OP(16, 4,41) 15 3 A2 1165,1381 1165,1385 1165,1385 10 0,597E+03 320 OP(16, 5,F) 15 5 5 F 1166,1390 1164,979 -38 0,597E+03 320 OP(16, 5,F) 15 5 5 F 1166,1390 1165,1385 -59 0,597E+03 320 OP(16, 5,F) 15 5 5 F 1166,1390 1165,1385 -59 0,597E+03 320 OP(16, 5,F) 15 7 7 E 1169,4489 1169,4284 16 0,361E+03 320 OP(16, 5,F) 15 7 7 E 1169,4489 1169,4284 16 0,361E+03 320 OP(16, 5,F) 15 7 7 E 1169,4489 1169,4284 16 0,361E+03 320 OP(16, 5,F) 15 9 9 A 1172,2481 1173,797 1174,78617 0,200E+03 320 OP(16, 12,8) 15 12 12 A 1177,791 1173,7929 1173,7929 -7 0,407E+03 320 OP(16, 12,8) 15 12 12 A 1177,791 1173,7929 1173,7928 -7 0,100E+03 320 OP(16, 12,8) 14 3 A1 1174,7791 1173,7929 -7 0,100E+03 320 OP(16, 12,8) 14 3 A1 1174,7791 1173,7929 -7 0,100E+03 320 OP(16, 12,8) 14 3 A1 1174,7791 1173,7929 -7 0,100E+03 320 OP(16, 12,8) 14 3 A2 1174,7799 1174,7753 -1 0,972E+03 320 OP(16, 16,8) 14 5 5 E 1176,4784 1176,4639 10 0,409E+03 320 OP(16, 16,8) 14 5 5 E 1176,4784 1176,4639 10 0,409E+03 320 OP(16, 16,8) 14 5 5 E 1176,4784 1176,4639 10 0,409E+03 320 OP(16, 17,8) 14 5 5 E 1176,4784 1176,4789 10 0,409E+03 320 OP(16, 17,8) 14 5 5 E 1176,4784 1176,4789 10 0,409E+03 320 OP(16, 17,8) 14 5 5 E 1176,4784 1176,4789 10 0,409E+03 320 OP(16, 17,8) 14 5 5 E 1176,4784 1176,4789 10 0,409E+03 320 OP(16, 17,8) 14 5 5 E 1176,4784 1176,4789 10 0,409E+03 320 OP(16, 17,8) 14 5 5 E 1176,4784 1176,4789 10 0,409E+03 320 OP(16, 17,8) 14 5 5 E 1176,4784 1176,4789 10 0,409E+03 320 OP(16, 17,8) 11 5 5 E 1188,6790 1188,6790 1198,6790 10 0,209E+03 320 OP(16, 17,8) 11 5 5 E 1189,4795 1198,6									0.0
OP(17, 1.4] 16									0.0
OP(17, 4.8, 1) 16									0.0
9P117. 6.4) 16 6 6 A 1158.3437									0.0
0P(17. 9.4) 16 0 0 A 1164.R356 0.270E+03 138 0P(16. 1.F) 15 1 1 F 1164.2590 1164.2860 -27 0.616E+03 130 0P(16. 2.F) 15 2 P 1164.5590 1164.8779 -38 0.597E+03 132 0P(16. 3.431) 15 3 3 A1 1165.1348 1165.1346 10 0.566E+03 132 0P(16. 3.431) 15 3 3 A1 1165.1348 1165.1366 10 0.566E+03 135 0P(16. 4.F) 15 4 E 1165.8004 1164.8786 -50 0.525E+03 130 0P(16. 5.F) 15 5 5 F 1166.8190 1166.8779 -7 0.476E+03 134 0P(16. 5.F) 15 5 5 F 1166.8190 1166.8712 7 0.476E+03 134 0P(16. 6. 1) 15 7 7 E 1169.4448 1169.4284 16 0.361E+03 134 0P(16. 7.F) 15 7 7 E 1169.4448 1169.4284 16 0.361E+03 134 0P(16. 7.F) 15 7 7 E 1169.4448 1171.1665 17 0.300E+03 1501E+03 1501E 0.6 16. 1171.1660 1771.1661 1771.1661 170.300E+03 1501E+03 1771.8617 0.200E+03 1501E+03 1501E+03 1771.8617 0.200E+03 1501E+03 1501E+03 1771.8617 0.200E+03 1501E+03									0.0
70 16.0									0.0
118									0.0
720									0.0
722									1.0
722									1.0
375									0.0
370									1.0
314				4 5					0.0
140									1.0
744									2.0
151 OP(16. 0.4) 15 2 9 4 1173.2458 1173.2597 -13 O.471E+03 OP(16.12.4) 15 12 12 1173.2597 -13 O.206E+03 OP(16.12.4) 15 12 12 1173.2791 1173.8014 -15 O.107E+04 OP(15.1.6) 1.6 1.6 1.7 1.7 1.7 1.7 0.276 -7 O.106E+04 OP(15.2.6) 1.6 2.6 1.7 1.7 1.7 0.272 -7 O.106E+04 OP(15.3.42) 1.4 3 3 3 1.7 1.7 0.7 0.7 0.106E+04 OP(15.3.42) 1.4 3 3 3 1.7 1.7 0.7									1.0
OP(16.12.4.) 15 12 12 A						1171.1465			1.0
753	351				1174.7448	1177.2597	-17		1.0
384									0.0
355									0.0
357									1.0
157									0.0
362						1174.7667			1.0
167									1.0
771									1.0
775									1.0
778									•1.0
384									0.0
388				2 .					1.0
389									1.0
390									1.0
392									1.0
392									1.0
373									1.0
394									
199									2.0
403									
20(15.12.A) 14 12 12 A 1188.4033 C.300E+03 408 OP(14. 7.F) 13 7 7 E 1188.6720 1188.6720 -5 0.963E+03 413 OP(14. 8.E) 13 9 9 F 1190.3431 1190.3206 22 0.780E+03 417 OP(14. 9.A) 13 0 0 A 1192.2039 1192.2295 -25 0.120E+04 420 OP(13. 0.A2) 12 0 0 A1 1192.9809 1103.0178 -27 0.282E+04 421 OP(13. 1.E) 12 1 1 E 1193.1356 1193.1238 11 0.279E+04 422 OP(13. 2.E) 12 2 2 E 1193.4270 1193.4423 -15 0.268E+04 423 OP(13. 3.A2) 12 3 3 A1 1193.9889 1193.9718 17 0.252E+04 423 OP(13. 3.A1) 12 3 3 A1 1193.9889 1193.9718 17 0.252E+04 424 OP(13. 4.E) 13 10 10 E 1194.5055 1194.5436 -35 0.416E+03 425 OP(14.10.E) 13 10 10 E 1194.5055 1194.5436 -35 0.416E+03 426 OP(13. 4.E) 12 4 4 E 1194.7325 1194.7208 11 0.230E+04 OP(14.11.E) 13 11 11 E 1195.4468 0.268E+03 428 OP(13. 5.E) 12 5 5 E 1195.7023 1195.6864 17 0.204E+04 430 OP(13. 6.A) 12 6 6 A 1196.6711 1195.8678 3 0.350E+04									•1.0
408	403				1107.2094		,		0.0
413	400				1188-6720				1.0
417									2.0
420					1102.2020	1102. 2205			1.0
421					1102.0000	1102.0170			2.0
472					1103.1356	1103.1238			1.0
423 QP(13, 3,A2) 12 3 3 A1 1193.9889 1193.9718 17 0.252E+04 423 QP(13, 3,A1) 12 3 3 A2 1193.9889 1193.9765 12 0.252E+04 425 QP(14,10,E) 13 10 10 E 1194.5055 1194.5436 -35 0.416E+03 426 QP(13, 4,E) 12 4 4 E 1194.7325 1194.7208 11 0.230E+04 QP(14,11,E) 13 11 11 E 1194.7325 1195.4468 0.268E+03 428 QP(13, 5,E) 12 5 5 E 1195.7023 1195.6844 17 0.204E+04 430 QP(13, 6,A) 12 6 6 A 1196.8711 1196.8678 3 0.350E+04									1.0
429 GP(13, 3,A1) 12 3 3 A2 1193,9889 1193,9765 12 0.252E+04 429 GP(14,10,E) 13 10 10 E 1194,5055 1194,5436 -35 0.416E+03 426 GP(13, 4,E) 12 4 4 E 1194,7325 1194,7208 11 0.230E+04 GP(14,11,E) 13 11 11 E 1195,4668 0.268E+03 428 GP(13, 5,E) 12 5 5 E 1195,7023 1195,6844 17 0.204E+04 430 GP(13, 6,A) 12 6 6 A 1196,6711 1196,8678 3 0.350E+04									1.0
425 GP(14+10.E) 13 10 10 E 1194.5055 1194.5436 -35 C.416E+03 426 GP(13+ 4+F) 12 4 4 F 1194.7325 1194.7208 11 0.230E+04 GP(14+11.E) 13 11 11 E 1195.4646 0.266E+03 428 GP(13+ 5+F) 12 5 5 F 1195.7023 1195.6644 17 0.204E+04 430 GP(13+ 6+A) 12 6 6 A 1196.6711 1196.8678 3 0.350E+04									1.0
426 QP(13. 4.E.) 12 4 4 F 1194.7325 1194.7208 11 0.230E+04 QP(14.11.E.) 13 11 11 E 1195.4468 0.268E+03 428 QP(13. 5.E.) 12 5 5 E 1195.7023 1195.6844 17 0.204E+04 430 QP(13. 6.A.) 12 6 6 A 1196.8711 1196.8678 3 0.350E+04 4									0.0
9P(14-11.E) 13 11 11 E 1195.4468 0.268E+03 428 9P(13.5.E) 12 5 5 E 1195.7023 1195.6844 17 0.204E+04 430 9P(13.6.A) 12 6 6 A 1196.8711 1196.8678 3 0.350E+04									1.0
478 GP(13. 5.E) 12 5 5 E 1195.7023 1195.6844 17 0.204E+04 430 GP(13. 6.A) 12 6 6 A 1196.8711 1196.8678 3 0.350E+04	4,0								0.0
450 OP(13, 6,A) 12 6 6 A 1196.8711 1196.8678 3 0.350E+04 4	428				1195-7022		17		1.0
434 ODI13- 7-F 1 12 7 7 F 1108-2474 1108-2748 -4 0 145F-04									*1.0
				7 6			-		
438 OP(14.12.A) 13 12 12 A 1198.7052 1198.7185 -15 0.370E+03					1198.7083				1.0
441 OP(17. N.E) 12 8 N E 1199.8936 1199.9090 -15 0.115E+04									1.0

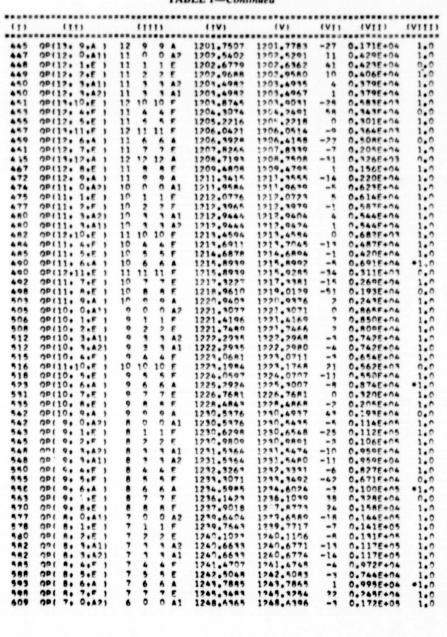
^a (I) Serial number. (II) Transition. (III) Values of J', |K'|, $|K'-l'_6|$, C' for upper levels of the transitions. (IV) Observed wavenumber in cm⁻¹. (V) Calculated wavenumber in cm⁻¹. (VI) (Expt – Calc) in 10^{-3} cm⁻¹. (VII) Calculated relative intensity. (VIII) Statistical weight.

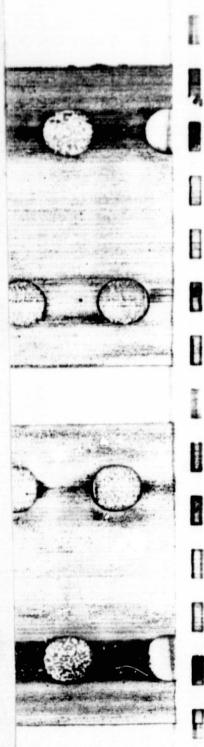


V3 BAND OF 12CH3D

TABLE I-Continued

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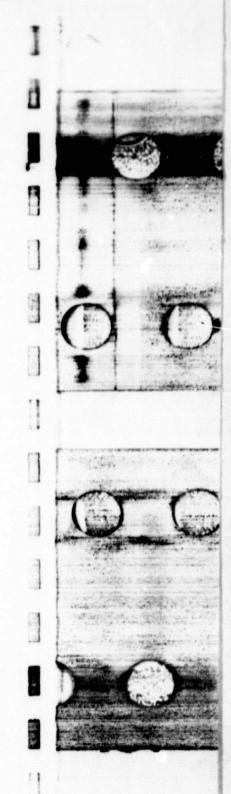


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TABLE I—Continued

*****	•••••	******		***********	***********	*****	•••••••	******
(1)	(11)		11)	(11)	(V)	(11)	(VII)	(ALLL)
610	QP(7. 1.E)	4 1	1 €	1248.7396	1248.7539	-14	0.167E+05	1.0
611	201 7. 2.F 1	6 1	2 5	1249.0912	1249.0973	-6	0.154E+05	1.0
614	201 7. 3.A2)	6 1	3 41	1249.6617	1249.6719	-10	0.132E+05	1.0
614	2P(7. 1.41)	6 1	3 42	1249.6617	1949.6719	-10	0.132E+05	1.0
617	201 7. 4.E 1	6 4	4 F	1250.4849	1250.4814	3	0.104E+05	1.0
619	OP1 7. 5.E 1	6 5	5 F	1251.5391	1251.5324	6	0.7136+04	1.0
623	OP1 7. 6.4 1	6 6	4 4	1252.8577	1252.9360	21	0.719E+04	*1.0
641	3P1 6. 0.411	5 0	0 42	1257.4662	1257.4729	-6	0.1936+05	1.0
642	201 A. 1.F 1	5 1	1 5	1257.5778	1257.5985	-10	0.186F+05	1.0
643	201 6. 7.5 1	5 2	2 F	1257.9310	1257.9361	-5	0.167E+05	1.0
644	201 6. 3.411	5 1	3 42	1258.5261	1258.5181	7	0.136E+05	1.0
644	OP1 6. 3.421	5 1	3 41	1258.5261	1258.5182	7	0.136E+05	1.0
647	301 A. 4.F 1	5 4	4 F	1259.3419	1259.3390	,	0.966F+04	1.0
652	OP! 6. 5.F 1	5 5	5 E	1260.3726	1260.4066	-33	0.501E+04	0.0
669	OP1 5. 7.421	4 0	0 A1	1266.1437	1266.1472	- 3	0.203E+05	1.0
670	701 5. 1.E 1	4 1	1 6	1266.2601	1266.2640	- 3	0.194E+05	1.0
671	OP1 5. 2.E 1	4 2	2 F	1266.6180	1266.6154	2	0.166E+05	1.0
673	201 5. 3.421		1 41	1267.2121	1267.2019		0.123E+05	1.0
673	OP! 5. 3.41)	4 1	1 42	1267.2121	1267.2040		0.123E+05	1.0
676	OP! 5. 4.5 1	4 4	4 1"	1268.0493	1268.0349	14	0.659E+04	1.0
447	OP! 4. 0.41)	3 0	n #5	1274.6616	1774.6574	•	0.198E+05	1.0
688	OP! 4. 1.E 1	1 1	1 5	1274.7719	1274.7702	1	0.184E+05	1.0
689	301 4. 2.E)	3 2	2 E	1275.1281	1275.1246	. ?	0.145E+05	1.0
691	OP! 4. 1.A1)		3 42	1275.7297	1275.7186	11	0.816E+04	1.0
601	OP! 4. 3.47)	2 2	3 41	1275.7297	1275.7186	11	G.816E+04	1.0
598	OP! 3. 0.A?)	2 0	2 41	1282.9971	1282.9798	12	0.174E+05	1.0
629	00(3. 1.E)	2 1	2 5	1283.4689	1283.4550	13	0.153E+05	1.0
7 11	20(17. 6.A.)		6 A	1287.0584	1287.0198	38	0.1506+03	0.0
716			4 4	1289.1251	1249.1021	22	0.3006+03	0.0
715	00(15. 5.5)	16 6	5 F	1290.0299	1290.0282	1	0.2185+03	1.0
, ,,	22(14. 3.42)	14 3	3 A1	124	1290.4359		0.164E+03	0.0
742	00114. 4.F 1	14 4	4 F	1291.1420	1291.1526	-10	0.2786+03	1.0
742	22(15. 6.A)	15 6	6 4	1291.1420	1291.1707	-28	0.5826+03	*1.0
742	2P(2. 0.A1)	1 0	0 42	1291.1420	1291.1226	19	0.1316+05	1.0
743	2P1 2. 1.E)	1 1	1 5	1291.2522	1291.2415	10	0.973E+04	1.0
746	00114. 5.F)	14 5		1292.0914	1292.0753	16	0.409E+03	1.0
748	20113. 3.411	19 9	1 42	1292.4438	1292.4328	10	0.298E+03	1.0
748	QQ(13. 3.A2)	13 3	3 A1	1292.4438	1292.4395	4	0.298E+03	1.0
749	20(15. 7.E)	15 7	7 F	1297.5970	1292.5527	44	0.362E+03	0.0
754	QC(13. 4.F)	13 4	4 F	1293.2126	1293.1535	59	0.506E+03	0.0
754	20114. 5.4 1	14 6	6 A	1293.2126	1293.2145	-1	0.109E+04	*1.0
	00(13. 5.E)	13 5	5 E		1294.0807		0.745E+03	2.0
757	00(15. B.F)	15 8	8 F	1294.2333	1294.2126	20	0.423E+03	1.0
757	00(17.12.A)	17 12		1294.2333	1294.7292	4	0.262E+03	1.0
758	20115. 3.423	12 3	9 A1	1294.3861	1294.3600	17	0.576E+03	1.0
758	22(12. 3.A1)	12 3		1294.3861	1294.3738	12	0.526E+03	1.0
758	00(16. 9.A)	16 9		1294.3861	1294.4098	-23	0.468E+03	0.0
799	20(14. 7.F)	14 7		1294.5814	1294.5514	0	0.681E+03	1.0
761	00(12. 4.E)	12 4		1295.1044	1295.0958	•	0.892E+03	1.0
762	00(13. 6.A)	13 6	.6 4	1295.2281	1295.2216	6	0.199E+04	•1.0
***	OP(14.11.E)	13 11		1204 704	1195.4468		0.268E+03	0.0
767	00(11: 2.F)	11 2		1295.7084	1295.7092	.0	0.413E+03	1.0
768	20(12. 5.E)	12 5		1296.0429	1296.0307	12	0.131E+04	1.0
769	00(11. 3.41)			1296.2391	1296.2333	:	0.899E+03	1.0
769	00(11. 3.A2)	14 5		1296.2381	1296.1985	19	0.7995+03	
769	20(15. 9.4)	15 9		1296.2381	1296.2598	-21	0.9255+03	
770	99(15.11.E)	15 11		1296.4855	1296.4735	11	0.404E+03	
771	20(13. 7.5)	13 7		1296.9876	1296.5825	•	0.1245+04	
,,,						,	**********	



V3 BAND OF 12CH3D

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TABLE I-Continued

(1)	(11)		11	111		(1V)	(V)	(VI)	(VII)	(VIII)
*****		***						*****	**********	
7	99111. 4.F 1	11	4		E	1296.9681	1296.9650	,	0.153E+04	1.0
715	20110. 1.5 1	10	1	1	F	1297.1873	1297.1540	32	0.175E+03	0.0
775	00(12. 6.4)	12			٠	1297.1873	1297.1790		0.351E+04	*1.0
777	20110 . 2 . 5)	10	2	. 2	•	1297.4844	1297.4724	11	0.685E+03	7.0
777	00(16.17.4)	16	12	12	•	1207.4844	1297.4760		0.5466+03	1.0
720	00111. 5.5 1	11	2		٤.	1297.9049	1297.9111	-6	0.224E+04	1.0
791	00(10. 3.47)	10	:	:	41	1298.0025	1298.0014	1	0.149E+04	2.7
781	00(10. 3.41)	10		9	47	1298.0025	1298.0014	0	0.149E+04	1.
793	20(13. 5.6)	13	Ä	ě	è	1298.1384	1298.1291		0.177E+04	1.0
796	20112. 7.5 1	12	,	7	è	1298.5400	1298.5442		0.146E+04	1.0
787	201 9. 1.5 1	' 6	1	1		1298.7523	1798.8085	-56	0.282E+03	0.0
787	99(10. 4.F)	10		ä		1298.7523	1298.7465	5	0.253E+04	1.0
789	001 9. 2.F 1	9	2	2		1299.0926	1299.1308	-38	0.111E+04	0.0
799	00(15.10.F)	15	10	10	٠	1299.0926	1299.1781	-85	0.479E+03	0.0
789	OP! 1. 0.471	0	0	0	41	1290,0926	1299.0756	16	0.709E+04	1.0
789	99111. 5.A 1	11	6	6		1299.0926	1299.0725	20	0.601E+04	*1.0
792	001 9. 3.411	9	•		42	1790.6747	1299.6686	6	0.241F+04	1.9
792	001 9. 3.AZ1	9	3	,	41	1299.6747	1299.6698		0.241E+04	1.0
792	00(10. 5.F)	10	•	. 5		1290.6747	1299.7069	-32	0.373E+04	0.0
793	00(14.11.E)	14	11	13	•	7.799.8933	1799.8481	45	0.816E+03	0.0
794	00(13. 9.4)	13	9	9		1100.0061	1400.0248	-18	0.726E+04	1.0
725	00(12. S.E.)	17			7	1300.1177	1300.1307	-12	0.258E+04	1.0
798	901 1. 1.E 1	6	1	1	:	1300.3389	1100.4429	-:	0.443E+03	1.0
799	90(11. 7.E)	5.5	*	4		1.00.4355	1300.4356	-16	0.409E+04	1.0
800	001 4. 2.5 1		2	2		1 *00 .6453	1300.6705	-25	0.1745+04	1.0
.00	22(14.10.5)	14	10	10	F	1100.6455	1100.6849	-39	0.8596+03	0.0
800	20(15.12.4)	15	12	12		1300.6453	1100.4152	10	0.1105+04	0.0
801	22(16.13.5)	16	12	19		1300.7702	1300.7489	21	0.288E+03	1.0
807	20110. 6.4 1	10				1900.8746	1900.8967	-12	0.9985+04	•1.0
803	Q01 8. 3.421		3	,	41	1301.2119	1301.2176	-5	0.378E+04	1.0
803	001 8. 3.411		3	•	42	1301.2119	1301.2182	-6	0.378E+04	1.0
874	001 9. 5.5)	9	,	•	E	1201.3914	1301.4079	-11	0.601E+04	1.0
	001 7. 1.5 1	7	1	1	E		1401.7451		0.681E+03	0.0
806	001 4. 4.5 1		4	4	•	1301.9490	1901.9876	-38	0.642E+04	0.0
806	00(12. 9.4)	12	0	9	*	1301.0400	1301.9454	.:	0.582E+04	1.0
807	901 7. 2.F 1	. ?	?	?	Ē	1302.0441	1302.0700	-33	0.267E+04	0.0
. 18	20(11. 8.5)	11	:	;	F	1302.7682	1102.0512	-21	0.443E+04	1.0
	90(13.10.F)	13	12	10	ř	1302.2682	1302.2734	-5	0.167E+04	1.0
809	001 7. 3.41)	7			A2	1107.6104	1302.6345	-24	0.5816+04	1.0
.09	991 7. 3.471	7			41	1302.6104	1302.6348	-24	0.5816+04	1.0
809	001 9. 6.A 1	9				1302.6104	1402.4044	4	0.161E+05	*1.0
811	001 6. 1.E 1	6	1	1	F	1302.9755	1303.0025	-27	0.1036+04	1.0
.11	991 A. S.E 1	8		4	E	1302.9755	1302.9835	-8	0.946E+04	1.0
*12	00(13.11.E)	13	11	11	E	1303.1502	1303.1039	46	0.165E+04	0.0
813	001 6. 2.5 1	6	2	2	F	1303.3733	1303.3407	-17	0.402E+04	1.0
.14	001 7. 4.E 1	. 7	. 4	. 4	E	1303.3894	1303.4182	-28	0.987E+04	1.0
815	00(14+12+4)	14	12	12	•	1303.5787	1303.5775	. 1	0.2121.+04	1.0
816	00(15+13+E)	15	13	13	٢.	1303.7217	1303.7349	-13	0.567E+03	1.0
817	201 6. 3.421	6	3	:	A1	1303.8981	1103.9064	-8	0.876E+04	1.0
817	20(10. 8.6)	10		;	Ē	1303.8981	1303.9046	-25	0.876E+04	1.0
817	20(11. 9.4)	11		-	Ä	1303.8981	1303.8785	19	0.100E+05	1.0
817	20(16.14.E)	16	14	14	Ē	1303.8981	1303.8947	.,	0.284E+03	1.0
819	221 9. 7.F 1		7	-7	E	1304.0319	1374.0405	-6	0.1016+05	1.0
819	20(12.10.E)	12	10	10	F	1304.0310	1204.0089	22	0.311E+04	0.0
820	201 5. 1.E 1	5	1	1	E	1304.0760	1904.1097	-27	0.153E+04	1.0
820	20117.15.4 1	17	15	15		1304.0760	1904.0857	-9	0.269F+03	1.0

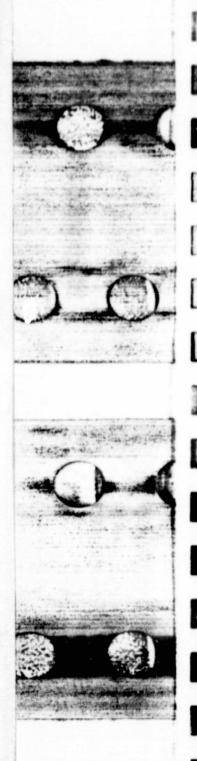
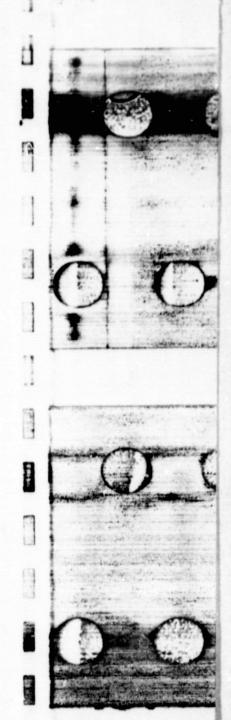


TABLE 1--Continued

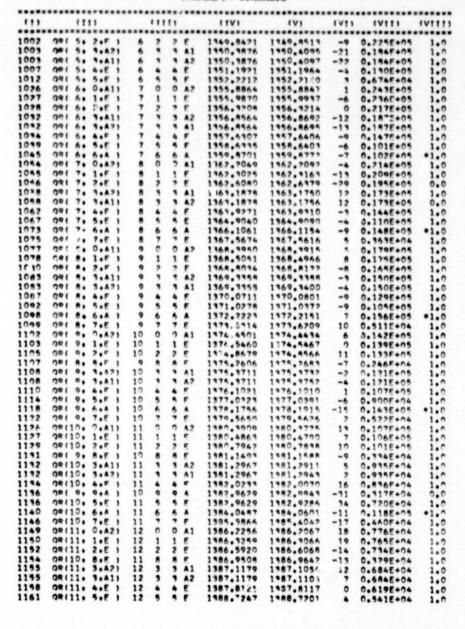
****	************		•••	***		**********			•••••	******
(1)	(11)		111	11		(11)	(V)	(11)	(VIII)	(VIII)
*****	************	****		***	****	**********	**********	*****	••••••	******
821	201 A. 6.4 1		6	6		1304.2096	1304.2120	-2	0.253E+05	.1.0
822	991 5. 2.5 1	5	2	2	•	1304.4384	1304.4468	-8	0.598E+04	1.0
822	201 7. 5.F 1	7			F	1304.4384	1204.4338		0.145E+05	1.0
874	201 6. 4.F 1	6		4		1104.7007	1204.7038	-1	0.149F+05	1.0
926	901 4. 1.F 1	4	1	1		1305.0240	1305.0382	-14	0.276E+04	1.0
826	221 5. 3.41)	5	4		A2	1305.0240	1105.0212	2	0.130E+05	1.0
876	701 5. 1.A21		1		41	1305.0240	1105.0211	2	0.1306+05	1.0
	001 4. 2.E 1	4	2	2	•	1305.4025	1305.3867	16	0.888E+04	1.0
828	991 7. 6.4 1	7	6	6		1305.7093	1105.6899	19	0.389E+05	91.0
828	991 A. 7.E)		7	7	•	1305.7093	1305.684	25	0.158E+05	1.0
828	901 9. A.E)	9	8			1405.7093	1305.7217	-12	0.119E+05	1.0
879	001 6. 5.F 1	6			E	1305.7566	1105.7300	17	0.219E+05	1.0
. 10	201 3. 1.5 1	3	1	1		1305.8102	1305.7970	13	0.342F+04	1.0
	20110. 9.4 1	10	9	0		1904.8102	1105.8020		0.167E+05	1.0
830	001 5. 4.E 1	5	4	4	E	1305.8102	1305.8316	-21	0.271E+05	1.0
812	221 4. 1.411	4			42	1305.9571	1305.9680	-10	0.1935+05	1.0
832	201 4. 3.42)	4	4		41	1305.9571	1305.9680	-10	0.193E+05	1.0
832	20111-10-E 1	11	10	10	E	1305.9571	1305.9246	12	0.548E+04	0.0
813	901 3. 2.E 1	,	2	2		1306.1410	1306.1484	-6	0.134E+05	1.0
833	99112.11.E 1	12	11	11		1906.1419	1106.0893	52	0.338E+04	0.0
834	00(13.12.4)		12	12		1306.2735	1306.2956	-22	0.392E+04	1.0
835	901 2. 1.E 1	2	1	1	E	1306.3951	1306.3729	22	0.547E+04	1.0
836	20114-11-5		19	1.		1306.5175	1206.5429	-25	0.107E+04	1.0
837	201 1. 1.F 1	1	1	1	Ē	1306.7713	1306.7600	11	0.106E+05	1.0
837	201 2. 2.E 1	2	2	2	E	1306.7713	1306.7272	44	0.214F+05	0.0
837	001 1. 1.411	•	3		42	1306.7713	1396.7374	33	0.292E+05	0.0
817	201 1. 1.421	•				1306.7713	1376.7374	**	0.297E+05	1.0
837	201 4. 4.E 1	4	4			1306.7713	1306.7902	-18	0.328E+05	1.0
837	20115-14-5 1		14	14		1306.7713	1306.8308	-59	0.554E+03	0.0
838	001 5. 5.E)	5	5			1306.8995	1306.8856	13	0.326E+05	1.0
819	201 6. 6.4 1	6			to	1307.0396	1307.0232	16	0.586E+05	•1.0
	00(16+15+4)		15	15			1307.1596		0.542E+03	0.0
840	201 7. 7.F 1	7	7	7		1307.3200	1307.2027	17	0.243E+05	1.0
841	221 8. R.E 1					1307.4439	1907.4298	20	0.187E+05	1.0
842	221 9. 9.4 1		9	0		1307.7097	1307.6860	23	0.249E+05	1.0
843	20110.10.F)	10	10	10	-	1308.0147	1407.9889	25	0.909E+04	1.0
845	00(11.11.E)		11	11	E	1308.3554	1308.3318	23	0.577E+04	1.0
847	00112+12+4 1		12	12		1308.7345	1108.7141	20	0.690E+04	1.0
	22(13.13.F)		13	13		1309.1452	1309.1352	9	0.1955+04	1.0
849	22114+14+F 1		14	14		1309.6168	1109.5944	22	0.104F+04	1.0
851	Q0(15.15.A)		15	15		1310,1039	1310.0909	12	0.105E+04	1.0
866	OP! 0. 0.41)	1	0	0	42	1314.4156	1314.4017	13	0.745E+04	1.0
893	OP(1. 0.42)	2	0	2	AI	1321.7830	1321.7741		0.144E+05	1.0
894	OP! 1. 1.E 1	2	1	1	F.	1321.8905	1121.8913	0	0.107E+05	1.0
918	291 2. 2.A11	3	ó	'n	A2	1328.9631	1128.9555	7	0.202E+05	1.0
919	OR1 2. 1.E 1		1	1	E	1329.0740	1329.0716	2	0.178E+05	1.0
921	OR1 2. 2.E 1	3	2	ż	Ē	1329.4305	1129,4257	9	0.109E+05	1.0
946	OR! 1. 7.421	4	ō	ñ	Al	1335.9591	1117,9414		0.2415905	1.0
947	QP(3. 1.F)	4	1	1	Ē.	1336.0644	1434.0640	8	2.2246.005	1.0
949	091 3. 2.F 1	4	ż	ż	F	1376,4160	1116.4465	6	5.1708+05	1.0
991	291 3. 3.421	4	i	3	41	1944,2944	1436.4868	6	0.9945+04	1.0
931	291 3. 3.A11		9	•	A2	1396.0944	1436.9868	6	0.994E+04	1.0
975	OP! 4. 0.411	•	0	2	A2	1942.7861	1942.7649	o	0.260E+05	1.0
976	001 4. 1.E 1	•	1	1	Ē	1342.8736	1342.8778	-4	0.248E+05	1.0
978	201 4. 2.E 1	•	ż	ż	Ē	1343.2113	1943.2171		0.213E+05	1.0
979	201 4. 3.A11	•	ŝ	i	A2	1343.7819	1343.7853	-9	0.157E+05	1.0
979	201 4. 1.421	ś		i	AT	1949.7818	1143.7854	-1	0.1975+09	1.0
983	OR! 4. 4.E 1	í	í	i	£,	1344.5945	1144.1869	-;	0.844E+04	
1000	201 5. 2.421		6	-	A1	1940, 1940	1949.4047	-11	0.2605+09	1.0
1001	ORI 5. 1.E 1		1	1	Ē,	1349,5020	1349.5177	-19	0.2516+09	1.0
		•	•			. ,		-19	0.15315444	1.0



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TABLE I-Continued



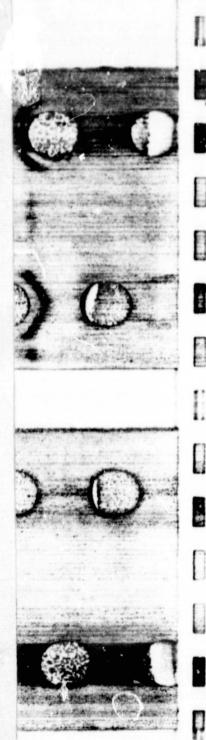


TABLE I—Continued

(1)	(11)		111	11		(17)	(V)	(VI)	(111)	(VIII)
	************	****		***	***				•••••	*****
1161	OR(10. 9.4)	11	9	9		1388.7247	1388.7429	-18	0.413E+04	1.0
1146	09(11. 6.A)	12	6	6		1389.8852	1389.8358	49	0.912F+04	0.0
1168	79(10.10.E)	11	10	17		1390.7500	1390.7367	13	0.9675+03	1.0
170	00 (11. 7.5)	12	7	7		1391.1547	1991.1674	-7	0.3675+04	1.0
173	20112. 2.411	13	0	2	42	1391.9574	1991.9441	13	0.536E+04	1.0
174	20(12. 1.5)	13	1	1		1392.0532	1392.0426	10	0.5295+04	1.0
175	28(12. 2.F)	13	2	2	F	1392.3553	1197.1197	16	0.510E+04	1.0
176	20111 . A.F)	12				1392.6906	1192.7047	-13	0.279E+04	1.0
177	OR(12. 3.41)	13	3		42	1397.8453	1392.9301	15	0.478E+04	1.0
177	20112. 1.421	13	3	3	41	1392.8453	1397.8368		0.478F+04	1.0
191	98(12. 4.E)	13	4	4		1191.5445	1191.5285	15	0.435E+04	1.0
183	OR(11. 9.A)	12	9	9		1394.4789	1394.4684	10	0.390E+04	1.0
183	20112. 5.E 1	13	•	•	€ .	1394.4789	1394.4270	51	0.385E+04	0.0
197	20(12. 6.A)	13	6	6		1395.5291	1395.5329	-3	0.660E+04	•1.0
193	29(11.10.F)	12	10	10	F	1396.5253	1396.4751	50	0.117F+04	0.0
195	OR(12. 7.F.)	13	7	7	E	1396.8570	1396.8522		0.272E+04	1.0
198	20(11. 0.42)	14	0	0	41	1397.5992	1297.4031	-1	0.355E+04	1.0
199	20(13. 1.5)	14	1	1	E	1397.7233	1397.7004	?2	0.351E+04	0.0
201	Q0(13. 2.E)	14	2	2	E	1397.9883	1497.9931	-4	0.338E+04	1.0
202	OR(12)	13			E	1399.3840	1298.1961	-12	0.2135+04	1.0
203	OP(13. 3.A2)	14	3	•	41	1398.4775	1398.4779	0	0.318E+04	1.0
203	OR(11. 1.A1)	14	3	1	A2	1308.4775	1398.4868	-9	0.318E+04	1.0
203	OR(11.11.E)	12	11	11	E	1398.4775	1398.4926	-15	0.561E+03	1.0
	OR(17. 4.F)	14	4	4			1399.1708		0.297E+04	0.0
206	09(11. 5.E)	14	5	- 5	F	1400.0912	1400.0627	50	0.261E+04	1.0
207	90112. 9.A 1	13	9	9		1400.1623	1400.1910	-29	0.3145+04	1.0
210	OR(13. 6.4)	14	6	5		1401.1829	1401.1644	18	0.4. *+04	*1.0
214	00(12.17.E)	13	10	17		1402.3239	1402.3792	-45	n.1006+04	0.0
215	OP(13. 7.E)	14	7	7		1402.5100	1402.4868	23	0.1895+04	1.0
219	20114. 0.A11	15	2	0	¥5	1403.1619	1403.1975	-35	0.224E+04	1.0
219	70(12.11.E)	13	11	11	E	1403.1629	1403.1418	20	0.625E+03	1.0
220	28114. 1.F 1	15	1	1	E	1403.2590	1401.2918	-35	0.223E+74	1.0
272	00114. 2.E 1	15	2	2	E	1403.5545	1403.5834	-58	0.215E+04	1.0
7 14	OR(14. 3.41)	15	,	3	A?	1404.0544	1404.0626	-8	0.203E+04	1.0
224	28(14. 3.A2)	15	3	3	41	1404.0544	1404.0737	-19	0.203E+04	1.0
274	OP(13. 9.E)	14	A		F	1404.0544	1404.0524	2	0.1516+04	1.0
	OR (14. 4.F)	15	4	4	E		1404.7514		0.187E+04	0.0
226	OR (14. 5.5)	15	5	5	E	1405.6479	1405.6396		0.168E+04	1.0
227	OP(13. 9.4)	14	9	9		1405.8946	1405.9244	-59	0.228E+04	1.0
229	20(12+12+A)	13	12	12		1406.2498	1406.2589	-9	0.618E+03	1.0
230	OR(14. 5.4)	15	6	4		1406.7701	1406.7420	28	0.294F+04	•1.0
232	20113-11-5 1	14	11	11		1407.5922	1407.5051	97	0.521E+03	0.0
235	OR(14. 7.5)	15	7	. 7	E	1408.1144	1408.0766	37	0.124E+04	1.0
236	OR(13.10.F)	14	10	10	E	1408.3797	1408.4147	-35	0.724E+03	
239	28(15. 0.A2)	16	0	0	AI	1408.7720	1408.7417	30	0.137E+04	1.0
238	28(15. 1.E)	16	1	1	F	1408.777	1408.8370	-55	0.136E+04	0.0
239	OR(15. 7.F)	16	2	2	E	1409.0475	1409.1237	-76	0.131E+04	0.0
241	OR(15. 3.47)	16	3	3	A 1	1409.5728	1409.5978	-65	0.124E+04	0.0
241	C" . 15. 3.41)	16	3		W.S	1409.5328	1409.6110	-78	0.124E+04	0.0
242	09(14. 9.F)	15	8		F	1409.7259	1409.6817	44	0.101E+04	0.0
245	OR(15. 4.E)	16	4	4	E	1410.2175	1410.2833	-65	0.115E+04	0.0
248	GR(15. 1.E)	16	. 5	. 5		14:1.1606	1411.1697	-9	0.104E+04	1.0
248	28(13.12.A)	14	12			1411.1606	1411.1545	6	0.736E+03	1.0
251	38(14. 9.A)	15	9	9		1411.6885	1411.6665	21	0.157E+04	1.0
251	28(14.11.E)	15				1411.7527	1411.7329	19	0.385E+03	
254	07(15. 6.A)	16	6			1412.3049	1412.2767	28	0.182E+04	
259	09(15. 7.E)	16		7		1413.6711	1413.6304	40	0.775E+03	1.0
1261	OR(16. 0.A1)	17				1414.1771	1414.2513	-74	0.801E+03	
1262	08(16. 1.F)	17	1	1	E	1414.3022	1414.3455	-49	0.793E+03	
1263	28(14.19.F)	15	10	19		1414.4955	1414.4192	-29	0.475E+03	1.0

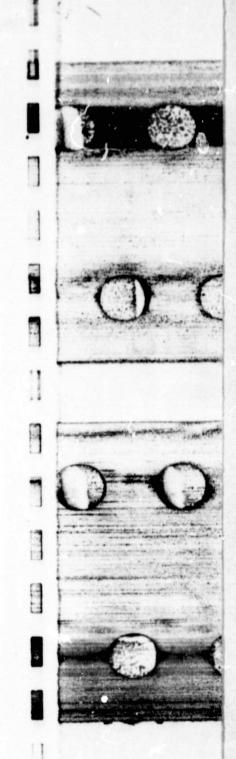


TABLE 1-Continued

(1)	(11)		11	111		(11)	(V)	(VI)	(VII)	IVITI
	********					**********				
	98(16. 2.E	1 1	7 2	2	E		1414.6291		0.768E+03	0.0
	SR(16. 3.A	1) 1	7 3		42		1415,0983		0.728E+03	
	OR(16. 3.4	21 1	7 3	3	A1		1415.1129		0.72RE+03	
1266	QR(15. 8.E	1 10			E	1415.3525	1415.2882	64	0.633E+03	-0.0
1267	OR(14.12.A	1 1	12	12		1415.7685	1415.7894	-20	0.630E+03	1.0
1247	98116. 4.E	1 1	7 4		5	1415.7685	1415.7800	-11	0.674E+03	1.0
	QR(15.11.E	1 10	11	11	E		1415.8919		0.261E+03	0.0
1272	28(16. 5.E	1 1	1 5		E	1416.6427	1415.6652	-22	0.612E+03	1.0
	OP(15. 9.A	1 1	9				1417.4099		0.961E+03	0.0
1275	38(16. 6.A	1 1	1 6	6		1417.8119	1417.7782	33	0.1080+04	0.0
	99(16. 7.E	1 1	7	7	E		1419.1547		0.461E+03	0.0
1201	20115.17.4	1 1	12	12		1421.,2789	1420.2295	49	0.461E+03	0.0
1283	OP (15.10.E	1 1	10	10	E	1420.6842	1420.6171	67	0.293E+03	0.0
	3P(16. 8.F	1 1	7 8		F		1420.8723		0.378E+03	0.0
1293	OR (16. 9.A	1 1	7 9	9		1423.1916	1423.1404	51	0.573E+03	0.0
1297	OR(16.12.A	1 1	1 12	12		1424.5292	1424.5432	-13	0.305E+03	1.0
1305	38116.10.E	1 1	10	10	E	1426.6595	1425.6710	-13	0.177E+03	1.0
V6 TR	ANSITIONS N	EWLY	1551	GNE	0					
35	PP(16. 9.A	1 1		9		1055.6194	1055.6275	-9	0.511E+03	1.0
63	PP(16.12.4	1 1	11	12		106 5848	1067.6641	-79	0.427E+03	0.0
67	PP(15.10.F	1 1			E	1068.8419	1068.8537	-11	0.45RE+03	1.0
77	PP(15+11+F	1 1		11		1073.0315	1073.0654	-33	0.434E+03	1.0
86	DD(15.12.A	1 1	11	12		1077.3742	1077.3384	-14	0.793E+03	1.0
93	PP(15.13.F	1 1		7.3		1081.3394	1081.3713	-31	0.323E+03	1.0
94	PP(14+11+F	1 1	10	11	F	1002.3541	1082.4020	-47	0.777F+03	1.0
103	PD(14+12+A	1 1	11	12		1096.9476	1086.8741	-25	0.144E+04	1.0
113	PP(14+13+E) 1	1 12	13	E	1091.3393	1091.3372	2	0.617E+03	1.0
176	PD(14+14+	1 1	111	14	F	1097.5698	1097.6238	-55	0.532E+03	0.0
357	PO116. 4.E	1 1			F	1173.3725	1173.4016	-29	0.294E+03	0.0
343	P7115.10.E	, 1		10	E	1182.4813	1182.5157	-34	0.283E+03	1.0
418	PO(14.17.A	1 1	11	12		1192.4664	1192.5126	-46	0.373E+03	0.0

RESULTS AND DISCUSSION

All experimental data assigned to the ν_3 band of $^{12}\text{CH}_3\text{D}$ are listed in Table I, column IV. ν_6 trans involving upper levels near the crossing, which could not be identified from the separated analysis of the ν_6 band are now reported at the end of Table I. Serial numbers of column I refer to the numbering of the lines in Fig. 1 of the present paper and in Fig. 1 of Ref. (1). Assignments and specifications of the upper state levels of the transitions are reported in columns II and III of the table.

In all, 342 transitions belonging to ν_3 are retained to be fitted besides 641 ν_4 transitions previously assigned in Ref. (1), or newly assigned in the present paper. Statistical weights equal to 1.0 are attributed to all data. For transitions going up to levels with $|K - l_6| = 0$ or 3 for $v_{\delta} = 1$, and with |K| = 3 for $v_{\delta} = 1$, A_1A_2 splittings are theoretically predicted, which corres d to observable effects; then, weights equal to 1.0 are assigned to each consent $A_1 \rightarrow A_2$ and $A_2 \rightarrow A_1$. For transitions going up to levels with $|K - l_6| = 6$, 9 ($l_6 = 0$ for $v_3 = 1$), no splitting is expected and the overall weights are equally distributed for the two components transitions with asterisk in column VIII of Table I.

The fitting of the data leads to 21 significant constants. Six of these constants are related to $v_3 = 1$, 12 are related to $v_6 = 1$, and 3 are interaction constants between the two states; these last, i.e., $C_{21}^{(2)}$, $C_{21}^{(3)}$, and $C_{21}^{(4a)}$ result from the second-order Coriolis interaction defined by Eq. (4). All constants involved in the first-order

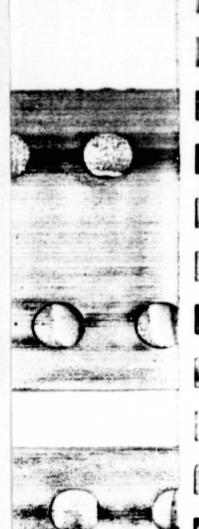
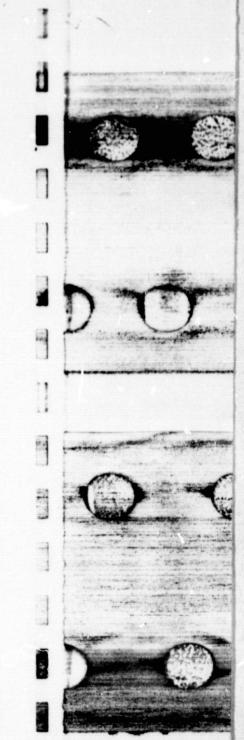


TABLE II Spectroscopic Constants for Vibrational States $v_3=1$ and $v_6=1$ of $^{12}{\rm CH_3D}$

CONSTANT	VALUE IN c	m ⁻¹	STANDARD D	EVIATION	99.5% CO	NFIDENCE IN cm
·v ₃	1306.836		0.002		0.02	
B _v 3	3.78294		0.00006		0.0004	
*A, 3	5.27208		0.00007		0.0005	
D 3	2.396	10-4	0.008	10-4	0.05	10-4
DJK 3	- 1.268	10-4	0.006	10-4	0.04	10-4
D _K 3	- 8.33	10-6	0.31	10-6	2.0	10-6
· v ₆	1161.096		0.002		0.011	
(AC) 6	3.1118		0.0002		0.0012	
B _v 6	3.83481		0.00003		0.0002	
A, 6	5.26735		0.00006		0.0004	
D _J 6	4.490	10-5	0.014	10-5	0.09	10-5
D 6	1.636	10-4	0.004	10-4	0.03	10-4
D _K	- 8.95	10-5	0.04	10-5	0.25	10-5
∙η _, 6	1.424	0-3	0.011	10-3	0.07	10-3
• 7 _K 6	- 4.94	10-4	0.11	111-4	0.72	10-4
·q ₂₂	- 1.742	10-2	0.002	10-2	0.011	10-2
f 22	4.03	10-6	0.08	10-6	0.49	10-6
•q ₁₂	€ 1.84	10-2	0.03	10-2	0.17	10-2
C ₂₁ ⁽²⁾	€'2.42	10-2	0.02	10-2	0.15	10-2
C ₂₁ (3)	€'6.18	10-4	0.09	10-4	0.60	10-4
C ₂₁ (4a)	-€'3.93	10-5	0.08	10-5	0.53	10-5
Constants	fixed to thei		state values	C		
н ^Ј - н _о ^Ј	1.172	10-9				
H _{J1K} - H ₀		10-8		6 • ± 1		
HJKK - HOJI	KK - 1.042	10-8		€'•±1		

Coriolis term, given by Eq. (2), are insignificant, verifying that no direct estimate of the Coriolis constant ζ_{36} can be obtained from available data.

The constants deduced from the fitting are reported in Table II, with their standard deviations and 99.5% confidence intervals. The fourth-order distortion constants H^{J} , H^{JJK} , and H^{JKK} , constrained in $v_3 = 1$ and $v_6 = 1$ to the ground state values determined elsewhere (7), are recalled at the bottom of the table. The 18 constants related to $v_3 = 1$ and $v_6 = 1$ are written using the usual notations, except the asterisks point out that they do not involve exactly the same contributions as in the individual analyses of the bands.



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ν₃ BAND OF ¹²CH₃D

The constants of Table II reproduce the 983 fitted data points with an overall standard deviation equal to 0.016 cm⁻¹. Calculated wavenumbers and deviations (obs - calc) are listed in columns V and VII of Table I for v3 and the newly assigned ν_6 transitions. For all other ν_6 transitions, these quantities do not differ significantly from those given in Table I of Ref. (1) and are not repeated here. Table I includes also predicted wavenumbers for unobserved ν_3 transitions with calculated relative intensities ≥0.200 × 103 in our arbitrary scale. This value corresponds to a limit for observation in the most favorable parts of our spectra. The transitions, for which no A_1A_2 splitting is expected, are reported in Table I with overall notation $A \rightarrow A$. The relative intensities of transitions are calculated according to Eq. (5). Unfortunately, no experimental measurements of line intensities are now available for getting an accurate value of the coefficient $\rho_{3,6}$ involved in Eq. (3). A theoretical estimate of this coefficient can be deduced from band strengths of ν_3 and ν_4 of $^{12}CH_4$ (8). Assuming that the variation of dipole moment with vibration is not significantly affected by the substitution of H by D, two values are expected for $\rho_{3.6}$, i.e.,

 $\rho_{3.6} = 1.11$ and $\rho_{3.6} = 1.16$

according as the derivatives $\partial M_{\alpha}/\partial Q_{3\alpha}$ and $\partial M_{\alpha}/\partial Q_{4\alpha}$ ($\alpha=x,y,z$) of dipole moment in methane have same or opposite signs. In fact, the value which seems to give the best qualitative agreement with observed intensities is $\rho_{3,6}=1.19$ which is consistent with previous results on opposite signs of the two quantities $\partial M_{\alpha}/\partial Q_{3\alpha}$ and $\partial M_{\alpha}/\partial Q_{4\alpha}$ (9,10). Relative intensities of the ν_3 transitions, calculated for $\rho=1.19$, are listed in column VII of Table I. Values for ν_6 transitions are not significantly different from those tabulated in Table I of Ref. (1).

Finally, it may be noted that:

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—* f_{12}^6 . * f_{42}^6 , and * H^K are found to be not significant, pointing out that the corresponding constants f_{12}^6 , f_{42}^6 , and H^K , in the individual analysis of ν_6 , contribute essentially in relation with the Coriolis interaction between $\nu_6 = 1$ and $\nu_3 = 1$.

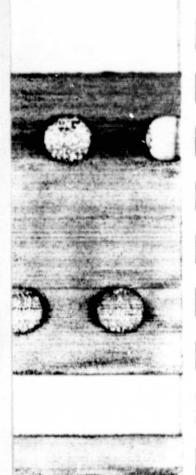
—The distortion constants related to $v_3 = 1$, i.e., ${}^*D_J^3$, ${}^*D_{JK}^3$, and ${}^*D_K^3$, show large deviations from their ground state values. The magnitude and the sign of these deviations are consistent with the expected effect of the Coriolis interaction between $v_3 = 1$ and $v_5 = 1$ when the predicted value of ζ_{35} , i.e., 0.57 is used. Unfortunately, the data now available on the ν_5 band do not allow us to have more information on this interaction.

—The present analysis predicts for ν_3 transitions belonging to the subband |K| = 3, A_1A_2 splittings up to 0.015 cm⁻¹ for J' = 17. Experimentally, such splittings are observed as broadenings of the lines, but are not resolved.

ACKNOWLEDGMENTS

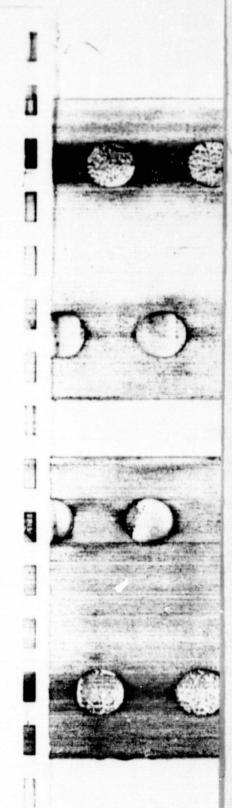
We would like to convey our thanks to M. Dang-Nhu and G. Poussigue for helpful discussions. One of us (K.N.R.) is also grateful to the National Aeronautics and Space Administration for support of some of this research.

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Analysis of the ve Band of 12CHaD at 8.6 µm

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The $v_6(E)$ fundamental vibration-rotation band of monodeuteromethane ($^{12}\mathrm{CH}_3\mathrm{D}$) has been recorded in the spectral range $1033-1270~\mathrm{cm}^{-1}$ with a resolution of approximately $0.04~\mathrm{cm}^{-1}$. Of the 669 transitions with $J' \leqslant 17$ identified, 633 have been retained for the determination of the rotational levels in the upper state $v_6 = 1$. The Coriolis interaction between the $v_6 = 1(E)$ and $v_2 = 1(A_1)$ vibrational states of $^{12}\mathrm{CH}_2\mathrm{D}$ results in large A_1A_2 splittings of levels with $v_6 = 1$ and $|K - l_6| = 0$ or 3; the mixing in K and l_6 also gives rise to some ten forbinden transitions observed in the spectra. These effects have been very well explained within the formulation based on the contact transformation method. Values of 15 molecular structure constants of the $v_6 = 1$ state have been determined from a least-squares analysis of the 633 retained transitions. These constants can be used to estimate values of the upper-state energies up to fourth order, and through them the spectral positions of the 633 retained transitions are reproduced with an overall stand... The action of 0.013 cm⁻¹, which is within experimental uncertainties.

I. INTRODUCTION

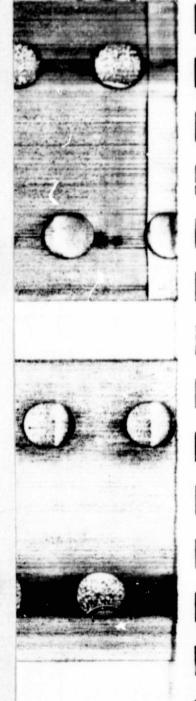
The absorption spectrum of $^{12}\mathrm{CH}_2\mathrm{D}$ in the range 1033–1270 cm⁻¹ is mainly due to transitions of the perpendicular-type $\nu_6(E)$ band, the lowest fundamental vibration-rotation band of $^{12}\mathrm{CH}_2\mathrm{D}$. This band, recently recorded (1) with a resolution of about 0.04 cm⁻¹, exhibits a very well resolved but rather complicated rotational structure. Indeed, because of large values of $A'-A_0$ and $B'-B_0$, very little extended J and K structure can be easily and reliably identified, except in the RQ_0 subbranch. Moreover, the structure of the spectrum is further complicated by A_1A_2 splittings for $J' \geq 8$; the effects of notably large second-order corrections to the energies in the $v_6 = 1$ vibrational state are observed. The Coriolis interaction with the nearest band, the parallel-type $\nu_3(A_1)$ band centered near 1306.8 cm⁻¹, is clearly responsible for the observed features. With the experimental data presently available, the analysis of the ν_6 band of $^{12}\mathrm{CH}_2\mathrm{D}$

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can be considered in two different ways: (a) In the first case, the Coriolis interactions are considered to be small and the analysis of the experimental data can utilize results from the usual contact transformation method (2). It is well known that such a formulation can satisfactorily explain giant A_1A_2 splittings as long as the numerical solution obtained from the analysis remains convergent; that is, as long as the values determined for the structure constants are consistent with their assumed order of magnitude in the expanded matrix energy. (b) In the second case, the Coriolis interactions are considered to be large and the ve band can no longer be analyzed separately from other bands. Then, the treatment becomes more complex: The dimensions of the matrices to be diagonalized are greatly increased; but also, in the present case, the contributions due to the two kinds of vibration-rotation interaction terms, diagonal and nondiagonal in ve, cannot be always separated unambiguously. This point will be discussed in the last part of this paper. Previously, Deroche et al. (3) reported results of the analysis of the ve band of 12CH2D which explicitly included diagonal contributions up to the third order and nondiagonal contributions between the $v_0 = 1$ and $v_2 = 1$ vibrational states. Their study led to an effective value of the coupling constant ζ_{10}^x which is very small, i.e., $|\zeta_{36}| = 0.041$; this indicates that the magnitude of the Coriolis interaction between the $\nu_6 = 1$ and $\nu_3 = 1$ vibrational states is relatively weak. Unfortunately, their formulation failed to explain the A1A2 splittings observed in the P3, O2, and R3 subbranches for J > 9.

In the present work, we show that the available data of the ν_6 band of $^{12}\text{CH}_3\text{D}$ (including J values up to 17) can be interpreted within the formulation based on contact transformations. Many previously unidentified transitions of the ν_6 band are assigned to lines recorded in the range $1033-1270~\text{cm}^{-1}$. Also, owing to the well-resolved spectra, it is now possible to determine values of molecular structure constants through fourth order.

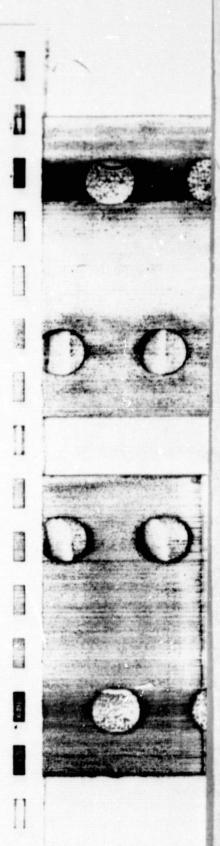
II. EXPERIMENTAL DETAILS

The 3.5-m Littrow-type vacuum spectrograph, operating in the single pass configuration, used to record spectra of ¹²CH₂D has been previously described by Mickelson (4). The data recorded from 1033 to 1270 cm⁻¹ have been obtained in the sixth and seventh orders of the grating with an operating resolution of approximately 0.04 cm⁻¹ over most of the range investigated.

All ¹²CH₃D spectra have been recorded with the gas sample contained in a 1-m-long absorption cell at room temperature. The data have been recorded twice with a gas pressure of 18 Torr and twice with a gas pressure of 30 Torr. Most of the stronger absorption lines have been measured at least three times and some have been measured six times.

All data have been measured relative to the 1–0 band of CO and several water vapor lines using the single pen technique described by Rao et al. (5). These calibration standards are considered accurate to at least ± 0.002 cm⁻¹ (5, 6). All data have been calibrated relative to the accepted standards by the polynomial regression and stepwise multiple regression analysis computer programs described elsewhere (1). The calculated spectral positions of the stronger unblended lines vary no more than ± 0.0025 cm⁻¹ from the corresponding mean values.

Figure 1 is a reproduction of the recorded spectrum; this region includes all identified



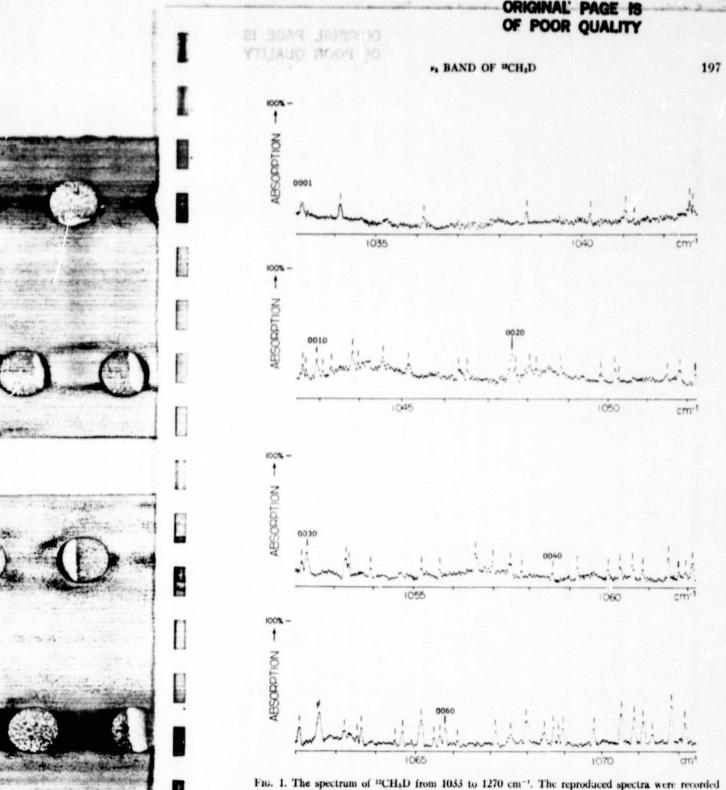
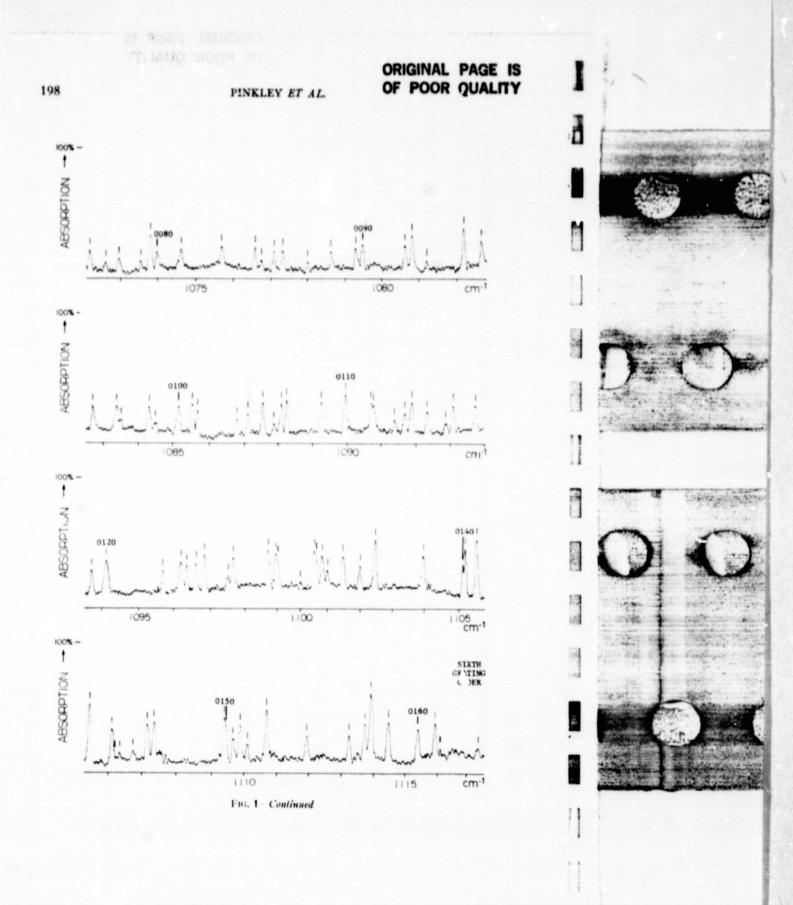
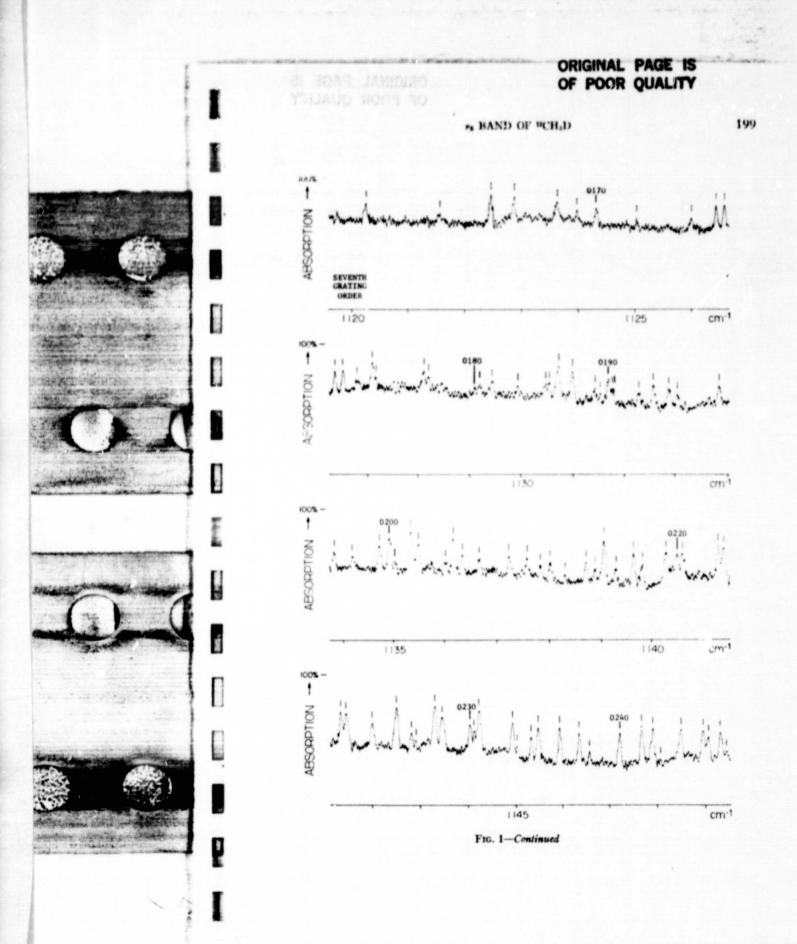
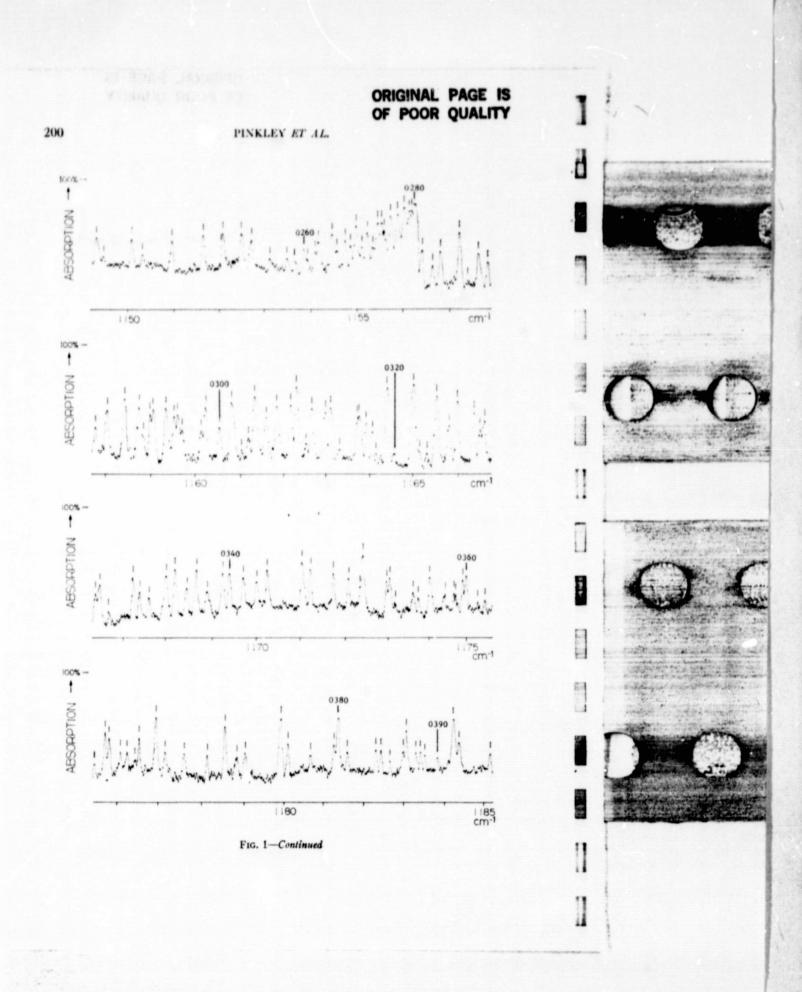
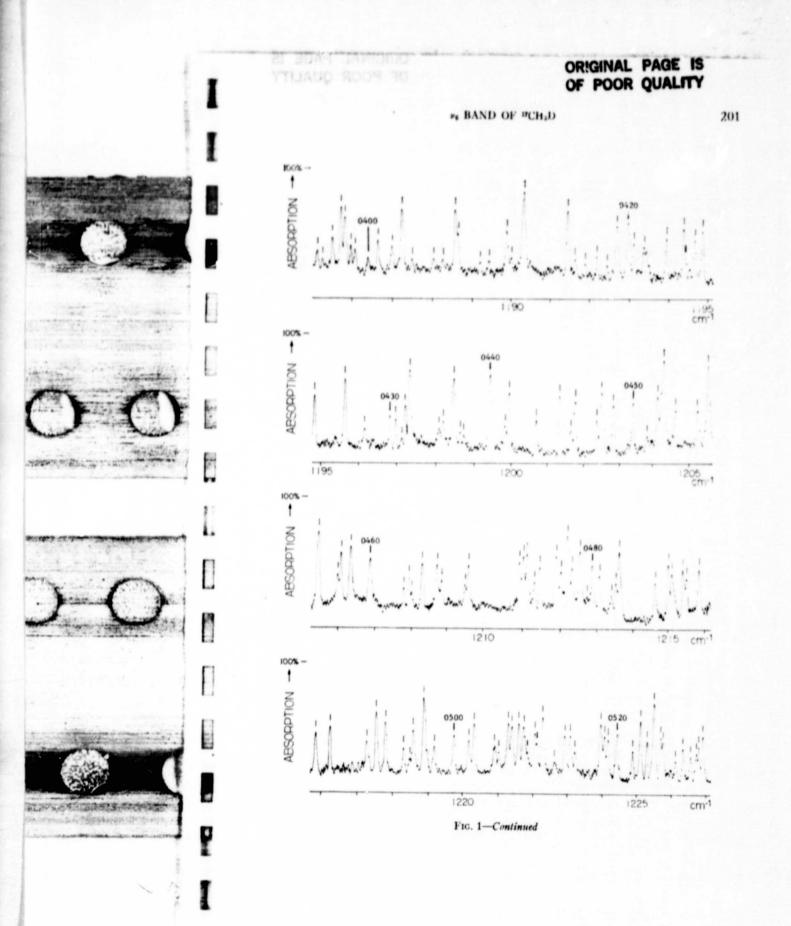


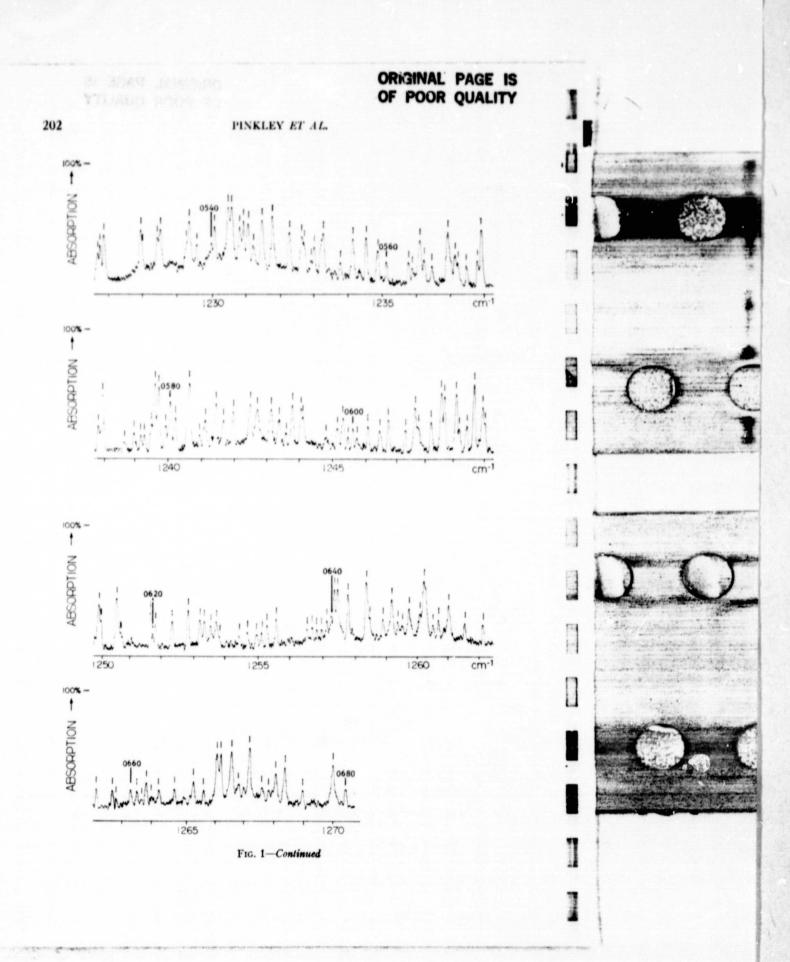
Fig. 1. The spectrum of ¹²CH₃D from 1033 to 1270 cm⁻⁷. The reproduced spectra were recorded with the gas sample contained within a 1-m-length absorption cell at a pressure of 30 Torr.











lines of ν_6 band, which is strongly overlapped on the high-frequency side by the parallel-type ν_2 band of $^{12}\text{CH}_2\text{D}$. The rotational structure in J and K is well resolved throughout the ν_6 band; even the very closely spaced components of the $^{12}\text{CO}_6$ subbranch could be measured from J=1.

III. COMPUTATIONAL PROCEDURE

Assignments of the transitions to observed lines and the determination of molecular structure constants were performed in an iterative, three-step computational procedure:

1. Experimental values of the energies of the upper-state levels, E'_i , with $v_6 = 1$ were calculated using the expression

$$E_i' = hc\sigma_i + E_i'', \tag{1}$$

where σ_i are observed (and assigned) wavenumbers of ν_i and E''_i are energies of the ground-state levels:

$$E''_{6} = E_{6}^{"0} + H_{0}^{K}K^{"6}. \tag{2}$$

Values of $E_i^{"o}$, including all but one term through the fourth order, were determined using the accurate values now available $(7, \delta)$ for the ground-state structure constants. The term involving the fourth-order distortion constant H_0^K remains an unknown to be determined. Thus, Eq. (1) can be expressed as

$$E_{i}^{\prime e} = E'_{i} - H_{0}^{K} K^{\prime\prime 6} \tag{3}$$

where $E_i'^e = hc\sigma_i + E_i''^o$ are considered to be experimental data and H_0^K is a constant to be determined. Because of the accuracy of the $E_i''^o$, the uncertainties of the $E_i'^e$ and $hc\sigma_i$ are equivalent.

2. Eigenvalues, E_i' , and eigenvectors were approximated for the energy matrix expanded to fourth order for the vibrational state with $v_6 = 1$ (see below). In this approximation it was assumed that $H_0^K = H^K$ ($v_6 = 1$); H^J , H^{JJK} , and H^{JKK} were constrained to the values determined for the corresponding ground-state constants (8).

The unknown upper-state structure constants $t_1 \cdots t_k \cdots t_n$, with $t_k = H^K = H_0^K$, were determined utilizing a trivial iterative process based on the least-squares method applied to the function

$$W(t_1 \cdots t_k \cdots t_n) = \sum_{i=1}^{N} g_i [E_i'^e - E'_i(t_1 \cdots t_k \cdots t_n) + t_k K''^e]^2, \tag{4}$$

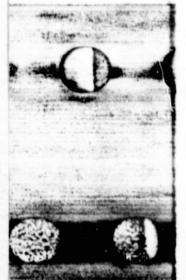
where N is the number of assigned transitions and g_i are suitable statistical weights defined so that $\sum_{i=1}^{N} g_i = N$.

3. Wavenumbers and relative intensities calculated from adjustable values of the molecular structure constants were used to extend progressively the assignments of the transitions to larger values of J and K.

IV. VIBROTATIONAL ENERGIES IN THE UPPER STATE

The energies of the levels of the $v_6 = 1$ vibrational state were calculated using the formulation developed in Ref. (2). The twice-transformed Hamiltonian H^+ , including all terms through h_4^+ , was utilized. The use of symmetrized functions, as defined in





Ref. (10) allows the energy matrix to be factored into four submatrices, i.e., A_1 , A_2 , E_a , and E_b , according to the irreducible representations of the C_{3x} group.³ The energy eigenvalues and eigenvectors were calculated by diagonalization of the A_1 , A_2 , and E_a submatrices.

The contributions to the energy which originate from the term r^2P^2 , in the second-order Hamiltonian h_2^+ , are enhanced by the Coriolis interaction between states with $v_6 = 1$ and $v_3 = 1$:

- (i) The totally diagonal elements are responsible for large vibrational corrections α_6^A and α_6^B to the rotational constants A_e and B_e .
 - (ii) The two nondiagonal elements

$$\langle v_6 = 1, J, K, l_6 = \mp 1 | r^2 P^2 | v_6 = 1, J, K \pm 2, l_6 = \pm 1 \rangle$$

$$= 2q_{22}{}^6 [J(J+1) - K(K \pm 1)] {}^4 [J(J+1) - (K \pm 1)(K \pm 2)] {}^4$$
 (5)

and

$$\langle v_6 = 1, J, K, l_6 = \pm 1 | r^2 P^2 | v_6 = 1, J, K \pm 1, l_6 = \mp 1 \rangle$$

= $2q_{12}{}^6 (2K \pm 1) [J(J+1) - K(K\pm 1)]^{\frac{1}{2}}$ (6)

are responsible for a significant mixing in K and l_6 . Even two fourth-order rotational corrections⁴ to the terms given by Eqs. (5) and (6), i.e.,

$$\langle v_6 = 1, J, K, l_6 = \mp 1 | r^2 P^4 | v_6 = 1, J, K \pm 2, l_6 = \pm 1 \rangle$$

$$= 2 f_{22}^{6,J} J (J+1) [J(J+1) - K(K \pm 1)]^{\frac{1}{2}} \times [J(J+1) - (K \pm 1)(K \pm 2)]^{\frac{1}{2}}$$
(7)

and

$$\langle v_6 = 1, J, K, l_6 = \pm 1 | r^2 P^4 | v_6 = 1, J, K \pm 1, l_6 = \mp 1 \rangle$$

= $2 f_{12}^{6,K} (2K \pm 1)^2 [J(J+1) - K(K \pm 1)]^{\frac{1}{2}}$ (8)

appear to contribute significantly to the energies for $J \ge 10$. Also, the fourth-order doubling element

$$\langle v_6 = 1, J, K = \mp 2, l_6 = \pm 1 | r^2 P^4 | v_6 = 1, J, K = \pm 2, l_6 = \mp 1 \rangle$$

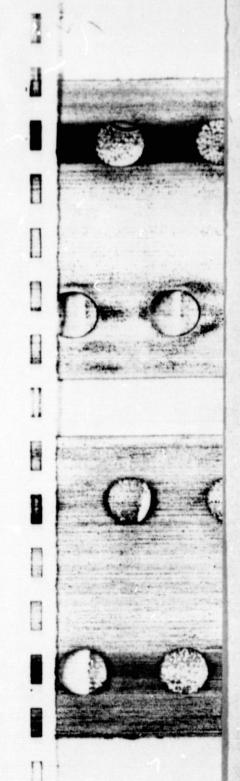
= $2 f_{12}{}^6 J (J + 1) [J (J + 1) - 2]$ (9)

gives significant contributions when $J \ge 8$. All these effects are observable in the spectra of the ν_6 band as an extended J and K structure, as well as by a rapidly increasing A_1A_2 splitting of the transitions with $|K'-l'_6|=0$ and 3, as the value of J increases. Nevertheless, no A_1A_2 splitting has been observed for transitions with $|K'-l'_6|=6,9,\ldots$

Since the mixing in K and l_6 is not extremely large the energy levels of the upper state will be labeled using the quantum numbers and indices $v_6 = |l_6| = 1, J, |K|, |K - l_6|$, and $C(C = A_1, A_2, E)$. The species A_1 and A_2 are defined according to Refs. (10, 11).

³ E_a and E_b are the two components of an E basis, respectively, symmetric and antisymmetric with respect to the operation σ_v of the C_{2v} group.

⁴ Note that the contributions due to $f_{12}^{\bullet,K}$ and $f_{12}^{\bullet,J}$ appeared to be insignificant. The fourth-order vibrational corrections due to r^4P^2 , in h_4^+ , are included in the effective values of A_* , B_* , $(q_{22}^{\bullet})_*$, and $(q_{12}^{\bullet})_*$ appearing in Table II.



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V. RELATIVE LINE INTENSITIES

The relative line intensities were calculated using the expression

$$I_{m \to n} \propto \sigma_{mn} S_{mn} \exp(-E_m/kT),$$
 (10)

where σ_{mn} is the wavenumber of the transition from E_m to E_n and S_{mn} is the square of the transition moment,

$$S_{mn} = 3 \sum_{ij} |\langle mi | M_{Z} | nj \rangle|^{2}. \tag{11}$$

 $|mi\rangle$ and $|nj\rangle$ are the eigenfunctions of H^+ which correspond to E_m and E_n , respectively. The subscripts i and j denote secondary quantum numbers: the magnetic quantum number M, the nuclear spin quantum numbers, and the index³ a or b for eigenfunctions with symmetry species E. M_Z^+ is the twice-transformed component of the dipole moment (12) on the space fixed axis Z.

In the computations of the relative line intensities, M_{z}^{+} has been approximated by retaining only the leading term

$$(M_{Z}^{+})_{0} = M_{Z} = \sum_{\alpha} \cos(Z, \alpha) M_{\alpha}.$$
 (12)

As usual, the molecule fixed components M_a of the dipole moment have been expanded in terms of the dimensionless normal coordinates q_n , q_{ta} , q_{tb} ; then, the part responsible for the ν_6 transitions can be written as

$$(M_x)_{r_6} = d_6 q_{6a} [1 + F^{A_1}(q_n, q_{ta}, q_{tb})],$$

$$(M_y)_{r_6} = d_6 q_{6b} [1 + F^{A_1}(q_n, q_{ta}, q_{tb})],$$

$$(M_z)_{r_6} = 0,$$
(13)

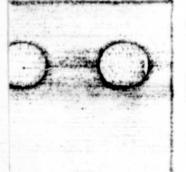
 $F^{A_1}(q_n, q_{ta}, q_{tb})$ is a totally symmetric expansion of q_n , q_{ta} , q_{tb} without a constant term. When using Eq. (12), all terms of M_Z^+ which depend on the angular momentum components are neglected. Nevertheless, the contributions of these terms to the intensities can be roughly estimated from theoretical values of the $\xi_{\sigma\sigma'\sigma'}^{\alpha}$ constants, to be generally less than a few percent of the contributions due to Eq. (12). Only the intensities of the weakest transitions, especially the forbidden transitions, with $\Delta |K - l_6| \neq 0$, may be seriously affected (up to 50%).

All transitions obey the strict selection rules (10, 11)

$$\Delta J = 0, \pm 1 \text{ and } A_1 \leftrightarrow A_2 \text{ or } E \leftrightarrow E.$$
 (14)

It should be noticed that the approximate selection rule $\Delta | K - l_6| = 0$ which holds for the allowed transitions in the zero-order approximation is not affected by the (2,2) coupling (see Eqs. (5) and (7)), which is diagonal in $K-l_6$. Thus, no additional transitions can appear when only such a coupling occurs. However, the (1,-2) coupling (see Eqs. (6) and (8)), which obeys $\Delta (K-l_6) = \pm 3$, can be responsible for ferbidden transitions. Of course, A_1 , splittings of energy levels with |K| = 3, 6, ... in the ground state might also provide a breakdown of the rule $\Delta |K-l_6| = 0$, but such splittings are insignificant for the values of J considered here (8).





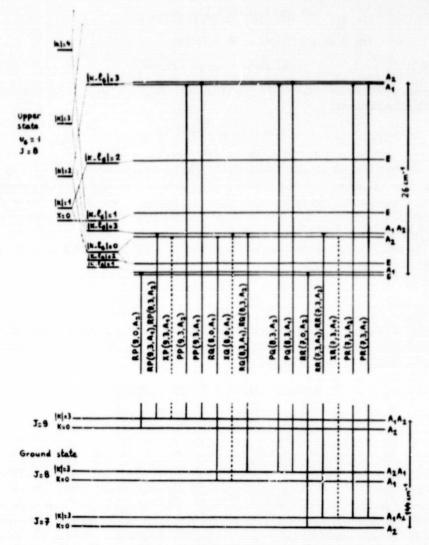
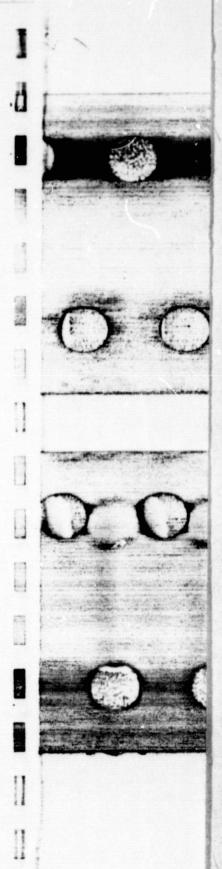


Fig. 2. The ν_6 band transitions of $^{12}CH_4D$ with J'=8 and $|K'-l'_6|=0$ and 3.

Actually, some forbidden transitions with $\Delta |K - l_6| = 3$, induced by the (1, -2) coupling, are observed in our spectra: they involve upper-state levels with small values of $|K' - l'_6|$, i.e., those which are the most strongly affected by the mixing in K and l_6 . In Fig. 2, as an example, the first few upper-state rotational levels with J' = 8 are represented along with the observed transitions having $|K' - l'_6| = 0$ or 3. The allowed transitions are represented by solid lines and are labeled PP(J'', |K''|, C''), PQ(J'', |K''|, C'')..., where the double prime indicates lower-state quantum numbers and PP, PQ, PR, ... have their usual meanings. The forbidden transitions are represented by broken lines and are labeled XP(J'', |K''|, C''), XQ(J'', |K''|, C'')



A more complete specification of the forbidden transitions is not necessary here, owing to the restricted number of observations.

VI. RESULTS AND DISCUSSION

Of the 689 lines measured in the range 1033–1270 cm⁻¹, 510 can be assigned to be totally or partially due to transitions of the ν_6 band of $^{12}\text{CH}_2\text{D}$, the other lines are due to transitions of the ν_4 band of $^{12}\text{CH}_2\text{D}$ or to transitions of $^{12}\text{CH}_4$ and H_2O impurities in the gas sample.

Assignments and observed wavenumbers of the transitions of the ν_6 band are listed in Table I, columns (II) and (IV); column (I) gives the correspondence with the lines of Fig. 1; column (III) gives the specification of the upper levels of the transitions. The transitions and upper-state energy levels are labeled as indicated in Part V of this paper, except that the pairs of transitions of the type

$$(J'', |K''|, A_1) \rightarrow (J', |K' - l'_6|, A_2)$$

and

$$(J'', |K''|, A_2) \rightarrow (J', |K' - l'_6|, A_1), \text{ with } |K' - l'_6| = 6, 9, \dots,$$
 (15)

which coincide exactly (doublets unresolved in the present work), are tabulated according to the global notation

$$(J'', |K''|, A) \rightarrow (J' |K' - l'_{6}|, A)$$
 (16)

instead of as in representation (15) above.

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Of the 680 transitions $E \leftrightarrow E$, $A_1 \leftrightarrow A_2$, or $A \leftrightarrow A$, which have been assigned to the 510 observed lines of the ν_6 band, 633 have been retained for calculation of molecular structure constants of the $\nu_6 = 1$ state. Statistical weights equal to 1.0 have been assigned to the 633 retained transitions; the weights for the transitions $A \to A$, have been distributed equally between the two components $A_1 \to A_2$ and $A_2 \to A_1$ and are indicated as *1.0 in Table I (see column VIII).

The 633 transitions were used to calculate values of molecular structure constants following the computational procedure outlined in Section III. The results of these calculations are listed in Table II. The standard deviations and 99.5% confidence intervals within the theoretical model adopted are also given. The values of the three constants H^J , H^{JJK} , and H^{JKKK} were constrained to the values listed at the bottom of Table II. The 633 values for the wavenumbers of the assigned transitions can be reproduced with an overall standard deviation of 0.013 cm⁻¹ using the stated values for the molecular structure constants, which is within the experimental uncertainty of the experimental data.

The values of the molecular structure constants listed in Table II have been used to predict the spectral positions and relative intensities for all transitions of the ν_6 band of $^{12}\text{CH}_2\text{D}$ with $J' \leq 17$. The predicted transitions include 745 allowed transitions, with 659 observed, and 12 forbidden transitions, with 10 observed, which have a calculated intensity ratio $\geq 1/250$ with respect to the most intense lines of the ν_6 band (i.e., having intensities $\geq 0.20 \times 10^3$ in our arbitrary scale). The intensity ratio $\geq 1/250$ was selected to approximately correspond to the limit for observation of lines in the most favorably recorded portions of the spectrum. Predicted wavenumbers and relative intensities are



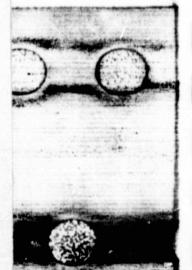
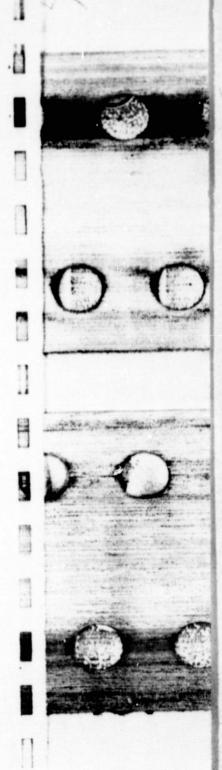


TABLE I

Observed and Calculated Wavenumbers of ν_6 of $^{19}{\rm CH_4D}$ Note: The explanations for Columns I-VIII are given at the end of the table as a footnote.

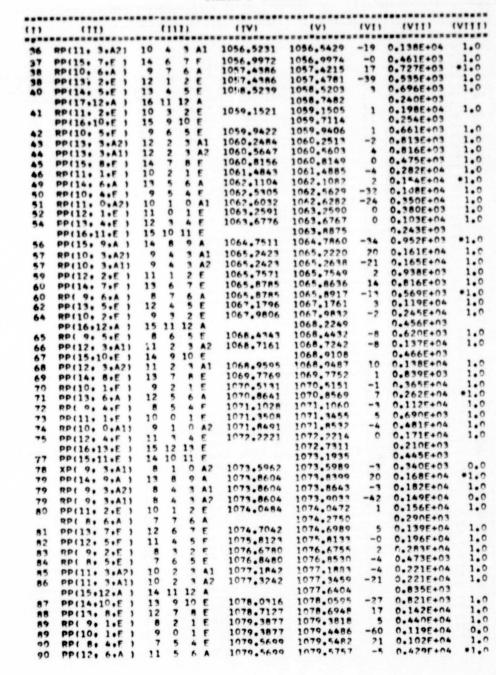
,	(11			(11	1)		(11)	(V)	(VI)	(VII)	(VIII)
•••			****	***				1001.4604		0.133E+03	
	RP(17.		16	4		Al				0.133E+03	
	RP(17.		16	4	3	A2		1001.4690		0.200E+03	
	RP(17.		16	1	0	Al		1010,9613		0.225E+03	
	PP(16.		15	4	3	Al		1010.9747		0.225E+03	
	RP(16.		15	4	3	A2		1012.8377		0.281E+03	
		2 .E)	15	3	2	E		1013.6675		0.296E+03	
	RP(15.		14	7	6	^ -		1014.7408		0.358E+03	
	RP(16.		15	1	0	A2		1015.9216		0.202E+03	
			14	6	5	E		1018.0373		0.254E+03	
	RP(15.		14		4	A2		1020.3314		0.362E+03	
	RP(15.		14	4	3	Al		1020.3572		0.361E+03	
	RF(15.		14	4				1022.3766		0.464E+03	
	RP(15.		14	3	2	E		1022.6139		0.4216+03	
	RP(14.		13	7	6	^		1023.9249		0.566E+03	
	RP(15.	1.E)	14	2	0	E A1		1024.5335		0.616F+03	
	RP(15.		14	1	5	Ê.		1024.9320		0.297F+03	
	RP(14.		13	5	4	E		1027.4271		0.289E+03	
	RP(14.		13	4	3	Al		1029.5737		0.553E+03	
	RP(14.			4	3	A2		1029.6051		0.551E+03	
	RP(14.		13	7	6	A		1031,4605		0.558E+03	
	RP(13.		12			E		1031.7830		0.726E+03	
	RP(14.		13	3	2		1033.5165	1033.5162	0	0.915E+03	1.0
1	RP(14.		13	2	5	F	1033.03103	1033.8394		0.408E+03	
_	RP(13.		12	6	0	A2	1034 3343	1034.2249	-0	0.102F+04	1.0
2	RP(14.		13	1	4		1034.2243	1035.7746		0.241E+03	
	PP(17.		16	5		F	1036.3069	1036.3015		0.557E+03	1.0
3	RP(13.		12		3	A2	1039.6987	1038.6850	13	0.799E+03	1.0
4	RP(13.		12	4	,	Al	1038.6987	1038.7175	-18	0.795E+03	1.0
4	RP(13.		12	7	6	A	1040.2039	1040.2089	-5	0.681E+03	.1.0
5	RP(12.		11	3	2		1041.0484	1041.0501	-1	0.108E+04	1.0
6	RP(13.		12	4	5	F	1041.2060	1041.1898	16	0.207F+03	1.0
	PP(16.			6	-		1042.6448	1042.6420	2	0.523E+03	1.0
	RP(12.		11	2	1	***	1042.9791	1042.9804	-1	0.141E+04	1.0
0	RP(13.		12	2	4		1043.3000	1043.3067	-6	0.246E+03	1.0
2	PP(15.	-	14	í	0		1041.8182	1043.8084	9	0.160E+04	1.0
3	RP(13.		12	2	3		1043.8182	1043.8443	-26	0.250E+03	1.0
3	PP(15.		15	5	6		1044.5584	1044.5584	-0	0.4665+03	.1.0
5	PP(16.		11	5	4		1045.1634	1045.1591	4	0.7575+03	1.0
6	PL (17.		16	8	9		1046.5907	1046.5533	27	0.271E+03	*1.7
	PP:15.		14	3	4		1046.5907	1046.5789	11	0.327E+03	1.0
,	RP(12.		11	4	3		1047.6750	1047.6640	11	0.109F+04	1.0
			11	4	3		1047.6750	1047.6957	-20	0.108E+04	1.0
?	RP(12.		15	6	,		1048.1149	1048.1094	5	0.250E+03	1.0
1	PP(16.		10	7	6		1048.8578	1048.8616	-3	0.752E+03	*1.7
3	RP(11.		13	1	2			1049.2198	-	0.290E+03	
	PP(14.		14	4	-		1049.8626	1049.8554	7	0.388E+03	1.0
•	PP(15.		11	3	-	E	1050.1790	1050.1732		0.151E+04	1.0
5	RP(12.		10	6	-		1051.3430	1051.3414	1	0.618F+03	1.0
7	PP(11.		13	2	9		1051.7913	1051.7771	14	0.459E+03	1.0
			15	7		F	1051.7913	1051.8230	-31	0.258E+03	1.0
	PP(16.		13	2	1		1052.1902	1052.1898	o	0.462E+03	1.0
9	PP(14.		11	2	,		1052.3030	1052.3075	-4	0.205E+04	1.0
0	RP(12.		11	1	ć		1053.2869	1053.2781	9	0.243E+04	1.0
1			14	5	ì		1053.3559	1053.3391	16	0.864E+03	.1.0
2	PP(15		10	5	1		1053,9116	1053.9147	-3	0.942F+03	1.0
3	RP(11.		13	3	7		1055.1563	1055.1263	29	0.594E+03	1.0
•	PP(13	4.E 1	12	0		E	1055.1563	1055.1770	-20	0.2000+03	1.0
	001114	1 .F 1	16	17		E.					
14		9.A 1	15	8		7 1		1055.6909		0.518E+03	

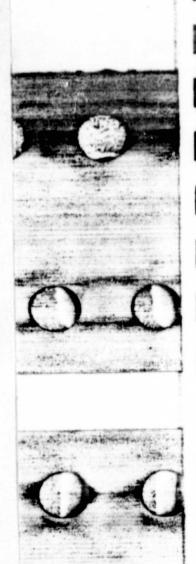


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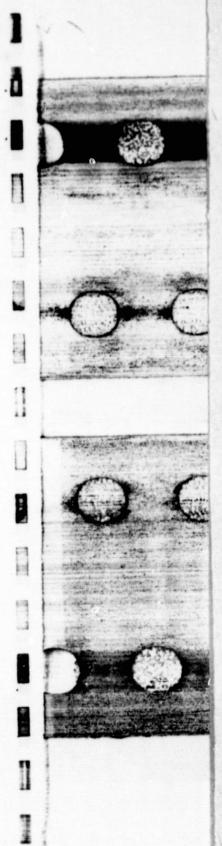




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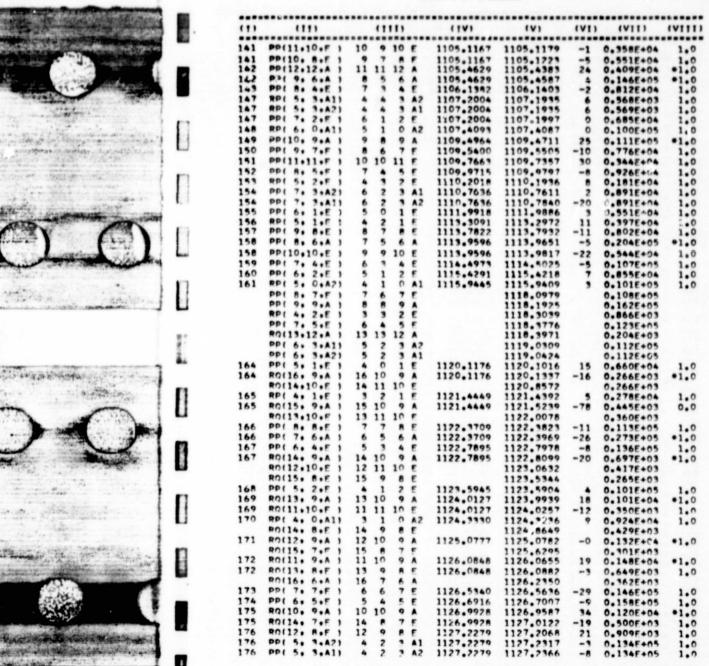
11	(11)		(1)	111		(11)	(V)	(VI)	(411)	(VIII)
	••••••					1000 7472	1000 7517		0.270F+04	1.0
71	PP(11. 4.E)	10	?	•	٠.	1080.7472	1080.7507	-6	0.630E+04	1.0
92	RP(9. 0.A2)		. 1	0	A1	1080.9422	1082.2595	-6	0.383E+03	1.0
	PP(15.13.E)	14	12	13	E		1082.2823		0.334E+03	
	PP(16.15.A)	15	14	2	Ê	1082.3541	1092.3486	5	0.246E+04	1.0
94	PP(10. 2.E)	,	1	5	A2	1082.3541	1082.3511	3	0.183E+04	1.0
94	RP(A. 3.A1)	7	4	3	Al	1082.3541	1082.3557	-1	0.180E+04	1.0
94	PP(14.11.E)	13	10	11	E		1082.4394		0.785E+03	
	PP(13. 9.A)	12				1082.8491	1082.8437	4	0.285E+04	.1.0
95	PP(17. 7.E)	11	6	7	Ē	1083.5002	1083.4942	5	0.225E+04	1.0
26	PP(11. 5.E)	10	4	5	Ē	1084.4195	1084.4222	-2	0.308E+04	1.0
00	RP(7. 5.F)	6	6	5	Ē	1085.2323	1085.1734	58	0.235F+03	0.0
00	DD(8. 2.E)	7	3	2	Ē	1085.2323	1085.2339	-1	0.302E+04	1.0
01	PP(10. 3.A1)	9	2	3	A2	1085.6266	1095.6344	-7	0.337E+04	1.0
02	PP(10. 3.A2)	9	2	3	Al	1085.7398	1095.7406	-0	0.337E+04	1.0
U.E.	PP(14+12+A)	13	11	12	A		1086.9864		0.147E+04	
	PP(15.14.E)	14	13	14			1087.0589		0.345E+03	
04	PP(13.10.F)	12	9	10	E	1087.1506	1087.1491	1	0.139E+04	1.0
25	PP(12. 8.E)	11	7		E	1087.5746	1087.5649	9	0.232E+04	1.0
05	PP(9. 1.E)	8	0	1	E	1087.5746	1087.5702	4	0.195E+04	1.0
06	PP(7. 4.F)	6	5	4	F.	1087.9027	1087.8941	8	0.758E+03	1.0
07	PP(9. 1.E)	7	2	1	E	1099.0955	1088.0872	-1	0.493E+04	1.0
08	PP(11. 6.A)	10	5	6	A	1088.2510	1088.2551	-4	0.672E+04	•1.0
09	PP(10. 4.E)	9	3	4	E	1089.2481	1089.2540	-5	0.408E+04	1.0
10	DD(8. 0.A1)	7	1	0	A 2	1080.9051	1089.9080	-3	0.781E+04	1.0
11	PP(9. 2.F)	9	1	2	E	1090.6623	1000.6494	12	0.367E+04	1.0
12	RP(7. 3.A2)	- 6	4	3	Al	1090.7203	1290.7307	-10	0.162E+04	1.0
12	PP(7. 3.A1)	6	4	3	A2	1090.7203	.090.7316	-11	0.162E+04	1.0
14	PP(13.11.E)	12	10	11	E	1091.6134	1091.6170	-3	0.133E+04	1.0
115	PP(14.13.E)	13	12	13	Ε		1091.7089		0.676E+03	
115	PP(12. 9.A)	11	. 8	,	A	1091.7990	1091.7890	9	0.465E+04	*1.0
	PP(15.15.A)	14	14	15	4		1092.0463		0.610E+03	1.0
1.1.6	PP(11. 7.E)	10		7	F	1092.2410	1092.2404	0	0.355E+04	1.0
118	PP(10. 5.E !	9	4	5		1092.9909	1092.9927	-1	0.464E+04	1.0
119	RP(7. 2.E)	6	3	2		1093.6652	1093.6671	-1	0.294E+04	1.0
120	PP(9. 3.A2)	8				1094.0927	1094.0523	30	0.490E+04	1.0
150	PP(9. 3.A1)	8				1094,0927	1094.1205	-37	0.298E+04	1.0
171	PP(A. 1.E)	7	0			1095.7018	1095.7074	56	0.370E+03	0.0
122	RP(6. 4.E)	. 5	-			1096.2050	1096.1481	33	0.228E+04	1.0
122	PP(12.10.E)	11		-		1096.2050	1096.2550	-49	0.250E+04	*1.0
122	DP(13.12.A)	12				1096.2050	1096.3769	3	0.365E+04	1.0
123	PP(11. 8.E)	10				1040.3000	1096.6149	,	0.609E+03	
	PP(14)14+E)	13		-		1004 4114	1096.6349	-1	0.508E+04	1.0
124	RP1 7. 1.E)	6				1096.6334	1096.8858	-5	0.101E+05	*1.0
125	PP(10. 6.A)	9		6		1096.8808	1097.7208	-8	0.588E+04	1.0
127	PP(9. 4.E)	8				1097.7120	1098.7301	-5	0.913E+04	1.0
128	OP(7. 0.A2)	6			-	1098.7248	1098.9376	13	0.516E+04	1.0
129	PP(A. 2.E)	5				1099.0118	1099.0096	2	0.118F+04	1.0
130	RP (6. 3.A1)	5				1090.0118	1099.0099	2	0.119E+04	1.0
130	PP(11. 9.A)	10				1100.6910	1100.6675	23	0.731E+04	*1.0
132		11				1100.6910	1100.7183	-27	0.2185+04	1.0
132	PP(12.11.E)	1 9				1100.9246	1100.9287	-4	0.535E+04	1.0
134	PP(13.13.F.)	12	- 40			1101.0577	1101.0714	-13	0.115E+04	1.0
135	PP(9. 5.E)	. 8				1101.5030	1101.5151	-12	0.669E+04	1.0
136						1101.9768	1101.9840	-7	0.2535+04	1.0
137	PP(8. 3.A1)	-			1 42	1107,4460	1102.4315	14	0.677E+04	1.0
138	PP(8. 3.A2)	,			1 1	1102.4460	1102.4726	-26	0.677E+04	1.0
138			, ;		E	1103.8537	1103.8514	2	0.4226+04	1.0
139	PP(7. 1.E)									



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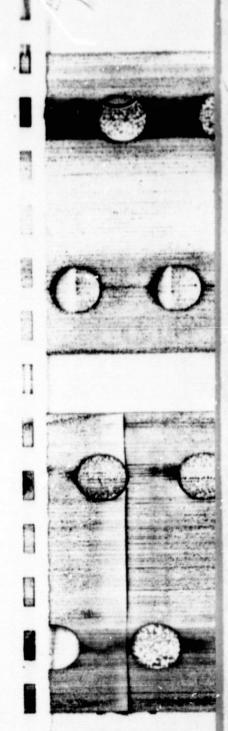
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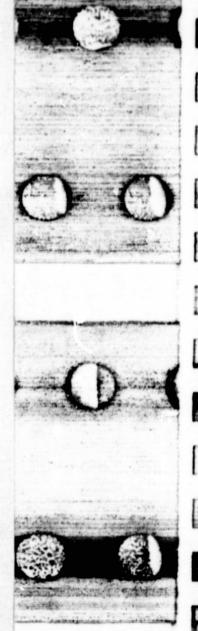
(1)	(11)		111	11		(1V)	(V)	(11)	(VII)	(VIII)
****	RO(15. 6.A)	15	7	6		•••••••••	1127.7956	•••••	0.649E+03	
	PP(4. 1.E)	3	0	1 1		1128.1901	1128.1717	18	0.718E+04	1.0
178			9			1128.1901	1128.2236	-33	0.115E+04	1.0
178	RO(11. 8.E)	11	Ä	7 1		1128.2893	1:28.2819	7	0.782E+03	1.0
	RO(13. 7.E)	10	9	8 1		1129.1506	1129.1417		0.125E+04	1.0
180			7	6		1129.2405	1129.2389	1	0.110E+04	*1.0
141	RO(14. 6.A)	14	é	7 1	-	1129.4615	1129.4411	20	0.115E+94	1.0
182	RO(12 . 7.E)	12					1129.4733	-11	0.134E+04	1.0
182	RP(3. 1.F)	2	2			1129.4615	1129.9644	12	C.978E+03	1.0
183	RQ(9. 8.E)		?	5 1		1129.9771	1130.0377		0.327E+03	0.0
183	RO(15. 5.E)	15	6			1129.9771	1130.4929	-60	0.155E+04	1.0
184	RO(11. 7.E.)	11	8 7	7.0		1130.4862		-10	0.178E+04	*1.0
185	RO(13. 6.A)	13	7	6		111 .5533	1110.5619	-	0.353E+05	*1.0
186	PP(6. 6.A)	5	2	6		1130.7255	1130.7469	-21	0 166E+05	1.0
187	PP(5. 4.F)	4	3	4 !		1131.0118	1131.0179	-	0.190E+04	1.0
188	RO110. 7.E)	10	8			1131.4417	1131.4406	1	0.575E+03	
189	RO(14. 5.E)	14	6			1131.5378	1131.5330	*		1.0
190	PP1 4. 2.F 1	. 2	7			1131.6990	1131.6934	-10	0.112E+05	1.0
101	RO(12. 6.A)	12			4	1131.7613	1131.7719	-10	0.269E+04	•1.0
	RO(15. 4.E)	15	5		-		1132.0865		0.284E+03	
193	RO(9. 7.F)	9	8		Ε.	1132.2873	1172.2881	-0	0.199E+04	1.0
	RO(16. 3.A2)	16	4		A 1		1132.3287		0.125E+03	
	RO(16. 3.A1)	16	4		42		1132.3372		0.126E+03	
194	RP(3. 0.A2)	2	1		^1	1132.5649	1132.5539	11	0.722E+04	1.0
195	RO(11. 6.A)	11	7	6	A	1132.8757	1132.8657	10	0.382E+04	•1.0
195	RO(13. 5.E.)	13	6		E	1132.8757	1132.9194	-43	0.951E+03	0.0
196	ROI R. 7.E 1	8	B	7	F	1133.0449	1133.0388	6	0.151E+04	1.0
	RO(14. 4.F)	14	5	4	F.		1133.6814		0.460E+03	
197	PO(10. 6.A)	10	7	15		1133.8485	1133.8492	-0	0.500E+04	•1.0
198	R0112. 5.E 1	12	6	5	E	1134.1924	1134.1857	6	0.149E+04	1.0
198	RO(15. 3.A2)	15	4	3	Al	1134.1924	1134.2399	-47	0.245E+03	0.0
198	RO(15. 3.A1)	15	4	3	12	1134.1924	1134.2533	-60	0.247E+03	0.0
199	RO(9. 6.A)	9	7	6	٨	1134.7229	1134.7263	-3	0.592E+04	.1.0
200	PP(5. 5.E)	4	4	5	E	1134.9235	1134.9416	-18	0.196E+05	1.0
202	RO(11. 5.E)	11	6	5	F	1135.3516	1135.3313	20	0.219E+04	1.0
202	PP1 4. 3.411	3	2	3	12	1135.3516	1135.3550	-3	0.154E+05	1.0
202	PP(4. 3.A2)	3	2	3	A1	1135.3516	1135.3566	-5	0.154E+05	1.0
202	PO(13. 4.F)	13	5	4	F		1135.4452		0.766E+03	
203	RO(8. 6.A)	8	7		٨	1135.4851	1135.5013	-16	0.602E+04	*1.0
204	PO(14. 3.A1)	14	4		A2	1136.0154	1136.0009	14	0.457E+03	1.0
204	RO(14. 3.42)	14	4	3	AI	1136.0154	1136.0267	-11	0.464E+03	1.0
205	ROI 7. 6.4 1	7	7		٨	1136.1837	1136.1784	5	0.444E+04	*1.0
205	PP(3. 1.E)	2	0		F	1136.1837	1136.1814	2	0.704E+04	1.0
206	RO(10. 5.E)	10	6		F	1136.3487	1136.3589	-10	0.300E+04	1.0
207	RO(12. 4.E)	12	5		F	1136.6780	1136.6765	1	0.151E+04	1.0
208	RO(9. 5.E)	9	6		E	1177.2706	1137.2727	-2	0.381E+04	1.0
209	RO(13. 3.A2)	13	4		Al	1137.6715	1137.6157	15	0.809E+03	1.0
209	RO(13. 3.A1)	13	4		A2	1137.6315	1137.6471	-15	0.828E+03	1.0
	RO(11. 4.E)	11	5		F	1137.8728	1277.8750	-2	0.232".04	1.0
210			3		F	1138.0714	1118.0642	7	0. 3 1E+03	1.0
211	RO(14. 2.E)	14	6			1138.0714	1138.0776	-6	J.437E+04	1.0
211		7	6		E	1138.7793	1138.7784	-0	0.431E+04	1.0
213	RO(7. 5.E)	10	5		F	1138.9575	1138.9567	0	0.332E+04	1.0
214	RO(10 . 4.F)		4			1139.1355	1139.0822	53	0.136E+04	0.0
215	RO(12. 3.A1)	12			A2					
215	RO(12. 3.A2)	12	4		A1	1139.1355	1139.1148	20	0.140E+04	1.0
215	PP(4. 4.E)	?	3		E	1139.1355	1139.1552	-19	0.196E+05	1.0
216	ROI 6. 5.E 1	6	6		E	1139.3687	1119.3800	-11		
217	PP(1. 2.F)	. 2	1		Ē	1139.7247	1139.7205	•	0.118E+05	1.0
	RO(13. 2.E)	13	3		E		1139.8420		0.584E+03	
218	ROL 9. 4.F. 1	. 9	•		Ε.	1139,9074	1139.9173	-9	0.443E+04	1.0
219	RO(11. 3.A2)	11	4	,	A1	1140.4136	1140.4005	13	0.215E+04	1.0



P. BAND OF "CH,D

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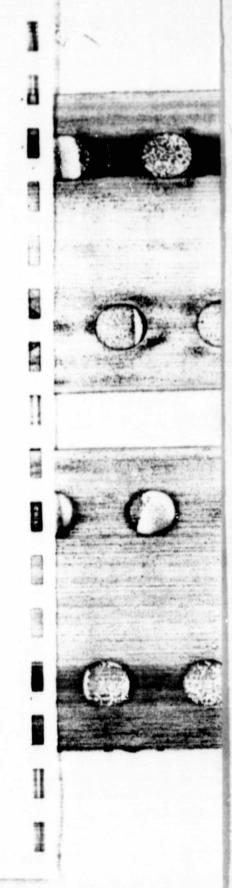
(1)		11		11	111		(11)	(V)	IVI	(VII)	IVIII
•••	••••••		****	••••				•••••			
219		3.A11	11			A2	1140.4136	1140,4777	- 14	0.222E+04	1.0
220	RP1 2.		1	1		A2	1140.6345	1140.6293	5	0.407E+04	1.0
221	ROI R.	4.5 1	. 8		-	-	1140.7552	1140.7605	-5	0.546E+04	1.0
222	R0(12+	2 .F 1	12	?			1141.4805	1141.4632	17	0.104F+04	1.0
222	RO1 7.		. 7				1141.4805	1141.4916	-11	0.608E+04	1.0
223	R0(10.		10	4		A2	1141.5830	1141.5721	10	0.317E+04	1.0
223	R0(13.	1 .F 1	13	?		Ε.	1141.5830	1141.5854	-2	0.223E+03	1.0
224	PO(10.		10	*		A1	1141.5830	1141.6039	-20	0.337E+04	1.0
225	ROI 6.	3.A21	6	5		E A1	1142.1194	1142.1165	-3	0.583E+04	1.0
225	ROI 9.	3.A11	9			A2	1147.6143	1142.5938	20	0.417E+04	1.0
225	ROI 5.			*		F	1142.6143	1142.6356	-71	0.466E+04	1.0
226	R0(11.	2.E)	11	,		F	1147.6143	1142.6407	-26	0.4075+04	1.0
	X21 9.	3.A21	9		-	A1	1142.9184	1142.9245	-6	0.176E+04	1.0
	XOL B.	3.41)	8	1		A2		1143.1798		0.500E+03	
228	PP(3.	3.421	2	2		Al	1143 1053			0.565E+03	
228	PP1 3.	3 . A11	2	2		A2	1143.3953	1143,3939	1	0.1725+05	1.0
228	R0(12.	1 .F)	12	2		E	1143.3953	1143.4030	-7	0.172E+05	1.0
229	ROL R.	3.421		4		Al	1143.5554	1143.5345	20	0.60RE+04	1.0
229	ROI B.		8	4		A2	1143,5554	1143,5735	-18	0.553E+04	1.0
230	PP(2.	1.5 1	1	0			1144.1764	1144,1166	9	0.6095+04	
231	R0(10.		10	•		F	1144.2196	1144.2251	-6	0.281E+04	1.0
232	RO1 7.	3.411	7	4		4.4	1144.3:59	1144.3085	-1	0.731E+04	1.0
232	ROI 7.	3.421	7	4		41	1144.3069	1144.3132	-6	0.7325+04	1.0
233	ROI 6.	3.421	6	4		Al	1144.9613	1144.9654	-4	0.793E+04	1.0
233	PO! 6.	3.A11	6	4			1144.9613	1144.9663	-4	0.794E+04	1.0
234	R0(11.	1 .F 1	11	2	- 5	F	1145.0791	1145.0677	11	0.809F+03	1.0
295	RO1 9.	2 .F 1	9	3			1145.3666	1145.3674	-0	0.420F+04	1.0
236	RO1 5.	3 . A11	5	4		42	1145.5140	1145.5127	1	0.741E+04	1.0
236	RO(5.	3.471		4	3	Al	1145.5140	1145.5129	i	0.741F+04	1.0
237	RO! 4.	3 . 421	4	4		41	1145.9561	1145.9576	-1	0.507E+04	1.0
237	ROI 4.	3 . A1)	4	4	3 /	42	1145.9561	1145.9576	-1	0.507F+04	1.0
238	ROI B.	2 .F)	8	3		F	1146.3544	1146.3560	-2	0.586E+04	1.0
239	R0(10.	1 .F 1	10	2		F	1146.5749	1146.5712	3	0.144E+04	1.0
240	RO! 7.	2 .F 1	7	3	2 1	F	1147.1989	1147.2013	-2	0.756F+04	1.0
241	PP(2.	2 .F 1	1	1	2 1	-	1147.6644	1147.6634	1	0.120F+05	1.0
242	BUI 6.	1 .F 1	9	2	1 1		1147.9051	1147.9067	-1	0.244F+04	1.0
242	RO(6.	2 .F)	6	3	2 1		1147.9051	1147.9105	-5	0.896F+04	1.0
244	RO(5.	2 .F. 1	5	3	2 1		1148.4903	1148.4946	-4	0.955E+04	1.0
45	RO! 4.	2 . F 1	4	3	2 1		1148.9677	1148.9640	3	0.876E+04	1.0
246	ROI R.	1.F)		2	1 1		1149.0665	1149.0699	-3	0.386E+04	1.0
47	RO(3.	2 .F 1	7	3	2 1		1149.3300	1149.3277	2	0.592E+04	1.0
49	ROL 7.	1 .E)		2	1 1		1150.0611	1150.0606	0	0.569E+04	1.0
1	BO! 6.	1 .F)	6	?	1 1		1150.8815	1150.8835	-2	0.767E+04	1.0
252	ROI 5.	1 .F 1	5	2	1 1		1151.5480	1151.5490	-1	0.935E+04	1.0
53	PP(1.	1.6)	0	0	1 6		1151.9797	1151.9670	12	0.441E+04	1.0
54	RO! 4.	1.F)	4	2	1 5		1152.0737	1152.0713	2	0.1015+05	1.0
35	BO1 3.	1 .F !	3	2	1 5		1152.4711	1152,4660	5	0.0335+04	1.0
*56	RO1 2.	1 .F 1	2	2	1 .		1152.7538	1152.7479	5	0.639E+04	1.0
57	R0(17.	0.A71	17	1		1	1153.1300	1153.1152	14	0.496E+03	1.0
258	RO(16.	O.A1)	16	1		12	1153.4663	1153.4927	-16	0.886E+03	1.0
150	PO(17.	1 . E)	17	0	1 5		1153.6257	1153.6255	0	0.475E+03	1.0
260	X0(10.		10	4		.2	1153.8202	1153.7901	*0	0.243E+03	0.0
260	R0(15.		15	1		11	1153.8202	1153.8259		0.152E+04	1.0
261	X0(14.		14	5	4 5		1153.8687	1153.8721	-3	0.249E+03	0.0
262	P0(16.		16	0	1 5		1154.0710	1154.0633	7	0.841E+03	1.0
263	RO(14.		14	1	-	.2	1154.1372	1164.1455	-8	0.251E+04	1.0
264	RO(13.		13	1		1	1154.4428	1154.4470	0	C.395E+04	1.0
165	PO(15.		15	0	1 5		1154.5140	1154.4943	19	0.141F+64	1.0
	R0(12.	UAATI	12	1	0	12	1154.716R	1154.7150	0	0.597F+04	1.0



PINKLEY ET AL.

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(1)	(11)		"	11	(14)	(V)	(11)	(111)	(1111)
****		9		3 A1	1154.8341	1154.8341		0.106E+04	0.0
267	XOL 9. 0.421		7		1154.9608	1154.9447	16	0.2135+04	1.0
268	PO(14. 1.F)	14	0	1 F	1154.9608	1154.9681	-7	0.863E+04	1.0
268	RO(11. 0.A2)	11	0	1 6	1155.0730	1155.0774	-4	0.276E+04	1,0
269	PO(13. 1.E.)	10	1	0 42	1155.2043	1155.2002	4	0.118E+05	1.0
270	RO(10. 0.A1)		-	2 F	1155.3359	1155.3214	14	0.662E+03	1.0
271	PO(16. 2.E)	16	1	2 €	1155.3359	1155.3299	6	0.293E+03	1.0
271	PO(17. 2.E)	17	i	0 41	1155.4323	1155,4208	11	0.149E+05	1.0
272	ROL 8. 0.A11		î	0 42	1155.5236	1155.5286	-5	0.174E+05	1.0
274	PO(12. 1.F.)	12	ô	1 €	1155.5896	1155.5996	-10	0.539E+04	1.0
275	X0(13. 1.E)	13	5	4 E	1155.7269	1155.6814	45	0.942E+03	0.0
275	ROL 7. 0.421	7	1	0 A1	1155.7269	1195.7296	-2	0.239E+05	1.0
276	X2(8. 0.A1)		â	3 A2	1155.8749	1155.8322	43	0.262E+04	0.0
276	ROL 6. 0.A11	6	1	0 A2	1155.8749	1155.8783	-3	0.271E+05	1.0
277	RO(5. 0.A2)	5	î	0 A1	1156.0149	1158.0025	12	0.288E+05	1.0
277	PO(11. 1.E)	11	ō	1 €	1156.0149	1156.0191	-4	0.775E+04	1.0
277	PO(15. 2.E)	15	1	2 €	1156.0149	1156.0360	-21	0.127E+04	1.0
278	ROL 4. 0.611	4	1	0 42	1156.1270	1156,1049	22	0.2855+05	1.0
279	RO(3. 0.42)	3	ī	0 A1	1156.1915	1156.1864	5	0.258E+05	1.0
280	RO1 2. 0.41)	2	1	0 A2	1156.2602	1156.2473	12	0.206E+05	1.0
280	RO(1. 0.42)	1	1	0 A1	1156.2602	1156.2879	-27	0.1346+05	1.0
	PO(17. 3.A1)	17	2	3 A2		1156,3637		0.3936+03	
281	PO(10. 1.F)	10	0	1 E	1156.4204	1156.4282	-7	0.104E+05	1.0
282	PO(14. 2.E.)	14	1	2 E	1156.6554	1156.6653	-9	0.208F+04	1.0
283	PO(9. 1.E)	9	0	1 5	1156.8410	1156.8402	0	0.132E+05	1.0
284	PO116. 3.421	16	2	3 A1	1157.2618	1157.2532		0.688E+03	1.0
284	PO(8. 1.E)	8	0	1 €	1157.2618	1157.2583	3	0.158E+05	1.0
284	PO(13. 2.E)	13	1	2 8	1157.2618	1157.2788	-17	0.12 E+04	1.0
285	PO(17. 3.A2)	17	2	3 A1	1157.4050	1157.4072	-2	0.381E+03	1.0
286	PO(7. 1.E)	7	0	1 E	1157.6870	1157.6807	6	0.177E+05	1.0
287	PO(12. 2.E)	12	1	2 E	1157.8972	1157.8913	5	0.467E+04	1.0
288	PO(6. 1.E)	6	0	1 E	1158.1056	1158.1001		0.186E+05	1.0
288	PO(16. 3.A1)	16	2	3 A2	1158.1056	1158.1077	-2	0.674E+03	1.0
298	PO(15. 3.A1)	3.5	2	3 A2	1158.1056	1158.1216	-15	0.115E+04	1.0
289	PO(5. 1.E)	5	0	1 E	1158.5046	1158.5038	0	0.183E+05	1.0
289	PO(11. 2.E)	11	1	2 €	1158.5041	1158.5061	-2	0.648E+04	1.0
290	PO(15. 3.A2)	15	2	3 A1		1158.8063		0.1135+04	
290	PO(4. 1.E)	4	0	1 E	1158.8825	1158.8758	. 6	0.168E+05	1.0
291	PO(14. 3.A2)	14	2	3 A1	1158.9629	1158.9762	-13	0.182E+04	1.0
292	PO(10. 2.E)	10	1	2 €	1159.1257	1159.1218	3	0.853E+04	1.0
293		. 3	0	1 6	1159.1994	1159.1985	0	0.142E+05	1.0
294	PO(17. 4.E)	17	3	4 E	1159.4697	1159.4227	47	0.344E+03	0.0
294		. 2	0	1 5	1159.4697	1159.4560	13	0.107E+05	
294		14	2	3 A2	1159,4697	1159.5138	-44	0.180E+04	
295		1	0	7 E	1159.6479	1159.6351	12	0.665E+04	
276		. ?	1	2 E	1159.7306	1159.7328	-2	0.107E+05	
297		13	2	3 A2	1159.8150	1159.8190	-4	0.276E+04 0.273E+04	
298		13	2	3 A1	1160.2349	1160.2318	3		
299		. 8	1	2 E	1160.3323	1160.3308	1	0.126E+05	
299		16	3	4 E	1160.3323	1160.3593	-27	0.595E+03	
300		12	2	3 41	1160.6403	1160.5486	-8 24	0.398E+04	
301		. 7	1	2 E	1150.9291	1160.9050	-28	0.395E+04	
301		12	2	3 A2	1160.9291	1160.9575		0.984E+03	
303		15	3	4 E	1161.2975	11/1.2933	:	0.149E+05	
304		. 6	1	2 E	1161.4481	1161.4431	-12	0.547E+04	
304		11	?	3 A2	1161.4481	1161.4608	-12	0.544E+04	
305		11	2	3 A1	1161.6858	1161.6853			
306			1	2 F	1161.9375	1161.9374		0.148E+05	
307		14	3	4 F.	1162.2417	1162.2230	18	0.716E+04	
	PO(10. 3.42)	10	2	3 A1	1162.2417	1162.2493	-7	7 A 1 A DE + (14	0

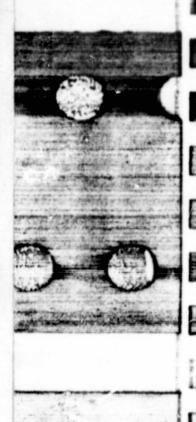


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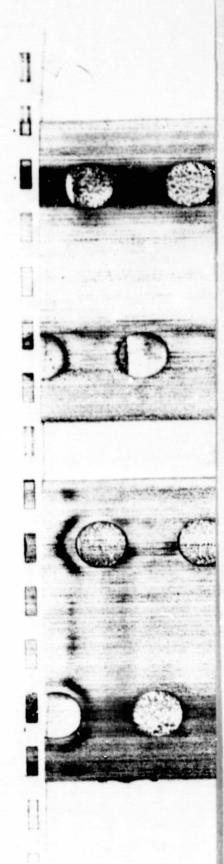
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"		11		"	11)	(11)	(4)	IVI	(VIII)	(VII
	•			••••		••••••		•••••		*****
07	P0(17.		17	•	5 E	1162.2417	1162.2535	-11	0.298E+03	1.0
08	PO! 4.		10	1	3 E	1162.3793	1162.3608	18	0.135E+05	1.0
10	PO(10.	3.A1)	3	2	2 E		1162.4069	-27	0.714E+04	1.0
11	PO(2.		2	î	2 E	1162.7175	1162.7172	.0	0.108E+05	1.0
ii		3.A11	,	ż	3 A		1162.9927	12	0.6745+04	1.0
12	PO! 9.		ő		3 4		1163.0062	-0	0.891E+04	1.0
12	PO(13.		13	3	A E	1163.1220	1163.1124	-22	0.889E+04 0.234E+04	1.5
13	PO(16.		16		5 E	1163.3256	1163.3341	-8	0.312E+03	1.0
14	PQ1 8.			2	3 A		1163,7226	7	0.1056+05	1.
15				2	3 A2		1163.7907	-13	0.105E+05	1.0
16	RR (0.		1	î	0 A2		1163.9084	17	0.931E+04	1.
17	PO(12.	4.F 1	12	•	4 F	1164.0534	1164.0517	0	0.336E+04	1.0
19	PO! 7.	3.A11	7	2	3 A2	1164.4066	1164.3889	17	0.117E+05	1.0
19	PQ(15.		15	4	5 E	1164.4066	1164.4068	-0	0.842E+03	1.0
19	PO! 7.		7	2	3 A		1164.4300	-23	0.117E+05	1.0
21	PO(11.	A.E I	11	3	4 E	1164.9964	1164.9373	59	9.461E+04	0.1
21	PO! 6.	3 . A21	6	2	3 A		1164.9957	0	0.122E+05	1.
21	PQ(6.		6	2	3 A2		1165.0187	22	0.122E+05	1.
23	PO(17.	6 .A 1	17	5	6 A	1165.3010	1165.3168	-15	0.510E+03	*1.
24	PO(14:	5.E 1	14	4	5 E		1165.4667		0.132E+f	
24	PO(5.		3	2	3 A2	1165.5366	1165.5340	2	0.116E+	1.
24	PO1 5.	3 . A21	5	2	3 A		1165.5455	-8	0.1165+05	1.
25	PG(10.	A.F.	10	3	4 F	1165.8049	1165.7927	12	0.600E+04	1.
26	PQ (4.	3 . A21	4	2	3 A1		1165.9957	5	0.973E+04	1.
26	PO! 4.	3 . A 1 1	4	2	3 A2	1166.0012	1166.0007	0	0.973E+04	1.
27	PQ(3.	3 . A11	3	2	3 A2	1166.3741	1166.3738	0	0.612F+04	1.
27	PQ(3.	3 . A21	3	2	3 A	1166.3741	1166.3754	-1	0.612E+04	1.
28	PQ(13.	5 .E 1	13	4	5 F	1166.5220	1166.5077	14	0.198E+04	1.
28	PO:16.		16	5	6 A	1166.5220	1166,5330	-11	0.870E+03	.1.
29	PQ(9.		9	3	4 E	1166.6036	1166.6085	-4	0.741E+04	1.
31	POI 8.	4.E 1		3	4 F	1167.3704	1167.3753	-4	0.862E+04	1.
32	PO(12.	5.F 1	12	4	5 E	1167.5238	1167.5224	1	0.283E+04	1.
33	PO(15.		15	5	6 A	1167.7372	1167.7330	4	0.142E+04	.1.
34	PO! 7.	4.E 1	7	3	4 E	1168.0755	1168.0838	-8	0.937E+64	1.1
35	RR(1.	1.E)	2	2	1 E	1168.2734	1168.2663	7	0.137E+05	1.
36	PO(11.	5.5 1	11	4	5 E	1168.509R	1168.5026	7	0.384E+04	1.
	P0(17.		17	6	7 E		1168.5681		0.214E+03	
37	-	4.E)	. 6	3	4 E	1168.7151	1168.7249	-9	0.931F+04	1.
30			14	5	6 A	1168.9218	1168.9104	11	0.222E+04	.1.
37	PO! 5.		. 5	3	4 F	1169.2837	1169.2904	-6	0.806E+04	1.
40	PO(10.		10	4	5 E	1169.4468	1169.4397	5	0.494E+04	1.
41	PO! 4.		. 4	3	4 E	1169.7676	1169.7732	-5	0.518E+04	1.0
	P0(16.		16	6	7 F		1169.9136		0.364E+03	
42	PO(13.		13	5	6 A	1170.0698	1170. 3580	11	0.3295+04	*1.
43	PO! 9.		9	4	5 F	1170.3242	1170.3248	-0	0.599E+04	1.
**	PO! 8.			4	5 E	1171.1640	1171.1494	14	0.677E+04	1.
44	PO(12.		12	5	6 A	1171.1640	1171.1681	-4	0.466E+04	•1.
	PO(15.	7.E 1	15	6	7 E		1171.2337		0.589E+03	
45	RRI 1.	0.A71	7	1	0 A1	1171.3567	1171.3481		0.135E+05	1.
	PQ(7.			5		1171.8919	1171.9051	-13	0.698E+04	1.0
47	PO(11.		11	?	6 A	1172.2383	1172.2324	5	0.624E+04	*1.
49	PO(6.	5.E 1	14	6	7 E	1172.5219	1172.5213	0	0.911E+03	1.
49	RR(2.	2.E 1	3	;	2 F	1172.5909	1172.5842	6	0.6255+04	1.
30	PO(5.		5	4	5 6	1172.5909	1172.6000	-9	0.206E+05	1.
91	PO(10.		10	5	6 A	1173.2458	1173.1707	-:	0.4135+04	1.
	PO(16.		-	7	RE	11.4.2448	1173.2427	3	0.787E+04	•1.0
53	PQ(13.		16	6	7 E	1177.7931	1173.4522		0.2985+03	
55	PO(9.		9	5	6 A		1173.7680	24	0.134E+04	1.0
17	LAI 41	DIN I		**	D 4	1174.1964	1174.1906		C. 923E+04	•1.





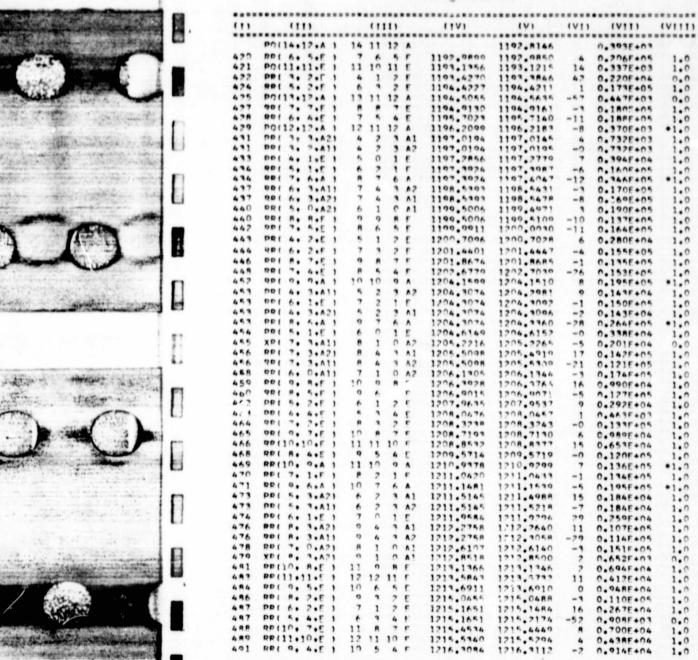
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	***********					(11)	(V)	(VI)	(111)	(VIII)
111						*********				
158	PO(15. 8.F)	15	,		E	1174.8796	1174.8892	-59	0.479E+03	0.0
150	PO(12. 7.F)	12	6		Ċ	1174.9840	1174.9687	15	0.186F+04	1.0
159	00 (1. 1.E)	2	C		F	1174.9840	1174.9744	9	0.197E+04	1.0
360	P2(A. 6.A)		5	å	A	1175.0548	1175.0684	-13	0.987E+04	*1.0
362	PO(17. 9.A)	17	8	0	Ä	1175.5188	1175.5399	-21	0.286E+03	.1.0
363	PP (2. 1.F)		2	1	F	1175.7505	1175.7405	9	0.147E+05	1.0
364	POL 7. 6.A)	,	6	ň	A	1175.8570	1175.8685	-11	0.914E+04	.1.0
365	PO(11 - 7.F)	11	6	,	E	1176.1182	1176.1125	5	0.244E+04	1.0
366	PO(14. 8.F.)	14	,		è	1176.2725	1176.2840	-11	0.731E+03	1.0
	the same of the latest the same of the sam		6	6		1176.5656	1176.5842	-18	0.6215+04	.1.0
369		6	4	4	41	1176.9803	1176.9764	3	0.249E+05	1.0
160	00(3, 3,42)	4		,		1176.9903	1176.9764	•	0.2492+05	1.0
340	not 3. 3.A1)	. 4	4	2	A.2	1177.1972	1177.1375		0.478E+03	0.0
370	PO116. 2.A 1	16	2	7	^	1177.1972	1177.1925	4	0.2986+04	1.0
370	00110. T.E)		6					21	0.106E+04	1.0
371	PQ(13. A.F.)	13	7		E	1177.6501	1177.6290	-0	0.330F+04	1.0
372	PO! 7. 7.E.	3	6	7	۴.	1178.6336	1178.2010	-6	0.168F+05	1.0
373	PP! 2. 0.A11		1	0	A.2		1178.6910	-57	0.757E+03	0.0
373	PO(15. 0.A)	15	7	9	A	1178.6336	1178.9165	6	0.144E+04	1.0
774	P7(12. A.E.)	12			ř			-19		1.0
375	D31 4 1	19	6	7		1179.1112	1179.1308		0.316F+04	
376	POL 7. 7.5 1	7	6	7	E	1179.9921	1179.9752	6	0.221E+04	1.0
376	DD(3, 2.5)	4	3	2	F.	1179.9821	1179.9878	-5	0.198F+05	1.0
377	PO(11. P.E)	11	7	9	£	1180.1522	1180.1386	13	0.182E+04	1.0
377	00(14. 0.A)	14	A	0	٨	1190.1522	1180.1928	-40	0.114F+04	•1.0
377	DQ(10. A.F.)	10	7		E	1181.3008	1181.2870	12	0.200F+04	1.0
CHE	RP(4. 4.5)	ε,	4	4	E	1181.3038	1181.3940	-2	0.263F+05	1.0
38:	PO(13. 0.A)	13	8	9	٨	1191.6247	1181.6352	-10	0.160E+04	•1.0
382	PO1 9. 8.E 1	9	7	9	E	1187.3580	1182.3572	1	0.207F+04	1.0
383	PR(2. 1.E)	3	0	1	6	1182.4813	1382.4731	8	0.3365.04	1.0
	PO(15.10.E)	15	9	10	F		1182.6374		0.290E+03	
185	PO(12. 2.A.)	12	я	9	٨	1183.0211	1183.0108	10	0.210E+04	•1.0
396	361 3. 1.E.)	4	2	1	E	1187.1014	1183.0981	3	0.157E+05	1.0
387	POL R. R.F.	8	7	я	F	1183.3333	1183.3398	-6	0.149E+04	1.0
	PO(14.10.F)	14	9	10	£		1184.2477		0.424F+03	
301	PRI 4. 3.411	- 6	4	3	A 2	1184.2873	1184.2769	10	0.224E+05	1.0
101	201 4. 3.421	5	4	3	A 1	1184.2873	1184.2770	10	0.224F+05	1.0
301	DO(11. 0.A)	11	8	0	A	1184.2873	1184.3120	-24	0.250E+04	•1.0
305	20(10. 9.A)	10	8	9	٨	1185.5379	1185.5318	6	0.257E+04	•1.0
306	301 1. 0.A21	4	1	0	A I	1185.7496	1185.7441	5	0.189E+05	1.0
306	DO(13.10.E)	13	9	10	F.	1185.7496	1185.7893	-39	0.576E+03	1.0
307	281 5. 5.E 1	6	*	5	E	1145.8514	1185.8500	-7	0.2516+05	1.0
308	PP1 2. 2.F 1	3	1	2	E	1185.9970	1185.0895	7	0.117E+04	1.0
401	DO(9. 9.A)	9	9	9	٨	1186.6619	1186.6635	-1	0.192E+04	*1.0
	PO115-11-F 1	15	10	11	F		1186.7313		0.213F+03	
403	PO112-10-F 1	12	9	10	E	1187.2654	1187.2540	10	0.711E+03	1.0
403	2P(4. 2.F)	5	3	2	F	1197.2654	1187.2050	0	0.188E+05	1.0
	PO(14.11.5)	14	10	11	E		1188.4528		0.100E+03	
407	281 5. 4.E)	6	5	4	E	1198.6120	1188.6001	2	0.225E+05	1.0
407	PQ(11.10.E)	11	9	10		1198.6120	1198.6374	-25	0.755E+03	1.0
	PQ116.12.A)	16	11	12	٨		1189.0621		0.203F+03	
411	DD(3. 1.5)	4	0	1	E	1189.9170	1189.9026	14	0.400E+04	
411	PO(10.10.F)	10	0	10	E	1189.9170	1189.9300	-13	0.583E+03	
412	PO(13.11.F)	13	17	11	•	1199.0366	1190.0964	-59	0.384E+03	
413	201 4. 1.F 1	- 5	2	1	F	1190.3431	1190.3232	19	0.167E+05	
413	201 6. 6.4 1	7	7	6	٨	1190.3631	1190.3656	-22	0.441E+05	
	PO(15.12.4)	15	11	12	٨		1190.9784		0.297E+03	
414	DR1 5. 3.421	6	4	*	A1	1171.4674	1191.4685	-1	0.198E+05	1.0
414	281 5. 3.A11	6	4	3	AZ	1191.4674	1191.4694	-1	0.19RE+05	
415	PO(12-11-F)	1.2	10	11		1101.6613	1191.6549	6	0.627E+03	1.0
412	991 4. 0.A11	5	1		A2	1192.7006	1102.7007	-0	0.196F+05	1.0



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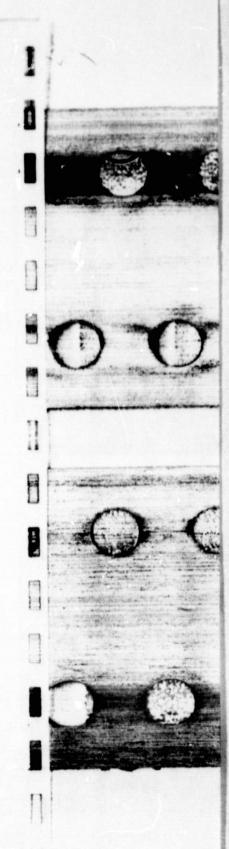
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TABLE I Continued

(1)	(11)		111	111		(1V)	(V)	(VI)	(VII)	(VIIIV)
	************						**********	*****	*********	*****
493	001 A. 1.F 1	9	2	1	F	1217.5994	1217.5948	4	0.114E+05	1.0
493	QR(11. 9.A)	12	10	9	٨	1217.5994	1217.6012	-1	0.915E+04	*1.0
494	RR(10. 6.A)	11	7	6	A	1217.8512	1217.8532	-2	0.139E+05	•1.0
405	RR(12.12.4)	13	13	12		1218.3644	1.718.3604		0.491E+04	•1.0
447	PRI 6. 3.411	7	2	3	12	1218.6394	1213.6235	15	0.193E+04	1.0
497	PP(6. 3.A2)	7	2	3	Al	1218.6394	1218.6647	-25	0.193E+04	1.0
408	RR(8. 0.A1)	9	1	0	A2	1218.9610	1218.9377	23	0.125E+05	1.0
498	90 (9. 3.41)	10	4	3	۸2	1218.9010	1218.9440	17	0.8735+04	1.0
408	PP(9. 3.A2)	10	4	3	Al	1218.9610	1218.9757	-14	0.884F+04	1.0
	PR 1 5. 5.F 1	6	4	5	F		1219.0632		0.290E+03	
499	PR(7. 1.F)	8	0	1	F	1219.2360	1219.2317	4	0.180E+04	1.0
500	OR(11. 8.F)	12	9		F	1219.7781	1219.7804	-2	0.471F+04	1.0
501	90(12.11.F)	13	12	11	F	1220.1980	1220.1769	11	0.266E+04	1.0
502	RP(10. 5.F)	11	*	5	F	1220.3453	1220.3488	-3	0.686E+04	1.0
507	RD(9. 2.F.)	10	3	2	6	1221.6104	1221.6003	1	0.874E+04	1.0
510	PR(11. 7.F)	12	. 8		F	1222.0577	1222.0593	8	0.479F+04 0.284E+04	1.0
511	B0112.10.E	13	11	10	E	1222.1194	1222.1137	-4	0.219E+04	1.0
512	PP(7. 2.E)	7	1	4	F	1222.2935	1222.3061	-12	0.116E+04	1.0
512	PP(6. 4.F)				*	1222.9127	1222.9170	-4	0.668E+04	1.0
514	QR(10. 4.E)	11	. 5	.4	E		1223.2029	-4	0.139E+04	1.0
516	89(13.13.F)	14	14	13	E	1223.1984	1223.9628	-2	0.914E+04	1.0
517	00(0. 1.F)	10	10	1	F	1224.1646	1224.1609	3	0.596E+04	*1.0
519	RP(12. 9.A)	13	10		A	1224.4254	1224.4286	-3	0.962E+04	•1.0
5,20	OP(11. 6.A.)	12		. 6		1224.8808	1224.8744	6	0.3055+65	*:.0
521	RP(13.12.A)	14	13	12	A 1	1225.1003	1225.1077	-7	0.980E+04	1.0
522	PR(10. 3.42)	11			Al	1225.4745	1225.4615	12	0.657E+04	1.0
524	RP(10. 3.A1)	11	4	,	AZ	1225.4745	1225.4933	-18	0.660E+04	1.0
524	DP(7. 3.A2)	8	,	,	Al	1225.6937	1225,6800	3	0.177E+04	1.0
526	PP(7. 3.A1)	8	2		A2	1225.7488	1225.7481	0	0.178E+04	1.0
527	PRI 6. 5.F)	7	4	- 5	F	1226.1009	1226.1117	-10	0.563E+03	1.0
528	RP(12. 8.F)	13	9		F	1226.3102	1226.3099	0	0.309E+04	1.0
529	PR(8. 1.F)	9	0	1		1226.5237	1226.5283	-4	0.115E+04	1.0
520	99(13.11.F)	14	12	11	F	1226.6908	1226.6733	7	0.166E+04	1.0
522	RR(11. 5.F)	12	6	5	E	1226.8802	1226.8751	5	0.479E+04	1.0
533	RR(10. 2.F)	11	3	2	E	1229.0032	1227.9991	4	0.665E+04	1.0
534	PP (14.14.F)	15	15	14	F	1228.0688	1228.1058	-36	0.734E+03	1.0
536	RP(12. 7.F.)	13	8	7	F	1228.5651	1228.5516	13	0.317E+04	1.0
6 76	RP(13.10.F)	14	11	10	F	1228.5651	1228.5870	-21	0.178E+04	1.0
537	DP(7. 4.F)	8	3	4	F	1229.3089	1229.3188	-9	0.121E+04	1.0
578	PD(11. 4.F)	12	- 5	4	E	1229.4056	1229.3924	13	0.464E+04	1.0
53A	DD (A. 2.F)	9	1	. ?	E	1229.4056	1229.4142	-8	0.165E+04	1.0
539	DD(14.13.F)	15	14			1229.6141	1729.6249	-10	0.826E+03	1.0
540	PR(6. 6.A)	7	5	-	^	1230.0239	1230.0557	-31	0.354E+03	*1.0
541	PR(10. 1.F)	11	?		E	1230.1425	1230.1504	-7	0.698E+04	1.0
543	99(13. 9.A)	14	10		^	1230.6298	1230.6053	24	0.376E+04	*1.0
544	QP(12. 6.A.)	13	7		^	1270.8755	1230.8751	0	0.642E+04	*1.0
546	00(10. 0.41)	11	. 1	12		1231.2348	1231.1266	-3	0.736E+04	•1.0
547	RP(14.12.4)	15	13	-				13	0.473E+04	1.0
549	RP(11. 3.A1)	12	4	-		1231.8322	1231.8188	-19	0.474E+04	1.0
549	PP(11. 3.A7)	12	2			1232.7025	1232.6764	26	0.147E+04	1.0
551	RP(13. 8.F)	14	9			1232.7025	1232.7188	-16	0.196E+04	1.0
552	PD(8. 3.A2)	9	2			1272.7653	1232.7827	-17	0.147E+04	1.0
554	90(14.11.F)	15	12			1273.0654	1233.0595	5	0.9996+03	1.0
554	PP(7. 5.F)	. 8	4			1277.0654	1733.0748	-9	0.712E+03	1.0
554	QP(15.15.4)	16	16			1233.0654	1233.0750	-9	0.738E+03	.1.0
555		13	4			1233.3071	1233.2658	41	0.322E+04	0.0
556		10	0		-	1233.8187	1233.8199	-1	0.691E+03	1.0
557		12	1			1234.2034	1234.2145	-11	0.483E+04	1.0

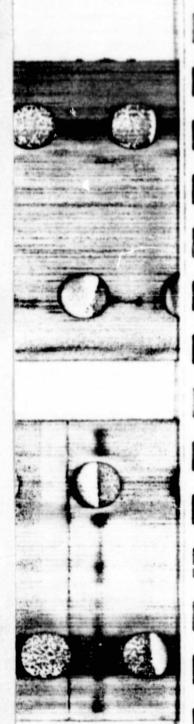


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(1)	(11)	11	11)	(17)	(٧)	(11)	(VII)	(1111)
••••	RP(15+14+E)	16 15	14 E		1234.4321		0.423E+03	******
550	RR(13. 7.E)	14 8		1234.9356	1234.9175	14	0.202E+04	1.0
559	PR(14.10.E)	15 11		1234.9356	1234.9460	-10	0.108E+04	1.0
560	XP(12. 4.E)	13 0		1235,2015	1235.2160	-15	0.840F+03	0.0
561	RR(12. 4.E)	13 5		1235.8228	1235.8203	2	0.240E+04	1.0
562	RR(15.13.E)	16 14		1235.9429	1235.9405	2	0. 477E+03	1.0
563	RR(11. 1.E)	12 2		1236.1423	1236.1632	-20	0.508F+04	1.0
564	PR(8. 4.E)	9 1		1236.2769	1236.2631	13	0.110F+04	1.0
565	PR(9. 2.F)	10 1		1236.5000	1236.5060	-5	0.115F+04	1.0
566	PR(14. 9.A)	15 10		1236.9774	1236.9306	46	0.2295+04	0.0
566	PP(7. 6.A)	8 5		1236.9774	1236.9718	•	0.6765+03	.1.0
566	RR(11. 0 A2)	12 1	0 41	1236.9774	1236.9973	-19	0.520F+04	1.0
567	PP(13. 6.A)	14 7		1237.1911	1237.1887	2	0.413F+04	*1.0
569	PR(15.12.A)	16 13	12 A	1237.584R	1237.5792	5	0.106F+04	*1.0
571	RP(12. 3.A2)	13 4	3 A1	1238.0264	1238.0130	13	0.3275+04	1.0
571	PD(12. 3.A1)	13 4		1238.0264	1238.0443	-17	J. 327E+04	1.0
573	RR(14. 8.F)	15 9		1239.0064	1239.0035	2	0.120E+04	1.0
575	RR(16.15.A)	17 16		1239.3313	1239.3003	31	0.410E+03	•1.0
575	RP(15.11.F)	16 12	11 E	1239.3313	1239.3374	-1	0.582E+03	1.0
576	RR(13. 5.F)	14 6	5 E	1239.5282	1239.5204	7	0.208F+04	1.0
577	PR(9. 3.A2)	10 2		1239.6404	1239.6211	19	0.111E+04	1.0
57B	PR (9. 3.41)	10 2		1230.7643	1239.7787	-14	0.112E+04	1.0
579	PR(A. 5.F)	9 4		1230.9701	1239.9592	10	0.728E+03	1.0
5 . 1	RP(12. 2.E)	13 3		1240.2564	1240.2551	1	0.3365+04	1.7
582	PR(16.14.F)	17 15		1240.6633	1240.4523	11	0.235E+03	1.0
584	DR(10. 1.F)	11 0		1241.1576	1241.1019		0.392E+03	0.0
584	RR(14. 7.E)	15 8		1241.1576	1241.1534	4	0.125E+04	1.0
584	RP(15.10.F)	16 11		1241.1577	1741.1875	-29	0.630E+03	1.7
586	RP(13. 4.F)	14 5		1241.7019	1241.6995	2	0.1916+04	1.0
587	RR(12 . 1 .F)	13 2		1242.0036	1242.0081	-4	0.353E+04	1.0
	RR (16 - 13 - F)	17 14			1242.1472		0.267E+03	
589	RP(12. 0.A1)	13 1		1242.7279	1242.7233	4	0.363E+04	1.0
589	XR(13. 4.F)	14		1242.7279	1242.7716	-44	0.214E+03	0.0
590	PR(15. 9.A)	16 .		1243.1416	1243.1338	?	0.1355+04	•1.0
590	PR(9. 4.F)	10 1		1247.1416	1243.1471	-5	0.900E+03	1.0
591	RR(14. 6.A)	15		1243.3736	1243.3669	-3	0.7395+03	•1.0
592	PP(10. 2.F)	11 1		1243.5776	1243.5807	23	0.595E+03	1.0
593	RR(16+12+A)	9 5		1243.7855			0.836E+03	•1.0
593	PR(R. K.A)				1243.8003	-11	0.2175+04	•1.0
594	PR(13. 3.A1)	14 4		1244.0661	1244.0687	-2	0.217F+04	1.0
597	RR(15. 8.F)	16 9		1245.1982	1245.1509	37	0.710F+03	1.0
599	RR(16.11.F)	17 12		1245.4967	1245.499?	7	0.328E+03	1.0
600		15		1245.6608	1245.6490	11	0.1285+04	
602	RR(14. 5.F)	14 3		1246 . 1220	1246.1232	-1	0.274E+04	1.0
603					1246.5218	-25	0.778F+03	1.0
604	PR(10. 3.A1)	11 7		1246.4959	1246.7463	16	0.786F+03	1.0
604	PR(9. 5.F)	10 4		1246.7625	1246.7718	-0	C.648F+03	
605	PP(15. 7.F)	16		1247.2825	1247.2570	26	0.7415+03	1.0
605	PP(16.10.F)	17 11	- 50	1247.2835	1247.1086	-25	C. 356F+03	1.0
606	RR(13. 1.F)	14 2		1247.7009	1247.6925	17	0.234F+04	1.0
606	RR(14. 4.F)	15		1247.7098	1247.7306	-20	0.130F+C4	1.0
608	RR(13. 0.A2)	14 1		1248.3120	1249.3082	1	1.230F+04	
608	PR(11. 1.E)	12		1248.3120	1248.3508	-47	0.2125+03	
612	RR(16. 9.A)	17 10		1249.1854	1249.2121	-26	0.763F+03	
613	RR(15. 6.A)	16		1249.4187	1249.4096	0	0.153F+04	
615	RR(14. 3.A2)	15 4		1249.9323	1249.9094	22	C.138F+04	
615	RP(14. 3.A1)	15 4		1249.9323	1249,9228	9	0.1385+04	
616	PR(10. 4.F)	11		1249.9893	1249.9793	2	0.675E+03	
	PR(9. 6.A)	10 .			1250.5474		0.836F+03	



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TABLE I-Continued

(1)	11	1)		111	111		(17)	(٧)	(11)	(711)	(VIII
		******					**********				
618	DB(11.		12	1	2	E	1250.6366	1250.6426	-5	0.447E+03	1.0
	pp(16.	8.E 1	17	0	A			1251.1888		0.405E+03	
620	RQ(15.	5.E 1	16	6	5	F	1251.7176	1251.7161	1	0.675E+03	1.0
621	RP (14.	2.E 1	15	3	2	E	1251.8317	1251.8232		0.143F+04	1.0
624	RR(14.	1 .E 1	15	2	1	F	1253.2264	1253.2240	2	0.1485+04	1.0
624	RR(16.	7.E 1	17	9	7	E	1253.2264	1257.2274	-0	0.424F+03	1.0
625	PR(11.	3 . A21	12	2	3	41	1253, 3701	1253.3852	-15	0.507E+03	1.0
626	PR(10.	5 .F 1	11	4	5	E	1253.5514	1253.5201	31	0.5196+03	1.0
626	RP(15.	4 .F 1	16	5	4	F	1253.1510	1257.6719	-20	0.803F+03	1.0
627	PR(11.	34611	12	2	3	12	1253.7125	1253.6941	38	0.516F+03	1.0
627	PP (14.	0.A11	15	1	0	12	1251.7125	1253.7561	-23	0.151E+04	1.0
629	pp(9.	7.F 1	10	6	7	F	1254.4603	1254.4549	4	0.236E+03	1.7
611	PP(16.	6 . A 1	17	7	4	٨	1255.3316	1255.3215	10	n. A74F+03	•1.0
634	PP (15.	3.421	16	4	3	A1	1255.6181	1:55.6073	10	0.8400+03	1.0
634	RP(15.	3.A1)	16	4	3	12	1255.6181	1255.6158	,	0.841E+03	1.0
636	pp(11.	4.E 1	12	3	4	E	1256.7610	1256.7676	-6	0.4695+03	1.0
637	RP (16.	5 .E 1	17	6	5	€	1256.9288	1256.9093	10	0.3055+03	1.0
639	PR(10.	6 . A)	11	5	4	٨	1257.2255	1257.2200	15	0.726E+03	•1.0
640	PR(15.	2 .E 1	16	3	2	5	1257.7676	1257.3412	2	C.872F+03	1.0
	PP(12.	2 .E 1	13	1	2			1257.6920		0.255E+03	
645	PR(15.	1 .F 1	16	2	1	F	1259.6376	1258.6008	27	0.903[+03	1.0
646	pp(15.	0.421	16	1	0	A !	1259.0632	1259.0716	-8	0.9165+03	1.0
	PP (16.	4 .F 1	17	5	4	e		1259.2470		0.4715+03	
651	PR(11.		12	4		•	1260.2194	1260.2117	7	0.3916+03	1.0
651		3.411	13	2	3	42	1260.2104	1260.2163	3	0.300E+03	1.7
653	pp(12.		13	2	3	A 1	1260.6446	1260.6200	15	0.318E+03	1.0
*****	pp(10.		11	6	7	F		1261.0645		0.229E+03	
655		3.41)	17	4	3	12	1261.1410	1261.1246	16	0.4925+03	1.0
655	RR(16.	3.421	17	4	3	41	1261.1410	1261.1674	-26	0.4925+03	1.0
658	PR(16.		17	3	2	F	1262.7432	1262.7446	-1	0.5115+03	1.0
661	PP(12.		13	3	4	Ē	1263,5418	1263.5195	22	0.304F+03	1.0
663	PP(11.		12	5	6	٨	1263.9345	1263.0749	9	0.567F+03	*1.0
663	PR (16.		17	2	1	F	1263.9345	1263.8574	-22	0.527F+03	1.0
665	PR(16.		17	1	2	A2	1264.2527	1264.2504	-6	0.5375+03	1.0
672	PR(12.		13	4	4	E	1266.8499	1266.9541	-4	0.259E+03	1.0
680	pp(12.		12	5	6	A	1270.3757	1270.3603	6	0.4065+03	.1.0
	PP(13.		14	5	6	^		1276.8602		0.2605+03	

(1) SERTAL NUMBER

(11) TRANSITION

(111) VALUES OF J. . K. | . K .- L. I . C. FOR UPPER LEVELS OF THE TRANSITIONS

THE ORSERVED WAVENUMBER IN CH

(V) CALCULATED WAVEMUMBER IN CM

INTO SEXPT-CALCO IN 10 CM

(VIII) CALCULATED RELATIVE INTENSITY

(VIII) STATISTICAL WEIGHT

given in the columns (V) and (VII) of Table I; column (VI) gives the deviations (expt - calc) corresponding to the observed transitions.

The 88 unassigned transitions either exc blended by lines of much larger relative intensity or occur in portions of the spectra unfavorable for the observation (a small part near 1118 cm⁻¹ could not be recorded). The transitions, assignments, upper-state

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energy levels, calculated spectral positions, and calculated relative intensities of predicted transitions which were not assigned to observed lines are included in Table I.

In the first column of Table II, the indices of the magnitude of the constants, expected from the expansion in power series of the transformed Hamiltonian H^+ (see Ref. (9), Chapt. VII) are listed. The comparison between these indices and the corresponding values determined for the constants shows that the expansion of H^+ is satisfactorily convergent, with the ratio $\langle h_{m+1}^+ \rangle / \langle h_m^+ \rangle$ between consecutive terms, being about $\frac{1}{10}$. However, this ratio is not small enough for the successive orders of magnitude to be very well separated. Thus, some neglected terms of H^+ , i.e., h_5^+ or h_6^+ , may slightly contribute to the effective values of the smallest constants appearing in Table II. In addition, the matrix element of h_2^+ of the form $\langle v_6 = 1, J, K, l_6 \rangle P^4 | v_6 = 1, J, K \pm 3, l_6 \rangle$, which could not be significantly determined, probably contributes in the same manner.

Finally, the validity of the formulation used in the present paper for analyzing ve

TABLE II

Structure Constants for the % = 1 Vibrational State of ¹²CH₄D

Index of Magnitude	Constant	Value in cm-1	Standard Deviation in cm	99.5 Confidence Interval in cm-1
0	v	1161.0975	0.0015	0.0085
2	(AC)	3.1123	0.0002	0.0012
2	B _v	3.83430	0.00003	0.00015
2	A _v	5.26896	0.00007	0.00039
6	D _v ^J	4.509 x 10 ⁻⁵	0.009 x 10-5	0.052×10^{-5}
6		1.935×10^{-4}	0.004 x 10-4	0.025×10^{-4}
6	DK.	- 1.098 x 10 ⁻⁴	0.005 x 10-4	0.029 x 10-4
6	J	1.608×10^{-3}	0.009 x 10 ⁻³	0.049×10^{-3}
6	D X X Y X X X X X X X X X X X X X X X X	- 8.20 x 10 ⁻⁴	0.12 x 10-4	0.66×10^{-4}
4	(q ₂₂) _v	-1.7452×10^{-2}	0.0015×10^{-2}	0.0085×10^{-2}
8	f ₂₂ ^{6, J}	4.31 x 10 ⁻⁶	0.06 x 10 ⁻⁶	0.35×10^{-6}
8	£642	2.99×10^{-6}	0.04 × 10-6	0.21×10^{-6}
10	$H^K = H^K$	1.89 x 10 ⁻⁷	0.08 x 10 ⁻⁷	0.45×10^{-7}
4	€(q ₁₂) _v	1.93 × 10 ⁻²	0.02 × 10 ⁻²	0.11×10^{-2}
8	€ (f ⁶ , K)	- 1.40 x 10 ⁻⁵	0.06 x 10 ⁻⁵	0.36 x 10 ⁻⁵
10	H ^J =H ^J	1.172 x 10 ^{-9(b)}		
10	HJJK=HJJK	1.200 x 10-8(b)		
10	HJKK=HJKK	- 1.042 x 10 ^{-8(b)}		

and the constants are defined in Ref. $(\underline{10})$; $\epsilon = \pm 1$ cannot be determined by the fitting.





 $^{^{}b}$ Constants constrained to their ground state values $(\underline{8})$.

can be discussed near the level crossings between the two rotational series of $v_6 = 1$ and of $v_3 = 1$. The elements of the untransformed energy matrix [A] responsible for the

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(a) a first-order term

$$\langle v_6 = 1, v_2 = 0, J, K, l_6 = \pm 1 | H | v_6 = 0, v_3 = 1, J, K \mp 1, l_6 = 0 \rangle$$

with a main part due to r^2P of H_1 , and

interactions between these two series are:

(b) a second-order term

$$\langle v_6 = 1, v_3 = 0, J, K, l_6 = \pm 1 | H | v_6 = 0, v_3 = 1, J, K \pm 2, l_6 = 0 \rangle$$

with a main part due to r^2P^2 of H_2 .

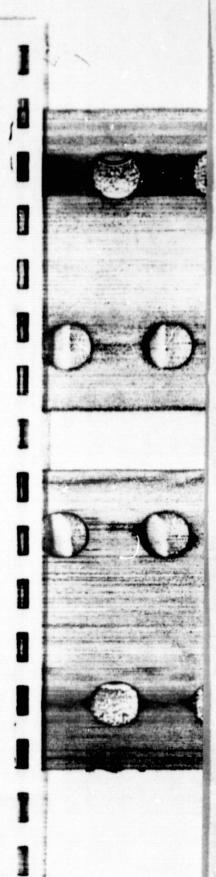
The crossing between the levels coupled by (a) occurs in the $v_6 = 1$ state for $Kl_6 \simeq -42$, i.e., very far from the observed values |K|. Such a situation explains that the first-order Coriolis interaction of $v_6 = 1$ with $v_3 = 1$ could not be clearly separated from all other vibration-rotation interactions. Therefore, the effective value of ζ_{36}^{*} determined in Ref. (3), i.e., $|\zeta_{36}^{*}| = 0.041$ is too small compared to the calculated value $(\zeta_{36}^{*})_{cal} \simeq -0.23$.

The crossing between the levels coupled by (b) lies between $Kl_6 = -12$ and $Kl_6 = -13$; the interaction, expected to be very localized, must affect mainly the PP, PQ, and PR transitions belonging to the two subbands |K| = 13 and |K| = 14. But these transitions are too weak to be observed in our spectra and no discrepancy due to this second-order interaction could be effectively detected. Let us note that the effects of this interaction should be more easily detected in the parallel band ν_3 , where the involved transitions correspond to the subbands with |K| = 10 and |K| = 11.

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Diode Laser Spectroscopy of the ν₆ Band of ¹²CH₃I PALASH P. DAS.* V. MALATHY DEVI.† AND K. NARAHARI RAO

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The vibration-rotation spectrum of the ν_6 fundamental of methyl iodide has been recorded in the 824 to 862 cm⁻¹ region by using a tunable semiconductor diode laser spectrometer. The rotational analysis performed for six Q branches ${}^{\nu}Q(J,3)$, ${}^{\nu}Q(J,4)$, ${}^{\nu}Q(J,5)$, ${}^{\nu}Q(J,6)$, ${}^{\nu}Q(J,7)$, and ${}^{\nu}Q(J,8)$ le 1 to accurate values for several molecular constants. The nuclear quadrupole splitting arising from the spin of iodine has been observed very clearly in the low J transitions and for various K values.

INTRODUCTION

For the methyl iodide molecule, as a result of extensive studies made in the past both in the microwave and the infrared regions, we now have available accurate values for the molecular constants of its ground state. However, not many microwave measurements have been made for the transitions in the ν_6 -state (1) which is located at 883 cm⁻¹. Also, in the infrared studies made with a grating spectrometer by Matsuura and Overend (2) the resolution was not sufficient to resolve the Q branches completely. The present work relates to the study of portions of the vibration-rotation fundamental ν_6 of $^{12}\text{CH}_3\text{I}$ with a tunable diode laser spectrometer having a spectral resolution of 0.002 cm^{-1} . Figure 1 displays the structure observed for one of the Q branches to illustrate the type of spectral data available in this work. Since we have observed the structure in the Q branches to J values as high as 70, several molecular constants could be determined for the ν_6 -state. The results are compared with the recent saturated absorption work of the ν_6 band by Arimondo and Glorieux (3) as well as the microwave and infrared results mentioned earlier.

EXPERIMENTAL DETAILS

The general experimental setup has been described elsewhere (4). In the present investigation a 2.54-cm-long germanium Fabry-Perot etalon was used to produce calibration fringes of about 0.049 cm⁻¹ separations. To minimize variation in fringe spacing due to temperature changes to which germanium is very susceptible, the following steps were taken: (i) The germanium etalon was enclosed in a thermally isolated box. (ii) A mode was scanned in a few minutes and several repetitive scans were obtained for each mode. (iii) Temperature of the etalon enclosure was monitored continuously and only those scans in which temperature remained steady were used in obtaining the final data. (iv) In each mode which was typically about 0.5 cm⁻¹ in extent, attempts were made to obtain at least two calibration lines so as to enable us to calculate the fringe spacings in addition to

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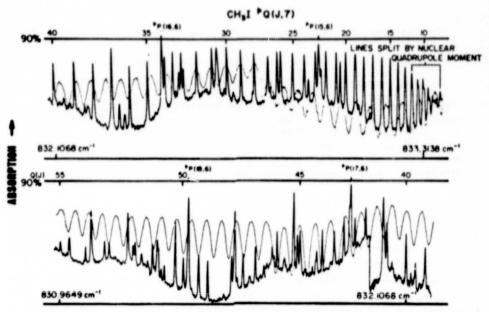


Fig. 1. ${}^{\nu}Q(J,7)$ structure in the ν_6 band of ${}^{12}CH_3I$ recorded with a diode laser spectrometer. Absorption path length = 12 m, pressure = 0.5 Torr approx.

providing absolute wavenumber standards. In this connection, the ν_1 band lines of OCS (5) were found useful.

Based on the quoted uncertainties of OCS lines, the absolute uncertainty of our measurements has been estimated to be about ± 0.002 cm⁻¹, whereas, the relative accuracy is about ± 0.0005 cm⁻¹.

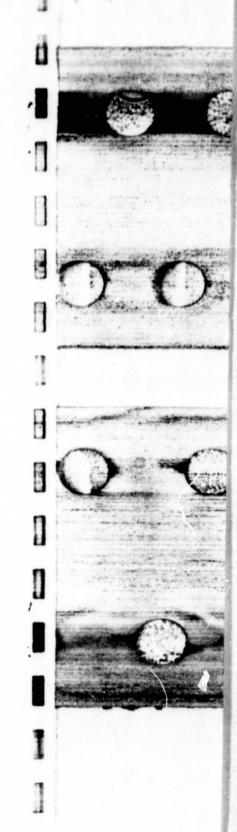
A commercial sample of CH₃I was used without further purification. A multipass cell of base length 1 m was used to provide a total pathlength of 12 m. Pressures of the gas sample ranged from 0.2 to 1 Torr. A Ge:Cu detector cooled to about 4 K was used to detect the signal from the multiple path absorption cell and a HgCdTe detector operated at 77 K was used in the etalon arm.

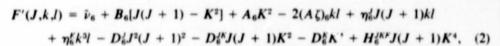
RESULTS

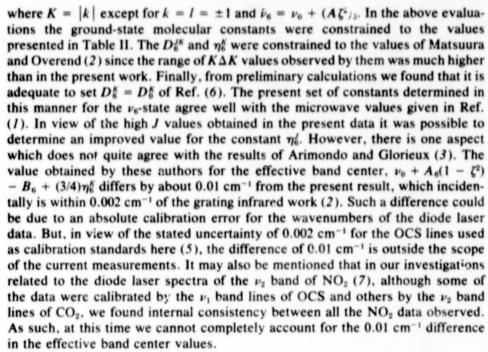
Molecular Constants

A total of 219 rotational transitions were identified in the Q branches recorded in the present investigation and their measurements are presented in Table I. These data were fitted by least-squares techniques making use of the term value expressions given respectively in Eqs. (1) and (2) below for the ground state and the ν_6 -state.

$$F''(J,K) = B_0[J(J+1) - K^2] + A_0K^2 - D_0^JJ^2(J+1)^2 - D_0^KJ(J+1)K^2 - D_0^KK^4 + H_0^{JKK}J(J+1)K^4,$$
 (1)







Along with the Q branch data for ${}^{12}CH_3I$ we also observed other transitions belonging to ${}^{p}P(J,K)$ and ${}^{p}R(J,K)$ transitions. The molecular constants of Table II were used to calculate the positions for these P and R lines and Table III compares these calculated values with the actual measured data and there is good agreement as shown by the observed – calculated values.

Nuclear Quadrupole Hyperfine Structure

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For many years now, it has been known from microwave studies (8) that CH₃I has a large quadrupole coupling constant. For v=0, the (eqQ) value is -6.45×10^{-3} cm⁻¹ (1) and for $v_6=1$ it is -6.47×10^{-3} cm⁻¹ (9). The nuclear spin (1) of 5/2 of iodine combines with J and splits the rotational levels represented in Eqs. (1) and (2) into 2I+1=6 components for J>2. The ¹²C and H nuclei have spins 0 and 1/2, respectively, and hence give no quadrupole effects. The contribution due to the quadrupole effect to Eqs. (1) and (2) is

$$\left(\frac{E_q}{hc}\right) = \frac{eqQ\left[\frac{3K^2}{J(J+1)} - 1\right][(3/4)C(C+1) - I(I+1)J(J+1)]}{2I(2I-1)(2J-1)(2J+3)},$$
 (3)

where C = F(F + 1) - I(I + 1) - J(J + 1) and F = I + J is the total angular



TABLE I Measured Q Branch Wavenumbers (vac. cm $^{-1}$) of the ν_8 Band of $^{12}{\rm CH_3I}$

×10	(0-C)	P _{Q(J,5)}	×10 ⁴	(0-C)	PQ(J,4)	(0-C)×10 ⁴	^p Q(J,3)	J
_			4		855.3243			6
13		847.9169	9		855.3147	- 11	862.7829	7
5		847.9051	2		855.3015	21	862.7737	8
3		847.8912	1		855.2870	11	862.7585	9
1		847.8760	5		855.2720	0	862.7417	10
0		847.8586	1		855.2543	- 8	862.7237	11
0		847.8397	4		855.2349	21	862.7079	12
2		847.8196	6		855.2143	- 2	862.6851	13
0		847.7973	4		855.1926	1	862.6636	14
1		847,7739	5		855.1690	- 5	862.6394	15
5		847.7481	3		855.1441	0	862.6148	16
7		847.7212	3		855.1180	- 7	862.5874	17
			6		855.0888	10	862.5610	18
4		847.6634	6		855.0590	7	862.5309	19
4		847.6320	12		855.0295	- 1	862,4986	20
1		847.5996				- 5	862,4653	21
1		847.5650				- 2	862.4310	22
3		847.5290	3		854.9249	- 6	862.3945	23
1		847,4908	4		854.8864	- 1	862.3574	24
1		847.4518				4	862.3187	25
5		847,4113	11		854.8056	8	862.2783	26
9		847.3673	5		854.7637	5	862.2356	27
1		847.3243	0		854.7203	- 4	862.1906	28
1		847.2787	3		854.6743	0	862.1455	29
2		847.2310	5		854.6269	- 5	862.0978	30
4		847.1820				0	862.0496	31
7		847.1313				- 1	661.9991	32
0		847.0801				- 3	861.9470	33
0		847.0264				4	861.8943	34
1		846.9711						35
7		846.9137						36
3		846.8558						37
4		846.7966						38
3		846.7342						39
0		846.6714						40
3		846.6070						41
0		846.5402						42

(0-C)×10	PQ(J,6)	J	(0-C)×10 ⁴	PQ(J.6)	J
1	839.5942	36	1	840.5862	8
5	839.5361	37	1	840.5721	9
0	839.4756	38	- 6	840.5556	10
- 6	839.4133	39	4	840.5394	11
1	839.3509	40	4	840.5205	12
- 4	839.2855	41	0	840.4997	13
- 5	839.2190	42	1	840.4778	14
- 2	839.1512	43	- 2	840.4538	15
2	839.0821	44	0	840.4289	16
- 1	839.0105	45	1	840.4024	17
			3	840.3743	18
3	838.2097	55	1	840.3442	19
- 2	838.1200	56	0	840.3125	20
- 1	838.0294	57	3	840.2799	21
- 4	837.9368	58	3	840.2453	22
- 2	837.8430	59	3	840.2091	23
10	837.7487	60	6	840.1716	24
- 4	837.6500	61	5	840.1322	25
0	837.5515	62	6	840.0913	26
- 1	837.4509	63	5	840.0487	27
11	837.3500	64	- 1	840.0039	28
0	837.245	65	4	839.9588	29
		66	5	839.9116	30
0	837.0324	67	1	839.8624	31
- 1	836.9235	68	1	839.8119	32
0	836.8132	69	0	839.7598	33
- 1	836.7009	70	4	839.7065	34
			0	839.6509	35

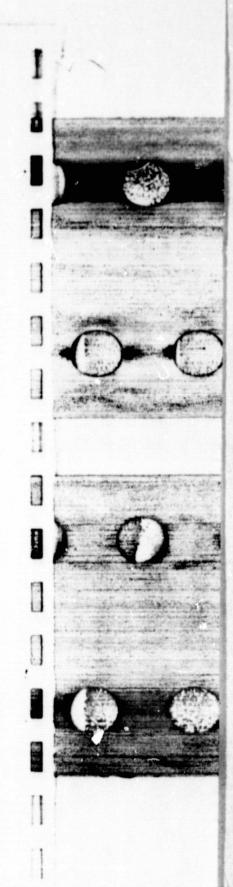


TABLE 1--Continued

(0-C)×10	PQ(J,8)	(0-C)×10 ⁴	PQ(J,7)	J
		- 9	833.3427	8
- 1	826.1654	. 3	833.3292	9
3	826.1502	0	633.3138	10
- 4	826.1321	- 5	833.2960	11
- 1	826.1135	-11	833.2765	12
6	826.0938	- 9	833.2563	13
0	826.0712	0	833.2352	14
8	826.04-4	0	833.2116	15
0	826.0224	- 1	833.1863	16
2	825.9958	2	833.1599	17
8	825.9680	0	833.1314	18
4	325.9377	2	833.1016	19
0	825.9057	- 7	833.0693	20
0	825.8726	0	833.0369	21
- 1	825.8379			22
- 6	825.8011			23
- 8	825.7631	3	832.9286	24
3	825.7248	4	832.8893	25
-14	825.6820	. 8	832.8487	26
		10	832.8063	27
1	825.5968	- 6	832.7605	28
10	825.5519	4	832.7159	29
4	825.5040	3	832.6684	30
- 1	825.4542	2	832.6194	31
0	825.4040	3	832.5690	32
3	8 25 . 35 22	6 8	832.5172	33
0	825.2982		832.4637	34
- 8	825.2420	0 2	832.4077	35
-10	825.1865	3	832.3510	36 37
10	825.1263	7	832.2926 832.2329	38
0	825.0662	4	832.1710	39
- 2	825.0055 824.9419	- 4	832.1068	40
-10	824.8762	0	832.0424	41
- 4	824.8102	4	831.9763	42
- 1	824.7424	7	831.9085	43
0	824.6727	- 7	831.8374	44
	014.0111	7	831.7675	45
		7	831.6946	46
		- 2	831.6191	47
		- 4	831.5428	48
		0	831.4654	49
		- 3	831.3857	50
		i	831.3052	51
		4	831.2229	5 2
		ï	831.1383	53
		- 8	831.0514	54
		1	830.9649	55

momentum of the molecule. The complete selection rules for the ${}^{p}Q$ branches are $\Delta v = 1$, $\Delta J = 0$, $\Delta K = -1$, and $\Delta F = 0$, ± 1 .

Figure 2 shows a comparison between the measured and calculated absorption spectra of the ${}^{\nu}Q(6,4)$ line located at 855.324 cm⁻¹. Using the selection rules given above, we find that this line splits into ten components corresponding to $\Delta F = \pm 1$ and six components corresponding to $\Delta F = 0$. The relative strengths of these lines (10) are indicated in Fig. 2a by the lengths of the vertical lines. The six $\Delta F = 0$ components account for about 85% of the line intensity. From these relative strengths and line positions determined from Eq. (3) the spectrum of the line was calculated assuming a Doppler absorption profile for each component convoluted by a Gaussian instrument function, which arises due to the finite linewidth of the laser. Under these conditions, the absorptance $A(\nu)$ can



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TABLE II

Molecular Parameters (in cm⁻¹) for the ν_6 Band of ¹²CH₃I

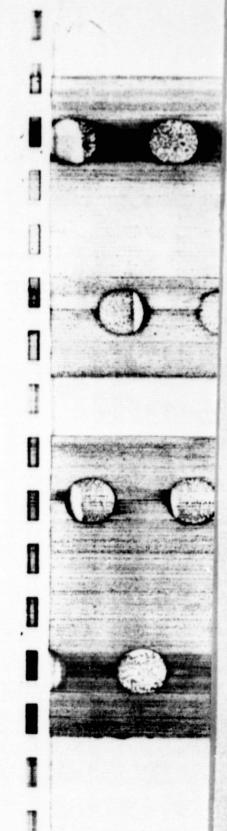
B ₀	0.2502156241 (Ref. 14)	D _J	2.103789	x 10°7 (Ref. 14)	
A _o	5.1734 (Ref. 6)	D_{o}^{JK}	3.29436	x 10 ⁻⁶ (Ref. 14)	
		D _O K	8.97	x 10 ⁻⁵ (Ref. 6)	
		H _O JKK	1,424	× 10 ⁻¹⁰ (Ref. 13)	
v ₆ State Con	stants:				
vo + (A52)6	882.91083(31)	v_6^J	$2.11970(26) \times 10^{-7}$		
B ₆	0.24943054(28)	D ₆ JK	3.29×10^{-6} (Ref. 2)		
A ₆	5.207589(16)	D6 -	D _O		
(AC) ₆	1.097131(72)	н ₆ ^{JKK} =	H _o JKK		
		76	7.227(47) × 10 ⁻⁶	
		76	1.524 x	10 ⁻⁴ (Ref. 2)	

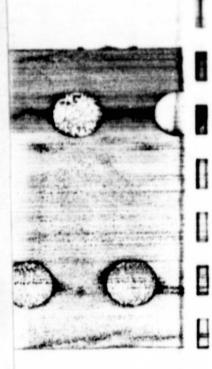
Number of lines 219; standard deviation of residuals is 0.00055 cm⁻¹. Error limits given in parentheses are standard deviations in the last digits.

TABLE III

Wavenumbers (vac. cm⁻¹) of Some Selected "P(J,K) and "R(J,K) Transitions Observed in the Q Branch Regions of CH_3I

LINE	obs.	(0-C)×10 ⁴	LINE	vobs.	(0-C)×10 ⁴
P _P (15,3)	855.1608	0	P _{P(21,5)}	837.1342	- 1
P _P (16,3)	854.6380	6	P _{P(6,6)}	837.6182	3
P _P (16,4)	847.1665	12	P _P (15,6)	832.9780	6
P _P (17,4)	846.6425	- 3	^p P(16,6)	832.4544	2
P _P (15,5)	840.2965	4	P _{P(17,6)}	831.9299	4
P _P (16,5)	839.7732	4	^p P(18,6)	831.4036	3
P _P (17,5)	839.2473	- 8	P _P (28,6)	826.0620	14
P _P (19,5)	838.1949	7	P _{R(12,6)} 846.9995		- 8
P _P (20,5)	837.6651	1	P _{R(13,6)}	847.4787	6





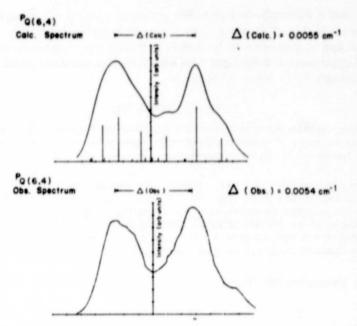


Fig. 2. Calculated and observed structure of ${}^{\nu}Q(6,4)$ line of ${}^{\nu}e$ of ${}^{12}\text{CH}_3\text{I}$ illustrating the nuclear quadrupole splitting. Absorption path length = 12 m, pressure = 0.2 Torr.

be written in a series as given below (11):

$$A(\nu) = \sum_{n=1}^{\infty} (-1)^{n+1} \frac{(S')^n}{n! n!^{1/2}} \left[\frac{1}{n} + \left(\frac{\delta}{\beta} \right)^2 \right]^{-1/2} \exp \frac{-(\nu^2)}{\beta^2 [1/n + (\delta/\beta)^2]}.$$

where

$$\beta = \frac{b_D}{(\ln 2)^{1/2}}, \qquad \delta = \frac{d}{(\ln 2)^{1/2}},$$
 (4)

 b_D = Doppler half width at half height, d = instrument half-width at half-height, and S' is related to the strength of the line.

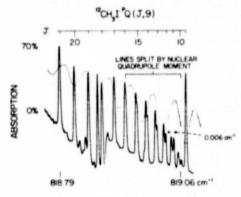


FIGURE 3

Both S' and δ were adjusted to fit the observed spectrum. The agreement, as demonstrated in Fig. 2 is good. The separation (Δ) between the two peaks of the split line is measured to be 0.0054 ± 0.0005 cm⁻¹ which agrees with the calculated separation of 0.0055 cm⁻¹. Similar splittings were observed for several low J transitions in the other pQ branches.

ACKNOWLEDGMENTS

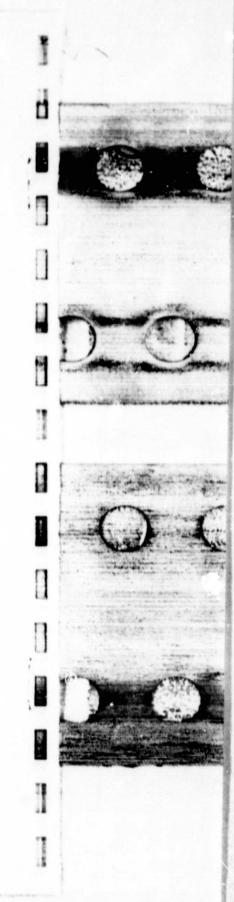
This research was done during the tenure of a grant from the National Aeronautics and Space Administration and one of us (KNR) is grateful for this support. We would also like to thank Drs. James E. G. Watson and T. Oka for reading the manuscript and making useful comments.

Note added after the above work was completed. The splittings due to nuclear spin molecular rotation interaction are more clearly seen in the spectrum of ${}^{p}Q(J,9)$ because of the large separation between the individual transitions. This is shown in Fig. 3. The observed values of ${}^{p}Q(J,9)$ for J=9 through 21 are 819.0814, 819.0648, 819.0483, 819.0300, 819.0084, 818.9871, 818.9639, 818.9391, 818.9319, 818.8833, 818.8527, 818.8214, and 818.7878 cm⁻¹, respectively. These wavenumbers agree well with those calculated from the constants given in Table II.

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Coriolis and I-Type Interactions in the ν_2 , $2\nu_2$, and ν_4 States of $^{14}NH_3$

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High-resolution infrared spectra have been remeasured for the ν_2 , $2\nu_2$, and ν_4 bands of $^{14}NH_3$ using a vacuum grating infrared spectrometer and a diode laser spectrometer. Farinfrared spectra of $^{14}NH_3$ have been measured with microwave accuracy in the 700–1100 GHz region by employing a submillimeter wave spectrometer (RAD) with acoustic detection. The pure inversion and inversion-rotation transition frequencies in the ν_2 excited state of $^{14}NH_3$ have been determined for the first time. T' .. vibration-inversion-rotation Hamiltonian of ammonia [Spirko, Stone, and Papousek, J. Mol. Spectrosc. 60, 159–178 (1976)] has been used for a precise parameterization of the energy levels of ammonia. The ground state rotational and centrifugal constants of $^{14}NH_3$ have been determined using a modified method of combination differences. Coriolis and l-type interactions between ν_2 , ν_4 , $2\nu_2$, $\nu_2 + \nu_4$, and $3\nu_2$ states have been analyzed and the band parameters have been obtained which reproduce the transition frequencies within the accuracy of the experimental data.

I. INTRODUCTION

In previous papers (1-5) a new vibration-inversion-rotation Hamiltonian for ammonia has been developed and applied to the available experimental transition frequencies in the infrared, submillimeter, and microwave regions for ¹⁴NH₃, ¹⁵NH₃, ¹⁴NH₂D, ¹⁴ND₂H, ¹⁴ND₃, and ¹⁴NT₃. The main purpose of this work is to

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obtain an accurate value of the inversion barrier in ammonia and to discuss certain anomalies in the spectra of this classic example of a nonrigid molecule (5).

In the present paper we have used this Hamiltonian for a precise parameterization of the energy levels of ammonia in the sense of obtaining the effective values of molecular parameters which reproduce the high-resolution infrared, submillimeter, and microwave data to within the accuracy of the experiments. Formulas which are obtained from this Hamiltonian to fit the experimental data are formally identical with those that would be obtained from the standard Darling-Dennison vibrational-rotational Hamiltonian. There are however two main advantages of our approach (Section II): (i) higher-order formulas can be obtained from the lower-order terms in the expansion of our Hamiltonian in terms of Q [cf. (6)]; and (ii) relations of the effective parameters to the basic molecular constants such as the molecular geometry and the potential energy function of ammonia are clearly defined in our treatment.

In the present paper we apply this treatment to the infrared data on $^{14}NH_3$ measured with the vacuum grating spectrometer at the Ohio State University in Columbus, Ohio, with the diode laser spectrometer at the Los Alamos Scientific Laboratory, and with the submillimeter wave spectrometer RAD at the Institute for Applied Physics at the Academy of Sciences USSR in Gorkii, (Section III). Although the infrared spectrum of ammonia has been studied in considerable detail (7-23), previous measurements (7-13) have been done with much lower resolution than achieved here. With the grating spectrometer, we have measured with resolution of about 0.03-0.06 cm $^{-1}$ the ν_1 , ν_2 , ν_3 , and ν_4 fundamental bands, the $2\nu_2$ and $2\nu_4$ overtone bands, and the "hot" bands for transitions from the ν_2 level to the $\nu_1 + \nu_2$, $\nu_2 + \nu_3$, ν_4 , and $2\nu_2$ levels of $^{14}NH_3$. The 10^{-4} -cm $^{-1}$ resolution of the diode laser spectrometer has made it possible to resolve certain features in the ν_2 band that remained unresolved in the grating measurements. We report here also for the first time the measured frequencies of the pure inversion and rotation—inversion transitions in the ν_2 state of $^{14}NH_3$ at 700-1100 GHz.

Combining these data with some other high- and ultra-high-resolution submillimeter and infrared data on the ν_2 (or $2\nu_2$) band of ammonia (14, 16-20, 22-24), we have determined the ground-state rotational and centrifugal distortion constants of ¹⁴NH₃ using a modified method of combination differences (Section IV). We have also analyzed in detail the Coriolis and *l*-type interactions between the ν_2 , ν_4 , $2\nu_2$, $\nu_2 + \nu_4$, and $3\nu_2$ states of ammonia (Section V). The results of analysis of the ν_1 , ν_3 , and $2\nu_4$ states including the perturbation-allowed transitions to the $2\nu_4$ level will be presented in a subsequent paper.

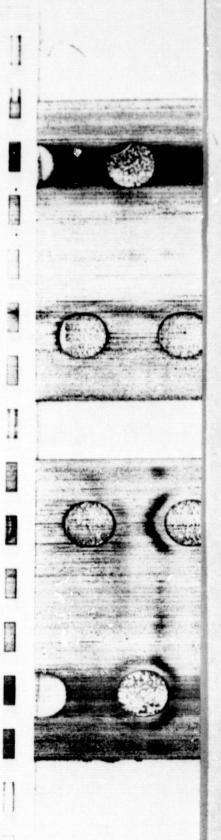
II. PARAMETERIZATION OF THE ENERGY LEVELS OF AMMONIA

If we expand the vibration-inversion-rotation Hamiltonian for NH₃ (1, 4) in the vibrational coordinates Q, and retain only terms of order of magnitude $\kappa^2 T_v$, we obtain

$$H = T_1^0 + T_r^0 + T_{\text{cent}} + T_{\text{cor}} + T_{\text{Vib}} + V, \tag{1}$$

where

$$T_{i}^{0} = (1/2)\mu_{\rho\rho}^{0}J_{\rho}^{2} + (1/2)(J_{\rho}\mu_{\rho\rho}^{0})J_{\rho} + (1/2)(\mu^{0})^{1/4}\{J_{\rho}\mu_{\rho\rho}^{0}(\mu^{0})^{-1/2}[J_{\rho}(\mu^{0})^{1/4}]\} + U_{0}(\rho), \quad (2)$$



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$$T_r^0 = (1/2)\mu_{xx}^0(J_x^2 + J_y^2) + (1/2)\mu_{zz}^0J_z^2, \tag{3}$$

$$T_{\text{Cent}} = (1/2) \sum_{\alpha,\beta=x,y,z,\rho} \left[\sum_{k} X_{k}^{\alpha\beta} Q_{k} + \sum_{k,l} Y_{kl}^{\alpha\beta} Q_{k} Q_{l} \right] J_{\alpha} J_{\beta}$$

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$$(1/2)$$
 $\sum_{\alpha = r, k \neq 0} [\sum_{k} (J_{\rho} X_{k}^{\rho \alpha}) Q_{k} + \sum_{k,l} (J_{\rho} Y_{k}^{\rho \alpha}) Q_{k} Q_{l}] J_{\alpha},$ (4)

$$T_{\text{Cor}} = -(1/2) \sum_{\alpha=x,y,z,\rho} \mu_{\alpha\alpha}^{0} (J_{\alpha} p_{\alpha} + p_{\alpha} J_{\alpha}) - (1/2) (J_{\rho} \mu_{\rho\rho}^{0}) p_{\rho}, \tag{5}$$

$$T_{\text{Vib}} = (1/2) \sum_{k} P_k^2 + (1/2) \sum_{\alpha = F, k \neq 0} \mu_{\alpha \alpha}^0 p_{\alpha}^2,$$
 (6)

$$V = V_0(\rho) + \sum_k \kappa_k(\rho)Q_k + (1/2) \sum_k \lambda_k(\rho)Q_k^2$$

$$+ \sum_{klm} k_{klm}(\rho) Q_k Q_l Q_m + \cdots . \quad (7)$$

All the symbols have the same meaning as in Refs. (1, 4, 5); k, l, m take on the values 1, 3a, 3b, 4a, 4b. It should be emphasized that all parameters occurring in Eqs. (2)-(7) are functions of the coordinate measuring the large-amplitude inversion motion.

In the theory of centrifugal distortion described in Ref. (4) we used second-order perturbation theory and evaluated the matrix elements of the vibrational and rotational operators occurring in T_r^0 , $T_{\rm Cent}$, $T_{\rm Vib}$, and V. Together with T_i^0 , this gives an effective vibration-inversion-rotation Hamiltonian for NH₃ that contains only functions of ρ and the operator $J_{\rho} = -i\hbar\partial/\partial\rho$ [Eqs. (14)-(16) in (4)]. If the resulting Schrödinger equation is solved numerically as an "inverse" eigenvalue problem, we can obtain physically reliable information on the molecular potential function of ammonia including the "true" double-minimum potential function for the inversion motion (5). This is, however, a treatment which is not suitable for fitting experimental data with an accuracy comparable to that of the high-resolution infrared and microwave data.

We have now modified the treatment so that the parameterization of the energy levels can be achieved relatively easily. We do not obtain direct information on the whole potential function of ammonia, but effective molecular parameters such as rotational constants, band origins, and interaction parameters are obtained by a least-squares fit.

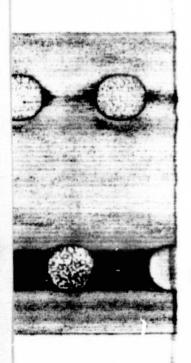
The basic idea of our treatment consists in the expansion of the ρ -dependent parameters in Eqs. (3)-(7) as a power series in the large-amplitude coordinate ρ in the point of the planar reference configuration of the atomic nuclei ($\rho = \pi/2$). The choice of the planar reference configuration in this expansion is the most natural from the point of view of the symmetry of the problem. All the ρ -dependent terms are either even or odd functions of $\bar{\theta} = \rho - \pi/2$ [A_1' or A_2'' species in the D_{3h} permutation-inversion group of NH₃ (1,5)]. Let us denote these parameters in general as $M_{\rho}^{(A_2')}$ and $M_{\rho}^{(A_2')}$. We can write

$$M_p^{(A_1')} = {}^{(0)}M_p^{(A_1')} + {}^{(1)}M_p^{(A_1')}\tilde{\theta}^2 + {}^{(2)}M_p^{(A_1')}\tilde{\theta}^4 + \cdots,$$
 (8a)

$$M_p^{(A_2^*)} = {}^{(1)}M_p^{(A_2^*)}\hat{\theta} + {}^{(2)}M_p^{(A_2^*)}\hat{\theta}^3 + \cdots,$$
 (8b)

where $^{(s)}M_p$ are ρ -independent coefficients depending only on the molecular geometry and atomic masses.





If we substitute Eq. (8) into Eqs. (3)-(7), Eq. (1) can be written in the following form:

$$H = [T_i^0 + V_0(\tilde{\theta})] + [(1/2) \sum_k P_k^2 + (1/2) \sum_k {}^{(0)} \lambda_k Q_k^2]$$

$$+ [(1/2)^{(0)} \mu_{xx}^0 (J_x^2 + J_y^2) + (1/2)^{(0)} \mu_{zz}^0 J_z^2] + H'$$

$$= H_0 + H',$$
(9)

where

$$H' = (1/2) \sum_{s=1} \left[{}^{(s)}\mu_{xx}^0 \tilde{\theta}^{2s} (J_x^2 + J_y^2) + {}^{(s)}\mu_{zz}^0 \tilde{\theta}^{2s} J_z^2 \right] + T_{\text{Cent}} + T_{\text{Cor}}$$

$$\div (1/2) \sum_{s=1}^{\infty} \mu_{\alpha\alpha}^0 p_{\alpha}^2 + \sum_{k=1}^{\infty} \sum_{s=1}^{\infty} \lambda_k \tilde{\theta}^{2s} Q_k^2 + \sum_{klm} k_{klm} Q_k Q_l Q_m.$$
 (10)

Wavefunctions ψ_0 that are solutions of the Schrödinger equation

$$H_0\psi_0 = E_0\psi_0 \tag{11}$$

can be written as the product functions

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$$\psi_0(\hat{\theta}; \theta, \Phi, \chi; Q) = [\psi_l(\hat{\theta})][S_{Jkm}(\theta, \Phi) \exp(ik\chi)][\prod \psi_v^0(Q)], \tag{12}$$

where v is summed over 1, 3a, 3b, 4a, 4b; here $\psi_i(\bar{\theta})$ are the inversion wavefunctions, $S_{Jkm}(\theta, \Phi)$ exp $(ik\chi)$ the symmetric rotor wavefunctions, and $\psi_i^0(Q)$ the harmonic oscillator wavefunctions. The inversion wavefunctions $\psi_i(\bar{\theta})$ are not obtained in the present treatment by a numerical integration of the Schrödinger equation with the operator $T_i^0 + V_0(\bar{\theta})$; they are assumed to be known.

The product functions (12) are basis functions in which the Schrödinger problem with the Hamiltonian H is solved either by standard perturbation methods or by a variational approach (if there is a close coincidence of the interacting levels). In this Section, only a simple example will be discussed to demonstrate the usefulness of this approach.

Let us consider the so-called rigid bender approximation which has been treated numerically in Refs. (1,2) to obtain information on the inversion potential function in ammonia. The rigid bender Hamiltonian H_{rb}^0 can be written in the form (1)

$$H_{\rm rb}^0 = [T_i^0 + V_0(\tilde{\theta})] + (1/2)\mu_{xx}^0(J^2 - J_z^2) + (1/2)\mu_{zz}^0J_z^2. \tag{13}$$

If we expand μ_{xx}^0 and μ_{xz}^0 in a power series according to Eq. (8a), H_{rb}^0 has diagonal and off-diagonal matrix elements in $\psi_i(\tilde{\theta})(\equiv |i\rangle)$:

$$(hc)^{-1}\langle v; J, k, m; i | H_{rb}^{0} | v; J, k, m; i \rangle$$

$$= E_{i}^{0}/hc + (h/4\pi^{2}c)[^{(0)}\mu_{xx}^{0} + \sum_{s=1}^{(s)}\mu_{xx}^{0}\langle i | \tilde{\theta}^{2s} | i \rangle][J(J+1) - k^{2}]$$

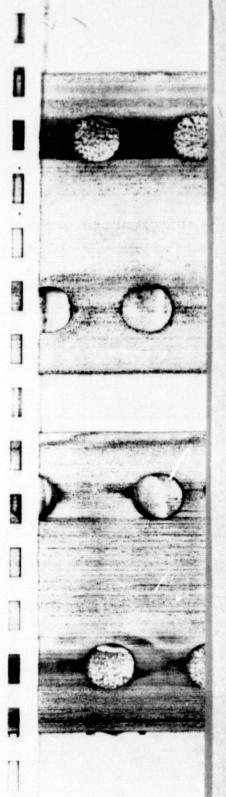
$$+ (h/4\pi^{2}c)[^{(0)}\mu_{zz}^{0} + \sum_{s=1}^{(s)}\mu_{zz}^{0}\langle i | \tilde{\theta}^{2s} | i \rangle]k^{2}$$

$$= E_{i}^{0}/hc + B_{i}J(J+1) + (C_{i} - B_{i})k^{2}, \qquad (14)$$

$$(hc)^{-1}\langle v; J, k, m; i | H_{rb}^{0} | v; J, k, m; i+2 \rangle$$

$$= (h/4\pi^{2}c)[\sum_{s=1}^{(s)}\mu_{xx}^{0}\langle i | \tilde{\theta}^{2s} | i+2 \rangle][J(J+1) - k^{2}]$$

$$+ (h/4\pi^{2}c)[\sum_{s=1}^{(s)}\mu_{zz}^{0}\langle i | \tilde{\theta}^{2s} | i+2 \rangle]k^{2}. \qquad (15)$$



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If we treat this problem by the standard perturbation theory up to third order, we obtain the well-known formula for the energy levels of a nondegenerate vibrational state and the *i*th inversion state including sex'ac centrifugal distortion coefficients:

$$\left(\frac{E_i}{hc}\right) = \left(\frac{E_i^0}{hc}\right) + B_i J(J+1) + (C_i - B_i)k^2 - D_J^{ij} J^2 (J+1)^2 - D_J^{ij} J(J+1)k^2 - D_K^{ij} k^4 + H_{JJJ}^{ij} J^3 (J+1)^3 + H_{JJK}^{ij} J^2 (J+1)^2 k^2 + H_{JKK}^{ij} J(J+1)k^4 + H_{KKK}^{ij} k^6.$$
(16)

It should be mentioned that from the mathematical point of view such a treatment is correct if the series expansions (8) are convergent. We have verified a rapid convergence of the series expansions of all parameters occurring in the Hamiltonian H in Eq. (1), which justifies our procedure.

Rotational and centrifugal distortion coefficients in Eq. (16), and in other relations which could be obtained from higher-order terms in the Hamiltonian by perturbation treatment, are effective parameters for the *i*th inversion state. Their relation to the coefficients of the expansion (8a) could be found in explicit form. The algebra involved in such a treatment is straightforward but rather lengthy and will not be carried out in this paper. Instead we concentrate on the determination of the effective molecular parameters by a least-squares fit to the experimental data (Sections IV-VI).

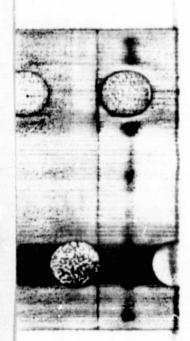
It should be noted that in our approach to the parameterization of the energy levels, NH_3 is treated as a planar molecule of D_{3h} symmetry with a large-amplitude motion. Because we use the model Hamiltonian described in Refs. (1,4,5), such an approach is physically correct and its advantage is that higher-order effects can be described in a lower-order treatment (see the above discussion of the rigid bender approximation). Previous approaches (e.g., (7,25)) considered ammonia as a molecule of C_{3v} symmetry and the vibration-rotation Hamiltonian was expanded in terms of Q in the point group of the equilibrium configuration. Strictly speaking this is not physically correct, and the relation between the effective molecular parameters and the true physical parameters is not clear in such a treatment, especially for the excited vibrational-rotational states.

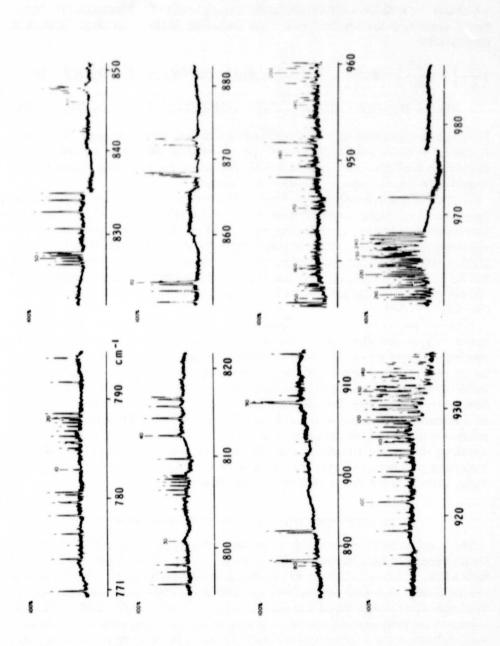
III. HIGH-RESOLUTION INFRARED MEASUREMENTS

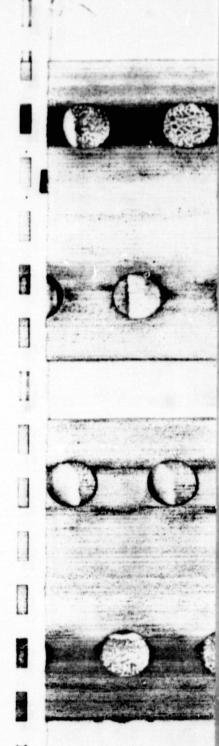
The ν_2 and ν_4 bands of ammonia in the infrared were recorded with a 3.5-m focal length vacuum grating infrared spectrometer at the Ohio State University equipped with a 9 in. \times 6 in. (22½ cm \times 15 cm) Bausch and Lomb grating with 40 grooves per millimeter and used echelle fashior. The gas sample used was supplied by Matheson and Company and was quoted to have a purity of 99.4% of ¹⁴NH₃. The source of continuous radiation was a carbon rod furnace. The measurements were made relative to the 1–0 band lines of CO and the accuracy is believed to be about 0.005 cm⁻¹. Other experimental details have been given in detail by Curtis (26). Figures 1a, 1b, and 1c display the spectra recorded, along with pressures and path lengths used for the ammonia gas.

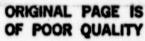
Portions of the ν_2 band were examined under Doppler-limited resolution using the diode laser spectrometer at Los Alamos Scientific Laboratory (27). This was











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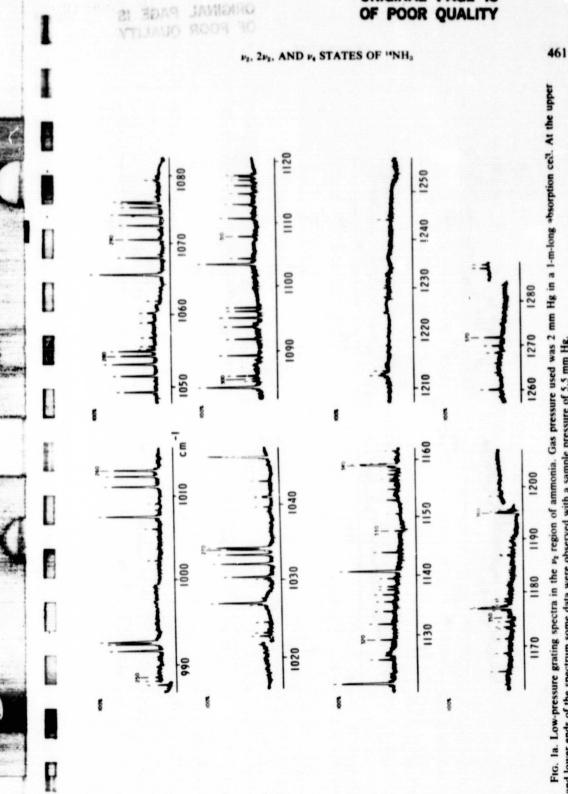
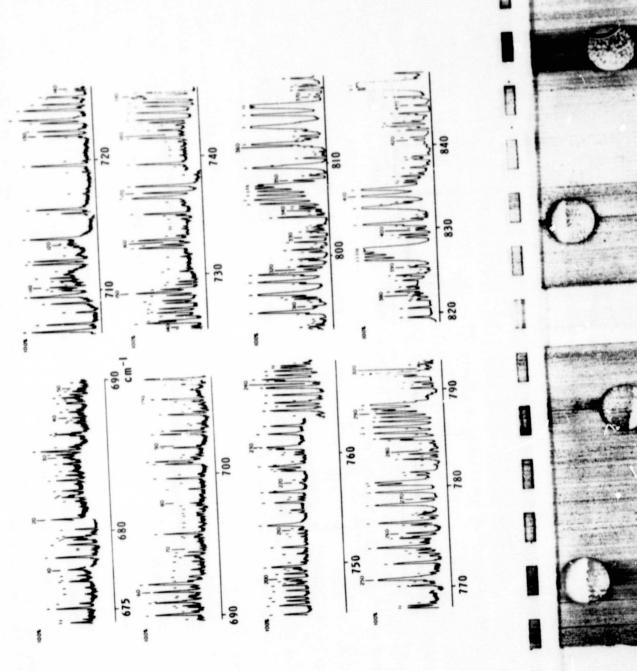
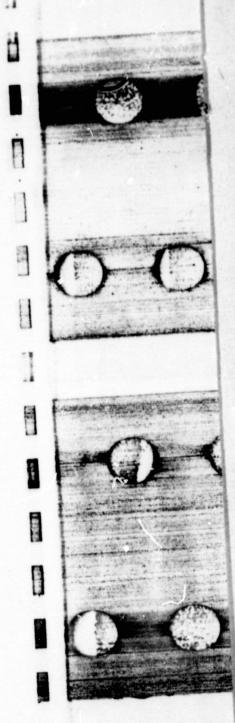


Fig. 1a. Low-pressure grating spectra in the ν_1 region of ammonia. Gas pressure used was 2 mm Hg in a 1-m-long *bsorption ce3. At the upper and lower ends of the spectrum some data were observed with a sample pressure of 5.5 mm Hg.





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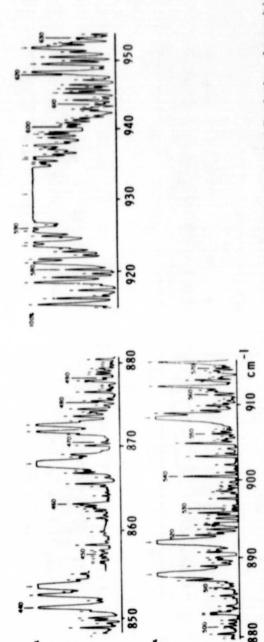
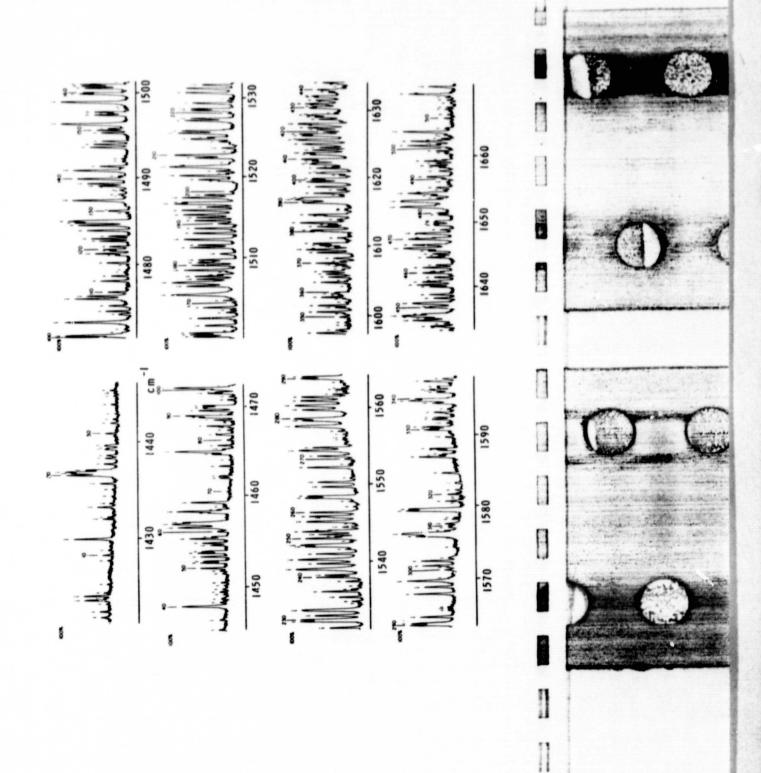
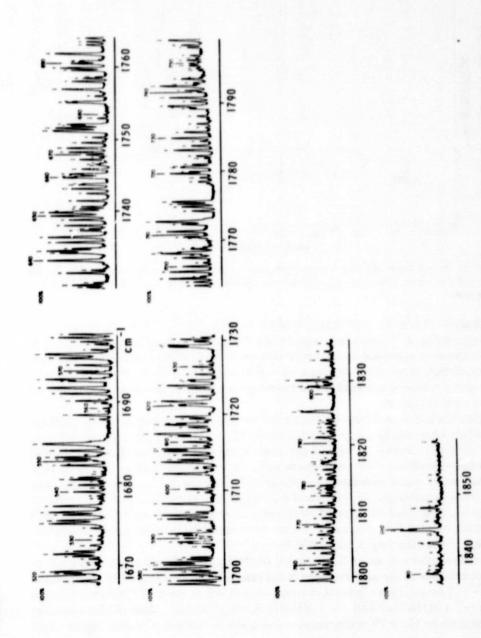


Fig. 1b. High-pressure grating spectra in the 12 region of ammonia. Gas pressure used was 1 cm Hg in an 11-m path. A line that has a dot over it is a blend of lines that are resolved in the low-pressure data.



ν₂, 2ν₂, AND ν₄ STATES OF "NH₃

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Fig. 1c. Grating spectra of the ν₄ region of ammonia. Gas pressure used was 1 cm Hg in a 1-m-long absorption cell. Line with a dot over it is an H₂O line or a blend with an H₂O line.

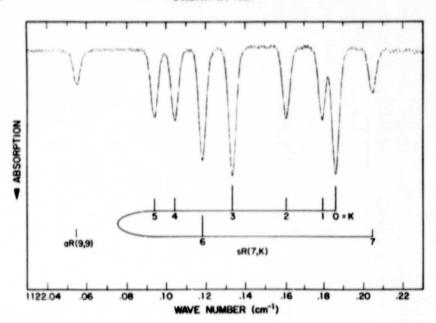


Fig. 2. The spectrum of $^{14}NH_3$ between 1122.03 and 1122.23 cm $^{-1}$ as recorded with a tunable semiconductor diode laser, showing sR(7,K) and aR(9,9). Sample pressure was ~ 0.1 Torr in a 40-cm cell.

particularly useful for resolving the sR(J,K) groupings; sR(7,K) is shown as an example in Fig. 2. Relative line separations in these spectra were measured against simultaneously recorded interference fringes from a germanium etalon. Absolute wavenumbers were then obtained by referencing the lines to one transition in each group whose wavenumber had been very accurately determined as described in the next paragraph.

Experimental transition wavenumbers for the ν_2 , $2\nu_2$, and ν_4 bands together with their assignments are given in Tables I to III. In the case of the ν_2 band, we have also used the precise values of the pure inversion transition frequencies and inversion-rotation transition frequencies in the ν_2 excited state as measured by a submillimeter wave spectrometer, RAD³ (28-30), the infrared-microwave two-photon measurements (16, 17, 31) of the vibration-inversion-rotation transitions to the ν_2 state, and the ground-state pure inversion transition frequencies (32) to obtain precise frequencies of certain vibration-inversion-rotation transitions to the ν_2 state (see Fig. 3 and Table I).

The submillimeter-wave line centers of $^{14}NH_3$ were found to depend linearly on the pressure of the ammonia gas in the range 0.2-2 Torr (e.g., the slopes of the J,K=1.1 and 2.2 pure inversion transitions in the ν_2 state of $^{14}NH_3$ were found to be -2.4 MHz/Torr and +1.5 MHz/Torr, respectively). Most of the transition frequencies in Table IV were measured at a single value of pressure (in the range

³ The particular system of frequency stabilization of the backward wave oscillator used (Belov, Gershstein, and Maslovskii, in "Proceedings of the IV All Union Symposium on Molecular Spectroscopy of High and Superhigh Resolution, Tomsk-Novosibirsk, 1978") makes it possible to measure frequencies to a very high degree of accuracy.

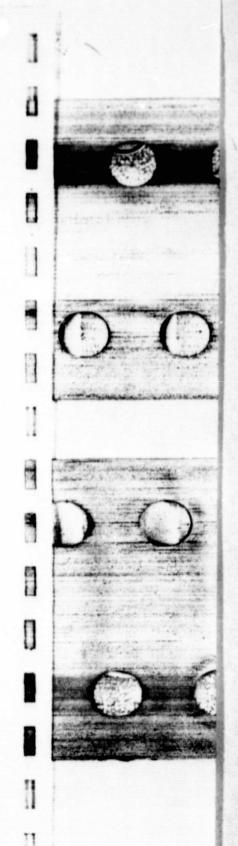
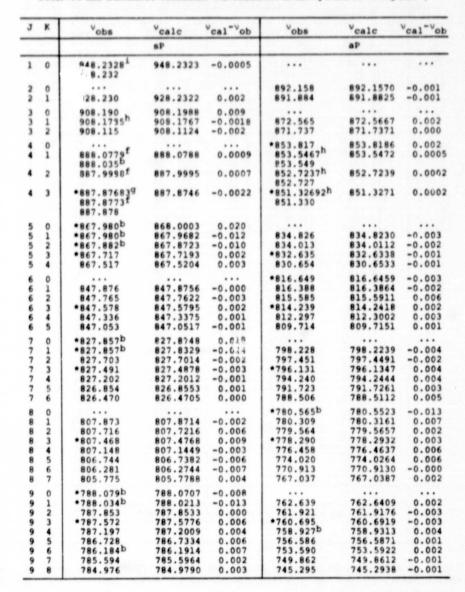
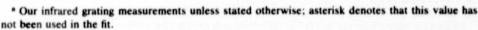


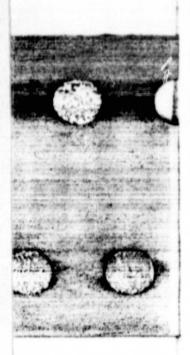
TABLE I Observed and Calculated Transition Wavenumbers in the ν_2 Band of ^{1*}NH₂ (cm⁻¹)*





b Blended line.

P







c Ref. (16).

d Ref. (14).

e Ref. (31).

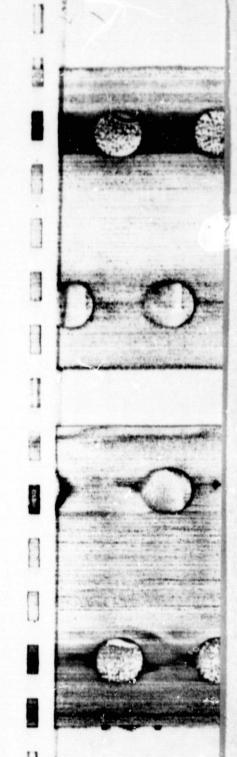
^{&#}x27; Ref. (17).

^{*} Ref. (19).

TABLE I-Continued

J K	Vobs	vcalc.	vcal vob	vobs	Veale	"cal"ot
		sP			a?	
0 0				*745.409b	745.3761	-0.033
0 1	•768.309b	768.311.	0.002	*745.194b	745.1760	-0.018
0 2	*768.107b	768.1258	0.019	744.490b	744.4809	-0.009
0 3	*767.812	767.8199	0.008	*743.312	743.3047	-0.007
0 4	767.408	767.3994	-0.009	741.622	741.6179	-C.004
0 5	766.877	766.8733	-0.004	739.384	739.3754	-0.009
0 6	766.241b	766.2545	0.013	736.513	736.5129	-0.000
0 7	765.565	765.5621	-0.003	732.949	732.9456	-0.003
0 8	764.824	764.8227	-0.001	728.580b	728.5700	-0.010
0 9	764.074	764.0734	-0.001	723.275	723.2714	-0.004
1 0	• 748.8100	748.8207	C.011			***
1 1	*748.810b	748.7689	-0.041	727.910	727.9026	-0.007
1 2	748.574b	748.5663	-0.008	727.251	727.2350	-0.016
1 3	•748.228b	748.2313	0.003	*726.109b	726.1077	-0.001
1 4	747.765	747.7687	0.004	724.517	724.4956	-0.021
1 5	747.179	747.1858	0.007	722.368	722.3585	-0.009
1 6	746.492	746.4935	0.001	719.647	719.6368	-0.010
1 7	745.703	745.7076	0.005	716.248	716.2486	0.001
1 8	744.869	744.8512	0.002	712.095	712.0894	-0.006
1 9	743.961	743.9555	-0.006	707.051	707.0372	-0.014
1 10	743.078	743.0644	-0.014	700.965	700.9650	0.000
2 0				•711.029	710.9625	-0.066
2 1	729.419	729.4196	0.001	*710.796b	710.8116	0.016
2 2	729.209	729.2006	-0.008	710.175	710.1682	-0.007
2 3	•728.822	728.8378	0.016	•709.098b	709.0851	-0.013
2 4	728.331	728.3350	0.004	*707.585b	707.5430	-0.042
2 5	727.699	727.6980	-0.001	*705.550b	705.5084	-0.042
2 6	726.943	726.9357	-0.007	702.943	702.9287	-0.014
2 7	*726.109b	726.0615	-0.047	699.727	699.7273	0.000
2 8	725.101	725.0940	-0.007	695.790	695.8018	0.012
2 9	724.052b	724.0600	0.008	691.030	691.0257	-0.004
2 10	723.007	722.9965	-0.011	685.281	685.2589	-0.022
2 11	721.978	721.9546	-0.023	678.376	678.3670	-0.009
3 0	•710.329b	710.3367	0.008			
3 1	•710.329b	710.2892	-0.040	•698.811b	693.9096	0.099
3 2	•710.069b	710.0545	-0.015	*693.233b	673.2841	0.051
3 3	*709.669b	709.6649	-0.003	*692.298b	692.2358	-0.062
3 4	*709.098b	709.1235	0.026	690.767	690.7523	-0.017
3 5	708.420	708.4349	0.015	*688.849b	688.8087	-0.040
3 6	*707.585b	707.6063	0.021	*686.391b	686.3621	-0.029
3 7 8	706.645	706.6485	0.003	*683.356b	683.3448	-0.011
	*705.550b	705.5769	0.027	679.653 •675.156b	679.6597	0.007
3 9	704.411	704.4136	0.003	-6/5.1560	675.1800	0.024
3 10	703.197	703.1896	-0.006			
3 12	701.962 700.774b	701.9477	-0.027		663.2297 655.4683	
		eQ			aQ.	
1 1	967.99774 ^C 967.995	967.9980	0.0003	931.62776h 931.6280f 931.632	931.6286	0.0008

h Value calculated from the microwave ground state inversion frequencies, the submillimeter ν_2 inversion frequencies, and the high-resolution infrared vibration-inversion-rotation transition frequencies (see text).



Ref. (20).

Value calculated from the frequency difference (resolved K structure) in Ref. (18).

^k Value calculated from the microwave ground-state inversion frequencies, the submillimeter ν_2 inversion-rotation frequencies, and the high-resolution infrared vibration-inversion-rotation transition frequencies (see text).

Our diode laser measurements.

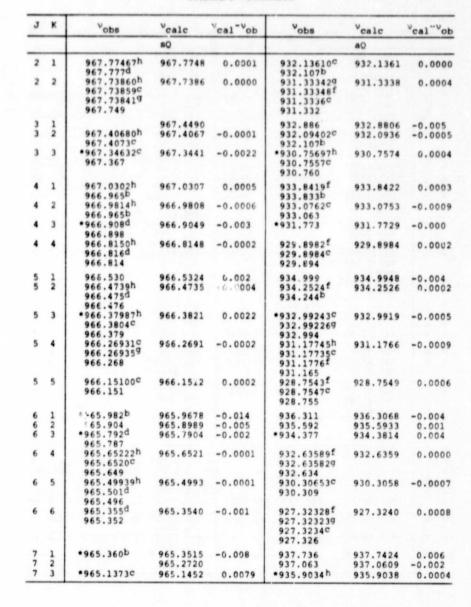
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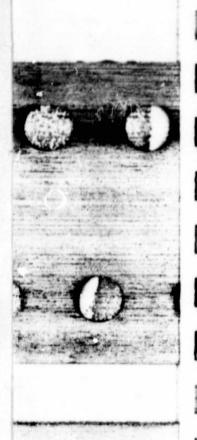
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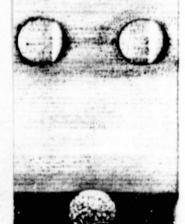
v2. 2v2. AND v4 STATES OF "NH3

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TABLE 1-Continued







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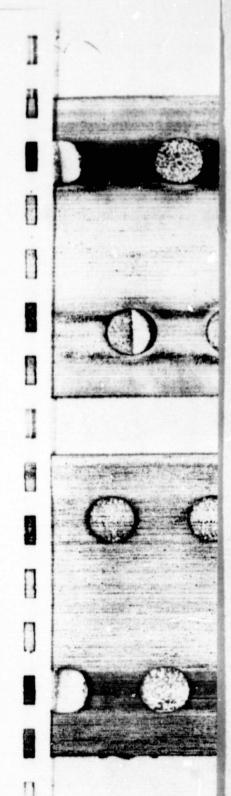
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TABLE 1-Continued

J	К	vobs	Vcalc	vcal vob	Vobs	Vcalc	Vcal -Vot
			sQ			aQ	
		965.140			935.914b		
7	4	964.9797h	964.9800	0.0003	934.2358f	934.2372	0.001
		964.982d			934.244b		
7	5	964.982 964.79006 ^C	064 7003	0.0003	033 01111h		
,	9	964.790	964.7903	0.0002	932.01111h	932.0117	0.000
7	6	964.5954C	964.5957	0.0003	933.014 929.1616f	929.1607	-0.000
		964.596b	304.3337	0.0003	929.161	929.1607	-0.000
7	7	964.42410C	964.4241	0.0000	*925.5977f	925.6016	0.003
		964.428b			925.598		
8	1		964.6981		939.256b	939.2637	0.008
8	2	*964.596b	964.6077	0.012	*938.625b	938.6164	-0.009
8	3	*964.428b	964.4621	0.034	*937.516	937.5182	0.002
8	4	964.271	964.2692	-0.002	*935.914b	935.9376	0.024
8	5	964.04115h	964.0413	0.0002	933.82600C	933.8275	0.001
		964.044			933.8265f		
_					933.833b		
6	6	963.79607h	963.7962	0.0001	931.12190°	931.1227	0.000
		963.651d			931.12271		
8	7	963.787 963.55939h	963.5587	-0.0007	931.124b		
0	,	963.55820	963.5587	-0.0007	927.74196 ^C 927.739	927.7403	-0.001
		963.555			927.739		
8	8	963.36262C	963.3627	0.0001	923.582	923.5827	0.001
	_	963.362	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	0.0001	,	223.3027	0.001
9	1		964.0219		*940.850b	940.8335	-0.016
9	2		963.9206		*940.235b	940.2212	-0.014
9	3	l	963.7560		*939.197	939.1837	-0.013
9	4	963.5343h	963.5352	0.0009	937.6989f	937.6929	-0.0060
	_				937.701		
9	5	963.276b	963.2691	-0.007	935.710	923.7053	-0.005
9	6	962.97368C	962.9733	-0.0003	933.15745h	933.1588	0.0013
		962.975			933.1571C		
9	7	962.6705h	962.6698	-0.0007	933.152 929.9711 ^f	929.9712	0.0001
,	'	962.6736°	902.0098	-0.0007	929.969d	929.9712	0.0001
		962.673			929.954		
9	8	962.38850h	962.3888	0.0003	926.04584C	926.0424	-0.0034
		962.3878C			926.047		
		962.389					
9	9	962.167	962.1715	0.004	921.255	921.2611	0.006
0	1		963.3374		942.42061	942.4201	-0.0005
0	2		963.2250		*941.872b	941.8415	-0.030
0	3		963.0415		*940.850b	940.8632	0.013
0	4	h	962.7931		*939.488b	939.4618	-0.026
0	5	962.4895h	962.4892	-0.0003	937.6119f	937.5988	-0.0132
0	6	962.1448h	962.1435	-0.0013	937.605	025 2150	0.004
U	0	302.1448"	902.1435	-0.0013	935.2216 ^f 935.223	935.2169	-0.0047
0	7	*961.760b	961.7752		935.223 932.233d	932.2373	0.004
		3011100	231.7732		932.233	332.23/3	0.004
0	8	961.4113h	961.4110	-0.0003	928.5581f	928.5590	0.0009
		961.409			928.559		



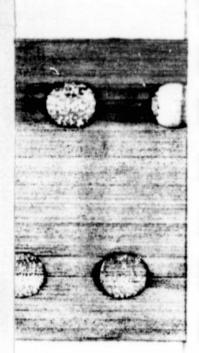
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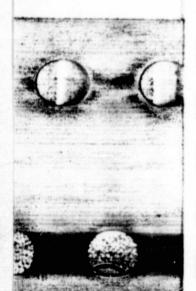
ν2, 2ν2, AND ν4 STATES OF "NH3

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TABLE I-Continued

J	K	Vobs	Vcalc	Vcal Vob	Vobs	Vcalc	"cal"ot
			sQ			aQ	
10	9	961.086	961.0870	0.001	924.068	924.0639	-0.004
10	10	960.85232 ^C 960.857	960.8522	-0.0001	918.6209f 918.620	918.6296	0.0087
11	1		962.6590			944.0025	
11	2		962.5357		943.4481 i 943.486b	943.4536	0.0054
11	3		963.3334		*942.56671 942.569	942.5290	-0.0377
11	4		962.0576		*941.255b	941.2108	-0.044
11	5		961.7165		*939.488b	939.4676	-0.020
11	6	*961.325b	961.3219	-0.003	937.273b	937.2494	-0.024
11	7	*960.857b	960.8908	0.034	934.471b	934.4829	0.011
11	8	960.446	960.4463	0.000	*931.124b	931.0693	-0.055
11	9	960.01990° 960.019	960.0199	0.0000	926.88456h 926.8844°	926.8868	0.0022
					926.887		
	10	959.656	959.6544	-0.002	921.8120 ^f 921.810	921.7999	-0.0121
11	11	959.408	959.4071	-0.001	915.665	915.6800	0.015
12	1		962.0032			945.5772	
12	3		961.8687		*945.0918ib	945.0511	-0.0407
12	3		961.6475		*944.18351	944.1696	-0.0141
12	4		961.3442		*942.9368ib	942.9217	-0.0151
12	5		960.9662			941.2849	
12	6		960.5237			939.2191	
12	7	1	960.0315			936.6600	
12	8		959.5098		*933.501b	933.5147	0.014
12	9	958.975	958.9858	0.011		929.6612	0.024
12	10	958.499	958.4968	-0.002	924.9499e	924.9547	0.0048
-					924.942		
12	11	958.089	958.0924	0.003	919.264	919.2443	-0.020
	12	957.835	957.8390	0.004	912.385	912.4029	0.018
_			s R			aR	
0	0		•••		951.7794 ¹ 931.775	951.7769	-0.0025
1	0	1007.5474 ³ 1007.544b	1007.5471	-0.0003			
1	1	1007.54068h 1007.544b	1007.5406	-0.0001	971.882249 971.882040 971.882	971.8821	-0.0001
2	0				*992.694	992.7003	0 000
2	1	1027.0467f	1027.0471	0.0004	992.45019k	992.7003	-0.0002
-	•	1027.040b	1027.04/1	0.0004		992.4500	-0.0002
2	2	1027.0331h	1027.0329	-0.0002	992.452 991.69069k	991.6903	-0.0004
•	•	1027.0335f 1027.040b	1027.0329	-0.0002	991.6914 ^C 991.690	991.6903	20.0004
3	0	*1046.4056 [£] 1046.392b	1046.4053	-0.0003			
3	1	1046.392b 1046.4008 ² 1046.392b	1046.4009	0.0001	1013.1758h 1013.174	1013.1757	-0.0001

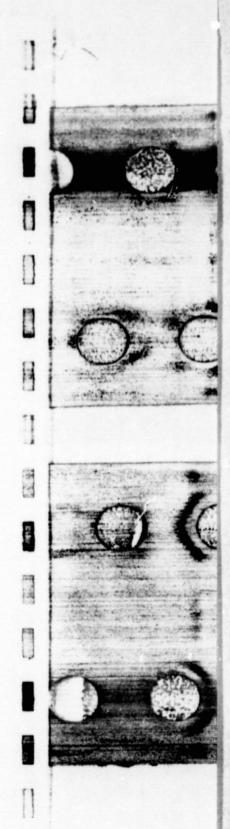




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TABLE I-Continued

J	K	Vobs	"calc	"cal"ob	Vobs	"calc	Vcal-Vob
			sR			aR	
3	2	1046.3881 ² 1046.3925	1046.3881	0.0000	1012.4452h 1012.442	1012.4449	-0.0003
3	3	*1046.37450h 1046.375d 1046.392b	1046.3745	0.0000	*1011.20350e 1011.2035c 1011.202	1011.2032	-3.0003
4	0				*1034.245	1034.2409	-0.004
4	1	1065.5944 ^f 1065.576 ^b	1065.5950	0.0006	1034.0128h 1034.011	1034.0141	0.0013
4	2	1065.5824f 1065.576b	1065.5820	-0.0004	1033.3165h 1033.316	1033.3166	0.0001
4	3	*1065.5655E 1065.576b	1065.5677	0.0022	*1032.1311h 1032.129	1032.1311	0.0000
4	4	1065.5636f 1065.576b	1065.5636	0.0000	1030.4222h 1030.421	1030.4217	-0.0005
5	0	*1084.62968	1084.6276	-0.0020			
5	1	1084.6244 ² 1084.608b	1084.6246	0.0002	1054.916	1054.9152	-0.001
5	2	1084.6082C	1084.6102	0.0020	1054.2511h 1054.254	1054.2548	0.0037
5	3	*1084.59306C 1084.608b	1084.5930	-0.0001	*1053.13044h 1053.135	1053.1320	0.0016
5	4	1084.58371C 1084.608b	1084.5837	0.0000	1051.51207h 1051.517	1051.5123	0.0002
5	5	1084.59924C 1084.608b	1084.5988	-0.0004	1049.34632h 1049.349	1049.3456	-0.0007
6	0				*1076.033	1076.0184	-0.015
5	1	1103.4846 ² 1103.452 ^b	1103.4864	0.0018	1075.8218h 1075.824	1075.8254	0.0036
5	2	1103.4686 ^g 1103.452 ^b	1103.4696	0.0010	1075.2024 [£] 1075.207	1075.2051	0.0027
5	3	*1103.4397 ² 1103.452b	1103.4478	0.0081	*1074.1477h 1074.152	1074.1505	0.0026
5	5	1103.4293 ² 1103.452b 1103.4324 ²	1103.4310	0.0017	1072.6262h 1072.631	1072.6288	0.0026
,	,	1103.452b	1103.4342	0.0018	1070.5909 [‡] 1070.594	1070.5914	0.0005
6	6	1103.4784 ² 1103.452b	1103.4792	0.0008	1067.9733h 1067.976	1067.9735	0.0002
7	0	*1122.1861 ² 1122.132 ^b	1122.1784	-0.0077		•••	
7	1	1122.1792 ² 1122.132 ^b	1122.1783	-0.0009	1096.6906h 1096.682	1096.6900	-0.0006
7	2	1122.1606 t 1122.132b	1122.1582	-0.0024	1096.1132f 1096.112	1096.1117	-0.0015
7	3	*1122.1332 ^k 1122.132 ^b	1122.1305	-0.0027	*1095.1296h 1095.124	1095.1288	-0.0008
	4	1122.1042 ² 1122.132 ^b	1122.1043	0.0001	1093.7119 ¹ 1095.715	1093.7111	-0.0008
,	5	1122.0937 ² 1122.132 ^b	1122.0934	-0.0003	1091.8120h 1091.816	1091.8128	0.0008
	6	1122.1181 ² 1122.132 ^b	1122.1175	-0.0006	1089.3704h 1089.370	1089.3704	0.0000



OF POOR QUALITY

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ν2. 2ν2, AND ν4 STATES OF 14NH3

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TABLE I-Continued

J	K	Vobs	Vcalc	vcal vob	vobs	Vcalc	Vcal Vol
			sR			aR	
7	7	1122.2046 [£] 1122.132 ^b	1122.2039	-0.0007	1086.3044 ^h 1086.295	1086.3032	-0.001
8	0	1140.697 ² 1140.628 ^b	1140.6988	0.002	*1117.642 *1117.453b	1117.6036 1117.4563	0.003
8	2	1140.628b	1140.6749	-0.002	1116.922	1116.9200	-0.002
8	3	*1140.624 [£] 1140.628 ^b	1140.6404	0.016	*1116.015	1116.0100	-0.005
8	4	1140.601 [£] 1140.628b	1140.6036	0.003	1114.703	1114.6992	-0.004
8	5	1140.575b2 1140.628b	1140.5768	0.002	1312.941b	1112.9457	0.005
8	6	1140.575bl	1140.5781	0.003	1110.687	1110.6893	0.002
8	7	1140.630 ¹ 1140.628 ^b	1140.6322	0.002	1107.843	1107.8503	0.007
8	8	1140.768 [£] 1140.628 ^b	1140.7725	0.004	1104.329	1104.3313	0.002
99999999	0 1 2 3 4 5 6 7 8 9	*1159.165b *1158.910b	1159.0418 1159.0479 1159.0198 1158.9776 1158.9289 1158.8851 1158.88624 1158.8830 1158.9771		1138.081 1137.597 *1136.759b 1135.559 1133.944 1131.868 1129.259 1126.025	1138.0777 1137.5818 1136.7423 1135.5368 1133.9287 1131.8628 1129.2630 1126.0314 1122.0536	-0.003 -0.015 -0.017 -0.025 -0.005 -0.004 -0.006
10	0 1 2 3 4 5 6 7 8 9	*1176.999b *1177.131b	1177.2275 1177.1944 1177.19436 1177.0820 1177.0199 1176.9719 1176.9583 1177.0061 1177.1515 1177.4422		*1158.663b *1158.493b *1156.049b *1157.301 1156.209b 1154.734b 1152.851b 1150.485 1147.535 1143.908 1139.471	1158.6099 1158.5200 1158.0600 1157.2846 1156.1770 1154.7079 1152.8295 1150.4716 1147.5389 1143.9135 1139.4646	-0.053 0.027 0.011 -0.016 -0.032 -0.022 -0.013 0.004 0.005 -0.006
	0 1 2 3 4 5 6 7 8 9 10 11	*1195.176b *1195.531b *1194.923b	1195.2250 1195.2426 1195.2038 1195.1431 1195.0669 1194.9847 1194.8609 1194.8620 1194.9458 1195.1547 1195.5449		*1178.703b 1178.319 *1177.591b *1176.618b *1175.293b *1173.583b 1171.435 1168.781 *1165.497b *1161.483b	1178.7681 1178.3665 1177.6135 1176.5895 1175.2441 1173.5398 1171.4156 1168.7823 1165.5222 1161.4957 1156.5574	0.065 0.018 0.023 -0.028 -0.049 -0.019 0.001 0.025 0.013

Note added in proof: ν_{obs} for aR(9,9) is 1122.0555 cm $^{-1}$ from our diode laser measurements (see Fig. 2).



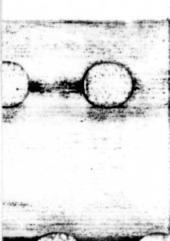


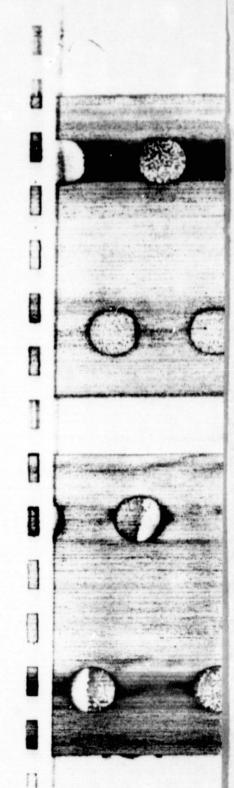
TABLE II

Observed and Calculated Transition Wavenumbers in the $2\nu_2$ Band of $^{14}NH_3$ (cm $^{-1}$)^a

J	K	Vobs	vcalc	Vcal Vob	Vobs	Vcalc	Vcal-Vol
			sP			aP	
1	0	1862.285 ^C	1862.2852	0.000			
2	0					1557.4402	
2	1	1842.099C	1842.0978	-0.001	1556.904	1556.8966	-0.007
3	0		1821.0697				
3	1	1821.255C	1821.2542	-0.001	*1538.012b	1538.0872	0.075
3	2	1821.814C	1821.8138	0.000	1536.458	1536.4411	-0.017
4	0				1520.442b	1520.4775	0.036
4	1	1799.9991d	1799.9994	0.0003		1519.8926	
4	2	1800.5218d	1800.5246	0.0028	1518.205	1518.1647	-0.040
4	3	1801.4303d	1801.4327	0.0024	*1515.337b	1515.4286	0.092
5	0	*1778.244C	1778.2226	-0.021			
5	1	1778.399C	1778.3929	-0.006		1502.3794	
5	2	1778.876C	1778.8763	0.000	1500.522b	1500.5565	0.034
5	3	1779.714C	1779.7129	-0.001	*1497.672b	1497.6352	-0.037
5	4	1780.958C	1780.9556	-0.002	1493.875	1493.8663	-0.009
5	0				1486.245b	1486.2533	0.008
6	1	1756.508C	1756.5051	-0.003		1485.6175	
5	3	1756.935C	1756.9399	0.004		1483.6886	
5	3	1757.689C	1757.6934	0.004		1480.5728	
5	5	1758.812C	1758.8155	0.003	1476.431b	1476.4561	0.025
		1760.390°	1760.3883	-0.002		1471.7713	
7	0	*1734.303C	1734.2412	-0.062			
	1	1724 0040	1734.4161			1469.6770	
,	2	1734.804° 1735.460°	1734.7962 1735.4564	-0.008	1467.645	1467.6287	-0.016
,	4	1/35.460-	1736.4429	-0.004		1464.3072	
,	5		1737.8324		1454.585b	1454.5813	-0.003
7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7	6	1739.739C	1739.7455	0.007	1449.227	1449.1740	-0.052
3	0					1455.3213	
3 3 3	1	1712.225C	1712.2137	-0.012		1454.6262	
3	0 1 2 3	1712.531°	1712.5341	0.003		1452.4369	
3	3	1713.084C	1713.0923	0.008	1	1448.8888	
3	4	1713.928°	1713.9301	0.003		1444.1324	
8	5	1714 7000	1715.1188			1438.3818	
3	7	1716.780° 1719.039°	1716.7706 1719.0563	-0.009		1431.9797	
	0	1719.039		0.017			
,	1	*1690.006b	1689.7693 1689.9922	-0.014		1440.5309	
,	2	1690.215b	1690.2488	0.034		1438.1608	
,	3	1690.664b	1690.6977	0.034	1434.318	1434.3468	0.029
,	4	1691.357b	1691.3764	0.019	1434.310	1429.2631	0.029
,	5	1692.354	1692.3494	-0.005		1423.1118	
,	6	1693.736	1693.7206	-0.015		1416.1252	
,	7	1695.6462d	1695.6492	0.0030		1408.6683	
,	8		1698.3710			1402.6953	
)	0					1428.2666	
)	1		1667.8509			1427.4485	
)	2		1668.0405			1424.8197	
)	3		1668.3745			1420.6748	
0	4	1	1668.8856			1415.2408	

^a Our infrared grating measurements unless stated otherwise; asterisk denotes that this value has not been used in the fit.

^c Calculated from the frequency of the hot band $2\nu_2 \leftarrow \nu_2$ transition (our measurements), ν_2 frequency, and the ground-state pure inversion transition.



^b Blended line.

ν2. 2ν2. AND ν4 STATES OF 14NH3

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TABLE II—Continued

J	K	Vobs	Vcalc	vcal-vob	obs	Vcalc	vcal-vot
			sP			aP	
10 10 10 10	5 6 7 8 9		1669.6316 1670.7066 1672.2579 1674.5058 1677.7695			1408.7246 1401.3108 1393.1949 1384.7213 1378.9995	
11 11 11 11 11 11 11 11	0 1 2 3 4 5 6 7 8 9		1645.5558 1645.8935 1646.0135 1646.5658 1647.0759 1647.8432 1649.0019 1650.7560 1653.4042			1415.3953 1412.3535 1407.8008 1415.2370 1409.8936 1402.8204 1393.9842 1383.3645 1370.7866 1355.1920	
12 12 12 12 12 12 12 12 12	0 1 2 3 4 5 6 7 8 9		1624.2283 1624.2769 1624.3694 1624.7966 1624.7966 1625.2479 1626.0030 1627.2489 1629.2630 1632.4426 1637.3413			1405.7387 1404.1348 1407.5765 1405.7684 1402.4624 1397.5366 1390.1382.4472 1372.2658 1360.3733 1346.6781	
			sQ			aQ	
1 2 2	1 1 2	1881.8646 ^d 1880.8542 ^d 1881.4389 ^d	1881.8637 1880.8523 1881.4400	-0.0009 -0.0019 0.0011	1596.634 *1597.594b 1596.046b	1596.6426 1597.6566 1596.0378	0.008 0.063 -0.008
3 3	1 2 3	1879.368° 1879.9299d 1880.8999d	1879.3695 1879.9319 1880.9022	0.001 0.0020 0.0023	*1597.594b *1594.810b	1599.2261 1597.5343 1594.8588	-0.060
4 4 4	1 2 3 4	1877.4586d 1877.9836d 1878.8987d 1880.2514d	1877.4555 1877.9848 1878.8985 1880.2501	-0.0031 0.0012 -0.0002 -0.0013	1596.790° 1593.157b	1601.3986 1599.6205 1596.7743 1593.1115	-0.016 -0.046
5 5 5 5 5	1 2 3 4 5	1875.168° 1875.648° 1876.4937° 1877.74712° 1879.4923°	1875.1619 1875.6511 1876.4960 1877.7471 1879.4879	-0.006 0.003 0.0023 0.0000 -0.0044	1604.181b 1602.347c 1595.341c 1590.859b	1604.2259 1602.3501 1599.3234 1595.3325 1590.8111	0.045 0.003 -0.008 -0.048
666666	1 2 3 4 5	*1872.561° 1872.994° 1873.7638d 1874.887° 1876.4821d 1878.6270d	1872.5510 1872.9937 1873.7590 1874.8938 1876.4764 1878.6290	-0.010 0.000 -0.005 0.007 -0.0057	1590.839 1607.775 1605.778° 1602.543° 1598.248° 1593.167° 1588.016	1607.7600 1605.7729 1602.5540 1598.2586 1593.1610 1587.9868	-0.015 -0.005 0.011 0.011 -0.006 -0.029

^d Calculated as indicated in footnote c but the diode laser measurements of the $2\nu_2 \leftarrow \nu_2$ frequency (20) have been used.





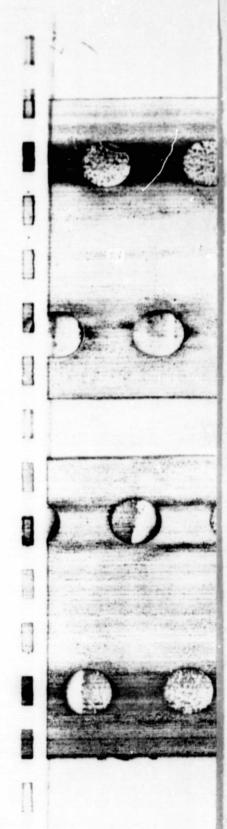


e Ref. (22).

^t Assigned as the $2\nu_2$ frequency although the corresponding upper state has the largest contribution from the $\nu_4(-l)$ state (see Section VI and Table XIII).

TABLE II-Continued

J	К	Vobs	Vcalc	vcal-vob	Vobs	Veale	Vcal Vob
_	\neg		sQ			aQ	
7 7 7 7 7	1 2 3 4 5	1870.081° 1870.735° 1871.763°	1869.6938 1870.0346 1870.7607 1871.7653 1873.1709	0.003 0.026 0.002	1612.064 ^C 1609.950 ^C 1606.501 ^C 1601.909 ^C 1596.373b	1612.0526 1609.9321 1606.4994 1601.9059 1596.3671	-0.011 -0.018 -0.002 -0.003 -0.006
	6 7	1875.102C 1877.681C	1875.0919 1877.7015	0.010	1590.225 ^C 1584.615	1590.2274 1584.6865	0.002
8 8 8 8	1 2 3 4 5 6 7 8	1866.660c 1866.979c 1867.559c 1866.410c 1869.661c 1871.343c 1873.6087d	1866.6690 1867.0032 1867.5822 1868.4447 1869.6571 1871.3254 1873.6116	0.009 0.024 0.024 0.035 -0.004 -0.018 0.0029	1611.161 *1606.297b *1600.297b 1593.652 1586.534 1581.014b	1617.1537 1614.8596 1611.1731 1606.2694 1600.3522 1593.6557 1586.5474 1580.9841	0.013 -0.028 0.074 0.004 0.013 0.029
9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9 9	1 2 3 4 5 6 7 8	1863.530 1863.846 1864.390b	1863.5615 1863.8353 1864.3105 1865.0214 1866.0275 1867.4255 1869.3657 1872.0720 1875.8676	0.031 -0.011 -0.079	*1620.601b 1605.018 1597.985 1582.219	1623.1060 1620.5600 1616.5538 1611.3158 1605.0545 1597.9567 1590.2206 1582.1937 1576.9893	-0.041 0.037 -0.028 -0.025
10 10 10 10 10 10 10 10 10 10 10 10 10 1	1 2 3 4 5 6 7 8 9		1860.4619 1860.6723 1861.C386 1861.5902 1862.3794 1863.4932 1865.0694 1867.3158 1870.5357 1875.1581		*1630.288b 1599.837 1587.831 1572.886	1629.9128 1626.9599 1622.5564 1630.2032 1625.1338 1618.4005 1609.9729 1599.8341 1587.8133	-0.085 -0.003 -0.017 -0.029
11 4 11 11 11 11 11 11 11 11 11	1 2 3 4 5 6 7 8 9	1869.1010 ^d	1857.4677 1857.6119 1857.8650 1858.2510 1858.8151 1859.6341 1860.8324 1862.6012 1865.222	-0.0005	*1640.823b *1607.502b	1637.3257 1640.8619 f 1639.2122 1636.1302 1631.4958 1625.2017 1617.2028 1607.5333 1596.2344 1583.2191	0.039
11 11	1		1974.7939 sR				
^	-				1616 2524	aR	
1 0	0	1920.6200d	1920.4180 1920.6191	-0.0019	1616.980b	1617.0601	0.080



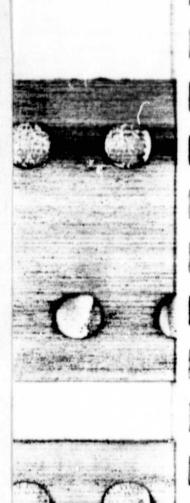
v2. 2v2. AND v4 STATES OF "NH3

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TABLE II-Continued

OF POOR QUALITY

,	K	obs	Calc	"cal"ob	Vobs	Vcalc	"cal "ol
			sR			aR	
	0				1659.336	1659.3593	0.024
١	1	1938.9681d	1938.9676	-0.0005	1658.777	1658.7954	0.018
	2	1939.5562d	1939.5580	0.0018	1657.128b	1657.1310	0.003
	0		1956.6277				
1	1	1956.8292d	1956.8256	-0.0036	1680.700	1680.7321	0.032
	2	1957.3904d	1957.3920	0.0016	1678.964	1678.9902	0.026
	3	1958.3681d	1958.3680	-0.0001	1676.219	1676.2046	-0.014
	0				1703.863	1703.8483	-0.014
	1		1974.2245		1703.236	1703.2451	0.009
	1 2 3	1975.6790d	1974.7596	0.0026	1701.399b	1701.4142	0.015
	4	1977.0421d	1975.6816	0.0026	1698.441b	1698.4626 1694.5776	-0.012
	0			-0.0003			
	1		1990.9939		1726.390b	1726.3683	-0.022
	2		1991.7050		1724.444	1724.4344	-0.010
	2		1992.5616		1721.289	1721.3046	0.016
	4		1993.8254		1717.132	1717.1350	0.003
	5	1995.5818d	1995.5760	-0.0058	1712.196b	1712.2008	0.005
	0					1750.7875	
	1 2 3 4 5	1	2007.8287		1750.154	1750.1356	-0.019
	2	1	2008.2821		1748.088 ^D	1748.0763	-0.012
	3		2009.0633		1744.733	1744.7461	0.013
	:		2010.2162 2011.8149		1740.312 1734.952	1740.2974	-0.015
	6	2013.9848d	2013.9755	-0.0093	1729.028	1729.0403	0.012
	G		2023.8772				
	ĩ		2024.1492			1774.5801	
	2		2024.5537		*1772.334b	1772.3548	0.021
	3		2025.2506		1768.769	1768.7837	0.015
	4		2026.2798		1764.034	1764.0429	0.009
	5	1	2027.7092		1758.376	1756.3375	-0.039
	0 1 2 3 4 5 6 7		2029.6467 2032.2568		1751.931 1745.084	1751.9034	0.028
	0				1800.503	1800.4941	-0.009
	1		2040.2383		1000.303	1799.7288	0.002
	2	1	2040.5896		*1797.338b	1797.2588	-0.079
	3		2041.1950			1793.3801	
	4		2042.0897		realist control	1788.3221	
	5		2043.3352		1782.291	1782.2949	0.004
	6		2045.0303		*1775.598b	1775.4872	-0.111
	8	1	2047.3281 2050.4556		*1768.133b 1760.428	1768.0996 1760.4826	0.033
		1					0.055
	0		2055.7768			1925 5703	
	2		2056.1725 2056.4670			1825.5703 1822.7002	
	2		2056.9746			1818.4355	
	4		2057.7260			1826.2782	
			2058.7753			1821.4637	
	5 6 7		2060.2121		1815.080	1815.0464	-0.034
	7		2062.1772		1806.919	1806.9986	0.080



YTUALID WART TO

TABLE II—Continued

J	К	Vobs	Vcalc	Vcal Vob	vobs	vcalc	"cal"of
	\neg		sR			aR	
9	8 9		2064.8819 2068.6338		1797.338 1785.824	1797.3065 1785.8030	-0.031 -0.020
10 10 10 10 10 10 10	0					1853.3861	
10	1		2072.0362			1851.8432	
10	2		2072.2707			1855.4684	
.0	3		2072.6752			1853.9678	
0	4		2073.2754	- 1		1851.0964	
0	5		2074.1185	- 1		1846.7360	
0	6		2075.2841	- 1		1840.7818	
0	7		2076.8999	- 1		1833.1915	
0	8		2079.1610	- 1		1824.0029	
0	9		2082.3544	- 1	*1813.351	1813.2611	-0.090
10	10		2086.8883		1800.850	1800.8838	0.034

0.3-0.6 Torr); the accuracy of those frequencies is about ± 1 MHz. The frequencies obtained by extrapolating to zero pressure from at least three measurements at different pressures are determined with the accuracy of ± 0.2 MHz (Table IV).

IV. GROUND-STATE ROTATIONAL AND CENTRIFUGAL DISTORTION CONSTANTS

Combination relations have been used in evaluating the molecular constants of the ground state of ¹⁴NH₃. In the standard application of this method to NH₃, rotational and centrifugal distortion constants are obtained separately for the symmetric and antisymmetric components of the inversion doublet. We have modified this treatment so that all the infrared data are processed simultaneously with the microwave pure inversion transition frequencies, using the following equation in addition to the standard combination relations:

$$\Delta_{i}(J,K) = 0.7934115 + [^{(a)}B_{0} - {^{(s)}B_{0}}]J(J+1)
- [^{(a)}D_{J}^{0} - {^{(s)}D_{J}^{0}}]J^{2}(J+1)^{2} - [^{(a)}D_{JK}^{0} - {^{(s)}D_{JK}^{0}}]J(J+1)K^{2}
+ [^{(a)}H_{JJJ}^{0} - {^{(s)}H_{JJJ}^{0}}]J^{3}(J+1)^{3} + [^{(a)}H_{JJK}^{0} - {^{(s)}H_{JJK}^{0}}]J^{2}(J+1)^{2}K^{2}
+ [^{(a)}H_{JKK}^{0} - {^{(s)}H_{JKK}^{0}}]J(J+1)K^{4} + \{[^{(a)}C_{0} - {^{(a)}B_{0}}] - [^{(s)}C_{0} - {^{(s)}B_{0}}]\}K^{2}
- [^{(a)}D_{K}^{0} - {^{(s)}D_{K}^{0}}]K^{4} + [^{(a)}H_{KKK}^{0} - {^{(s)}H_{KKK}^{0}}]K^{6}. \quad (17)$$

The values of the molecular parameters obtained in this way are then consistent also with the experimental inversion frequencies; they are listed in Table V. In this and all other fits the weights of the experimental data have been chosen to be inversely proportional to the squares of their estimated dispersions.

V. CORIOLIS AND I-TYPE INTERACTIONS IN THE EXCITED VIBRATION-INVERSION STATES OF "NH₃

The lowest-lying vibration-inversion energy levels of ¹⁴NH₃ (Fig. 4) form a complicated system of interacting levels. In order to keep the problem in a

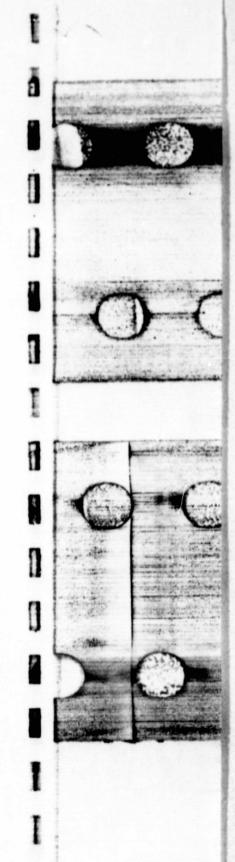
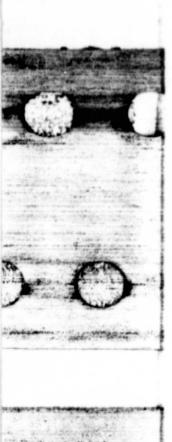


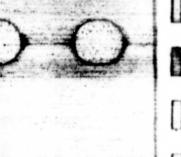
TABLE III

Observed and Calculated Transition Wavenumbers in the v4 Band of 14NH3 (cm-1)8

J	K	Vobs	Veale	vcal vob	Vobs	Vcalc	Vcal Vot
			s ^P P			a ^P P	
1	1	1610.089	1610.0987	0.010	*1610.361b	1610.4571	0.096
2 2	1 2	1590.684 1594.810b	1590.6940	0.010	1591.085 1595.072b	1591.1419	0.057
3	1	1571.823b	1571.8419	0.019	1572.483	1572.5064	0.024
3	2	1575.852 1579.364	1575.8580	0.006	1579.621b	1576.3409	0.052
4	1	1553.622b	1553.6469	0.025	1554.725	1554.7387	0.014
4	3	*1557.532b	1557.4948	-0.035	*1558.270b	1558.2431	-0.027
4	4	1563.829	1560.8956 1563.8094	0.003	1561.373 1564.075	1561.3889 1564.1027	0.016
5	1	1536.207	1536.2316	0.025	1538.012b	1538.0321	0.020
5	2	1539.756 1542.973	1539.7715 1543.0126	0.016	1541.002 1543.844	1540.9782 1543.8055	-0.024
5	4	1545.796	1545.8165	0.020	1546.324	1546.3065	-0.018
5	5	1548.188	1548.1644	-0.024	1548.421	1548.4278	0.007
6	1 2	1519.659b *1522.712b	1519.6938 1522.7758	0.035	1522.388 1524.736	1522.4910	0.103
6	3	1525.761	1525.7670	0.006	1527.055	1527.0520	-0.003
6	4	1528.375	1528.4040	0.029	1529.287	1529.2215	-0.065
6	5	1530.613 1532.445	1530.6304 1532.4265	0.017	1531.153	1531.1241	-0.029
7	1	1504.025	1504.0737	0.018	1532.666	1532.6630	-0.003
7	2	*1506.604b	1506.5935	-0.010		1509.5679	
7	3	*1509.138b	1509.2316	0.094	1511.314	1511.2770	-0.037
7 7	5	1511.599 1513.654	1511.6277 1513.6786	0.028	1513.054b 1514.599	1512.9703 1514.5373	-0.084
ŕ	6	1515.337	1515.3466	0.010	1515.871	1515.8837	0.012
7	7	*1516.663b	1516.6045	-0.058	1516.840	1516.8229	-0.017
8	1	*1489.304b	1489.3635	0.059	1405 736h	1494.9140	-0.075
8	2	1491.320	1491.2920 1493.4829	-0.028	1495.726b	1495.6511 1496.6293	-0.075
8	4	1495.527	1495.5537	0.027	1497.672b	1497.7002	0.028
8	5	1497.332	1497.3607	0.029	*1498.822b	1498.7882	-0.036
8	6	*1498.822b	1498.8457	0.024	1499.806 1500.522b	1499.8198 1500.6392	0.014
8	é	1500.746	1500.7068	-0.039	1500.933	1500.9200	-0.013
9	1		1475.5394			1482.8682	
9	2	1476.879	1476.9114 1478.5927	0.032	*1483.249b	1483.0076 1483.2377	-0.011
9	4	1480.191	1480.2554	-0.064	-1403.249	1483.5720	-0.011
9	5	1481.718	1481.7408	0.023		1484.028€	
9	6	1482.955	1482.9743	0.019		1484.5927	
9	8	1483.911 1484.496	1483.9131 1484.5187	0.002		1485.1588 1485.4564	
9	9	1484.779	1484.7417	-0.037	1484.950	1484.9608	0.011
0	1		1462.5856			1472.0371	
0	2	1464 642	1463.4693	-0.022	1471.623b	1471.6845	0.062
0	4	1464.643	1464.6209 1465.8093	-0.022	*1471.147b *1470.754b	1471.2078	-0.008
o	5	1466.939b	1466.8940	-0.045	14.01734	1470.4355	0.000

^a Our infrared grating measurements unless stated otherwise; asterisk denotes that this value has not been used in the fit.





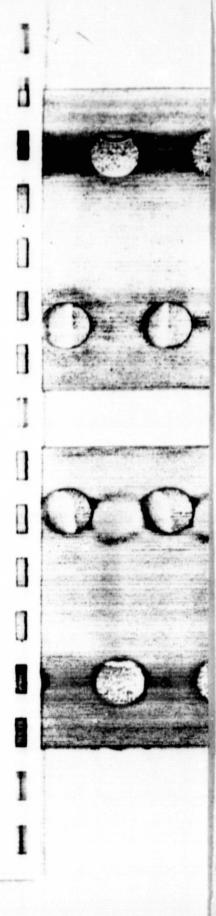


^b Blended line.

^c Assigned as the ν_4 ^RX (X = P, Q, R) frequency although the corresponding upper state has the largest contribution from the ν_4 (-l) state (see Section VI and Table XIII).

TABLE III - Continued

J K	Vobs	vcalc	vcal ob	Vobs	calc	"cal "ot
		s ^P P			a ^P P	
10 6 10 7 10 8 10 9	1467.783 1468.466 •1468.920b •1468.920b 1468.753b	1467.7981 1468.4752 1468.8873 1468.9890 1468.7170	0.015 0.009 -0.033 0.069 -0.036	1468.920	1470.3631 1470.5058 1470.6699 1470.4116 1468.9406	0.021
11 1 11 2 11 3 11 4 11 5 11 6 11 7 11 8 11 9 11 10	*1452.828b *1453.335b 1453.743b 1453.907b 1453.335b 1452.662	1450.5043 1450.9757 1451.6141 1452.2890 1452.9040 1453.3978 1453.8682 1453.7729 1453.8682	0.076 0.063 -0.013 -0.039 0.030 0.055 -0.022	1452.828	1462.5162 1461.7552 1460.6378 1459.3724 1458.1999 1457.3275 1456.48519 1456.6879 1456.5894 1455.5894	0.011
12 1 2 2 2 3 12 4 12 5 12 6 12 7 12 8 12 9 12 10 12 11 12 12	*1439.621b *1439.621b *1439.208b 1438.563 1437.684 *1436.483b	1439.3178 1439.4463 1439.6113 1439.7628 1439.8582 1439.8573 1439.5399 1439.1554 1438.5725 1438.5725	-0.010 -0.081 -0.053 0.010 0.042 0.037		1408.5551 1453.3373 1451.6431 1447.5176 1447.5176 1444.4163 1443.6952 1443.3408 1442.8008 1442.8008 1442.810792 1436.6113	
		s ^P Q			a ^P Q	
1 1	1630.444b	1630.4598	0.016	1630.878	1630.8879	0.010
2 1 2 2	1631.427 1635.520b	1631.4400 1635.4842	0.013	1632.028b 1635.896b	1632.0758 1635.9376	0.048
3 1 3 2 3 3	1632.998 1636.905 1640.358	1633.0170 1636.9020 1640.3651	0.019 -0.003 0.007	1634.076 *1637.544b 1640.823	1634.0721 1637.6128 1640.8191	0.004 0.069 -0.004
4 1 4 2 4 3 4 4	1635.268 1638.881 1642.176b 1645.117	1635.2942 1638.8801 1642.1982 1645.1109	0.027 -6.001 0.022 -0.006	1637.023b 1640.075 1643.003 1645.588	1637.0513 1640.0422 1642.9447 1645.5516	0.028 -0.033 -0.058 -0.036
5 1 5 2 5 3 5 4 5 5	1638.325 1641.515 1644.588 *1647.406b 1649.731	1638.3506 1641.4870 1644.5696 1647.3356	0.025 -0.028 -0.018 -0.070	1643.382 1648.183	1641.0993 1643.3683 1645.8026 1648.0979	-0.014
6 1 6 2 6 3 6 4	*1649.731 *1642.176b 1644.827 *1647.592b 1650.062	1649.7299 1642.2086 1644.7910 1647.5341 1650.0786	-0.001 0.033 -0.036 -0.058 0.017	1650.213 1649.582 1651.455	1650.1639 1646.2071 1647.7121 1649.5238 1551.3618	-0.049 -0.058 -0.093



ν₂, 2ν₂, AND ν₄ STATES OF "NH₂

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TABLE III—Continued

J	K	Vobs	Vcalc	"cal"ob	Vobs	Vcalc	Vcal-Vo
Ξ			s ^P Q			a ^P Q	
6	5	*1652.418b	1652.3226	-0.095		1653.1170	
6	6	1654.217b	1654.2301	0.013	1654.703	1654.6965	-0.007
7	1		1646.8437			1652.3403	
777777	2	1648.857	1648.8425	-0.015		1653.1463	
7	3	1651.201	1651.1513	-0.050	*1654.217b	1654.2399	0.023
7	4 5	•1653.311b	1653.3888	0.078	*1655.431b	1655.4737	0.043
7	6	*1655.431b 1657.128b	1655.4128	-0.018	1656.839	1656.7735	-0.065
7	7	1658.601	1657.1670 1658.6187	0.039	1658.071	1658.0675	-0.003
8	1		1652.2163		*1659.528b	1659.4910	-0.037
8	2		1653.6658			1659.7064	
8	3	*1655.431b	1655.4772	0.046		1660.0640	
8888	5	1659.056b	1657.3237	-0.007		1660.5783	
ě	6	1660.560	1659.0485 1660.5791	0.019		1661.2690	
8 8 8	7	1661.833	1661.8755	0.043		1663.0379	
8	8	*1662.813b	1662.9024	0.089		1663.7453	
999999	1		1658.2962			1667.6946	
9	2		1659.2641			1667.4248	
9	4	1669.560	1660.5569 1661.9451	-0.003		1667.0869	
9	5	1663.270	1663.2899	0.020		1666.7655	
é	6	1664.510	1664.5170	0.007		1667.0090	
9	6	1665.565b	1665.5830	0.018		1667.5315	
9	8	1666.379b	1666.4534	0.075		1668.1423	
9	9	1667.066	1667.0871	0.021		1668.4014	
0	1	*1665.565b	1665.0728			1677.0337	
0	3	*1666.379b	1665.6345 1666.4244	0.070		1676.3616	
ŏ	4	*1667.349b	1667.3134	-0.036		1674.3386	
0	5		1668.2075			1673.4402	
0	6	1668.975b	1669.0478	0.074		1672.9076	
0	7		1669.7978			1672.8407	
0	8	1670.423	1670.4280	0.005		1673.1575 1673.5212	
0	10	1670.873 •1671.197b	1670.9044 1671.1777	-0.019		1673.2540	
1	1		1672.5572			1641.2047	
1	2		1672.7814			1686.6227	
1	3		1673.1069			1685.0870	
1	4 5		1673.4854			1683.2691	
1	6		1673.8767 1674.2535			1681.4768	
i	7		1674.5974			1679.1719	
î	8		1674.8921			1678.9627	
1	9		1675.1153			1679.2018	
1	10		1675.2304 1675.1785			1679.3418 1678.3922	
-	**		s ^P R			a ^P R	
_			s R			a R	
1	1	1671.197b	1671.2058	0.009	1671.793	1671.8218	0.029



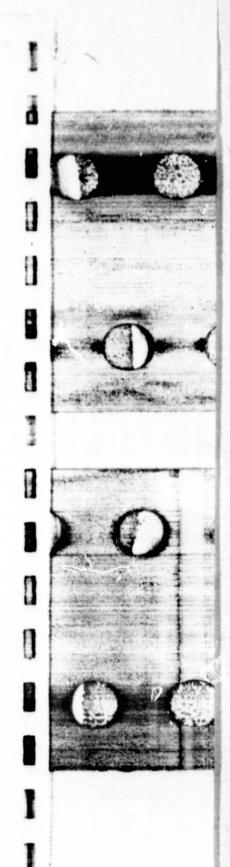


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TABLE II! - Continued

J	К	Vobs	Calc	vcal-vob	Vobs	vcalc	vcal-vot
_			s ^P R			a ^P R	
2	1 2	1692.615 1696.516	1692.6151 1696.5282	0.000	1693.643 1697.224	1693.6415 1697.2095	-0.001 -0.015
3 3	2 3	1714.639 1718.281 •1721.728b	1714.6643 1718.2873 1721.6678	0.025 0.006 -0.060	1716.357 1719.430 1722.425	1716.3848 1719.4119 1722.3749	0.028 -0.018 -0.050
4 4 4	1 2 3 4	1737.388 1740.601 1743.727 1746.605b	1737.4132 1740.5956 1743.7552 1746.6301	0.025 -0.006 0.028 0.025	*1739.982b 1742.433 1744.978 1747.416	1740.1186 1742.4324 1744.9418 1747.3430	0.137 -0.001 -0.037 -0.073
5 5 5 5 5	1 2 3 4 5	*1760.802b 1763.527 *1766.252b 1768.980 *1771.287b	1760.8654 1763.5022 1766.3367 1769.0103 1771.4221	0.063 -0.024 0.085 0.030	1764.804b •1766.253b 1768.294 •1770.340b •1772.287b	1764.8155 1766.3736 1768.2744 1770.2382 1772.1568	0.011 0.121 -0.019 -0.102 -0.130
6 6 6 6 6	1 2 3 4 5 6	*1784.943b 1787.051b 1789.512b 1791.843 1794.030	1784.9786 1787.0400 1789.4539 1791.8397 1794.0568 1796.0505	0.036 -0.011 -0.058 -0.003 0.027 -0.042	*1793.775b 1795.443 1796.921b	1790.4233 1791.2904 1792.4866 1793.8652 1795.3532 1796.8803	-0.090 -0.090 -0.041
777777777	1 2 3 4 5 6 7	1811.191b 1813.121 1815.080b	1809.6964 1811.2162 1813.1456 1815.1589 1817.1006 1818.9004 1820.5207	0.025 0.024 0.079		1816.9173 1817.2016 1817.6746 1818.3518 1819.2543 1820.3709 1821.6008	
8 8 8 8 8 8 8	1 2 3 4 5 6 7 8	1837.457 1840.601 •1842.147b	1834.9730 1536.0185 337.4414 1839.0134 1840.5976 1842.1217 2845.5454	-0.015 -0.003 -0.025		1844.3174 1844.1236 1843.9132 1843.3270 1844.0058 1844.5395 1845.4106	
9 9 9 9 9 9	1 2 3 4 5 6 7		1860.7833 1861.4293 1862.3604 1863.4492 1864.6034 1865.7667 1866.9055			1872.6912 1872.1019 1871.2724 1870.4136 1869.7701 1869.5535 1869.8663	
9 10 10 10 10 10	8 9 1 2 3 4 5		1867.9941 1869.0027 1887.1257 1887.4401 1887.9171 1888.5098 1889.1802			1870.6298 1871.5110 1855.7222 1901.2291 1899.8425 1898.231 1896.7171 1895.6109	



vs. 2vz, AND v4 STATES OF "NH3

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TABLE III - Continued

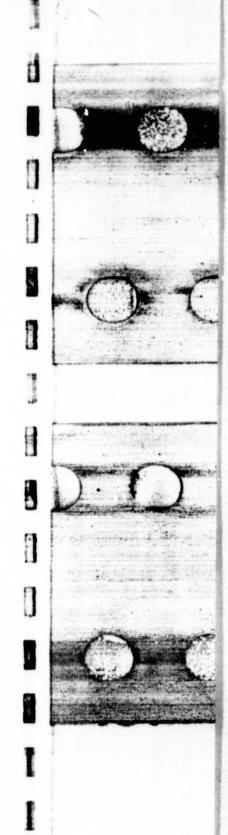
J	K	Vobs	Vcalc	"cal"ob	obs	Vcalc	Vcal-Vot
			s ^P R			a ^P R	
10 10 10	7 8 9		1890.6649 1891.4519 1892.2469 1893.0182			1895.1606 1895.4323 1896.2285 1897.0065	
			s ^R p			a ^R p	
2	0				1586.859	1586.8905	0.031
3	0	1567.990 1561.756	1567.9820 1561.7647	0.009	1562.369	1562.3783	0.009
4 4	0 1 2	1543.213 1537.464	1543.2388 1537.4717	0.026 0.008	1552.147 *1543.976b *1538.288b	1552.0983 1544.0147 1538.3108	-0.049 0.039 0.023
5 5 5	0 1 2 3	*1533.361b 1524.948 1519.527 *1513.054b	1533.7662 1524.9773 1519.5267 1513.0786	-0.095 0.029 0.000 0.025	*1525.761b 1520.747 1514.236	1525.8208 1520.7207 1514.2343	0.060 -0.026 -0.001
6 6 6 6	0 1 2 3 4	*1501.859b 1495.726 1488.596b	1506.8824 1501.9012 1495.7247 1488.5850	0.042 -0.001 -0.011	1521.264 *1507.809b 1503.431 1497.518 1490.191	1521.2243 1507.7154 1503.3869 1497.4829 1490.1889	-0.040 -0.094 -0.044 -0.035 -0.002
7 7 7 7 7 7	0 1 2 3 4 5	1502.754 1488.962b 1484.496 1478.715 *1471.889b 1463.997	1502.7137 1488.9154 1484.5084 1478.7080 1471.8475 1463.9890	-0.040 -0.047 0.012 -0.007 -0.041 -0.008	1486.245b 1481.083 1474.397 1466.212	1489.7439 1486.2209 1481.0309 1474.3912 1466.2159	-0.024 -0.052 -0.006 0.004
888888	0 1 2 3 4 5 6	1471.147 1467.344 1461.969 1455.495b	1471.0952 1467.2891 1461.9553 1455.4543 1447.8958 1439.2895	-0.052 -0.054 -0.013 -0.041	1494.241 1469.180 1458.961 1451.530	1494.3012 1472.0249 1469.2347 1464.8287 1458.9233 1451.4957 1442.3400	0.060 0.055 -0.038 -0.034
9999999	0 1 2 3 4 5 6 7	1474.737 1450.180 1445.414 *1439.310b 1432.170 1423.901	1474.7816 1453.4751 1450.2295 1445.4154 1439.3445 1432.1498 1423.8678 1414.4866	0.045 0.049 0.001 -0.035 -0.020 -0.033	1428.824	1454.7163 1452.5523 1448.9270 1443.8018 1437.1169 1428.7759 1418.5357	-0.048
10 10 10 10 10	0 1 2 3 4 5		1436.1282 1433.3655 1429.0762 1423.4811 1416.7038 1408.7949			1471.6725 1438.0228 1436.4106 1433.5140 1429.1507 1423.1851 1415.5329	





TABLE III—Continued

J	K	obs	Vcalc	"cal"ob	Vobs	Vcalc	Vcal Vo
			sRp			a ^R p	
10	7		1399.7610			1406.1186	
10	8		1389.5813			1394.7183	
11	0		1420.2757				
11	1		1419.1469			1422.2458	
11	2		1416.7782 1412.9757			1421.1994	2
11	4		1407.8661			1418.9497	
11	5		1401.5382			1395.1671	
11	6		1394.0395			1387.4224	
11	7		1385.3862			1378.8971	
11	9		1375.5727			1369.7325	
12	0						
12	1		1402.6499			1454.1513	
12	2		1400.5891			1400.5114	
12	3		1397.2061			1395.5822	
12	4		1392.5536			1389.4992	
12	5		1386.6760			1382.3974	
12	7		1371.3495			1374.3996	
12	8		1361.9180			1356.0518	
12	9		1351.3002			1345.8586	
. 2	10		1339.4772			1335.2400	
			s ^R Q			a ^R Q	
1	0	1625.487b	1625.4747	-0.013			
2	0	1621.351	1621.3628	0.012	1626.133b 1621.934	1626.1827 1621.9477	0.050
3	0	1626.340b	1626.4059	0.066			
3	1	*1622.623b	1622.6089	-0.014	1623.350	1623.3482	-0.002
3	2	*1616.833b	1616.8789	0.046	1617.668	1617.6804	0.012
4	0				1627.314	1627.4063	0.092
4	1 2	1624.005 1618.613	1624.0399	0.035	1624.879b	1624.8401	-0.039
4	3	1612.250	1618.6352 1612.2642	0.022	1619.790 1613.336b	1619.7848 1613.3735	-0.005 0.038
5	0		1627.9994	0.014			
5	1	1625.599b	1627.9994	-0.059	1625.341b	1626.3237	-0.017
5	2	1620.601	1620.6125	0.011	1622.062	1622.0484	-0.017
5	3	1614.511	1614.5273	0.016	1616.273b	1616.2335	-0.039
5	4	1607.502	1607.5166	0.015	1609.045	1609.0653	0.021
6	0	1627.009b	1637 0503		1629.279b	1629.4055	0.127
6	2	*1627.009b	1627.0503 1622.7060	0.041	*1627.875b 1624.387	1627.8269 1624.3651	-0.046 -0.022
6	3	1616.980b	1617.0106	0.031	1619.320	1619.2776	-0.022
6	4	*1610.326b	1610.2984	-0.028	1612.801	1612.7827	-0.019
6	5	1602.627	1602.6330	0.006	1604.763	1604.7956	0.033
7	0	*1630.288b	1630.1886	-0.099			
	1 2	1628.608 *1624.879b	1628.5753 1624.8396	-0.032	1626.664	1629.4513 1626.7299	0.066
7							



ν2, 2ν2, AND ν4 STATES OF "NH3

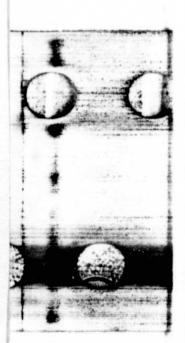
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TABLE III-Continued

J	K	Vobs	Vcalc	Vcal Vob	Vobs	Vcalc	Vcal Vot
			s ^R Q			a ^R Q	
7	4	1613.336b	1613.2894	-0.047	*1616.716b	1616.6968	-0.019
7 7 7	5	1605.950	1605.9479	-0.002	*1609.458b	1609.4810	0.023
-	6	*1597.594b	1597.6108	0.017	1600.567	1600.5877	0.021
8	0		1620 1510		*1621 204h	1632.5383	
8	2	1627.009b	1630.1519 1626.9839	-0.025	*1631.284b 1629.279	1631.3391 1629.2511	0.055
8	3	*1622.383b	1622.2999	-0.083	*1625.599b	1625.7533	0.154
8	4	*1616.378b	1616.4128	-0.035	*1620.709b	1620.8081	0.099
8	5	*1609.458b	1609.4575	0.000	1614.367	1614.3572	-0.010
8	6 7	1601.469b 1592.432	1601.4726 1592.4490	0.004	1606.297 1596.373b	1606.3064 1596.4147	0.010
	- 1						
999999	0	1632.998b	1633.0390 1631.8388	-0.040		1633.6804	
9	2	1629.171b	1629.1603	-0.011		1632.1509	
9	3		1625.0123	0.011		1629.3931	
9	4	1619.603b	1619.6169	0.013		1625.2257	
9	5	1613.059	1613.0997	0.041		1619.5150	
9	6 7	1605.527b	1605.5138	-0.013	*1612.136b	1612.1788	0.043
9	é	1596.840 1587.115b	1596.8688 1587.1474	0.029	1603.130 1592.178b	1603.1443 1592.1907	0.014
10	0					1637.7580	
0	ĭ		1633.7154			1636.76329	0
10	2	*1631.427b	1631.4369	0.010		1635.8058	
LO	3		1627.7859			1633.7053	
10	4	aaab	1622.8905		*1616.980b	1616.9754	~0.005
0	5	*1616.833b	1616.8416	0.008	*1610.326b *1602.926b	1610.4074	0.081
10	7	1601.469b	1609.6895 1601.4537	-0.015	*1594.810b	1603.0025 1594.8858	0.076
10	8	1592.178	1592.1325	-0.046	1,,,,,,,,,,	1586.2020	0.070
10	9	*1581.615b	1581.7082	0.093	*1577.276b	1577.2549	-0.021
11	0		1636.9352				
11	1		1635.8893		12	1687.6396	C
11	2		1633.9242			1633.7968	
11	3 4		1630.7017 1626.2762			1629.0261 1623.1670	
ii	5		1620.6945			1616.3566	
ii	6		1613.9892			1608.7203	
11	7		1606.1789			1600.3525	
11	8		1597.2703			1591.3194	
11	10	1587.269 *1576.198b	1587.2601 1576.1351	-0.009		1581.7197 1571.7810	
	10	-15/6.1985		-0.063			
			sRR			a ^R R	
1	0	1667.349	1667.3303	-0.019	:":	!**	
1	1	1661.140	1661.1286	-0.012	1661.685	1661.6937	0.009
2	0				1691.037	1690.9800	-0.057
2	1	1682.167	1682.2070	0.020	1682.925	1682.9176	-0.007
2	2	1676.509	1676.5051	-0.004	1677.263	1677.2771	0.014
3	0	1712.196b	1712.1712	-0.025			
3	1	1703.382	1703.4100	0.028	1704.202	1704.1735	-0.028

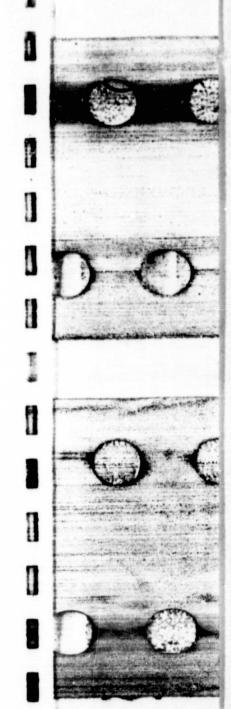


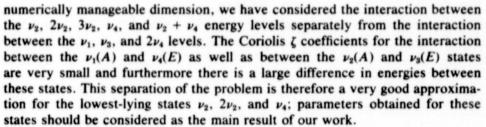


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TABLE III—Continued

J	K	Vobs	Vcalc	vcal vob	Vobs	Vcalc	Vcal Vo
			s ^R R			a ^R R	
3	2	1698.026	1698.0425	0.016	1699.181	1699.1544	-0.026
3	3	1691.746	1691.7338	-0.012	1692.792	1692.8037	0.012
4	0				1738.843	1738.8193	-0.024
4	1	1724.581	1724.6017	-0.021	1725.378	1725.3430	-0.035
4	3	1719.704	1719.7210	0.017	1721.151b	1721.1125	-0.038
4		1713.711	1713.7129	0.002	1715.405	1715.3727	-0.032
4	4	1706.818	1706.8110	-0.007	1708.306	1708.3104	0.004
5	0	1759.503	1759.4664	-0.037			
5	1	*1745.743b	1745.7070	-0.036	1746.427	1746.4352	0.008
5	3	1741.417 1735.817	1741.4172 1735.8132	0.000	1743.050b 1738.075b	1743.0266 1738.0282	-0.023 -0.046
5	4	1729.226	1729.2300	0.004	1731.694	1731.6591	-0.035
5	5	1721.728b	1721.7325	0.005	1723.821	1723.8354	0.014
6	0				1789.684	1789.7673	0.083
6	ĭ	1766.751	1766.7102	-0.041	1767.504	1767.5343	0.030
6	2	1763.105	1763.0371	-0.068	1764.804	1764.8741	0.070
6	3	1757.932	1757.9263	-0.006	1760.693	1760.6861	-0.007
6	4	1751.749	1751.7403	-0.009	1755.100	1755.0883	-0.011
6	5	1744.604	1744.5919	-0.012	1748.088	1748.0607	-0.027
6	6	1736.504	1736.4944	-0.010	1739.386	1739.4010	0.015
7	0	1808.848	1808.8894	0.042			
7	1		1787.6321			1788.7654	
7	2	1784.507	1784.5343	0.027		1786.7464	
7	3	1779.960	1779.9683	0.008	1770 F13b	1783.3639	0 070
7	5	1774.245 1767.504	1774.2479 1767.5096	0.003	1778.512b 1772.334b	1778.5816 1772.3425	0.070
7	6	1759.812	1759.7939	-0.018	1764.538	1764.5541	0.016
7	7	*1751.027b	1751.0942	0.067	1754.986	1754.9777	-0.009
8	0					1843.9000	
8	1	1	1808.5156			1810.3031	
8	2		1805.9146			1808.8497	
8	3	1801.872	1801.8967	0.025		1806.2194	
8	4	*1796.635b	1796.6852	-0.050		1802.2320	
8	5	1790.398	1790.4074	0.009	1700 cosb	1796.7554	0 025
8	6	1783.150b	1783.1186 1774.8316	-0.031 -0.017	1789.684b 1781.028	1789.7093 1781.0233	0.025
8	8	1774.848 1765.504	1765.5311	0.027	1770.498	1770.4796	-0.018
9	0	1703.304		0.027			
9	1	*1829.411b	1860.2487 1829.4260	0.015		1832.4208	
á	2	1029.411-	1827.2317	0.013		1831.5461	
9	2		1823.7219			1829.5844	
9	4		1819.0263			1813.0504	
9	6		1813.2375			1806.7373	
9	6	*1806.393b	1806.4084	0.015	*1799.625b	1799.6484	0.023
9	7	1798.585	1798.5615	-0.024		1791.9115	
	8	*1789.684b	1789.6986	0.014	1775 270	1783.6744	0 026
9	9	*1779.727b	1779.8062	0.079	1775.270	1775.2446	-0.026
0	0		1850.4577			1901.7987	c
0	2		1848.5829			1848.4033	
0	3		1845.5119			1843.7816	
0	4	i	1841.3006			1838.1332	
0	5		1835.9979			1831.5969	
0	6		1829.6392			1824.3004	
0	7	*1822.154	1822.2464	0.092		1816.3412	
0	8	1813.852	1813.8301	-0.022		1807.7890	
0	9	1804.413	1804.3917	-0.022		1798.7464	
0	10	1793.775	1793.9120	0.137	*1789.512b	1789.4456	-0.066





In our scheme, Coriolis interactions between the ν_2 , $2\nu_2$, $3\nu_2$, ν_4 , and $\nu_2 + \nu_4$ energy levels are described by the operator (4, 5)

$$H_{2,4} = (-i/2)[X_{4a}^{\nu\rho}J_{\rho} + (1/2)(J_{\rho}X_{4a}^{\nu\rho})](Q_{4}^{-}J_{+} - Q_{4}^{+}J_{-}), \tag{18}$$

where $Q_4^{\pm} = Q_{4a} \pm iQ_{4b}$, $J_{\pm} = J_x \pm iJ_y$. Besides this, there is a centrifugal distortion operator

$$H_{\text{Cent}} = (1/8)(Y_{4a4a}^{xx} - Y_{4a4a}^{yy})[(Q_4^+)^2 J_-^2 + (Q_4^-)^2 J_+^2]$$
 (19)

which connects the $\pm l$ levels in the ν_4 and $\nu_2 + \nu_4$ states. The energy matrix of these interactions is thus the 7×7 matrix given in Table VI. For the nondegenerate vibrational states, the diagonal matrix elements of this matrix are given by Eq. (15). For the inversion-rotational energy levels in the degenerate vibrational states the following expression has been used:

$$\left(\frac{E_i}{hc}\right) = \left(\frac{E_i^0}{hc}\right) + B_i J(J+1) + (C_i - B_i)k^2 - 2C_i \zeta_4^{(i)}kl_4 - D_J^{(i)}J^2(J+1)^2
- D_{JK}^{(i)}J(J+1)k^2 - D_K^{(i)}k^4 + \eta_{4J}^{(i)}J(J+1)kl_4 + \eta_{4K}^{(i)}k^3l_4
+ H_{JJJ}^{(i)}J^3(J+1)^3 + H_{JJK}^{(i)}J^2(J+1)^2k^2 + H_{JKK}^{(i)}J(J+1)k^4 + H_{KKK}^{(i)}k^6
+ \tau_{4JJK}^{(i)}J^2(J+1)^2kl_4 + \tau_{4JKK}^{(i)}J(J+1)k^3l_4 + \tau_{4KKK}^{(i)}k^5l_4 + \tau_{4KK}^{*(i)}k^3l_4^3.$$
 (20)

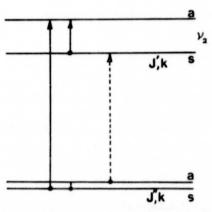
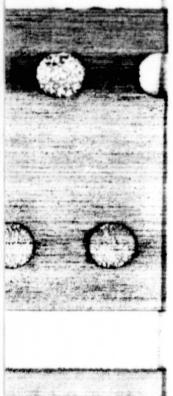


Fig. 3. Energy-level diagram illustrating the calculation of the ν_2 transition frequency (dashed line) from the known ν_2 frequency for a transition starting from a level with opposite parity and from the known pure inversion transitions in the ground and excited ν_2 states (see Table I).



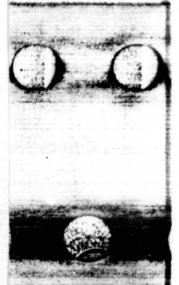


TABLE IV

Submillimeter Wave Inversion and Rotation-Inversion Transitions of ¹⁴NH₃ in the ν_2 State (MHz)

			Pure	Inver	sion	Transit	ion Freque	encies			
J	K		vinv.	J	K		'inv.	J	K		'inv.
1	1	1066	650.8ª	7	1	810	918.7	10	6	790	818.2
_	-			7	2		521.9	10	7	867	343.5
2	1	1045	318.7	7 7 7 7 7	1 2 3 4 5 6		394.2	10	8	964	060.3
2	2	1067	676.8ª	7	4	902	459.2	-	7		
				7	5	961	885.4	11	6	706	899.0
3	1	1014	084.0	7	6	1039	360.8	11	7	775	717.6
3 3	2	1035	816.2					11	8	862	810.
3	3	1073	050.7	8	1	747	286.8	11	9	972	301.6
				8 8 8 8	1 2 3 4 5	763	583.3				
4	1	973	826.5	8	3	791	531.4	12	8	763	996.
444	2 3	994	747.8	8	4	831	969.3	12	8	861	672.5
4	3	1030	531.2	8		887	018.8				
				8	6	958	627.9	13	9	755	447.6
5	1	925	657.4	8	7	1050	521.5	13	11	1000	151.6
5	2	945	604.8								
5 5 5	1 2 3 4	979	650.0	9	3	721	263.7	14	12	1019	895.5
5	4	1029	374.5	9	4	759	001.2				
				9	5	809	481.9				
6	2	889	710.9	9 9 9 9	6	875	368.8				
6 6	3	921	940.3	9	7	959	569.0				
6	4	968	810.0	9	8	1065	868.2				
6	5	1032	323.0								

Rotation-Inversion	Transition	Frequencies

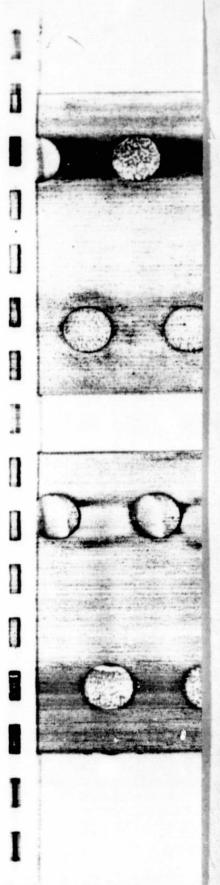
(J',K') - (J,K)	'invrot.
$a(0,0) - s(1,0)^{a,b}$	466 245.1
$s(2,1) - a(1,1)^{c}$	140 143.5
$s(3,2) - a(2,2)^a$	741 788.1
$s(3,1) - a(2,1)^a$	762 852.3
$s(3,0) - a(2,0)^a$	769 710.2
$a(3,3) - a(2,0)^d$	772 594.9
$s(3,0) - s(3,3)^{d}$	1070 166.6

- ^a Value obtained by extrapolating to zero pressure.
- b Ref. (29).
- c Ref. (30).
- ^d Forbidden (or perturbation-allowed) transition.

The precision of our ν_2 data would require the additional introduction of operators that connect levels with $\Delta k = \pm 3$ in the ground and ν_2 excited state. A detailed study of this interaction using submillimeter wave data on the inversion and inversion-rotation transitions in the ν_2 excited state will be published in a separate paper. Therefore in the present paper we have not included in our fit transitions from the ground-state levels with k'' = 3 or 0 to those ν_2 excited states that are perturbed by this interaction (cf. Table I).

VI. SPECTROSCOPIC PARAMETERS OF 14NH₃

In Tables VII to XII, spectroscopic parameters of ¹⁴NH₃ are presented that have been obtained by a damped least-squares fit (33) to the experimental data



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TABLE V

Ground-State Molecular Parameters of 14NH3 (in cm-1)

		8	٨	
B _o	9.9466529	± 0.0000004	9.9416356 ± 0.00	000004
B _o	(8.4721	± 0.0002) × 10 ⁻⁴	(8.3184 ± 0.0	002) × 10 ⁻⁴
DOK	(-15.6907	± 0.0007) × 10 ⁻⁴	(-15.2647 ± 0.0	007) × 10 ⁻⁴
H _{JJJ}	(2.298	\pm 0.001) \times 10 ⁻⁷	(2.071 ± 0.0	$01) \times 10^{-7}$
H _{JJK}	(-8.554	\pm 0.006) \times 10 ⁻⁷	(-7.604 ± 0.0	06) × 10 ⁻⁷
H _O	(11.49	± 0.02) × 10 ⁻⁷	(10.15 ± 0.0)	2) × 10 ⁻⁷
((a) c	o - (a)Bo] .	- [(s)C _o - (s)B _o]	(6.9989 ± 0.000	1) × 10 ⁻³
(a) D	C - (s) DC)		(2.9512 ± 0.000	2) × 10 ⁻⁵
[(a) H	KKK - (s) HC	KK]	(6.3031 ± 0.000	B) × 10 ⁻⁸

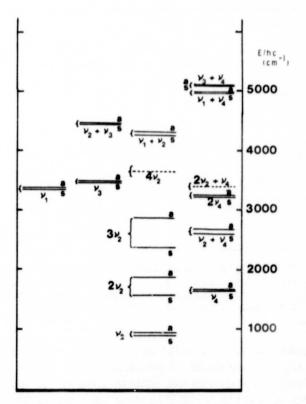


Fig. 4. The lowest vibration and inversion energy levels in NH₃.



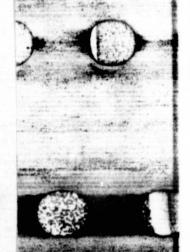


TABLE VI

Energy Matrix of the Coriolis and *l*-Type Interaction Between the ν_2 , $2\nu_2$, ν_4 , $\nu_2 + \nu_4$, and $3\nu_2$ States of NH_3^a

\v_2, \v_4*; J, k>	3v2 **	(v2 + v4)*	-t (v ₂ + v ₄) [‡]
3 [±] ,0°;J,k)	E3v2 (J,k,£4=0) *	(3 [±] a 1 [∓]) f (J, k+1)	-{3 [±] a 1 [∓] }f(J,k-1)
1 ⁺ ,1 ⁺¹ ;J,k+1)		$E_{v_2+v_4}(J,k+1,\ell_4=1)^{T}$	(1 [†] 8 1 [†])q(J,k)
1 ⁷ ,1 ⁻¹ ; J, k-1)			E _{v2+v} (J,k-1,£4=-1)
2 [±] ,0°;J,k)			
0 ⁷ ,1 ⁺¹ ;J,k+1)	Hermi	itian	
0 ⁺ ,1 ⁻¹ ;J,k-1)			
1 [±] ,0°;J,k)			

2v [±] ₂	+2 v+	- & ∨ ∓ 4	v ₂
0	(3 [±] α 0 [∓])f(J,k+1)	$-(3^{\pm} \alpha 0^{\mp})f(J,k-1)$	0
$\langle 1^{\mp} \alpha 2^{\pm} \rangle f(J,k+1)$	0	0	0
- (1 [‡] a 2 [±]) f (J, k-1)	0	0	0
E _{2\2} (J,k,£ ₄ =0) [±]	(2 [±] a 0 [∓] >f (J, k+1)	- \(2^\frac{1}{2} \alpha 0^\frac{1}{2} f (J, k-1)	0
	E _{V4} (J,k+1,£4=+1) +	<0 [₹] 8 0 [₹] >g(J,k)	$\langle 0^{\bar{t}} \alpha 1^{\pm} \rangle f (J,k+1)$
		E _{v,} (J,k-1, £4=-1) [‡]	$-(0^{\bar{i}} \alpha 1^{\pm}) f(J,k-1)$
			E _{v2} (J,k,£4=0) [±]

a $\alpha(\text{cm}^{-1}) = h^{3/2}[2(2\pi)^{5/2}c]^{-1}[X_{4a}^{*a}\partial/\partial\rho + \frac{1}{2}(\partial X_{4a}^{*a}/\partial\rho)]\lambda_{4}(\rho)^{-1/4}$, where the first operator $\partial/\partial\rho$ operates outside the square brackets while the second one operates only within the brackets; $\beta(\text{cm}^{-1}) = h^{2}[8(2\pi)^{3}c]^{-1}(Y_{4a}^{*a} - Y_{4a}^{*a})\lambda_{4}(\rho)^{-1/2}; f(J, k \pm 1) = +[J(J+1) - k(k \pm 1)]^{1/2}; g(J,k) = f(J, k + 1) \times f(J, k - 1); \langle 3^{\pm}|\alpha|1^{\pm}\rangle$ means $\langle v_{2} = 3^{\pm}|\alpha|v_{2} = 1^{\pm}\rangle$, etc.; and $E_{n}(J,k,l_{4})^{\pm} = E_{i}/hc$, where E_{i} is defined in Eq. (20).

on the ν_2 , $2\nu_2$, and ν_4 bands given in Tables I to III and on the data on the $\nu_2 + \nu_4$ and $3\nu_2$ bands (11). Because the fitted parameters are in general correlated, considerable care had to be taken to use suitable damping factors during the first cycles of iteration in order to reach the convergence region where the damping

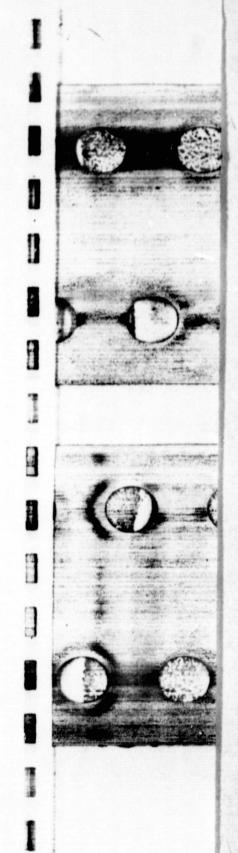


TABLE VII

Independent Linear Combinations of the Parameters of the v2 Band of 14NH3 (in cm-1)

Vo (8 + a)*	968.12224 # 0.00006	v _o (a + s)*	931.64155	t	0.00007
AB	10.21521 ± 0.00007	*B	10.50682	:	0.00003
(ac - aB) - (sco - sBo)	-0.33864 ± 0.00007	("C - "B) - ("C" - "B")	-0.65066		0.00003
spo - ap,	(0.655 ± 0.001)10-4	*D0 - *D3	(-2.325		0.002)10-4
SDO - ADJK	(1.875 ± 0.002)10-4	ADO - BDJK	(7.396		0.004)10-4
s _D ° - a _D _{JK}	(1.095 ± 0.001)10 ⁻⁴	apo - apk	(-5.500	:	0.002)10-4
AHJJJ - SHO	(-1.950 ± 0.004)10"7	SHJJJ - SHO	(-0.58		0.02)10-7
aH _{JJK} - BHOJJK	(6.50 ± 0.02)10 ⁻⁷	SHJJK - AHO	(0.54		0.04)10-7
aHJKK - SHO	(-7.59 ± 0.02)10 ⁻⁷	SHJKK - AHO	(-1.41	:	0.06)10-7
aHKKK - BHO	(3.14 ± 0.01)19-7	PHKKK - PHO	(1.46		0.03)10-7

Inversion splitting in the J = k = 0 v_2 state: 35.68726 ± 0.00020

factor could be put equal to zero. Calculations have been done with the CDC Cyber 172 computer using double-precision arithmetic (120 bits).

Because in the investigated system of interacting levels not k but $k-l_4$ is a good quantum number, the standard assignment of the transition frequencies to the ν_2 , $2\nu_2$, ν_4 , etc., bands is only approximate. This is especially true for the $s2\nu_2$, $a\nu_4$ levels because of the level crossing in this system. The only objective criterion on which such assignments can be based is to consider the coefficients of mixing of the wavefunctions in the system of interacting levels,

$$\psi_i = \sum_k c_{ik} \psi_{ik}^0, \qquad (21)$$

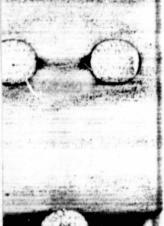
TABLE VIII

Independent Linear Combinations of the Parameters of the 2v2 Band of 14NH3 (in cm 1)

o (s - a) .	1882.1751 ± 0.0	001 v _o (a - s) •	1596.6753 :	0.0020
a _B	9.73747 ± 0.0	0009 SB	10.5676 ±	0.0002
(ac - aB) - (sc - sB)	0.15203 ± 0.0	0008 (SC - SB) - (AC - AB	-0.9084 ±	0.0002
Do - anj	(3.860 ± 0.0	06)10 ⁻⁴ a _D ° - s _D	(3.23 :	0.04)10-4
D _{JK} - AD _{JK}	(-10.225 ± 0.0	10)10-4 apo - spjk	(-6.56 ±	0.007)10
DK - aDK	(6.057 ± 0.0	07)10 ⁻⁴ a _D ° - s _D _K	(4.94 ±	0.06)10-4
H _{JJJ} - BHO	(-0.478 ± 0.00	05)10 ⁻⁶	(-0.227 ±	0.030)10
H _{JJK} - SHO	(1.868 ± 0.0)		(3.22 ±	0.08)10-6
H _{JKK} - SHO	(-4.179 ± 0.0		(-14.67 ±	0.20)10-6
HKKK - BHCKK	(4.596 ± 0.0		(13.87 ±	0.10)10-6

Inversion splitting in the J = $k = 0.2v_2$ state: 284.7064 ± 0.0020





^{*} The band origin ν_0 is defined as $\nu_0 = (E_F^0 - E_F^0)/hc$ [Eq. (16)].

^{*} The band origin ν_0 is defined as $\nu_0 = (E_F^0 - E_F^0)/hc$ [Eq. (16)].

TABLE IX

Independent Linear Combinations of the Parameters of the v₄ Band of ¹⁴NH₃ (in cm⁻¹)^a

	, n	A A, n - A
Vo - (nco - ngo) + 1(nng + nreg) - 2nrege	1629.99072 ± 0.00021	1630.3391 ± 0.0000
"C - "B - "C"C4 + ""K + "TEK - 4TEK	-2.43295 ± 0.00005	-2.41663 ± 0.00017
$^{1}C - ^{1}B - (^{1}C^{\circ} - ^{1}B^{\circ}) - ^{5}n_{KKK} + ^{3}2(^{1}n_{K} + ^{1}r_{KK})$	-0.160203 ± 0.000035	-0.15235 ± 0.00008
's	10.040597 ± 0.000027	10.02747 ± 0.00006
'p' - 'p _j	(-2.235 ± 0.006)10 ⁻⁴	(-0.871 ± 0.019)10-4
DO - DOK	(5.503 ± 0.014)10-4	(1.393 ± 0.044)10-4
DO - DK + STTKKK	(-2.706 ± 0.011)10-4	(2.580 ± 0.041)10-4
H _{JJJ} - ⁿ H _{JJJ}	(1.2014 ± 0.0040)10-6	(0.2486 ± 0.0018)10"
H _{JJK} - ⁿ H _{JJK}	(-3.685 ± 0.015)10-6	(-1.167 ± 0.007)10 ⁻⁵
H _{JKK} - ⁿ H ^o _{JKK}	(3.096 ± 0.025)10-6	(2.545 ± 0.012)10-5
HKKK - DHO	(-0.557 ± 0.013)10-6	(-1.753 ± 0.006)10 ⁻⁵
n _J	(-0.967 ± 0.038)10-4	(5.08 ± 0.17)10-4
, 11k	(1.1278 ± 0.0042)10 ⁻⁵	(0.2702 ± 0.0025)10"
TJKK	(-2.708 ± 0.006)10 ⁻⁵	(-1.3090 ± 0.0028)10"
$D_{K}^{O} + \frac{5}{6} T_{KKK} - \frac{1}{4} (^{n} \eta_{K} + ^{n} \tau_{KK})$	(1.0438 ± 0.0011)10 ⁻³	(1.0501 ± 0.0038)10
HO + 6 TRKK	$(2.3124 \pm 0.0030)10^{-6}$	(1.6732 ± 0.0015)10"
n _K - n _n (v ₂ +v ₄) + n _t - n _t (v ₂ +v ₄)	(-4.7168 ± 0.0046)10 ⁻³	(1.905 ± 0.016)10 ⁻³
THE THE TENT	0 b	0 b

Inversion sp_itting in the $J = k = 0 v_4$ state: 1.1488 ± 0.0010

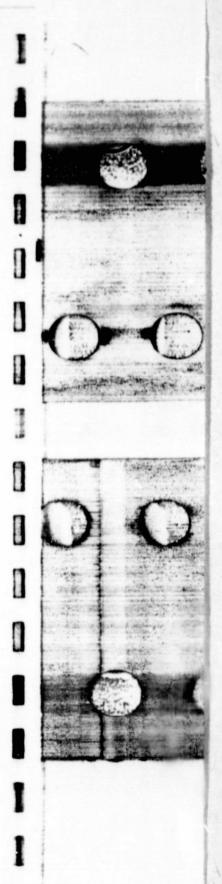
where ψ_{ik}^0 are the unperturbed wavefunctions. Our assignments in Tables I to III are essentially based on the principle of the maximum contribution of the unperturbed wavefunction ψ_{ik}^0 to the perturbed state. This is illustrated by Table XIII which, however, also shows that in some cases this principle may not work.

TABLE X

Off-Diagonal Matrix Elements of the Coriolis and I-Type Interaction^a

(3" a 1")	6.6934 ± 0.0018	(3 ⁺ a 1 ⁻)	-1.484	±	0.021
(3 a 0 +)	0.504 ± 0.011	(3+ a 0->	-5.9031	±	0.0010
(2" a 1")	5.2619 ± 0.0032	(2+ a 1-)	11.3684	±	0.0040
(2" a 0")	-2.194 ± 0.006	(2+ a 0-)	-1.2827	±	0.0025
(1- a 0+)	10.338 ± 0.031	(1 ⁺ a 0 ⁻)	11.461	±	0.009
(1 ⁺ 8 1 ⁺)	-0.244 ± 0.029	(1" 8 1")	0.00466	±	0.00041
<0+ 8 0+>	-0.00291 * 5.00011	(0 B 0)	-0.00701	±	0.00021

^{*} Matrix elements are defined in Table VI.



^{*} The band origin ν_0 is defined as $\nu_0 = (E_F^0 - E_F^0)/hc$ [Eq. (20)].

b Constrained value.

TABLE XI

Independent Linear Combinations of the Parameters of the 3v2 Band of 14NH3 (in cm-1)

v _o (s + a)*	2895.6063 ± 0.0012	v _o (a + s) •	2383.3804 + 0.0026
*B	8.91779 # 0.00026	*8	9.42827 ± 0.00031
(AC - AB) - (BCO - BBO)	1.10027 # 0.00018	(8c - 8B) - (8c0 - 8B0)	0.47510 + 0.00028
*D° - *DJ	(1.179 ± 0.042)10 ⁻³	*D° - *D3	(0.963 + 0.019)10-3
"DO - "DJK	(-3.165 ± 0.058)10 ⁻³	ADO - ADJK	(-2.528 ± 0.043)10 ⁻³
DO - DIK	(1.994 ± 0.038)10-3	*D° - *DK	(1.661 + 0.038)10-3
aH _{JJJ} - BH _{JJJ}	(-1.573 ± 0.049)10 ⁻⁵	SHJJJ - SHOJJJ	(-0.769 ± 0.026)10 ⁻⁶
"HJJK - "HJJK	(4.618 ± 0.059)10 ⁻⁵	SHJJK - SHOJJK	(2.269 + 0.061)10-6
AHJKK - SHOJKK	(-4.096 ± 0.058)10 ⁻⁵	SHJKK - AHOJKK	(-2.222 ± 0.038)10 ⁻⁶
aHKKK - BHC	(1.039 + 0.029)10-5	SHKKK - SHO	(0.737 ± 0.032)10-6

Inversion splitting in the $J = k = 0.3v_2$ state: 511.4325 ± 0.0040

For example, the 1855.468 cm⁻¹ and 1901.229 cm⁻¹ transitions should both be assigned as $a^PR(10,2)$, $\nu_4(-l)$. To avoid this, we assign the 1901.229 cm⁻¹ transition as $a^PR(10,2)$, $\nu_4(-l)$ because it has a larger contribution from the $|0^-,1^{-1};11,1\rangle$ state than the 1855.468 cm⁻¹ transition. The latter is then assigned as aR(10,2), $2\nu_2$ because its second-largest contribution is from the $|2^+,0^0;11,2\rangle$ state. The few such cases that occurred in our data are indicated in Tables II and III by special labels.

TABLE XII

Independent Linear Combinations of the Parameters of the ν_2 + ν_4 Band of ¹⁴NH₃ (in cm⁻¹)^a

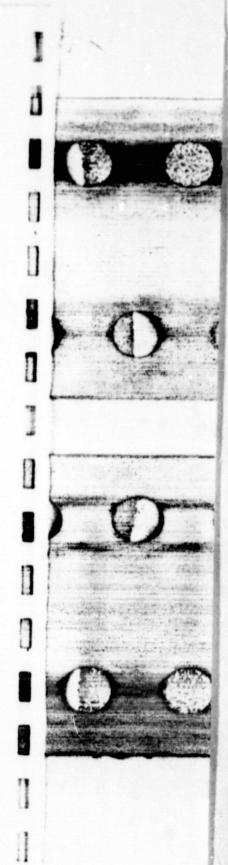
	s ++ 1	١,	n = s	a ↔ a,	n = a
$v_O - (^{\overline{n}} e^{O} - ^{\overline{n}} B^O) + \frac{1}{4} n_K$	2544.2587	±	0.0012	2588.9592 ±	0.0017
nc - nB - nent4 + nnK	-3.0029	±	0.0006	-2.4137 ±	0.0008
$^{n}C - ^{n}B - (^{n}C^{0} - ^{n}B^{0}) + \frac{3}{2}n_{K}$	-0.6471	ŧ	0.0006	-0.1187 ±	0.0006
n _B	10.38786	±	0.00021	9.97176 ±	0.00046
no - no	(5.302	±	0.011)10-4	(-1.759 ±	0.009)10-4
DO - DJK	(-4.539	±	0.007)10 4	(11.711 ±	0.047)10-4
nDO - nDK	(-3.314	±	0.021)10-4	(-15.881 ±	0.051)10-4
nny	(2.369	±	0.005)10-3	(-6.503 ±	0.009)10-3

Inversion splitting in J = k = 0 $v_2 + v_4$ state: 45.4993 ± 0.0030

[•] The band origin ν_0 is defined as $\nu_0 = (E_i^0 - E_i^0)/hc$ [Eq. (16)].

^{*} The band origin is defined as $\nu_0 = (E_i^0 - E_i^0)/hc$ [Eq. (20)]; sextic parameters have been neglected.

Calc. freg.				>	v2.v4";3,k)			
(cm ⁻¹)	Assignment	(3*,00;11,2)	13,00;11,2 1-,1+1;11,3 1-,1-1;11,1 2+,00;11,2 0-,1+1;11,3 0-,1-1;11,1 1+,00;11,2	(1,11,1-1,1)	12,00;11,2)	10-,1+1;11,3	(0,11-1,11,1)	11,00,111,
1158.060	aR(10,2),v2	0.0	0.0	0.0	0.0	3.3	3.3	93.4
1848.403	aR(10,2),v4(+E)	0.2	8.0	0.8	42.5	51.9	2.6	1.3
1855.468	aR(10,2),2v2	0.0	0.7	0.7	38.1	19.7	40.6	0.2
1901.229	a^R(10,2),v4(-t)	1.7	0.4	•.0	15.8	24.1	52.6	5.2
2557.356	aR(10,2),3v2	97.4	0.3	0.3	0.1	6.0	1.0	0.0
2797.258	aR(10,2), v2+v4 (+£)	0.1	73.9	95.8	0.2	0.0	0.0	0.0
2838.523	a R(10,2), v2+v4 (-E)	9.0	23.9	72.1	3.3	0.0	0.0	0.0



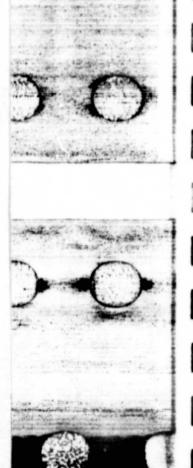
ν₂, 2ν₂, AND ν₄ STATES OF "NH₃

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It should be emphasized that such assignments are only formal if there is strong mixing of the upper-state wavefunctions. Strictly speaking, we should give the same label to a series of lines originating from the same ground inversion-rotational level J'', k'' and terminating on upper-state levels with the same J'', $k'' - l_4$ quantum numbers.

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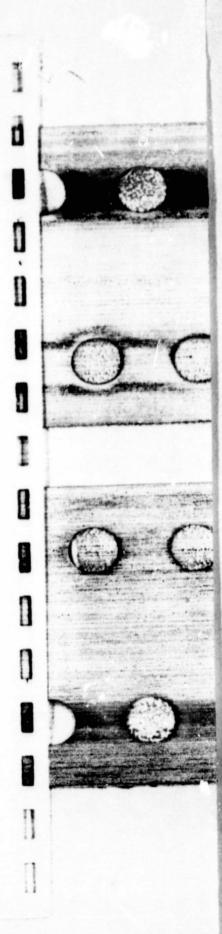


JOURNAL OF MOLECULAR SPECTROSCOPY 91, 273 (1982)

ERRATUM

Volume 79, No. 2, February (1980), in the article "Coriolis and I-Type Interactions in the ν_2 , $2\nu_2$, and ν_4 States of ¹⁴NH₃," by S. Urban et al., pp. 455-495:

The value of $\{^{(a)}D_k^a - {}^{(i)}D_k^a\}$ in Table V should be considered with the negative sign. The value of 'B in Table VII should be 10.44847 \pm 0.00003.



JOURNAL OF MOLECULAR SPECTROSCOPY 88, 293-299 (1981)

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The Diode Laser Spectrum of the v2 Band of 14ND3 and 15ND3

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The spectra of the ν_2 bands of ¹⁴ND₃ and ¹⁵ND₃ were measured under Doppler-limited resolution with a diode laser spectrometer in order to resolve the K structure of the sR(J,K) and aR(J,K) multiplets. By a simultaneous least-squares analysis of these data, the Fourier spectra of the ν_2 band measured by Jones [J. Mol. Spectrosc. 74, 409 (1979)], and the ground-state microwave transitions, sets of improved ν_2 -band parameters were obtained, including the sextic centrifugal distortion coefficients.

INTRODUCTION

In our previous papers (1-3) we have measured and analyzed the high-resolution spectra of $^{14}NH_3$ in the submillimeter wave, far-infrared, and medium-infrared spectral regions. The ultimate goal of these studies is the determination of a complete anharmonic potential function of ammonia by a least-squares fit to the experimental data of various isotopically substituted ammonia molecules [cf. Ref. (4)].

There are, however, very few high-resolution infrared data on the deuterated ammonia. We have therefore undertaken a study of the spectra of the ν_2 bands of $^{14}ND_3$ and $^{15}ND_3$ (cf. Fig. 1) under Doppler-limited resolution with a diode laser spectrometer.

The Fourier spectra of the ν_2 fundamentals of $^{14}ND_3$ and $^{15}ND_3$ were recently measured by Jones (5) with 0.055-cm^{-1} resolution with apodization or slightly less than $0.04~\text{cm}^{-1}$ without apodization. This has made it possible to resolve many of the P, Q, and R lines and to determine the ν_2 band parameters including the quartic centrifugal distortion coefficients. Among the transitions which remained unresolved in the Fourier spectra were especially the sR(J,K) and aR(J,K) transitions which appear as multiplets within the $0.2-0.3~\text{cm}^{-1}$ interval.

In the present paper, we have resolved the K structure of these multiplets using a diode laser spectrometer operating between 818 and 852 cm⁻¹. In a simultaneous least-squares fit to the ν_2 -band transition wavenumbers of Jones (5), the ground-

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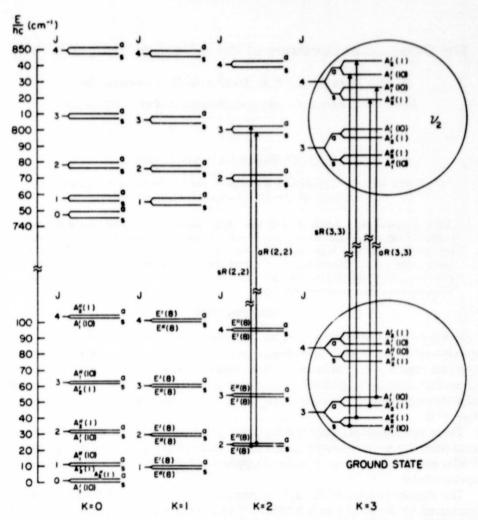


Fig. 1. Energy level diagram of the inversion-rotation levels in the ground and ν_2 state of ND₃ with examples of the sR(J, K) and aR(J, K) transitions. The inversion splittings and the A_1 , A_2 splittings are not drawn to scale. Values in parentheses behind the symbol of the symmetry species denote the spin statistical weights. The species of the energy levels in the ν_2 state are the same as in the ground state. The symmetry species of the K=4, 5, 6 levels are the same as for the K=1, 2, 3 levels, respectively, except that ' should be changed to " and vice versa (the rotational quantum number K should be considered with modulo 6 for obtaining the species of energy levels with higher K values).

state microwave frequencies of the inversion-rotation and inversion transitions (6-8), and our diode laser measurements, we arrived at an improved set of spectroscopic parameters for $^{14}ND_3$ and $^{15}ND_3$ including the sextic centrifugal distortion coefficients.

EXPERIMENTAL DETAILS

The diode laser spectrometer has been described earlier (9, 10). The ν_1 band lines

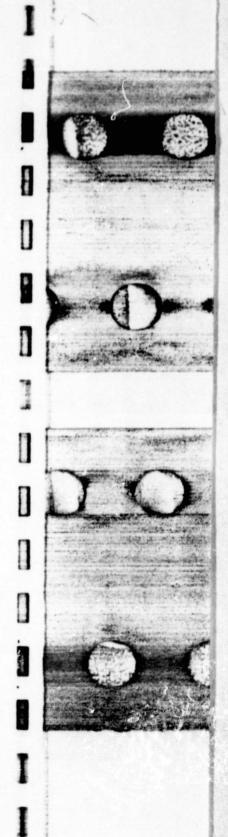


TABLE I

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Calculated and Observed Wavenumbers (cm-1) of the v2 Band of 14ND3

3,K	Veels	· obe	Feels	Vobe .	3,K	cels	V obs	reals	Veb
	-	(B,K)		(8,K)		-	(B,K)		(0,K)
0.0	019,2100	.2102	816, 1535		0.0	848,0419	.0416	845, 4256	. 424
.1	010,2144	.2141	816, 1444		1.1	843,0470	.0473	845,4193	.410
	015,2277	.2272	016, 1160	1 1	- 23	848.0623	.0633	645, 4003	. 397
			(816,0701			[848,0878		/845,3682	
	1017,2502	.2499	1616,0703		9.3*	1848,0006	.0891	1845, 3674	. 366
	019,2826	.2820	616,0029			040, 1251	. 1279	845,3206	. 321
	019, 3255	. 3255	815,9130			040, 1730	1772	845,2572	.257
	019,3000	.3796	815,7979		- 1	040, 9352	. 9359	845,1750	.173
••			****		4.5	848, 3106	.3120	845,071)	.000
		(7.K)		1(7,K)		848, 4014	. 3996	844, 9416	. 941
	was a manage		was an armed a		,	848, 5093	. 5000	844, 7815	. 782
.1	926, 9060	. 9072	825,9883	. 9664	*,*		. 2000		
•	626,9107	. 9101	825,9802	9553		-	(10.K)		(10.K)
	828, 9248	. 9246	825,9555	. 9553		Carrier Committee	(40,0)		(*** **)
.30	828,9485	. 9401	(825, 9135	.0112	10,0	057,4787		055,0196	
	1028,9408		1025,9133		10,1	057,484		055,0142	
	828,9628	. 9821	825.8527	. 8505	10,2	857,499/		054,9981	
	629,0281	.0272	825,7715	.7701	10,3	1857,5270		J054, 9697	
.0	629,0854	.0839	825,6672	. 6665		1057,5256		1854,9710	
.7	629,1560	. 1553	825, 5365	.5357	10,4	057,5642		054, 9301	
-					10,5	057,6141		854,8756	
- 1	•R(0,K)		R(8,K)	10,6	857,6770		854, 8047	
0.0	836,5172	.5176	235,7471	.7478	10,7	057,7540		054,7144	.716
	630,5221	. 5219	835,7399	.7412	10.8	857,8465		054,6012	. 600
.1	030,5360	. 5370	835,7100	.7186	10,9	857,9563		854,4602	. 459
	(630, 5620		(835,6805		10,10	858,0852		854. 2858	. 205
.30	1838.5616	. 5618	1835, 6809	, 6606					
1,4	838,5975	. 5972	835,6266	. 62 66					
. 5	838.6447	. 6440	835,5544	. 5537					
.6	836,7043	7035	835,4611	4603					
7	838,7775	.7758	835, 3437	,3433					
	838,0660	. 0664	835, 1980	.1970					
	,		, 1900						

of OCS (11) were used for calibration purposes. Both $^{14}ND_3$ and $^{15}ND_3$ were obtained from Prochem Ltd. and were stated to be 98% pure. These gases were contained in 1-m-long Pyrex cells with KBr windows. The sample pressures ranged from 0.3 to 0.5 Torr. The relative uncertainty of the observed wavenumbers is about ± 0.0005 cm $^{-1}$, while the absolute uncertainty due mainly to OCS calibration is about ± 0.002 cm $^{-1}$.

The observed wavenumbers of the R(J, K) multiplets obtained in this work are given in Tables I and II for $^{14}ND_3$ and $^{15}ND_3$, respectively. Figure 2 illustrates the resolution of the sR(6, K) multiplet in the ν_2 band of $^{14}ND_3$. The corresponding FTS scan at 0.04-cm⁻¹ resolution is also shown in this figure.

RESULTS

The experimental data which have been processed in a simultaneous least-squares fit for $^{14}ND_3$ were the transition wavenumbers in the ν_2 band measured

3,K	'.elc	Vobe.	Peale	Vobe	3,K	celc	Pobe.	Vcelc.	Pobe
		(7,K)		R(7,K)		- 0.5	(9,K)	•8	(9,K)
7.0	822, 1094		819,7656	.7651	9.0	841,5567	. 5594	839,1983	. 1989
7.1	622, 1117		819,7559	.7557	9,1	841,5596		839, 1905	. 1906
7.0	822,4187		619,7267	.7267	9.0 9.1 9.2	841,5684	. 5680	839, 1669	. 1675
7,30	1022,43.5		(019.6773	6775	9,3	(841.5823	. 5824	(839, 1273	. 126
	1022,430b		1819.6771			1841.5836		1839, 1265	
7.4	822,4482		819,6061	, 6062	9,4	841,8044	. 6035	839,0694	.0692
7.6	022,4721		819,5114	.5107	0.5	841,6325	.6326	836,9927	. 994
7.6	022,5035		819,3906	.3697	9,6	841,5686	. 6689	838, 8944	. 8950
7.7	822,5440		619,2403	.2393	9,7	841,7136	.7141	838,7717	.7700
					9,0	841,7695	.7697	630,6206	.6190
	98	(8,K)		R(8,K)	0,0	841,8384	.8372	636, 4361	, 4360
.1	832,0249	.0247	829,5200	.5196			(10,K)		(10,K)
0,1	832,0275		829,5113	.5109			(** '*')	The same of the sa	(*** ***)
	832,0354	.0345	829, 4850	.4844	10,0	951,0019	.0060	848,7970	(.790
0,3	[832,0490	.0487	1829,4400	,4396	10,1	851,0051		846,7901	
	1832,0485		829,4405		10,2	851.0146	.0166	848,7693	,7689
. 6	832,0681	.0680	829.375	.3754	10,3	(051,0318	.0320	(848,7332	.7326
	832,0941	.0946	629,7903				*	1048,7345	
	632, 1277	.1290	629,1608	.1782	10,4	851.0545	.0543	848,6829	. 682
	832,1705	.1709	829,0443	.0464	10,5	051,0051	.0845	848,6148	. 6140
•,•	632,2243	.2227	829, 8767	.0767	10,6	051,1237	. 1230	848,5276	. 526
					10,7	051.1713	.1707	848,4184	.410
	1		1		10,0	051,2294	.2267	848,2837	,282
	ł		i		10,9	851,3000	, 2900	848,1186	. 1179
					10,10	051,3850	.3856	847,9175	.9166

[·] Upper value of Vanta corresponds to the more intense line in the A. A. double





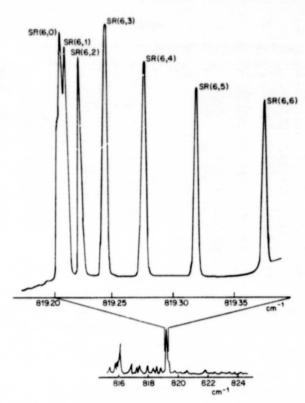
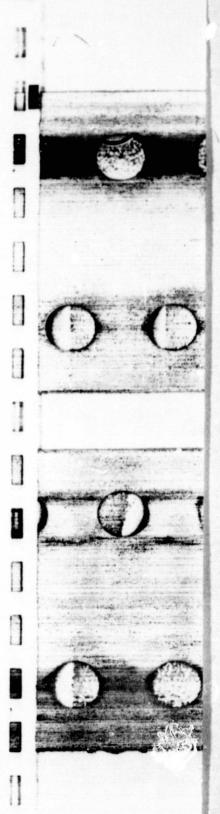


Fig. 2. Resolution of the sR(6, K) multiplet in the ν_2 band of ¹⁴ND₃ along with the FTS scan by Jones (5).

by us with the diode laser spectrometer and by Jones with a Fourier spectrometer (5), the ground-state inversion frequencies (6), and the ground-state $J=1\leftarrow 0$ and $2\leftarrow 1$ inversion-rotation transition frequencies (7, 8). For $^{15}ND_3$ only the inversion-rotation transition frequencies $J=1\leftarrow 0$ and $2\leftarrow 1$ are available (7, 8) and these have been combined with the infrared data. A more detailed discussion of the simultaneous analysis of the infrared and microwave data of ammonia is given in Ref. (3).

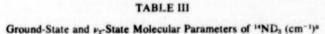
The statistical weights assigned to the experimental data were $w_j = \delta_j^{-2}$, where δ_j is the estimated uncertainty of the measurement. We have used $\delta = 2 \times 10^{-3}$ cm⁻¹ for our diode laser measurements, $\delta = 5 \times 10^{-3}$ cm⁻¹ for the data of Jones (5) (except for few blended lines which have not been considered in the fit), and $\delta = 0.05$ MHz for the microwave inversion frequencies (6).

We have used a standard polynomial expression for the inversion-rotation energy levels in the ground and ν_2 excited states including the sextic centrifugal distortion constants. The A_1 , A_2 splitting of the K=3 levels was described by the parameter η_3 in the matrix element of the corresponding interaction (cf. Ref. (3)),



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OF POOR QUALITY



	Ground-state	V2-010
(*) _E (J=K=0)	0	745.5997(7)b
(•) _B	5,142763(2)	5, 12251(5)
(e) _{D3} ×10 ⁴	1,983(3)	1.964(9)
(e)D_XX104	-3,501(9)	-3,685(20)
(•) D _{3K} ×10 ⁴ (•) H ₃₃₃ ×10 ⁶ (•) H ₃₃₃ ×10 ⁶	2,39(15)	2.11(48)
Many XIO	-5,69(63)	-8.7(17)
(•)H _{DKK} ×10 ⁸	4.9(11)	12.1(20)
Inversion point ting \(\alpha_{inv} \) \[\{ \alpha_{i} \\ \alpha_{i}	0.0530882(3) -1.7337(12)×10 ⁻⁴ 2.6138(14)×10 ⁻⁴ -2.14(17)×10 ⁻⁷ 6.54(43)×10 ⁻⁷ -4.66(28)×10 ⁻⁷ 0.36(7)×10 ⁻¹⁰ -1.54(30)×10 ⁻¹⁰ 2.25(40)×10 ⁻¹⁰ -1.11(20)×10 ⁻¹⁰ 7.14(6)	3.5484(3) -1.047(5)×10 ⁻² 1.567(10)×10 ⁻⁵ -1.62(10)×10 ⁻⁵ 4.64(26)×10 ⁻⁵ -3.41(15)×10 ⁻⁵ -1.26(52)×10 ⁻⁸ 5.6(19)×10 ⁻⁸ -8.8(24)×10 ⁻⁸ 4.5(11)×10 ⁻⁸ (7.14) ⁶
(•)c - (•)B	-0.014	61(7)
(•) D(×2) - (•) DO	2.49(12)×10-5
(*)H(N) - (*)HO		78)×10 ⁻⁸

Values in parentheses are standard deviations of the parameters, given in units of the last digit quoted; s and a denote respectively the lower and upper component of the inversion doublet.

Band origin for the s \rightarrow s transition is 749,1481(8), for the s \rightarrow s transition 745,5466(7)cm⁻¹. Constrained value.

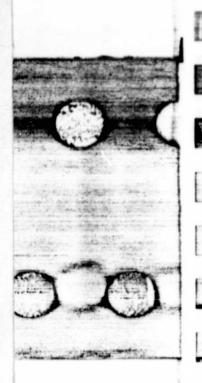
$$\frac{1}{2}\left(\left(\frac{s}{a}\right)\middle|\left\{\langle J, +3\right| \pm \langle J, -3\middle|\left\}(H'/hc)\middle|\left(\frac{s}{a}\right)\right\}\left[\left|J + 3\right\rangle \pm \left|J - 3\right\rangle\right]\right]$$

$$= \pm \eta_3(J + 3)!/(J - 3)!, \quad (1)$$

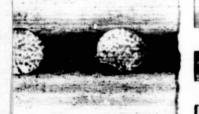
where s and a denote the parity of the corresponding inversion wavefunctions.

The calculated wavenumbers are compared with the measured wavenumbers for the R(J,K) multiplets in Tables I and II; in Tables III and IV the values of the corresponding molecular parameters obtained for $^{14}ND_3$ and $^{15}ND_3$ by the least-squares fit are presented. The microwave data (6-8) and the Fourier data (5) have been described within their experimental accuracy.

The ground-state parameter η_3 [Eq. (1)] was determined for ¹⁴ND₃ from the ground-state inversion frequencies (6). This parameter (Table III) was assumed to have the same value in the ν_2 state of ¹⁴ND₃ and in the upper and lower states of ¹⁵ND₃. The magnitude of the splitting of the R(J, 3) lines is given by the difference between the value of the splitting of the upper and lower state K = 3 energy







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TABLE IV Ground-State and ν_2 -State Molecular Parameters of $^{15}ND_3$ (cm $^{-1}$)^a

	Ground-state	V2-state
(*)E(J-K-0)	0	739,5343(9) ^b
(*) _B	5,129658(4)	5, 10554(4)
(•) _{D-×10} 4	1.977(5)	1,986(6)
DJK*104	-3,500(18)	-3,733(18)
(•)H333×108	2,1(3)	2,75(25)
H33K×10	-5.9(9)	-10.2(10)
(•)HJKK×10 ⁶	6.9(14)	13,3(15)
Inversion	G.047791(5)	3,2454(18)
oplitting Aim	-1.60(6)×10-4	-1.008(7)×10-2
(0)c-(0)8] - [(0)c-(0)8	2.23(11)×10-4	1.499(8)×10-2
(0)0 (0)0	0.05(70)=10=7	1.495(0)×10"5
(*) _{D3} -(*) _{D3} (*) _{D3K} -(*) _{D3K} (*) _{DK} -(*) _{D3K} (*) _K -(*) _{M333}	0.95(70)×10-7	-1.62(12)×10"5
JK(a) JK	-25.2(32)×10 ⁻⁷	5.18(25)×10-5
K (SK	0-	-3.32(12)×10
(a) Hooo - (a) Hooo	0.37(16)×10 ⁻¹⁰	-2,52(42)×10-6
Hanne Hann	-3.0(8)×10 ⁻¹⁰	8.0(14)×10-8
(a) HJKK (a) HJKK	4.1(9)×10 ⁻¹⁰	-6.6(21)×10-8
(a)HKKK-10HKKK		oe .
73 × 1010	(7.14)°	(7.34)°
[(•)c(×)-(•)B(×)] -		
-[(*)c - (*)B]	-0.01	632(5)
(*)DK(*1) (*)DO	2,44	(7)×10 ⁻⁵
(*)HKKK - (*)HO		5)×10 ⁻⁸

[•] See footnote in Table III. b Band origin for the $s \rightarrow a$ transition is 742,7797(18), for the $s \rightarrow a$ transition is 739,4865(9) cm⁻¹. c Constrained value.

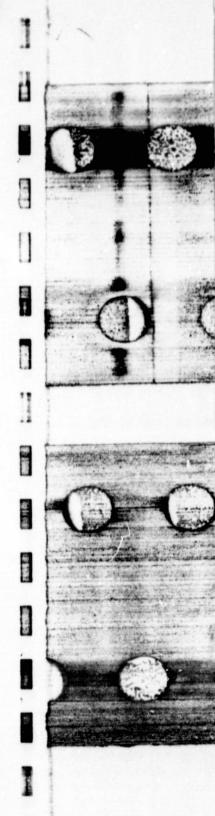
levels; due to the spin statistical weights, the ratio of the intensities of the split lines should be 1:10 (see also Fig. 1).

It can be seen from the calculated values of the R(J, 3) transition wavenumbers (Tables I and II) that for $J \ge 10$ the splittings might be observable with the 0.002-cm^{-1} resolution of the diode laser spectrometer. We have observed shoulders

TABLE V

Comparison of Ground-State Constants (in cm⁻¹) Obtained from Vibration-Rotation Data and Pure Rotation Data

		14 _{NO3}			15 _{NO3}	
	Ref.7	Ref.5	This work	Ref. 7	Ref,5	This work
8,	5.14266(1)	5,14246(8)	5,142677(2)	5,12357(1)	5,12330(11)	5,123578(8)
05×104	1.97(4)	1,930(3)	1.982(3)	1.98(4)	1,933(5)	1.978(5)
00K×104	-3,50(2)	-3.407(6)	-3,498(10)	-3,52(1)	-3,434(5)	-3,513(10)



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in the R(10, 3) lines which probably correspond to this effect but this could be confirmed only by measuring transitions with $J \ge 10$.

In Table V, the ground-state constants of $^{14}ND_3$ and $^{15}ND_3$ obtained in our work are compared with those obtained previously by Helminger and Gordy (7) and Jones (5). Our values are much closer to the values obtained from pure rotation data (7) than those of Jones (5). The reason of the inconsistency in Ref. (5) is obvious: Jones (5) assumed that $^{(8)}B_0 = ^{(8)}B_0$ in his simultaneous treatment of the observed rotational and vibrational—rotational transition frequencies which is not adequate to the accuracy of the experimental data.

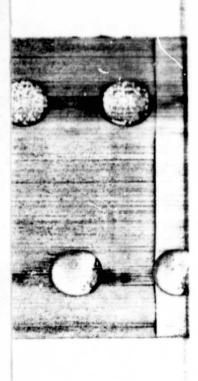
ACKNOWLEDGMENT

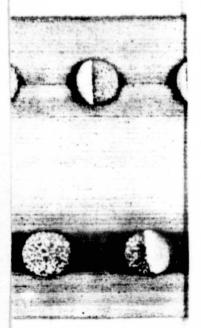
This work was done during the tenure of a grant from the National Aeronautics and Space Administration. One of us (K.N.R.) acknowledges this support gratefully.

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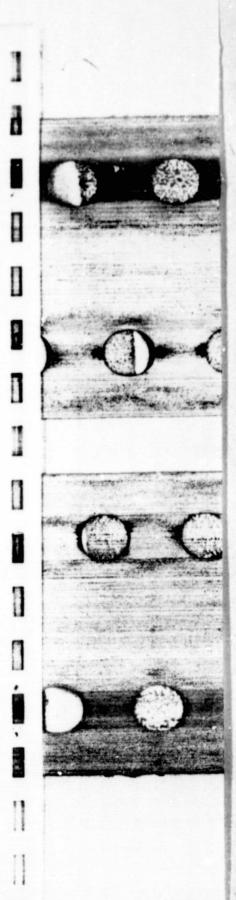


JOURNAL OF MOLECULAR SPECTROSCOPY 92, 282 (1982)

ERRATUM

Volume 88, No. 2 (1981), in the article "The Diode Laser Spectrum of the P2 Band of 14ND3 and 15ND3" by V. Malathy Devi, P. P. Das, K. Narahari Rao, \$. Urban, D. Papoušek, and V. Špirko, pp. 293-299:

In Tables III and IV, the ground state values of the differences $^{(a)}H - ^{(s)}H$ should be with the factor 10^{-8} instead of 10^{-10} .



JOURNAL OF MOLECULAR SPECTROSCOPY 72, 135-142 (1978)

Interpretation of the 13C16O2 Spectrum at 4.4 µm

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The overlapping structure observed in the region of the ν_2 fundamental of $^{13}C^{14}O_2$ at 4.4 μ m has been assigned to several transitions belonging to not only the 13-C variety of carbon dioxide ($^{12}C^{14}O_2$) but also to $^{12}C^{14}O^{12}O$ and $^{12}C^{14}O^{14}O$ species occurring in the sample. Molecular constants have been evaluated for the assigned transitions.

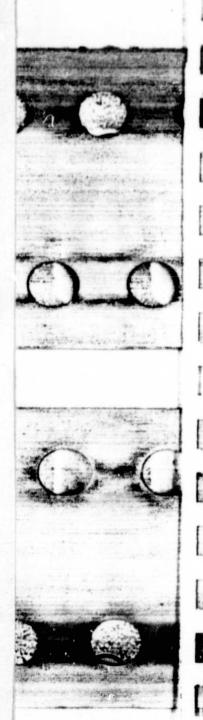
In a recent note (1) we reported the measurements and analysis of the ν_2 band of $^{13}C^{16}O_2$ recorded with a high resolution vacuum infrared spectrograph by using an enriched sample of this isotopic species. We now present the interpretation of most of the structure observed in the region where this ν_2 band occurs. The data were obtained in the spectral interval 2173 to 2334 cm⁻¹ and the experimental conditions have already been described before (1). In all, 12 bands have been observed in the region, 10 belonging to $^{13}C^{16}O_2$ and the remaining two are the ν_3 fundamentals of $^{13}C^{16}O^{17}O$ and $^{13}C^{16}O^{18}O$. A few specific details are given below for some of the data pertaining to the current investigation.

Σ+u(e) - Σ+e(u) Bands of 13C16O2

Three bands belonging to this category were observed and the observational data are presented in Table I. For these three Σ - Σ bands, the lower states of the transitions observed are the levels 10°0, 02°0, and 00°1. Since the rotational constants for these lower levels are known with high accuracy from the laser studies (2), it was considered more appropriate to make use of these values and then determine the upper state constants of the three Σ - Σ bands studied here. Table V gives the rotational constants so obtained along with the band centers ($\nu_0 - B'U'^2 + B''U''^2$).

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TABLE I Measured Line Positions (cm⁻¹-vac.) of $\Sigma^+_{u(\varphi)} \leftarrow \Sigma^+_{2(u)}$ Bands of $^{12}C^{16}O_2$

1001	- 10°0	0201 -	020	00°2 -	0001
P OBS	# 08S	P OBS	A nes	P Des	A Des
			2262.6882		
	2265.1504	2260.3443	2264.2208		
2259.6930	2266.6552	2258.7448	2265.7330		
2258.0827	2266.1368	2257.1324	2267.2214		2264.5830
2256.4458	2269.5931	2255.4958	2268.6834	2254.5210	2266.0499
2254.7857	2271.0245	2253.8325	2270.1329	2252.6826	2267.4850
2253.1023	2272.4382	2252-1482	2271.5525	2251.2154	2268.8955
2251.3904		2250.4424	2272.4466		
	2275.1802	2248.7130	2274.3175	2247.8226	
2247.9041	7276.5167	2246.9584	2275.6707		2272.9895
2246.1257	2277.8282	2245.1822	2276.9938		
2244.3243	2279.1168	2243.3856	2276.2936	2242.5589	
2242.4974	2280.3788	2241.5642	2279.5751	2240.7573	2276.8691
2240.6449	2281.6196	2239.7209		2238.9319	227A.1153
2236.7695	2282.8335	2237.8578		2237.0818	2279.3335
	2284.0256	2235.9696	2263.2710	2235.2059	2280.5324
	2285.1955		2284.4578	2233.3126	2281.7041
2233.0044	2286.3360		2285.6199		
2231.0376			2286.7587	2229.4501	2283.9801
2229.0474	2288.5493		2287.8742		2785.0793
2227.0277	2289.0188	2226.1895	2289.9634		
2224.9929		2224.1623	2290.0343	2223.4827	2287.2051
2222.9303		2222.1175	2291.0779	2221.4506	
2220.8413	2292.6798	2220.0484	2292.0995		
2218.7323				2217.3136	
2216.6049	2294.5994	2215.8490	2294.6721		2291.1751
2214.4485	2295.5236	2213.7133	2295.0220		
2212.2648	2296.4249	2211.5552		2210.9303	
2210.0636	2297.2958	2209.3756	2296.8490		
2207.8385		2267.1719	2297.7255	2206.5606	
2205.5902	2298.9724	2204.9464	2298.5757	2264.3376	
2203.3198		2202.7012	2299.4049		
2201.0233		2200.4375		2199.5360	
2198.7071					
2196.3642	2302.0218	2195.8352			
2193.9991			2302.4887		
2191.6079		2191.1432	2303.2035		
2189.2023	2304.0539	2186.7725			
2186.7630	-30 100031	2186.3729			
		2183.9498			



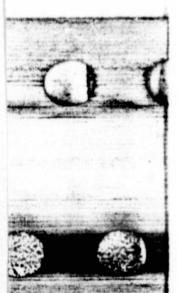
ISOTOPIC BANDS OF CARBON DIOXIDE AT 4.4 µm

TABLE II $\label{eq:table_energy} Measured Line Positions (cm^{-1}-vac.) of II_o \leftarrow II_u \ Bands of {}^{13}C^{16}O_2$

	***	1+	
• 085	# 085 ·	P 085	• 055
	2273.3056	2276.1900	2274.0727
2269.199	5 2274.0203	2266.5915	2275.5850
2267.796	2270.3264	2200.9755	********
2266.168	5 2211.7995	2265.3371	2278.5364
2204.015	9 2279.2501	2263.6085	2279.9755
2262.645	5 2284.6754	2261.9819	2281.3931
2761.150		2260.2705	2202.7858
	2263.4580	2250.5342	2284.1544
2257.683	2 2284.8136	2256.7772	2265.5000
2255.916	1 2286.1478	2254.9939	2286.8227
2254.128	8 2287.4555	2253.1903	2288.1212
2252.315	3 2288.7442		2289.3928
2250.480	6 2290.0067	2251.3585	2290.6457
22-9.621	9 2291.2413	2247.6315	2241.0712
2240.144	2 2242.4562	2249.7317	**********
2244.831	5 2293.0089		****
2242.905	0 2294.8121	2243.6121	2295.4048
2240.453	5 2795.9525	2239.8976	2296.5354
2234.476	8 2297.0744	2237.9074	2297.0401
2236.981	3 2298.1682		2296.7231
2234.963	9 2299.2380	2235.8934	2240.7231
2232.414	0 2300.28-5	2233.8542	2360.9127
2230.052	7 2301.3073		2301.8230
2224.764	9	2229.7117	2307.9092
2220.053	5 2301.2631	2227.6653	2303.7699
2224.517	5 2354.2343	2225.4744	7303.7644
2222.305	7 2305.1601	2223.322-	****
2220.15-	4 230c.0e35	2221.1504	2305.6208
	2326.9399	2218.9525	2366.5061
2215.757	. 2307. PS3	2216.7325	2367.3728
2213.511	2308.6237	2214.4664	
2211.239	8 2304.4334	2212.2211	****
2208.943	9 2316.2152		2309.6177
2200.032	9 2311.9698	2207.6176	2310.5921
2204.290	- 2311.7032	2205.2678	****
2201.933	7	2202.9265	2112.0-29
2194.550	8 2313.1003	2200.5463	2312.7346
	2313.7579	2198.1434	****
2194.717	5	2195.7187	2314.0476
2192.271	0 2315.0105	2193.2720	2314.6700
2189.801		2190.6021	****
		2166.3136	2315.#365
2164.788		2185.8106	
		2183.2628	

(continued)





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TABLE II-Continued

	11'1	- 11'0				- 03'0		
		,	-1		-	1-1		
P ves	× 005	P 085	# 085	P nes	4 085	+ 00S		
2246.6479	2250.0307	2249.0355 2247.4446 2245.8268 2242.5172	2252.4145	7743.4680 2741.8013	2252.2535	2747.5224 2744.2548 2742.5941	2255.0065 2256.0172 2256.025	
2241.6991 2240.0065 2238.2927 2236.5499	226C.9141 2262.2923 2263.6449	2239.1107	2261.6152	2230.4105	2761.0969	2237.459	2261.75	
£234.7883	2700.2854	2235.6227	2205.0484	2233.2090	2265.2562	2233.4207	2264.4927 2265.8266 2267.1314	
2229.3585	2200.0747	7220.0774 7220.0794 2224.0728	2270.6670	2227,0079	2270.6261	2220.4507	2270.0147	
2223.7206 2221.7471 2214.4486	3276383 22747516	2222.0734	2273.0520	2722.6907		2222.7718		
2217.8816	2276.0639	2212.6666	2276.4353			2212.0000	2276.768	
2209.7725	2261.1358	2210.6192 2208.5340 2206.4222	2780.5966 2281.5755			7210.8069 2708.7364	2280. 9934	
2/01.4454 2701.2494 2194.1204	2283.0666	2202.1331	2284.3794			2202.3842	2763.4301	
2190.4235		2197.7570				2193.6062		
2147.9184		2198.7180				2164.0411 2184.802C		

Π_a ← Π_a Bands

Table II gives the observational data for the three II-II bands measured in the present investigation. Two of the bands, namely the ones from the 11¹⁰ and 03¹⁰ have been identified for the first time, whereas in the case of the third band, 01¹¹-01¹⁰, the rotational structure has been observed to J numbers about 30 higher than what was recorded in the emission work of Steiner, et al. (3). The band constants have been evaluated by using standard least squares programs and are included in Table V.

$\Delta_u \leftarrow \Delta_u$, $\Phi_u \leftarrow \Phi_u$ and $\Gamma_u \leftarrow \Gamma_u$ Bands

Again, in the case of the $v_3 + 2v_2^2/2v_2^2$ band we have observed the rotational structure to very high J values. In fact, it has been possible to observe the rotational structure in

ISOTOPIC BANDS OF CARBON DIOXIDE AT 4.4 µm

TABLE III

Measured Line Positions (cm⁻¹-vac.) of Some Other Bands of ¹³C¹⁶O₂

	02'1	- 02'0		63'1 -	03 0	64'1 .	0.0	0."1 -	***
			-1		1,				
* 1185	. 005	+ 0+5	4 085	* 645	* 005	* 0*5	. 005	. 0*5	
	2202.303v	*******	*********		***********	***********			
2750.0020					2252.1954				
2255.2004	2205.3050	****	2200.0708		\$525:4225	5512:2144	33:2:53:2		
		2254,4428	2700.1030	55:1:2:2:					
2251.4537	2268.2808	2252.7933	2207.5014		131.301		2747.5417		77mm . 2151
2250.2042	2209.0005	2251105	2768.4417		1127.3320		2240.9731	2236-0120	
2244.5542	2271.0986	2244.4 (0)	2270.400	3339:3725		3335:9235		3550-0140	\$5:7:3262
	2272.4701		2271.7643		7766.8687	3330.0007	121:32		2/44.1632
2245.6409	227 5. 01 01	2245.944			2707.1010		3757:4246	3331:9925	774: . 474
2243.2788	2275.1489	2244.1769	2774.4884	3337:3977				3334:3341	
2241.4752	2270.444	22-7.1021	2275.7000	3330.4734	}}::::::/	>>>1,0100	\$127: Can		
2234.6527	2277.7247	2240.5657			2700.0000		127:5201	\$\$\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	
2237.8051	2274.9631	7736.7311	2270. 1500	3334:9771			2254.2780	33 12:2:2:	
2235.4266	2240.2124	2230.007	2274.5003	******	\$1:4:1:4:);	7716.1797		111111111	
	2201.4220		2700.0744	3331:5345	7764.0004	5511:5212			
2232.1105		22,3.0745	2702.0179	2220.4054	2774.9963	3312:2141			
2236.1001	2283. 1009	2231.1-50	2203.1000					2007 - 71-07 2004 - 2071 2004 - 2071	
	2209050	2274.1417	2/44.3363	13 13 14 00 00 00 00 00 00 00 00 00 00 00 00 00	\$545:4343	2700.002		2202.2073	
2224./224		2227.2150	2205.4029	35 (2:2160	5544:2521	1205.5034		2203.6104	
2224.2166		2225.2172	2200.50fc		\$372:2727			5521:2:21	
2222.1754	2200.1750	2223.1954	2207.0010	## :: 1 · 1		3363::008			
2220.1220	2704.2100		2200.0986			3122:1202			
7718.0445		2714.0007	2204.7250	2267.3436	2771.1031	7100.1000			
2215.9001		2216.0477	2246.7316		3,12:1731				
2213.0199	2747.7084	2214.0030			3340:1131				
2211.6769		2212.7500		2761.6473					
			2243.0675	2141.4749					
2207.3152	2294.9764	2260.0100		} :::::::					
		2206.2131		2143.4374					
2202.8621	2190.7078	2203.4859	7740.7008						
2200.0100	2297.5296	2701.7300		31:1:25:1:					
7198.3301	2294.3352	7100.0677	2241.4114						
2140.0250	2200.1205	2197.1609							
2143.4444		2194.0010							
	2300.0121								
21 60. 9856		2140.1711							
2180.5958		2187.7000	2361.6762						
2104.1016		\$10. '3050							
			2362.4444						
		118: .5146							

both the P and R branches to P(79) and R(77) as compared to P(39) and R(39) of the emission data (3). The data for this band as well as for the Φ - Φ and Γ - Γ bands identified in this region are included in Table III. The Q branches belonging to the $\nu_3 + 2\nu_2^2 - 2\nu_2^2$ and $\nu_3 + 3\nu_2^3 - 3\nu_2^3$ transitions could be distinguished without any ambiguity. However, in the case of the $\nu_3 + 4\nu_2^2 - 4\nu_2^2$ and $\nu_3 + 4\nu_2^4 - 4\nu_2^4$ bands, it was somewhat difficult to locate the band centers; therefore, the assignments had to be based upon calculated values for the band centers employing the molecular constants available in the literature (4, 5). Thus, the J assignments for these bands should be regarded as tentative.

The oxygen-17 and oxygen-18 varieties of the carbon-13 carbon dioxide were detected in the sample used in this work. The ν_3 fundamental vibration rotation bands of these two isotopic species were identified and the observational data for them are given in



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TABLE IV

Measured Line Positions (cm⁻¹-vac.) of the P₂ Bands of ¹²C¹⁶O¹⁷O and ¹³C¹⁶O¹⁸O

	6017		13 _C 16 ₀ 1	
ou°	1 -	00°0	00°1 -	oc°0
P 08	s	R OBS	P 085	R 085
2272.56	68			2267.4254
			2263.7434 2262.9911 2262.2369 2261.7608 2259.9253 2259.3665 2257.5619	2269.5718
2269.45 2268.66 2267.06 2267.06 2265.44	20	2279.9374	22062 - 2349 22062 - 23709 22266 - 7008 22259 - 14609 22259 - 3609	2271.6565 2272.3480 2273.0293
2260.25	56		2258.3605	
		2283.4076	2255.9624	2274 - 3741 2275 - 0402 2275 - 6940
2263.79 2262.96 2262.12		2285.4196	2255-9659 2255-35249 22553-69642 22553-69642 22551-8656 22256-13665 22248-4883	2277.0359
2259.57	98	2286.7300 2287.3781 2288.0229	2252.6959 2251.8642 2251.0252	2278.2705
2259.57	28	2288.6583	2250.1867	2280.1408
		2290.5331	2246.7631	2778-9755 2778-9259 52778-9259 52278-52059 22280-1408 73608-7862-7862 22281-3569 22281-3559 22283-15559 22284-3136
2252:53		2242.3575	2245.8967 2245.0212 2245.0212 2244.1407 2243.2544	2283.7354
2250.7 2249.80 2246.61 2247.91	191 040 858	22.72.72.72.72.72.72.72.72.72.72.72.72.7	2243.2544	
2247.0	212			2287.6687
2245.13	886	2296.4038 2296.9568 2297.5054	2236.8962	2288.2084
2241.30	23		2235 - 0336 2234 - 0914 2233 - 1451 2232 - 1435 2230 - 2785 2230 - 2785 2229 - 3061	2289.7966 2290.8208 2291.3268 2291.3268 2292.3254 2292.3254 2293.772
2238.3			2235 - 0336 2234 - 0914 2233 - 1451 2232 - 1936 2231 - 2365 2230 - 2785 2229 - 3061	2289.7966 2290.8208 2291.8268 2291.8308 2292.3254 2292.3117 2293.2945
2235.3	155			2293.2945
			2227.3557 2226.3736 2225.3797 2224.3851 2223.3844	
2231.3			2223.3851	2295.1693
2007.20	,		2221.3682 2220.3517	2296.9496 2297.3807 2297.6037
			2221.3682 2220.3517 2219.3267 2216.3031 2217.2716 2216.2327	2298.6411
				2270.0411
			2214.1360 2213.0770 2212.0205	

Table IV and the molecular constants of the levels involved are included in Table VI.

These constants compare well with the ones calculated by Chedir (6).

These constants compare well with the ones calculated by Chedin (6). The molecular constants obtained in this work for the levels 10°1 and 11°0 of $^{15}C^{16}O_2$ are consistent with the earlier results for the $(\nu_1 + \nu_2)$ and $(\nu_1 + \nu_3)$ bands $(7, \delta)$.

Molecular Constants* (cm-1) of 13C14O; Bands TABLE V

$\begin{array}{cccccccccccccccccccccccccccccccccccc$	93	SYMMETRY	vo-8'r'2,8'.'?		•	P. * 10	, s 10°	"je
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	001 - 100	***	2 262.8 0 0 . 4	0.386 727 5 + 23	0.389 721 4* + 18	12.14 . 2	12.149 : 11	78. 78
T _k - T _k	201 - 6200	+12	2 261.910 3 : 5	0.388 041 3 · 24	0.390 920 3 + 19	6.8.3	15.992 : 12	78. 72
Fig. T _k - T _k = T _k - T _k = 2271.759 9 : 3	1,00 - 2,0	*12	2 280.064 3 - 7	0.384 314 1 + 30	0.387 277 7 1 17	13.39 · 5	13.465 . 7	63, 49
F-F T _k = T _k = T _k = 230.666 2 · 9 0.387 178 · 16 0.390 997 · 16 13.66 · 5 13.64 · 5 13.64 · 5 15.65 · 9 0.387 188 · 16 0.390 997 · 16 13.66 · 5 13.64 · 6 13.7 · 12 13.65 · 8 0.387 188 · 17 0.390 997 · 18 13.64 · 60 12.52 · 60 12.7 · 18 13.65 · 8 13.7 · 18 13.65 · 8 13.90 · 17 13.65 · 18 13.90 · 18 13.9				0.387 670 · 5	0.340 601 . 5	13.43 · 5	13.48 : 5	
			2 271.759 9 : 3			13.60 · 5	13.44 . 5	88. 82
						13.05 - 42	13.15 + 42	67, 51
F-F T _k - T _k 2 250.653 4 · B 0.385 164 · 118 0.392 041 · 118 15.75 · 43 15.84 · 43 F-F A _k - A _k 2 280.069 1 · 4 0.385 696 · B 0.391 611 · B 13.62 · 12 12.77 · 12 F-F A _k - A _k 2 239.300 0 · 12 0.385 7 · 4 0.392 17 · 4 4 F-F T _k - T _k - T _k 2 235.614 2 · 12 0.385 390 · 12 0.392 275 · 12 12.81 · 73 12.86 · 73 F-F T _k - T _k - T _k 2 235.614 2 · 12 0.385 390 · 14 0.392 82 · 4 4			2 250.606 2 · 9	0.387 985 • 17		12.44 · 40	12.52 : 40	64. 34
1. 1. 1. 1. 1. 1. 1. 1.								35, 29
F-F A _u -A _c 2 280.069 1 · 4 0.388 696 · 8 0.391 611 · 8 13.62 · 12 13.70 · 12 13.50 · 12 0.389 25 · 4 0.392 13 · 4 d d d d d d d d d d d d d d d d d d			2 250.693 4 . 8			15.75 - 43	15.84 . 43	66, 52
						12.62 - 12	11.11	76. 70
7-7 54-54 2 239,300 0 · 12 0.389 25 · 4 0.382 13 · 4 4 12,81 · 33 12.88 · 33			2 260.069 1 . 4	0.388 696		13.84 · 8	13.90 . 8	11.11
f-f f _e f _o - f _o 2 286.357 1 · 4 0.389 390 · 12 0.392 275 · 12 12.81 · 33 12.96 · 33 f-f f _e f _o - f _e 2 226.676 2 · 12 0.389 59 · 4 0.392 82 · 4 d			2 239.300 0 • 12		9.362 13 4			8.3
t.f			7 248.357 1 · 4			12.81 · 33	12.56 · 33	60. 43
			2 236.674. 2 : 12	0.389 59 : 4	0.392 82 + 4			40, 18

The following P-tyre doubling constants q and .. are given as defined by Baldacci, Nalathy Devi. Chen, Rac & Fridowich, J. Mol. Spectrot. (in press)

 $q_{0}l_{0} = (0.635 + 0.0 \times 10^{-3} ; q_{11}l_{0} = (0.653 + 23) \times 10^{-3} ; q_{01}l_{1} = (0.615 + 7) \times 10^{-3} ; q_{11}l_{1} = (0.655 + 58) \times 10^{-3} ; q_{01}l_{1} = (0.17 + 7) \times 10^{-3} ; q_{11}l_{1} = (0.615 + 58) \times 10^{-3} ; q_{02}l_{0} = (0.13 + 16) \times 10^{-3} ; q_{02}l_{0} = (0.12 + 16) \times 10^{-3} ; q_{01}l_{0} = (0.12 + 16) ;$

uncertainties quoted are standard deviations in units of the last digit.

a) see ref. (3).

b) constants for the e-e sub-band obtained by the least squares fitting are (8'-8') = (-3.030 · 34) x 10^-3; (0'-0') = (-102 · 3) x 10^-8 value obtained from the f-f sub-band.

d) not determinable.

Note: The complete reference by Baldacci et al. cited in the footnote to this table is J. Mol. Speatrosc. 70, 143-159 (1978).

TABLE VI

Molecular Constants* (cm-1) of the v2 Band of 12C16O12O and 12C16O18O

MOLECULAR SPECIES	v ₀ -B'1'2+B*1"2	B*	8'	D' x 10 ⁸	D" x 10 ⁸	J _{max} P,R
13 _C 16 _O 17 _O	2274,0859 ± 4	0.375773 : 11	0.378641 ± 11	10.8 : 5	10.8 ± 5	50,35
13 _C 16 _O 18 _O	2265.9717 : 4	0.365403 ± 9	0.368193 : 9	11.71 : 20	11.78 ± 20	62,56

Uncertainties quoted are standard deviations in units of the last digit.

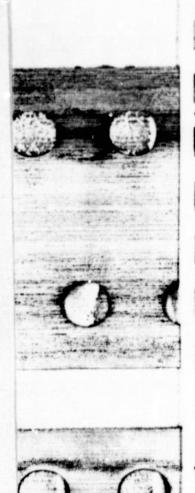
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RECETTED: March 1, 1978

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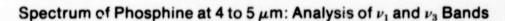
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The absorption spectrum of phosphine has been investigated in the region $2087-2482~{\rm cm}^{-1}$. About 1200 transitions belonging to the bands ν_1 and ν_3 were assigned. A strong Coriolis interaction between these bands gives rise to many "forbidden" transitions and large A_1A_2 splittings. The simultaneous analysis of the two bands enabled us to get a set of vibration-rotation constants for the vibrational states $\nu_1 = 1$ and $\nu_3 = 1$, and to obtain a value for the ratio S_1/S_3 between the band strengths of ν_1 and ν_3 .

INTRODUCTION

Studies on the four fundamental bands of phosphine were reported in the past, under low resolution, by McConaghie and Nielsen (1), using a vacuum grating spectrometer. Subsequently, Hoffman et al. (2) and later Yin and Rao (3) measured and analyzed the two low-frequency fundamental vibration-rotation bands occurring at 8--12 μ m with higher resolution. In attempting to obtain some information on the inversion barrier of PH₃, Maki et al. (4) studied the 3.4- μ m region, where the $3\nu_2$ and $4\nu_2 - \nu_2$ bands were resolved. Molecular constants of the ground state are now well determined from the analysis of the "forbidden" rotational transitions (5, 6) as well as from the above-mentioned infrared studies.

The present high-resolution study has been undertaken with a view to identifying the transitions involved in Jupiter's spectrum at $4-5~\mu m$ (7). The most prominent feature of the absorption by phosphine in this spectral range can be ascribed to the two strong overlapping fundamental bands ν_1 and ν_3 centered at 2321.131 and 2326.877 cm⁻¹, respectively. But weaker bands, $2\nu_2$, $\nu_2 + \nu_4$, and $2\nu_4$ also contribute to the absorption on the lower-frequency side.

This investigation relates to the two strongest bands ν_1 and ν_3 . Their unusual and rather complicated rotational structures are well explained up to J' = 10-11 by a strong Coriolis-type interaction between them. Unfortunately, for higher J' values, Fermi-type interactions with the overtone $2\nu_4$ give rise to rapidly increasing dis-



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crepancies between observed and calculated wavenumbers, so that the assignments became difficult beyond J'=13. Nevertheless, more than 1200 transitions belonging to ν_1 and ν_3 could be assigned in the spectral interval 2184-2446 cm⁻¹.

EXPERIMENTAL METHOD

The entire spectrum was recorded at the Ohio State University by using a 10-m double-pass focal length Czerny-Turner-type vacuum infrared spectrograph equipped with a 31 groove/mm, $40 \times 20 \text{ cm}^2 (16 \times 8 \text{ in.}^2)$ echelle grating. The continuous source of radiation was provided by a carbon furnace heated to 2500 K. Argon gas was passed through the source tank in order to avoid the accumulation of CO during the scans. An InSb detector cooled to liquid nitrogen temperature was used and the data were obtained with a spectral resolution ranging between 0.025 and 0.035 cm^{-1} .

Interferometric fringes, which are recorded simultaneously with the observation of infrared spectra and absorption standards, were used in determining the spectral positions of the lines. A dispersion curve, for estimating the wavenumbers of the PH₃ lines, was then obtained from a least-squares fit of the internal calibration standards. 1–0 and 2–0 transitions of ¹²C¹⁶O were used as reference lines; wavenumbers (vac.-cm⁻¹) for the rotational structure of these two bands were taken from Rao (8).

In scanning the total spectral region, measurements were made with absorption path lengths of 10 and 100 cm at pressures ranging between 0.2 and 6.0 Torr. Under these conditions, data were recorded in the interval 2087-2482 cm⁻¹ and about 3100 absorption lines were observed and measured.

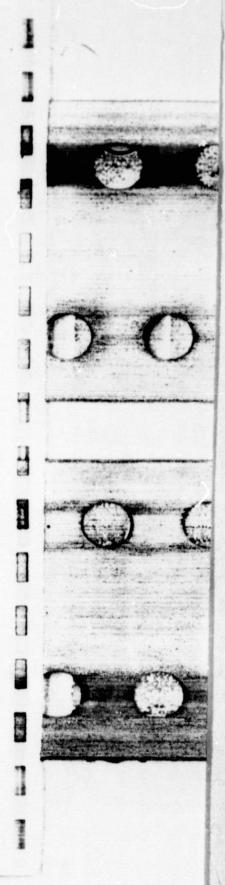
The average of the stronger lines was taken over four or more measurements. The estimated accuracy is limited to ± 0.005 cm⁻¹; however, for well-resolved lines the values are about ± 0.003 cm⁻¹.

The spectrum showed lines due to atmospheric absorption of carbon dioxide, in the Q-branch region and among the lower first and R-branch lines of the ν_1 and ν_3 bands of PH₃. This overlapping CO₂ structure was easily identified (9).

ASSIGNMENTS—COMPUTATIONAL PROCEDURE

The two bands ν_1 and ν_3 are very close to each other. The strong Coriolis-type interaction between them is responsible for many "forbidden" (perturbation allowed) transitions as well as large A_1A_2 splittings. The first assignments were performed on the basis of the ground-state combination relations among transitions. The accurate values already available for the ground-state constants (4,6) were used to calculate the differences.

As a second step, we used an iterative computation as indicated in Ref. (10). This improved the assignments and allowed us to find the vibration-rotation constants related to the upper states $v_1 = 1$ and $v_3 = 1$. The energies in these states were obtained as eigenvalues of an effective twice-transformed Hamiltonian $(v_n, v_t)H^+$, diagonal in v_s except for the two Coriolis terms T_1 and T_2 , given by Eqs. (3) and (4) in Ref. (11). Besides these vibrational interactions (v_1, v_3, \ldots, v_t)



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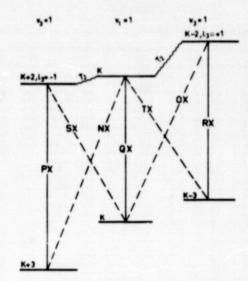


Fig. 1. "Forbidden" transitions induced by T_2 in ν_1 and ν_3 . In the indicated notation, the first symbol refers to K' - K'', the second one, X = P, Q, R, to J' - J''.

 \pm 1, $v_3 \mp$ 1, . . .) all essential resonances within $v_1 = 1$ and $v_3 = 1$ were also taken into account (10, 12).

It may be noted that the first-order term T_1 , which has only matrix elements in $\Delta(K-l_3)=0$ cannot induce "forbidden" transitions, whereas, the second-order term T_2 , with elements in $\Delta(K-l_3)=\pm 3$, breaks down the usual approximate selection rule $\Delta(K-l_3)=0$, as sketched in Fig. 1. In the present analysis of ν_1 and ν_3 , the great number of "forbidden" transitions observed in the spectrum is related to a large contribution of the term T_2 .

LINE STRENGTHS

The line strengths were approximated from the leading operator responsible for the observed transitions, i.e.,

$$(M_{\Gamma})_{1,3} = d_3[\cos{(\Gamma,x)}q_{3a} + \cos{(\Gamma,y)}q_{3b} + \rho_{1,3}\cos{(\Gamma,z)}q_1],$$

where q_1 , q_{3a} , and q_{3b} are dimensionless normal coordinates related to v_1 and v_3

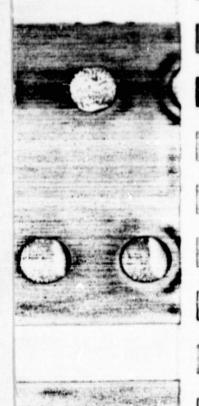
$$d_3 = \left(\frac{\partial M_x}{\partial q_{3a}}\right) = \left(\frac{\partial M_y}{\partial q_{3b}}\right); \qquad \rho_{1,3} = \left(\frac{d_1}{d_3}\right) \qquad \text{with} \qquad d_1 = \left(\frac{\partial M_z}{\partial q_1}\right).$$

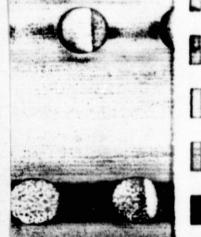
 $M_{\Gamma}(\Gamma = X, Y, Z)$ and $M_{\alpha}(\alpha = x, y, z)$ are the usual dipole moment components on space- and molecule-fixed frames.

Therefore, the ratio $\rho_{1,3}$, as well as the eigenvectors arising from the diagonalization of the matrix $\binom{c}{\nu_{1,\nu_3}}H^+$, are involved in the relative intensity calculations; absolute line strengths would also depend on the coefficient d_3 and experimental conditions (13).

The rigorous selection rules for the transitions are (12):

$$\Delta J = 0, \pm 1;$$
 $A_1 \leftrightarrow A_2$ or $E \leftrightarrow E$.





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TABLE I Wavenumbers and Intensities in the Jands v1 and v3 of PH3

*****	***************************************	***************************************		*********	*****	*********	******
(1)	(11)	(111)	(IV)	(V)	(IA)	(VII)	(VIII)
	RP(14.12.4)	13 13 1 4	2184.0785	3038.9929	-38	2.17-03	•1.0
1	AP(14.11.E)	13 12 1 6	2184.8219	3051.9604	-10	4.22-03	1.0
3	AP(14.10.E)	13 11 1 6	2185.5191	3063.7081	35	1.01-02	1.0
- 4	HP(14. 9.4)	13 10 1 A	2186.1397	3074.2357	47	3.83-02	•1.0
5	*RP(14. 1.E)	13 2 0 €	2186.3028	3115.1635		3.02-02	1.0
6	*SP(14. 4.E)	13 6 -1 €	2187.0589	3108.3930	317	1.19-01	.0
7	AP(14. 7.E)	13 8 1 6	2187.2029	3091.7888	-46	4.58-02	1.0
	QP(14. 3.A)	13 3 0 4	2187.5198	3112.3730	145	3.87-01	0
9	QF(14. 2.E)	13 2 0 €	2187.8257	3115.1854	30	2.46-01	1.0
10	QP(14. 0.41)	13 0 0 AZ	2187.9485	3117.3090	38	1.95-01	1.0
11	OP (14. 1.E)	13 1 0 €	2188.0250	3116.8857	3	2.83-01	1.0
12	QP(14. 5.E)	13 5 0 €	2188.5799	3105.3721	73	6.28-02	1.0
13	*UP(14. 0.A1)	13 4 1 42	2189.1974	3118.5579	13	1.45-01	1.0
14	OP(14. 4.E)	13 4 0 E	2189.4704	3110.8045	136	4.69-02	.0
15	43P(14. 2.E)	13 1 0 €	2189.5152	3116.8748	-7	6.81-02	1.0
16	OP(14:13.E)	13 13 0 €	2190.5773	3032.0609	18	2.01-01	1.0
17	3P(14.12.4)	13 12 0 4	2190.6406	3045.5550	15	7.31-01	•1.0
18	QP(14.11.E)	13 11 0 €	2190.7520	3057.8905	-10	5.01-01	1.0
19	OP(14.10.E)	13 10 0 €	2190.9201	3069.1091	-26	6.11-01	1.0
50	UP(14. 9.A)	13 9 0 4	2191.1436	3079.2395	-26	1.40.00	•1.0
21	OP(14. 8.E)	13 8 0 €	2191.4279	3088.3146	0	7.70-01	1.0
55	QP(14. 7.E)	13 7 0 €	2191.7667	3096.3525	40	8.23-01	1.0
53	9P(14. 6.A)	13 7 1 4	2192.1366	3103.3514	66	1.72.00	*1.0
24	*NP(14. 3.41)	13 0 0 AZ	2192.4647	3117.3179	47	4.71-01	1.0
25	HP(14. 5.E)	13 6 1 €	2192.5476	3109.3398	74	8.85-01	1.0
26	•SP(13. 8.E)	12 10 -1 E	2192.8043	2966.1350	959	3.57-03	.0
27	RP(14. 4.E)	13 5 1 €	2193.0179	3114.3520	82	9.15-01	1.0
28	AP(14. 3.42)	13 4 1 A1	2193.5781	3118.4313	21	8.88-01	1.0
29	AP(14. 3.41)	13 4 1 A2	2193.7247	3118.5779	33	4.84-01	1.0
30	AP (14. 2.E)	13 3 1 €	2194.1772	3121.5368	21	8.10-01	1.0
31	HP(14. 1.E)	13 2 1 €	2194.8744	3123.7350	-30	7.62-01	1.0
32	*NP(14. 4.E)	13 1 0 E	2195.5490	3116.8831	0	7.06-02	1.0
33	RP(13. 9.A)	12 10 1 A	2195.9651	2960.4284	-17	3: 16-02	•1.0
34	*SP(14. 0.41)	13 5 -1 WS	2196.4834	3125.8439	51	,	1.0
35	AP(13. 8.E)	17 € 1 €	2196.5351	2969.8659	-70		1.0
36	*SP(13. 4.E)	12 4 -1 E	2196.8369	2994.8244	4	1.	.0
37	AP(13. 7.E)	15 ' 1 E	2197.1873	2978.2836	11.0	602	1.0
38	PP(14. 1.E)	13 0 1 E	2197.3020	3126.1627	3.5	7.60-01	1.0
39	QP(13. 3.A)	12 3 0 4	2197.5054	2999.0416	4.3	6.75-01	•1.0
40	OP(13. 0.4)	15 6 0 7	2197.7538	2985.5359	-10	2.11-01	•1.0
41	3+(13. S.E.)	15 5 0 €	2197.9172	3001.9810	7	4.69-01	1.0
42	QP(13. 0.A2)	15 0 0 VI	2198.0067	3004.0881	39	2.46-01	1.0
43		15 1 0 E	2198.1290	3003.7063	-4	5.47-01	1.0
44	OP(13. 5.E)	12 5 0 E	2198.3484	2991.7555	-4	1.42-01	1.0
45	*NP(14+ 5.E)	13 2 0 €	2198.3846	3115.1768	21	2.22-01	1.0
46	PP(14. 2.E)	13 1 -1 €	2198.5850	3125.9446	34	7.02-01	1.0
47	PP(14. 3.42	13 2 -1 A1	2198.5850	3123.4382	-19	8.60-01	1.0
44		13 0 1 E	2198.7812	3126.1408	-2	1.32-04	1.0
49	OP(13. 4.E)	12 4 0 E	2199.0665	2997.0540	. 9	1.38-01	1.0
50	*OP(14. 3.A1)	13 1 1 AZ	2199.5158	3124.3690	-64	6.43-01	1.0
51	•PP(13. 2.€)	15 1 0 E	2199.6475	3003,7112	0	1.00-01	1.0
52		12 5 -1 A	2200.0440	3001.5802	33	1.64-01	•1.0
53		15 15 0 4	2200.2948	2931.2813	16	6.84-01	•1.0
54	OP(13.11.€)	12 11 0 €	2200.3176	2943.6381	-6	6.26-01	1.0
55	3P(13+10+€)	15 10 0 E	2200.4164	2954.8855	-15	8.61-01	1.0
56	OP(13. 9.A)	15 6 0 V	2200.5511	2965.0144	-42	2.11.00	•1.0
57	PP(14. 4.E)	13 3 -1 E	2200.6984	3122.0325	55	7.71-01	1.0
58	UP(13. 8.E)	12 8 0 E	2200.7944	2974.1251	-11	1.21.00	1.0

- (1) SEWIAL NUMBER
- (II) THANSITION
- (III) VALUES OF J' + [K'] + ℓ_3' +C' + FOR UPPER LEVEL OF THE TRANSITION (IV) DESERVED WAVENUMBER IN CM
- (V) UPPER LEVEL ENERGY IN CM
- (VI) (OHS-CALC) WAVENUMBER IN 10 CM
- (VII) CALCULATED HELATIVE LINE STRENGTH
- (VIII) STATISTICAL WEIGHT



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(1)	(11)		111	1)		(IV)	(V)	(VI)	(VII)	CVIII

	*OP(14+		13	1		41	2200.9580	3125.8112	36	8.01-02	1.0
60	PP(14.		13	5		45	2201.0078	3125.6610	36	2.17-01	1.0
61	QP(13.		15	7	0	E	2201.0578	2982.1541	-15	1.32.00	1.0
65	•NP(14.		13	3	0		2201.1636	3112.3784	150	9.38-01	0
63	•SP(13.		15		-1	E	2201.2761	3005.3398	. 1	5.16-05	1.0
64	RP(13.	6+A)	15	7	1		2201.3885	2989.1706	-17	5.80.00	*1.0
65	RF(13.	5.E)	14		1	E	2201.7959	2995.2030	-55	1.44.00	1.0
66	HP(13.		12	5	1	E	2202.3249	3000.3123	3	1.48+00	1.0
67	PP(14.		13		-1	E	2202.3249	3119.1171	78	9.34-01	1.0
68	·0+(14.		13	5	1	E	2505.3885	3123.7333	-31	5.07-03	1.0
69	•NP(13.		12	0	0	Al	2202.5585	3004.0947	46	1.18.00	1.0
70	•SP(13.	1.€)	14	3	-1	E	2202.7009	3008.2782	-28	6.28-03	1.0
71	RP(13:	3.A1)	15	4	1	45	2202.9450	3004.4812	-12	1.46.00	1.0
72	RP(13.	3.A21	12		1	Al	2203.1633	3004.6995	-9	4.20-01	1.0
73	RP(13.	2.E)	12	3	1	E	2203.6552	3007.7190	5	1.41.00	1.0
74	PP(14.	7.E)	13		-1	E	2203.8177	3108.4035	328	8.93-01	.0
75	PP (14.		13	5	-1		2204.1887	3115.4035	164	1.69.00	
76	RP(13.	1.E)	15	5	1	£	2204.4286	3010.0059	-13	1.39+00	1.0
77	RP (13.		12	ī	i	Al	2204.6294	3010.7108	-29	6.91-01	1.0
78			13	3	i	E	2204.7373	3121.5295	14	8.31-04	1.0
79			10	5	-1	AI	2205.7294	3011.8108	5	1.95.00	1.0
80			12	2	i	E	2205.9432	3010.0070	-12	1.50-01	1.0
81	PP (14.		13	7	-1	Ē	2206.2400	3103.1267	503	1.38-00	. (
82	RP(12.		ii		i	È	2206.4450	2864.8685	-15	2.89-02	1.0
83			ii	6	-1	È	2206.5692	2889.8480	38	2.18-01	1.0
84	PP(13.		12	ő	i	È	2206.7413	3012,3186	7	1.16.00	1.0
85	RP (12.		ii		i	Ē	2207.0699	2873.3221	-14	7.60-02	1.0
86	*0P(14+		i.	4	i	AI	2207.2331	3118.4479	37	5.93-03	1.0
87	*OP (14.		.3	-	i	SA	2207.3527	3118.5675	23	4.61-03	1.0
86	•SP(12+		ii	5	-i		2207.4581	2894.3138	15	1.08-00	*1.0
89	RP(12.		ii	7	-i		2207.6513	2880.6432	-3	3.26-01	.1.0
90			ii	Š	ò	Ē	2207.9546	2897.3578	7	8.49-01	1.0
91	PP 13.		12	î	-1	È	2207.9807	3012.0445	ó	1.22.00	1.0
					-0		2208.1741	2899.1029	7	1.01.00	1.0
92			11	5	0	E		2886.8361	-18	2.80-01	1.0
93			11		0	E	2208.1741		59	1.08-00	
94			11	0		AZ	2208.2640	3009,8852	-55	1.51.00	1.0
95			15	5	-1	YS.	2208.3490		-		1.9
96			12	5	0	E	2208.5823	3001.9894	15	4.38-01	1.0
97			13	8	-1		2208.6560	3096.7520	795	3.65.00	
98			11	4	0	E	2208.7621	2892.0409	-14	3.44-01	1.0
99			15	1	1	Al	2209.1721	3010.7083	-32	1.08-00	1.0
100	3P(12.	-	11	3	0		2209.5940	2896.4497	-6	4.84-01	•1.0
101			11	1	0	E	2209.6945	2899.0977	. 1	1.34-01	1.0
102			11	11	0	E	2209.9289	2838.0938	-10	5.59-01	1.0
103			11	10	0	E	2210.0045	2849.4111	34	1.03.00	1.
104	QP(12+	9.A)	11	9	0		2210.0815	2859.5649	20	2.84+00	•1.0
105	QP(12.	8.E)	11	8	0	E	2210.2191	2868.6425	4	1.74.00	1.0
106	PP(13.	3.421	12	5	-1	Al	2210.2668	3011.0030	-2	3.63-01	1.
107	PP(13.	4.E)	12	3	-1	E	2210.3092	3008.2967	-10	1.50+00	1.
108	QP(12.	7.E)	11	7	0	E	2210.4254	2876.6776	-11	1.79.00	1.0
109			14	1	1	SA	2210.4791	3012.0153	46	8.85-02	1.1
110			11	6	0		2210.7246	2883.7165	-21	4.29.00	.1.
iii			ii	4	-1		2210.7901	2900.1934	-32	8.04.02	1.
112			ii	6	i	Ē	2211.1376	2889.7996	-22	2.22.00	1.
113			13	9	-i		2211.1376	3089.3266	1173	2.22.10	
114			iz	3	ō		2211.2631	2999.0452	57	1.96.00	•1.0
115			ii	5	ĭ	Ē	2211.6753	2894.9541	-17	2.24.00	1.0
116			iż		-i		2211.9387	3005.3458	7	1.60.00	1.0
	1131	315 ,	16					2002.2430		1.00.00	



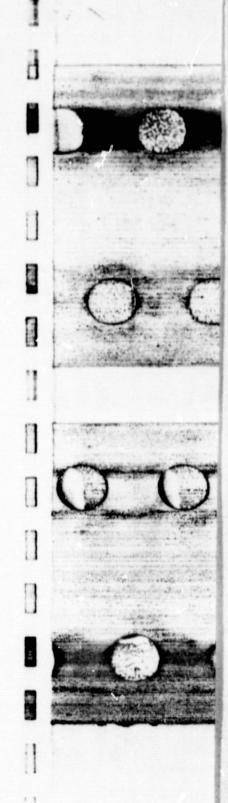
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TABLE I-Continued

		E I—Conn				
(1) (11)	(111)	(IV)	(4)	(IV)	(VII)	(VIII)
117 AP(12. 3.A1)	11 4 1 42	2212.2858	2899.1415	26	2.47.00	1.0
118 AP(12. 3.AZ)	11 4 1 41	2212.3377	2899.1935	-21	2.22.00	1.0
119 AP(12. 2.E)	11 3 1 5	2213.1208	2902.5240	14.23	2.21.00	1.0
120 PP(14.11.E)	13 10 -1 €	2213.5413	3080.6798	1432	1.88.00	.0
121 PP(13. 7.E)	12 6 -1 E	2213.7465	3001.5806	165	2.71.00	.1.0
122 PP(13. 6.A)		2213.798	2904.8932	0	2.30.00	1.0
		2214.2186	2905.6554	-12	7.89-01	1.0
124 •SP(12. 0.41) 125 •OP(13. 5.E)	11 2 -1 42	2214.3049	3007.7120	-12	2.94-03	1.0
126 *TP(11. 0.A2)	10 3 0 A1	2214.5322	2799,9953	-8	5.90-02	1.0
127 AP(12. 0.A1)	11 1 1 42	2215.0276	2906.4644	-9	3.39.00	1.0
128 *NP(12. 4.E)	11 1 0 E	2215.8135	2899.0923	-3	2.99-01	1.0
129 *NP(13. 7.E)	12 4 0 E	2215.9712	2997.0675	23	8.07-01	1.0
130 PP(13. 8.E)	12 7 -1 €	2216.0574	2989.3881	359	2.74.00	.0
131 PP(12. 1.E)	11 0 1 E	1622.9122	2907,1579	14	1.76.00	1.0
132 *SP(11. 4.E)	10 6 -1 €	2216.2455	2793.4902	-17	2.29-01	1.0
133 AP(11. 8.E)	10 9 1 E	2216.2455	2768.4480	13	1.56-02	1.0
134 PP(14.12.A)	13 11 -1 A	1682.9122	3071.2035	1968	6.18.00	0
135 .AP(11. 1.E)	10 2 0 €	2216.3553	2401.3066	-9	6.13-02	1.0
136 PP(11. 7.E)	10 8 1 E	2216.9076	2776.9981		6.96-02	1.0
137 •SP(11. 3.A)	10 5 -1 4	2217.3269	2798.1751	-12	1.55.00	•1.0
138 PP(12. 2.E)	11 1 -1 €	2217.4393	2906.8426	0	2.03.00	1.0
139 RP(11. 6.A)	10 7 1 4	2217.4897	2784.3707	-7	4.33-01	.1.0
140 *OP(12. 2.E)	11 0 1 E	2217.7438	2907.1471	3	2.20-02	1.0
141 OP(11. 2.E)	10 S O E	2217.8985	2801.3130	-3	1.45.00	1.0
142 OP(11. 5.E)	10 5 0 €	2218.0128	2790.6063	. 5	4.86-01	1.0
143 PP(12. 3.AZ)	11 5 -1 VI	2218.0745	2904.9302	-19	2.53.00	1.0
144 OP(11. 1.E)	10 1 0 €	2218.1116	2803.0629	-1	1.78-00	1.0
145 OP(11. 0.A2)	10 0 0 A1	2218.1853	2803.6484	-6	1.93.00	1.0
146 PP(13. 9.A)	12 8 -1 A	2218.1853	2982.6486	463	3.60-01	• .0
147 •NP(13. 8.E)	10 4 0 E	2218.4278	2795.7283	-1	7.47-01	1.0
149 PP(14-13-E)	13 12 -1 €	2218.5923	3060.0759	1980	3.62.00	1.0
150 .NP(12. 5.E)	11 2 0 E	2218.6872	2897.3492	-1	7.93-01	1.0
151 PP(12. 3.A1)	11 2 -1 AZ	2218.8003	2905.6560	-11	1.86.00	1.0
152 OP(11. 3.A)	10 3 0 4	2219.1502	2799.9984	-5	1.24.00	*1.0
153 QP(11-10-E)	10 10 0 €	2219.5386	2752.5784	-3	8.77-01	1.0
154 OP(11. 9.A)	10 9 0 4	2219.5795	2762.7736	15	3.24.00	.1.0
155 .0P(12. 3.A1)	11 1 1 42	2219.6164	2906.4721	-1	4.92-01	1.0
156 OP(11. 8.E)	10 8 0 €	2219.6653	2771.8678	14	2.24.00	1.0
157 QP(11. 7.E)	10 7 0 E	2219.8200	2779.9106	3	2.74.00	1.0
158 PP(12. 4.E 1	11 3 -1 E	2219.9182	2903.1970	-20	2.61.00	1.0
159 *OP(12. 3.A2)	11 1 1 41	2219.9600	2906.8157	16	9.02-02	1.0
160 OP(11. 6.A)	10 6 0 A	2220.0784	2786.9594	-8	6.13.00	•1.0
161 *SP(11+ 2+E)	10 4 -1 €	2220.2960	2803.7105	-55	2.17-01	1.0
162 RP(11. 5.E)	10 6 1 E	2220.4784	2793.0719	-16	3.19.00	1.0
163 PP(13.10.E)	12 9 -1 €	2220.4784	2974.9475	718	4.14.00	.0
164 PP(14.14.E)	13 13 -1 €	2220.8888	3047.6989	1903	4.25.00	.0
165 AP(11. 4.E)	10 5 1 €	2221.0392	2798.2839	-15	3.17.00	1.0
166 •NP(13. 9.A)	12 6 0 4	2221.0808	2985.5441	-5	2.89-01	•1.0
167 PP(12. 6.A)		2221.3187	2894.3106	12	3.93.00	-1.0
		2221.5445	2900.2065	- 8	2.58-04	1.0
170 RP(11. 3.4)	10 4 1 4	2221.7465	2802.5947	13	6.68-00	-1.0
171 *OP(13. 8.E)	12 6 1 6	2221.8787	2995.2094	-10	1.80-02	1.0
172 RP(11. 2.E)	10 3 1 6	2222.5738	2805.9883	-10	3.17.00	1.0
173 *NP(11. 3.A2)	10 0 0 A1	2222.7842	2803.6324	-24	1.74-01	1.0
174 PP(13.11.E)	12 10 -1 €	2222.8106	2966.1311	955	4.87-00	
						• •



VI AND VS OF PHS

YTHAUG ROOF TO

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TABLE I-Continued

1)	(11)	(111)	(IV)	(4)	(AI)	(VII)	IVII
	************					*********	
175	RP(11) 1:0 1	10 5 1 6	2223.4705	2808.4218	5	3.47-00	1.
176	ONP(120 Ent)	11 3 0 A	2223,4705	2896.4624	5	4.06.00	•1.
177	PP.12. 726 1	11 6 -1 €	2223.6040	2889.8562	46	3.69.00	1.
178	•SP(11. 0.42)	10 2 -1 A1	2223.7558	2809.2189	-1	5.49-01	1.
179	*OP(12. 5.E)	11 3 1 6	2223.8607	2902.5227	0	5.10-03	1.
180	RP(11. 0.A2)	10 1 1 A1	2224.4132	2809.8763	5	5 71+00	1.
181	PP(13.12.4)	12 11 -1 A	2225.2408	2956.2273	1213	1.14.01	
182	PP(12. 8.E)	11 7 -1 €	2225.6700	2884.0935	116	5.05.00	
183	PP(11+ 1+E)	10 0 1 €	2225.7135	2810.6648	9	2.68+00	1.
184	ONP(12. 7.E)	11 4 0 E	2225.7860	2892.0382	-16	9.38-01	1.
185	•NP(11+ 4+E)	10 1 0 €	2225.8165	2803.0612	-2	4.66-01	1.
186	*SP(10. 4.E)	9 6 -1 6	2225.8558	2705.7755	-36	1.99-01	i.
187	RP(10. 8.E)	9 9 1 5	2225.9374	2680.6405	24	3.62-03	i.
188		11 4 1 4	2226.1417	2899.1336	18	2.20-02	•1.
189	AP(10. 7.E)	9 8 1 6	2226.6620	2689.3085	17	4.33-02	1.
190	PP(11. 2.E)		2226.9199				
191				2810.3344		3.26-00	1.
			2227.1071	2710.6550	-28	1.91.00	*1.
192	.OP(11. 5.E)	10 0 1 E	2227.2590	2810.6735	18	3.28-02	1.
193	RP(10. 6.4)	9 7 1 A	2227.2784	2696.7628	-5	4.73-01	.1.
194	PP(13.13.E)	15 75 -1 E	2227.3821	2944.8157		6.72.00	
195	PP(12. 9.A)	11 8 -1 A	2227.7217	2877.2051	551	1.24.01	٠.
196	PP(11. 3.A1)	10 5 -1 WS	2227.7605	2808.6087	-7	4.06.00	1.
197	OP(10. 2.E)	9 5 0 E	2227.7605	2713.8923	-10	2.34.00	1.
198	OP(10. 5.E)	9 5 0 E	2227.7895	2703.0259	5	7.56-01	1.
199	OP(10. 1.E)	9 1 0 E	2227.9657	2715.6448	1	2.92.00	1.
200	QP(10. 0.A1)	9 0 0 42	2228.0195	2716.2139	0	3.14.00	1.
201	QP(10. 4.E)	9 4 0 €	2228.1613	2708.1010	3	1.41+00	i.
202	PP(11. 3.42)	10 2 -1 A1	2228.3722	2809.2204	0	3.29.00	i.
203	QP(10. 3.A)	9 3 0 A	2228,6765	2712.2244	ō	2.78.00	•1.
204	*NP(11. 5.E)		2228.7166	2801.3102	-6	1.34.00	i.
205	*OP(11. 3.42)	10 2 0 E	2229.0298	2809.8780	6	4.56-01	i:
206	QP(10. 9.A)	9 9 0 4	2229.0729	2674.7039	-14	2.64.00	•1.
207	QP(10. 8.E)	9 8 0 6	2229.1140	2683.8171	6	2.45.00	i.
208	OP(10. 7.E)	9 7 0 6	2229.2209		13		
209	QP(10. 6.A)			2691.8674		3.38.00	1.
			2229.4307	2698.9151	6	8.08.00	•1.
210	PP(11. 4.E)	10 3 -1 E	2229.4953	2806.7400	-50	4.29.00	1.
212	*SP(10. 2.E)	3 6 -1 E	2229.7443	2715.8761	-17	4.54-01	1.
			2229.8113	2705.0478	-4	4.25.00	1.
213	PP(12.10.E)		2227.8619	2869.2685	374	7.34.00	
214			2530.3960	2710.3157	-7	4.14.00	1.
215	PP(11. 5.E)	10 4 -1 E	2231.1180	2803.7115	-21	4.28.00	1.
216	RP(10. 3.A)	9 4 1 4	2231.1430	2714.6909	-1	8.38+00	.1.
217	*OP(11. 4.E)	10 2 1 E	2231.1826	2808.4273	7	1.11-05	1.
218	PP(11. 6.4)	10 5 -1 A	2231.2999	2798.1809	-6	7.56.00	.1.
219	RP(10. 2.E)	9 3 1 5	2232.0163	2718-1481	2	4.20.00	1.
220	PP(12+11+E)	11 10 -1 E	2232.0852	2860.2501	539	8.66.00	
221	•NP(10. 3.41)	9 0 0 AZ	2232.6660	2716.2139	0	3.74-01	1:
222	RP(10. 1.E)	9 2 1 6	2232.9595	2720.6386	7	4.78.00	i.
223	•NP(11 . 6.A)	10 3 0 A	2233.1182	2799.9992	-4	5.57.00	
224	*SP(10. 0.A1)	9 2 -1 A2					.1.
225			2233.2269	2721.4213	10	1.48-01	1.
		10 6 -1 E	2233.4030	2793.4936	-14	6.75.00	1.
922	RP(10. 0.A1)	9 1 1 AZ	2233.8554	2722.0498	17	8.68.00	1.
227	PP(12.12.A)	11 11 -1 A	2234.2242	2849.9511	539	2.04.01	٠.
228	PP(10. 1.E)	9 0 1 6	2235.1932	2722.8723	7	3.97.00	1.
558	PP(11. 8.E)	10 7 -1 E	2235.3260	2787.5285	14	8.74.00	1.
230	•SP(9. 4.E)	8 6 -1 E	2235.3848	2626.7207	-41	1.39-01	1.
231	*NP(11. 7.E)	10 4 0 E	2235.6423	2795.7328	1	9.38-01	1.
232	*OP(11. 6.A)	10 4 1 4	2235.7177	2802.5987	17	3.28-02	.1.



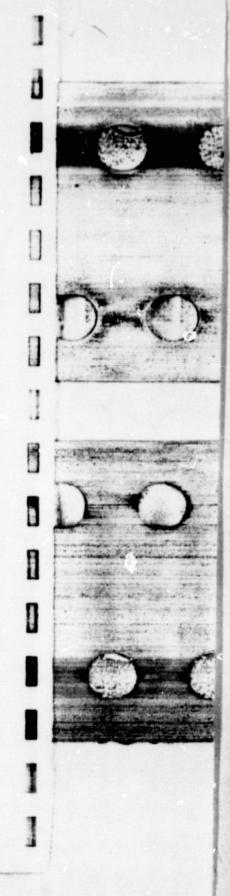
OR POOR QUALITY

BALDACCI ET AL.

TABLE 1-Continued

(1)	(11)		(11	1)		(IV)	(V)	(IV)	(VII)	(VIII

	*NP(10. 4.E)	9		0		2235.7177	2715.6374	-5	6.35-01	1.0
234						2235.9976	2635.1417	1	6.29-02	1.1
235		9			-	2236.3913	2722.5231	0	4.99.00	1.0
536		*	0		E	2236.7550	272.1.8868	55	3.81-02	1.0
237						2236.7896	2631.7766	-32	1.92.00	.1.0
538						1586.9851	2617.8165	-13	3.99-0	.1.
239	PP(11. 9.4)	10		-1		2237.2616	2780.4557	58	2.11.01	.1.
240	PP(10. 3.42)	9	5	-1	41	2237.3752	2720.9231	-14	6.21.00	1.1
241	QP(9. 5.E)		- 5	. 0		2237.5078	2624.1308	-5	1.05.00	1.
242	QP(9. 2.E)		5	0	E	2237.5450	2635.1321	-8	3.58.00	1.
243	QP(9. 1.E)				Ē	2237.7170	2636.8611	-1	4.59.00	1.0
244	154.0 .4 14D				Al	2237.7542	2637.4168	o	4.81.00	1.0
245				0		2237.8326	2629.1685	-2	2.38.00	1.0
246		¥				2237.8732	2721.4211	10	5.52.00	1.0
247						2238.1757	2633.1627		5.45.00	•1.0
248		10				2238.2039	2798.2944		2.92-02	
249		.,				2238.5038	2722.0517	19	2.42-01	1.
250	QP(9. 8.E)	i	å			2238.5557	2604.5140	-16	1.91.00	1:
251	OP(9. 7.E)		7			2238.6103	2612.5631	-10	3.54+00	1.
252						2238.6625	2713.8990	-4	2.09.00	
253			6	ő		2238.7661	2619.6005	13		.!.
254	PP(10. 4.6)		3			2239.0318	2718.9515		9.57.00	•1.
255	RP(9. 5.E)		6					-15	6.75-00	1.
256			4			2239.1209	2625.7439	6	5.19.00	1.
257	PP(11.10.E)	10			è	2239.1493	2636.7364	-2	8.70-01	1.
258			5				2772.3167	123	1.25.01	!
259	RP 1 9. 3.4 1					2239.7312	2631.0671	-5	4.91-00	1.0
260		3	:			2340.5278	2635.5146	-5	9.78.00	.1.
		3				£ 40.6435	2715.8799	-13	6.68.00	1.0
261			5			40.7216	2720.6413	10	1.41-05	1.1
595	PP(10. 6.4)		.5		•	2241.1796	2710.6640	-19	1.40-01	•1.0
563		10				2241.3695	2763.0800	187	1.48-01	.1
264	HP(9. 2.E)	8	3		E	2241.4359	2639.0230	0	5.12.00	1.0
265	HP(9. 1.E)		5	1	E	2242.4177	2641.5618	7	6.08.00	1.0
266	*5P(9. 0.A2)		2			2242.6247	2642.2873	8	2.74-04	1.0
267		9	3			2242.7459	2712.2303	5	6.75.00	.1.0
268	PP(10. 7.E)	4	6		E	2243.1381	2705.7846	-56	1.15.01	1.0
598	RP(9. 0.A2)	8	1	1	Al	2243.3206	2642.9832	19	1.18.01	1.0
270	PP(9. 1.E)		0	1	E	2244.6571	2643.8012	10	5.66.00	1.0
271	*SP(8. 4.E)	7	6	-1		2244.8504	2556.3734	-27	7.09-02	1.0
272	PP(10. 8.E)	9	7	-1	E	2244.9642	2699.6673	-19	1.43.01	1.0
273		9	4	1		2245.2140	2714.6984	6	3.25-02	.1.0
274	•NP(10 . 7.E)	4	4	0	E	2245.4616	2708.1082	10	8.02-01	1.0
275	•NP(9. 4.E.)	8	1	0	Ē	2245.5323	2636.8682	5	7.55-01	1.0
276	PP(9. 2.E)		1	-!	E	2245.8622	2643.4493	9	7.27.00	1.0
277	*SP(8. 3.4)	7	5	-1	Ā	2246.3709	2561.5658	-24	1.48.00	.1.0
278	RP(8. 6.4)	7	7	1		2246.6387	2547.5997	5	2.05-01	*1.0
279	PP(10. 9.A)		8	-1		2246.8380	2692.4691	3	3.44.01	*1.0
280	PP(9. 3.41)	8	2	-1	42	2246.9493	2641.9363	-5	9.09.00	1.0
281	3P(8. 5.E)	7	5	ō	•	2247.1803	2553.9633	-8	1.30.00	1.0
282	QP(8. 2.E)	7	S	ō	È	2247.2434	2565.0533	-1	5.13.00	1.0
283	PP(9. 3.42)		S	-1	AL	2247.2996	2642.2866	- 6	8.49-00	1.0
284	QP(8. 1.E)	7	î	ō		2247.3790	2566.7547	7	6.60.00	
285	OP(8. 0.A1)	7	ò	0	A2	2247.3991	2567.2963	5		1.0
286	QP(8. 4.E)	,	4	0	Ē	2247.4466	2558.9696	0	6.68+00	1.0
287	UP (8. 3.4)	,	3	0	-				3.61.00	1.0
288	*OP(10. 7.E)	÷	5	i		2247.6444	2562.8393	-	9.26.00	•1.0
289	OP(8. 7.E)	ï	7	0		2247.6746	2710.3211	-1	3.08-02	1.0
290	QP(8. 6.A)	,	6	0	E		2542.0251	-15	2.64.00	1.0
	urt no nes /	,		0		2248.0725	2549.0335	5	9.67+00	.1.0



VI AND VI OF PH

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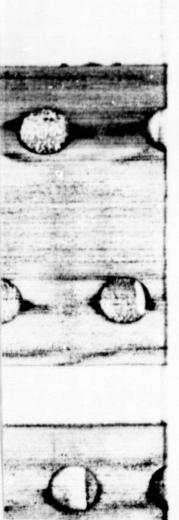


DESIGNAL PAGE 15

OF POOR QUALITY

(1)	(11			11)		(IV)	(4)	(11)	(VII)	IIIIV

292		5.E)	7	6 1		2248.3893	2555.1724	9	5.66.00	1.0
593	*SP(8:					2248.4926	2566.3025		1.21.00	1.0
294	PPI 9		:		E	2248.5148	2635.1378	-8	3.04.00	1.0
295	PP(10.		÷		È	2248.7948	2684.1990	37	2.04.01	1.0
296		4.6	i		È	2249.0397	2560.5627	0	5.10.00	1.0
297		1.6		4 0		2249.1083	2497.5092	i	6.13-03	1.0
298	RPI 8		7	- i		2249.8852	2565.0801	-3	1.03-01	•1.0
299		5.E)			Ē	2250.1111	2636.7341	-4	1.01-01	1.0
300	*OP (9				è	2250.2293	2641.5652	11	1.22-04	1.0
301		2.6)	7		Ē	2250.8282	2568.6381	-2	5.76-00	1.0
302	PP (9				Ā	2250.9539	£631.7883	-20	2.47.01	.1.0
303	RPI B	1.E 1	7	2 1	E	2251.8388	2571.2145	5	7.16.00	1.0
304	*SP (8.	0.A1)		2 -1	SA	2251.9957	2571.8929	21	4.76-02	1.0
305	.MP (8.		7	0 0		2252.1040	2567.2989		5.19-01	1.0
306	•NP (9			3 0		2252.3355	2633.1699	13	6.83.00	•1.0
307	*OP 1 9				ε	2252.4060	2639.0240		2.89-03	1.0
304	PP (9.					2252.7813	2626.7342	-28	1.83-01	1.0
309	RPI B		!		A2	2252.7813	2573.6785	53	1.46-01	1.0
310	PPI B		7	0 1		2254.0866	2573.4623	10	7.71.00	1.0
311	PP (9.				E	2254.5532	2620.5115	-55	2.24.01	1.0
312	•OP (9			• 1		2254.6873	2635.5217	5	2.74-02	•1.0
313			8	4 0		2255.2118	2629.1646	-6	5.72-01	1.0
314	•RP1 7		6		E	2255.2357	2566.7587	11	7.79-01	1.0
316	PP (8		7		-	2255.2692	2503.6701	0	4.16-02	1.0
317	*SP (7.		6	5 -1		2255.8457	2573.1055	10	7.68-01	•1.0
318	PP 1 9.			8 -1		2256.3915	2613.2190	-6	5.39.01	*1.0
319	PP (8		7	2 -1		2256.4619	2571.6568	5	1.27-01	1.0
320	PP (8.		7	2 -1		2256.6889	2571.6838	12	1.22-01	1.0
321		5.€)		6 1		2256.8126	2492.5564	-4	1.33.00	1.0
322	QP (7.		6		Ē	2256.8444	2503.6715	1	6.94+00	1.0
323	QP (7.	0.42)	6	0 0	Al	2256.9375	2505.8625	4	9.14.00	1.0
324	QP (7.	1.€)	6	1 0	E	2256.9375	2505.3384	13	8.81.00	1.0
325	QP (7.	4.E)	6		E	2256.9969	2497.5049	-3	4.95.00	1.0
356	QP (7.		6	3 0		2257.0816	2501.2804	7	1.35.01	•1.0
327		7.E)	8	5 1		2257.1308	2631.0836	14	2.70-02	1.0
328		6.4	6			2257.3498	2487.2415	-12	7.01-00	•1.0
329	*OP (8.		?	1 1		2257.4809	2572.6758	50	1.20-02	1.0
330		5.E !	6	5 0		2257.6024	2493.3462	9	5.24.00	1.0
331		2.E 1	7	3 -1	٤	2257.7905	2504.6176	6	1.42.00	1.0
332		5.E)	'n	2 0		2258.2775	2569.4814	5	4.05.00	1.0
334		4.6)	6		Ė	2258.3092	2498.8172	-1	4.35.00	1.0
335	RP (7		6			2259.2100	2503.4088	-6	9.47.00	•1.0
336		5.E)	7	4 -1		2259.5240	2566.3070	4	1.50-01	1.0
337	*0P1 8.		7		Ē	2259.6909	2571.2139	4	2.60-04	1.0
338		2.6)	6	3 i		2260.1836	2507.0197	-6	5.66.00	1.0
339		6.A)	7	5 -1		2260.6148	2561.5758	-14	4.07.01	.1.0
340	RP (7.		6	2 1		2261.2129	2509.6138	-2	7.78.00	1.0
341	•NP (7.	3.421	6	0 0		2261.6709	2505.8697	11	4.49-01	1.0
342	*TP(6.	0.411	5	3 0	24	2261.7312	2448.5014	7	2.50-02	1.0
343		5.E 1	7	3 1		2261.8525	2568.6356	-5	1.66-03	1.0
344		6.A 1	7	3 0		2261.8852	2562.8462	15	5.51.00	.1.0
345		(SA.0	6	1 1		2262.1923	2511.1173	15	1.74.01	1.0
346	PP (8.		1	6 -1		2262.3432	2556.3828	-17	2.76.01	1.0
347	•QP (7.			1 -1	3	2263.1228	2511.5237	19	7.19-02	1.0
348	PP (7.	1.E)	6	0 1	E	2263.4687	2511.8696	4	9.96.00	1.0



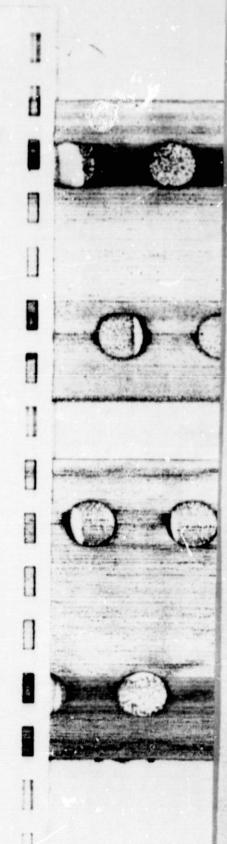


BALDACCI ET AL.

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TABLE 1-Continued

1)		11)			11			(IV)	(4)	(VI)	(111)	IIIVI
				****	**			**********				
349			8.E)	7	7			2264.0794	2550.0779	-12	3.36.01	1.0
350	.0P(6 .A)	7				2264.1229	2505.0839	0	1.76-02	*1.0
351	991	7.	5.E)		1	-1	E	2264.6964	2511.5135	9	1.33-01	1.0
352	MPI	7.	4.E 1		1	0	•	2264, 8325	2505.3405	15	6.76-01	1.0
353	·SP(6.	3.4)	5	5	-1		2265.2192	2447.2420	6	2.19-01	*1.0
354	PPI	7.	3.41)		2	-1	42	2265.9074	2510.1062	6	1.70-01	1.0
355	991	7.	3.421	6	2	-1	41	2266.0417	2510.2404	14	1.66.01	1.0
356	QP (6.	2.6 1	5	5	0		2266.3461	2451.0090	. 5	8.78+00	1.0
357	OPI		0.41)	5	ō	0		2266.3849	2453.1551		1.12.01	1.0
358	GPI		1.€)	5	1	0	•	2266.3849	2452.6287		1.08-01	1.1
359	391	6.	3.4 1	5	3	0		2266.4775	2448.5003	6	1.66-01	\$1.0
360	QP (6.	4.E)	5	4	0		2266.4775	2444.7930	-10	5.91.00	1.0
361	190.	8.	7.E)	7	5	ī		2266.5283	1560.5679	5	1.83-92	1.0
362	39 (6.	5.E)	5	5	ō		2266.6559	2440.1856	-9	4.45.00	1.0
363	*0P(7.	3.621	6	ĩ	ĩ		2266.9226	2511.1214	16	4.49-04	1.0
364	.SP (6.	2.6)	5		-i		2267.0433	2451.7062	12	1.28.00	1.0
365	.OP(7.	3.41)	6	ī		42	2267.1587	2511.3575	-7	2.12-02	1.0
366	PPI		4.E)	6	3	-i		2267.3482	2507.8562		2.07.01	1.
367	991	6.	4.E 1	5	5	-i		2267.5363	2445.8517	-4		
368	·NP (7.	5.E)	6	2	ò		2267.9327	2503.6765	-	4.77-00	1.
369	901	6.	3.4 1	5	-	ĭ		2268.4933	2450.5161	-10		.1.
370	PPI		5.E)	6	-						7.06-00	•1.
P 1 P		7.			2	-1		2268.8791	2504.6224	11	2.17.01	1.
371	*0P1	7.	4.E)	5	3	1	Ē	2269.1087	2509.6167	-9	3.79-04	1.
	AP (***		5	1		2269.4945	2454.1574		5.23.00	.!.
373	PPI	7.	6.A)	6		-1		2270.1618	2500.0535	0	6.19.01	•1.
374	RPI		1.€	5	5	1		2270.5415	2456.7853	-4	7.73.00	1.
375	·NP (3.417	5	0	0		2271.1356	2453.1584	11	3.21-01	1.
376	•0P(7.	5.E 1	6	3	1	E	2271.2678	2507.0116	-5	8.82-04	1.0
377	.MP (7.	6 . A)	6	3	0		2271.3988	2501.2905	17	3.17.00	•1.
378	RP	6.	0.41)	5	1	1	75	2271.5619	2458.3321	10	1.92.01	1.
379	991	7.	7.E)	6	6	-1		2271.8281	2494.7625	5	3.97+01	1.
380	.0P(-	1.E)	5	1	-1		2272.4437	2458.6875	5	4.35-02	1.
381	99 (6.	1.6)	5	0	1		2272.8079	2459.0517	4	1.21.01	1.0
382	.0P(7.	6.A)		*	1		2273.5240	2503.4157	0	9.37-03	.1.
383	991	6.	5.E)	5	1	-1		2274.0285	2458.6914	8	1066.01	1.
394	.NP		4.E)	5	1			2274.3218	2452.6373	17	4.80-01	1.
385	.0P(6.	2.E)	5	0	1		2274.3912	2459.0541	7	1.63-02	1.
366	PPI	6.	3.451	5	2		AL	2275.2824	2457.3052	7	2.18.01	1.
387	PPI	6.	3.41)	5	2	-1	24	2275.3537	2457.3765	13	2.15.01	1.0
388	991	5.	(. A2)		0	0	41	2275.7352	2409.1895	10	1.26.01	1.
389	QP (5.	1 .E)	4	1	0		2275.7352	2408.66:1	3	1.21.01	1.
390	OP (5.	2.E)		5	0	E	2275.7352	2407.0741	-4	1.02.01	1.0
391	OPI	5.	3.A)		3	0		2275.8076	2404.4961	-4	1.70.01	.1.
392	QP (5.	4.E 1		4	0		2275.8946	2400.8614	-11	5.41.00	1.0
393	·SP (5.	2.6)			-1	Ē	2276.2528	2407.5917	16	6.48-01	1.0
394	*0P(6.	3.A1)	5	1	i	42	2276.3096	2458.3324	11	2.81-04	1.0
395	.OP(6.	3.A2)	5	ī	i	Al	2276.4781	2458.5009	-9	1.01-02	1.0
396	PPI	6.	4.E)	5	3	-1		2276.6869	2455.0023	11	2.75.01	1.0
397	·NP (6.	5.E)	5	2	0		2277.4814	2451.0111	4	4.64.00	1.0
398	ap (5.	3.A 1	4		1	Ā	2277.7275	2406.4160	-15	3.46.00	*1.0
399	PPI	6.	5.E)	5		-i		2278.1790	2451.7086	14	3.07.01	1.0
400	*0P1	6.	4.E 1	5	2	i		2278.4651	2456.7805	-9	2.00-04	1.0
401		5.	2.6)	-	3	i	È	2278.7455	2410.0844	-19	3.87+00	1.0
402	PPI	6.	6.A)	5	5	-i		2279.5986	2447.2476	12	8.75.01	•1.0
403	RPI	5.	1.€)	4	2	-i	Ē	2279.8058	2412.7317	-13	6.86.00	1.0
404	.NP	5.	3.42)		ō	ò		2280.4951	2409.1836	-13	1.81-01	1.0
405	.0P		5.E)	5	3	ĭ		2280.6261	2454.1558	-11	5.49-04	1.0



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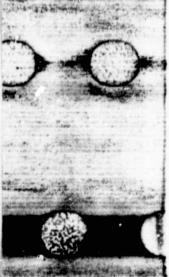
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TABLE I-Continued

	11			(11			(IV)	(V)	INT	(111)	(1114)
				;							
											*1.0
The second second											•1.0
	5.	2.E)									1.0
		2.E)	4								1.0
PPI	5.	3.41)	4	2	-1						1.0
PPI	5.	3.421		5	-1		2284.6167	2413.3052	9		1.0
3P (4.	0 . A1)		0	0	42	2284,9921	2373.9878	9		1.0
		1 .E)	3	1	0		2284.9921	2373.4577	0	1.22.01	1.0
-			3	2	0	-	2285.0186	2371.8921	-6	1.00-01	1.0
									-8	1.25.01	.1.0
									-		1.0
											1.0
				3	-1						1.0
				-	-1						1.0
			-	5	-;						1.0
			- 1	5	;						1.0
			3	2	i						1.0
		5.E)		ā							1.0
		0.A1)	3								1.0
PP(4.	1.E)	3	0	1						1.0
		2.E)		1			2292.5270		3		1.0
		2.E)					2292.8852	2379.7587	-7		1.0
		3.A)					2293.8236	2378.0381	4	6.31+01	.1.0
-						-			0	1.17.01	1.0
			2						-10	1.05+01	1.0
			- 5						-		1.0
											1.0
			- 1								1.0
											1.0
			5								1.0
											1.0
			2								1.0
			2								1.0
											-1.0
QP (2.	0.41)									1.0
QP (2.	1.E)	1		0	•					1.0
•101 ·	9.	3.E		5	0	E	2305.4363	2703.0234			1.0
				3	0	AZ	2307.6029	2999.0397	31	1.26-01	1.0
				13			2307.9975	3038.9840	-47	1.59-01	.1.0
				*	0	E	2308.0672	2795.7463	14	9.00-02	1.0
								2335.0353	-17	8.17-00	1.0
									-21	1.31-01	1.0
											1.0
										1.22.01	1.0
										2.49-03	.0
											1.0
											-1.0
											1.0
											1.0
											1.0
											0
											1.0
						-					1.0
			11	3		A1					1.0
	**************************************	**PP(6: PP(6:	**NP(6* 6* A) **PP(5* 1* E) **PP(5* 2* E) **OP(5* 2* E) **OP(5* 3* A2) **OP(5* 5* E) **OP(5* 3* A2) **OP(5*	**NP(6* 6* A) 5 **PP(5* 1* E) 4 **PP(5* 1* E) 4 **PP(5* 2* E) 4 **PP(5* 3* A1) 4 **PP(5* 3* A1) 4 **PP(5* 3* A2) 4 **PP(5* 5* E) 3 **PP(4* 2* E) 3 **PP(4* 2* E) 3 **PP(4* 3* A) 3 **PP(3* 1* E) 2 **PP(3* 1* E) 1 **PP(2* 1*	***P(6* 6* A) \$ 3 3 PP(5* 1* E) 4 0 PP(5* 2* E) 4 1 PP(5* 3* A 2) 4 2 PP(5* 3* A 2) 4 1 PP(6* 5* 5* E) 4 2 PP(6* 6* 6* A 2) 2 PP(6* 6* 6* A 2) 2 PP(6* 6* 6* A 3* A 3) 3 PP(6* 6* A 3* A 3* A 3* A 3* A 3* A 3* A 3	**NP(6* 6*A)	**NP(6* 6* A)	**NP(6* 6* A) 5 3 0 A 2280.8577 **PP(5* 1.E) 4 0 1 C 2282.86837 **OP(6* 6* A) 5 4 1 A 2282.86832 **PP(5* 2.E) 4 1 -1 E 2283.3129 **OP(5* 2.E) 4 0 1 E 2283.6832 **PP(5* 3.A1) 4 2 -1 A1 2284.6921 **PP(5* 3.A1) 4 2 -1 A1 2284.6921 **OP(5* 3.A1) 3 0 0 A2 2284.9921 **UP(4* 1.E) 3 1 0 E 2284.9921 **UP(4* 1.E) 3 1 0 E 2285.0781 **OP(5* 3.A2) 4 1 1 A1 2285.6219 **OP(5* 3.A2) 4 1 1 A1 2285.6219 **OP(5* 3.A1) 4 1 1 A2 2285.77343 **OP(5* 5.E) 4 2 0 E 2286.9225 **OP(5* 5.E) 4 2 1 E 2287.4325 **OP(5* 5.E) 4 2 1 E 2287.7679 **OP(5* 4.E) 3 3 1 E 2287.9445 **OP(5* 5.E) 4 2 1 E 2287.945 **OP(5* 5.E) 4 3 1 E 2289.014 **OP(5* 5.E) 4 3 1 E 2289.014 **OP(5* 5.E) 4 3 1 E 2289.014 **OP(5* 5.E) 4 3 1 E 2289.014 **OP(5* 5.E) 3 1 1 E 2292.8852 **OP(4* 2.E) 3 1 1 E 2292.8852 **OP(4* 3.A1) 3 1 1 A2 2292.8852 **OP(4* 3.A1) 3 2 -1 A 2293.8236 **OP(4* 3.A2) 2 0 0 A1 2294.1527 **OP(4* 3.A2) 3 0 1 E 2292.8852 **OP(4* 3.A2) 3 2 -1 A 2293.8236 **OP(4* 3.A2) 3 1 1 A2 2294.1527 **OP(3* 0.A2) 2 0 0 A1 2294.1527 **OP(3* 0.A2) 2 1 A 2293.8236 **OP(4* 3.A2) 3 1 1 A2 2294.852 **OP(3* 0.A2) 2 1 A 2293.8236 **OP(4* 3.A2) 3 1 1 A2 2294.1527 **OP(3* 0.A2) 2 1 A 2302.9531 **OP(3* 0.A2) 2 1 A 2302.9533 **OP(2* 0.A1) 1 0 0 A2 2303.2303 **OP(2* 0.A1) 1 0 0 A2 2303.2303 **OP(2* 0.A1) 1 0 0 A2 2303.2303 **OP(2* 0.A1) 1 1 A2 2308.3256 **OP(3* 0.A2) 2 1 A 2302.9531 **OP(3* 0.A2) 2 1 A 2302.9533 **OP(3* 0.A2) 2 2 A 2 2302.9533 **OP(3* 0.A2) 2 2 A 2 2302.9533 **OP	**NP(6* 6* A) 5 3 0 A 2280.8577 2448.5087 PP(5* 1.E) 4 0 1 C 2282.8683 2415.0096 0066 6* A) 5 4 1 A 2282.8683 2450.5192 PP(5* 2.E) 4 1 -1 E 2283.3129 2414.6518 0066 0066 5* 2.E) 4 0 1 E 2283.6832 2415.0221 PP(5* 3.41) 4 2 -1 A2 2284.5927 2413.2081 PP(5* 3.41) 4 2 -1 A2 2284.5927 2413.2082 2414.6518 0066 0066 0066 0066 0066 0066 0066 00	***P(6. 6. A 5 3 0 A	**MP(



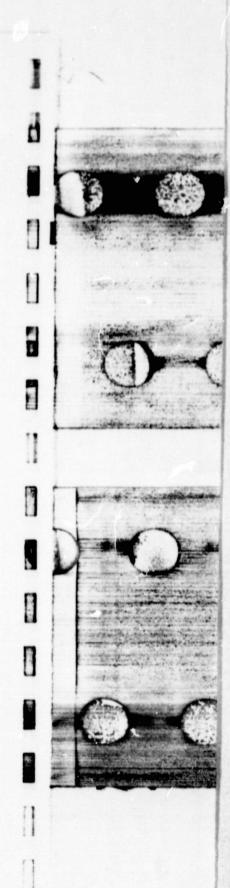


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TABLE I-Continued

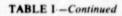
(1)	(11)	(111)	(14)	(V)	(IA)	(VII)	(VIII
	************		2411 0515				
	*RQ(12. 1.E)	15 5 0 €	2311.0515	3001.9803	. 6	1.28-01	1.0
467	00(13. S.E)	13 5 0 €	2311.1055	3115.1693	14	59-05	1.0
468	QQ(13. 6.A	13 6 0 A	2311.2544	3099.0365	-50	3.58-01	*1.0
469	HU(11-10-E)	11 11 1 €	2311.4535	2844.4931	35	2.78-01	1.0
470	HG(12. 8.E)	15 6 1 E	2311.4535	2969.8770	-58	3.19-01	1.0
471	*50(12. 4.E)	15 9 -1 E	2311.5581	2994.8369	159	3.56-03	
472	*TQ(10. 0.A1)	10 3 0 AZ	2311.7987	2799.9931	-51	2.22-01	1.0
473	. vo(8. 0.41)	8 5 -1 AZ	2311.8985	2631.7957	-14	5.3A-02	1.0
474	QQ(13. 5.E)	13 5 0 €	2311.9622	3105.3693	70	1.64-01	1.0
475	40(12. 7.E)	14 8 1 E	2312.0254	2978.2776	-12	3.03-01	1.0
476	RG(11. 9.A)	11 10 1 4	2312.0754	2855.2695	. 1	8.78-01	•1.0
477	00(12. 3.A)	15 3 0 W	2312.1756	2999.0313	43	5.15-04	*1.0
478	OP(1. 0.A2)	0 0 0 A1	2312.2150	2321.1193	-12	4.98-00	1.0
479	00(12. 6.A)	15 6 0 W	2312.5343	2985.5262	-50	5.34-01	*1.0
480	00(15 € €)	15 5 0 E	2312.5781	3001.9814	7	6.15-05	1.0
481	RO(11. 8.E)	11 9 1 E	2312.6670	2864.8695	-14	4.97-01	1.0
482	00(13. 4.E)	13 4 0 E	2312.8121	3110.7996	131	1.44-01	. (
483	RQ(10. 9.A)	10 10 1 A	2313.0957	2754.7267	38	9.07-01	.1.0
484	96(12. 5.E)	12 5 0 E	2313.0957	2991.7577	-2	2.23-01	1.0
485	RQ(11. 7.€)	11 8 1 6	2313.2383	2873.3289	-6	4.66-01	1.0
486	*TQ(8. 0.41:	8 3 0 AZ	2313.2016	2633.1767	18	2.44-01	1.0
487	*50(11. 3.A)	11 5 -1 A	2313.4568	2894.3050	6	1.04-02	•1.0
488	*RQ(10. 1.E)	10 5 0 €	2313.6337	2801.3128	-3	1.74-01	1.0
489	RG(10. 8.E)	10 9 1 €	2313.7396	2768.4427	8	6.72-01	1.0
490	RQ(11. 6.A)	11 7 1 4	2313.7578	2880.6388	-7	7.44-01	•1
491	Qu(12. 4.E)	12 4 0 E	2313.7578	2997.0366	-7	1.83-01	1.0
492	*SQ(13. 3.A)	13 5 -1 A	2313.8630	3115.3992	160	2.38-01	
493	00(11. 2:5)	1: 2 0 E	2313.9193	2897.3338	-16	1.80-03	1.0
494	*TU(7, 0.42)	7 3 0 A1	2313.9193	2562,8443	11	2.05-01	1.0
495	90(11. 5.E)	11 5 0 €	2314.2493	2886.8428	-11	2.63-01	1.0
496	**************************************	12 1 0 €	2314.2991	3003.7024	-8	1.83-01	1.0
497	RG(10. 7.E)	10 8 1 E	2314.3376	2776.9841	-9	6.65-01	1.0
498	*501 9. 4.E)	9 6 -1 E	2314.4351	2705.7710	-40	5.19-02	1.0
499	"TUI 6. 0.A1)	6 3 0 AZ	2314.5273	2501.2975	23	1.45-01	1.0
500	QQ(13.12.A)	13 12 0 4	2314.5674	3045.5539	10	6.85+00	•1.0
501	90(13-11-E)	13 11 0 €	2314.5674	3057.8879	-13	3.08+00	1.0
502	QQ(13.10.E)	13 10 0 €	2314.6196	3069.0887	-46	2.81.00	1.0
503	QQ(13.13.E)	13 13 0 E	2314.6196	3032.0532	10	3.86.00	1.0
504	RQ(9. 8.E)	9 9 1 6	2314.6966	2680.6549	38	6.58-01	1.0
505	*SO(12. 3.A)	12 5 -1 A	2314.7140	3001.5697	23	2.99-01	•1.0
506	QQ(13. 9.A)	13 9 0 A	2314.7768	3079.2401	-25	5.20.00	•1.0
507	00(11. 4.E)	11 4 0 E	2314.8049	2892.0496	-5	1.84-01	1.0
508	RQ(10. 6.A)	10 7 1 4	2314.8939	2784.3783	0	9.65-01	•1.0
509	QQ(13. 8.E)	13 8 9 6	2314.9851	3088.3158	ő	2.43.00	1.0
510	*TO(5. 0.42)	5 3 0 A1	2315.0577	2448.5120	17	7.94-02	1.0
511	00(13, 7.E)	13 7 0 €	2315.2445	3096.3409		2.29.00	1.0
512	RQ(9. 7.E)			2689.3037	28		
513			2315.3509	2790.5873	12	8.44-01	1.9
514			2315.3509		-13	2.57-01	1.0
			2315.4704	2954.8770	-24	4.66.00	1.5
515		15 11 0 E	2315.4704	2943.6353	-9	5.17.00	.! • 9
516	00(12. C.A.)	12 9 0 A	2315.5299	2965.0133	-43	8.54+00	.1.0
517	RG(13, 6.A)	13 7 1 4	2315.5555	3103.3376	52	4.34.00	•1.0
518	QQ(12.12.A)	12 12 0 A	2315.5555	2931.2824	18	1.17-01	•1.0
519	QQ(11. 3.A)	11 3 0 4	2315.5971	2896.4453	-11	2.89-01	•1.0
520	*50(9. 3.A)	9 5 -1 A	2315.6675	2710.6545	-58	1.42-01	•1.0
521	00(12. B.E)	15 8 0 E	2315.6956	2974.1190	-17	3.98+00	1.0
522	00(10. 4.E)	10 4 0 E	2315.8061	2795.7258	-5	1.20-01	1.0
523	*TO(3. 0.42)	3 3 0 41	2315.8651	2369.2751	-55	7.05-03	1.0



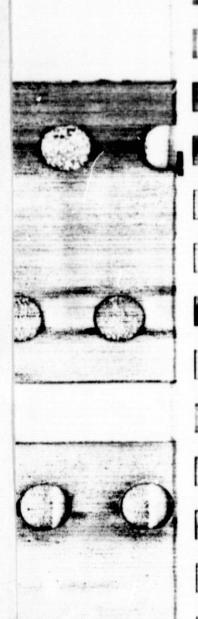
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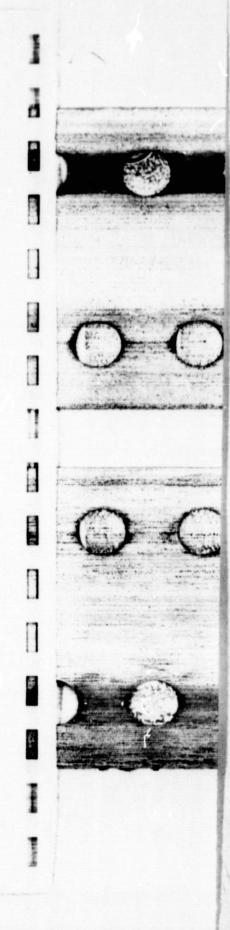
1)	(11)			111		(IV)	(V)	(VI)	(VII)	IVI
						2215 0017	2982.1449	-24	3.77-00	
524	00(12.		15	7	9 €	2315.8427	2696.7588	-6		.!
525	HQ (9.				1 4	2315.9244			1.16.00	•1
526	RG(13.		13	:	1 €	2315.9244	3109.3315	-25	9.47-02	!
527	•TG(10.	* * * * *	10	;	-1 E	2316.0288	2803.7079		7.21.00	-1
528	BG(12.		15		1 4	2316.1681	2989.1600	-27		.;
530	RO1 8.		11	8	1 E	2316.2347	2610.2744	18	1.35-01	•1
531	90(11.		ii	10	OE	2316.3686	2849.4082	31	7.45+00	i
532	90(11.		ii	11	OE	2316.3686	2838.0790	-25	8.42.00	i
533	RO(13.		13	'5	ı€	2316.3686	3114.3560	86	1.85+00	;
534	90(10.		10	3	0 4	2316.4374	2799.9853	-18	1.74-01	•1
535	00(11.		ii	8	OE	2316.4374	2868.6399		6.22+00	i
536	·50(13.		13		-1 E	2316.4374	3122.0147		8.16-02	i
537	80(12.		iż	6	i E	2316.5456	2995.2076	-17	3.47-00	i
538	00(11.		ii	7	0 E	2316.5836	2876.6741	-15	5.90+00	i
539	*TO: 9.				. i E	2316.7362	2715.8823	-11	8.68-02	i
540	901 9.		ě	4	OE	2316.7840	2708.1199		2.21-02	i
541	00(11.		11	6	0 4	2316.8296	2883.7106	-27	1.14.01	•i
542	ROL A.			7	1 4	2316.8650	2617.8260	-3	1.22.00	• i
543	RC(13.		13	4	i a		3118.4405	30	1.90.00	i
544	P.3(13.		13	-	i a		3118.5431	-1	2.66.00	i
545	RU(12.		12	5	iε	2317.0069	3000.2857	-23	3.27.00	i
546	00(10.		10	9	G A	2317.1403	2762.7713		2.05.01	•i
547	00(10.		10		OE	2317.1625	2771.8656	12	9.33+00	î
548	00(10.		10	10	OE	2317.1625	2752.5667	-15	1.16.01	i
549	RO(11.		ii	6	iE	2317.1923	2889.7858		5.54.00	i
550	00(10.	7.E 1	10	7	ÔE	2317.2579	2779.9044	-2	8.78-00	i
551	001 9.	3.A)		3	0 4	2317.2579	2712.2449	20	1.85-02	•1
552	901 8.				OE	2317.3165	2635.1264		1.56-01	i
553	901 8.	5.E)	8	5	OE	2317.3650	2624.1480	11	5.86-02	i
554	90(10.		10	6	0 4	2317.4724	2786.9568	-11	1.70.01	• i
555	RU(13.	-	13	3	1 E	2317.4724	3121.5362	20	1.29.00	i
556	.50(10.	2.E)	10	-		2317.5775	2803.7094		8.05-02	i
557	001 8.		. 8	4	0 E	2317.6255	2629.1485	-22	4.42-02	i
558	HU112.		15	4	1 4		3004.4812	-12	3.23.00	i
559	RO1 7.		7	7	i A	2317.7044	2547.5961	1	9.56-01	• i
560	90(11.		11	5	iε	2317.7044	2894.9491	-22	5.34.00	i
561	RQ(10.		10	6	iε	2317.8391	2793.0756	-12	8.38+00	i
562	001 9.		. 4		OE	2317.8654	2683.8237	13	1.35.01	i
553	901 9.		9		0 4	2317.8654	2674.6928	-25	3.05+01	•1
504	R0(12.	3.421	12	4	1 4		3004.7211	11	4.18-00	i
565	901 9.	7.E)	. 9	7	0 E	2317.9104	2691.8633		1.25.01	i
566	001 8.	3.A)		3	0 4	2317.9595	2633.1544	-1	1.19-01	• i
567			13		-1 E	2317.9595	3122.0233		8.69-01	i
568	001 9.		. 9	6	0 4	2318.0794	2698.9138		2.42.01	•1
569	RQ(13.	1.E)	13	S	1 €	2318.1557	3123.7330		1.46+00	i
570	901 7.	2.6)	7	2	0 E	2318.2327	2565.0598		4.09-01	i
571	901 7.	5.E)	7	5	OF	2318.2327	2553.9765	4	1.09-02	i
572	80(11.	3.41)	11	4	1 4		2899.1275	12	3.09+00	i
573	20(12.		iż	3	iε	2318.2950	3007.6983		2.76.00	i
574	RO(10.	4.E)	10	5	iξ	2318.3570	2798.2767	-22	8.17.00	i
575	HO(11.	3.421	ii	4	i A		2899.2052	-9	5.22+00	i
576	Quí 7.	1.E)	* 7	ī	ÔĒ	2316.3570	2566.7579		1.26-01	i
577	RO 1 9.	5.E)	9	:	ΪĒ	2318.4260	2705.0490		1.20-01	i
578	001 7.		7	640	Ĉ €	2318.4542	2558.9622		6.23-01	i
579	PP (1.	1.5)	ó	0	1 E	2318.4978	2326.8688		8.90.00	i
580	901 8.	7.E 1		7	ÖE	2318.5135	2612.5531	-3	1.70-01	i
581	UU (8.			8	OE	2318.5135	2604.5120		1.91-01	i



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TABLE 1-Continued

(1)	(11)	- (11	1)		(IV)	(4)	(VI)	(111)	(VIII
	*******					****	**********			*********	
582	901 8.		8	6	0		2318.6395	2619.6005	13	3.25.01	.1.0
583	961 7.		7	3	0		2318.6395	2562.8383	7	1.15 00	*1.0
584	•NQ(11.	3.41)	11	0	0	45	2318.8816	2899.7298	-30	3.44+00	1.0
585	•Ru(12.	3.E	12	3	-1	Ε	2318.8816	3008.2849	-55	6.94-01	1.0
586	*SO! 8.		8	4	-1	€	2318.9364	2636.7463	7	3.04-03	1.0
587	401 B.			6	1	E	2318.9635	2625.7466	9	1.61.01	1.0
588	HQ1 9.	4.E)	9	5	1	E	2318.9635	2710.2994	-53	1.17.01	1.0
589	991 6.	2.E)		5	0	E	2319.0148	2503.6777	7	9.71-01	1.0
540	RQ(6.	5.E)	6	6	î	E	2319.0328	2492.5625	1	4.96-01	1.0
591	RU(10.	3.A)	10	4	1		2319.0328	2802.5807	0	1.41.01	.1.0
592	80(12.	1.E)	12	5	1	E	2319.0574	3009.9862	-33	2.49.00	1,0
593	991 7.	7.E)	7	7	0	E	2319.0880	2542.0224	-14	2.25.01	1.0
594	991 6.	1.E)	6	1	0	E	2319.0880	2505.3318	6	2.89-01	1.0
595	R0(11.	2.E)	11	3	1	E	2319.0880	2902.5025	-21	4.80.00	1.0
596	991 7.	6.A)	7	6	0		2319.1425	2549.0342	5	4.14.01	.1.0
597	991 6.	4.E)	6	4	0	E	2319.1782	2497.4937	-14	2.79.00	1.0
598	201 6.	3.A)	6	3	0		2319.2571	2501.2799	7	4.14.00	.1.0
599	901 7.	5.E)	7	6	1	E	2319.4292	2555.1730	9	2.03-01	1.0
600	.50(11.	0.A2)	11	2	-1	Al	2319.4699	2904.9330	-17	3.54-04	1.0
601	RG(8.	4.E)	8	5	1	E	2319.5483	2631.0713	1	1.56.01	1.0
605	991 6.	6.A)	6	6	0		2319.5859	2487.2369	-16	5.01.01	.1.0
603	001 5.	2.E)	5	2	0	E	2319.6663	2451.0052	0	2.12.00	1.0
604	RQ1 9.	3.A)	9	4	1	۸	2319.7018	2714.6888	-3	2.18.01	.1.0
635	901 5.	1.E)	5	1	0	E	2319.7018	2452.6277	7	5.94-01	1.0
206	RQ(13.	(SA.0	13	1	1	AL	2319.7402	3125.8216	46	4.56.00	1.0
607	901 5.	3.A)	5	3	0		2319.8180	2448.5065	12	1.05.01	.1.0
608	001 5.	4.E)	5	4	0	E	2319.8180	2444.7848	-19	8.90+00	1.0
609	991 61	5.E)	6	5	0	Ē	2319.8180	2493.3477	10	2.33.01	1.0
610	RQ(10.	2.6)	10	3	1	E	2319.8540	2805.9858	-2	7.51.00	1.0
611	RQ(11.	1.E)	11	2	1	Ē	2319.9350	2904.8863	-6	4.08-00	1.0
612	991 5.	5.E)	5	5	0	E	2320.0202	2440.1826	-12	2.60.01	1.0
613	HQ (7.	4.E)	7	5	1	E	2320.0521	2560.5601	-1	1.90.01	1.0
614	•NU(10.	3.A2)	10	0	0	41	2320.0902	2803.6381	-18	2.39.00	1.0
615	301 4.	1.E)	4	1	0	E	2320.2010	2408,6666	9	1.11.00	1.0
616	991 4.	2.E)	4	5	0	Ē	2320.2010	2407.0745	-4	4.33.00	1.0
617	991 4.	3.A)	4	3	0	Ā	2320.2812	2404.4957	-5	2.21.01	.1.0
618	RQ (8.	3.A 1	8	4	1		2320.3190	2635.5139	-2	2.98.01	.1.0
619	991 4.		4	4	ō	E	2320.3791	2400.8598	-12	2.49.01	1.0
620	*NQ(12.	4.E)	12		0	E	2320.4096	3003.6884	-22	9.05-01	1.0
621	·50(10.		10	5	-1	54	2320 396	2808.6040	-12	2.33-04	1.0
622	RQ(6.		6	5	i	E	2320.4979	2498.8134	-5	2.02.01	1.0
623	RQ1 9.		9	3	i	E	2320.5652	2718.1523	6	1.09.03	1.0
624	991 3.	1.E)	3	ī	ō	F	2320.5652	2373.4437	-14	1.99.00	1.0
625	RQ(12.		12	i	i	42	2320.5652	3012.0020	33	7.55+00	1.0
626	991 3.	2.E)	3	ż	ò	E	2320.6078	2371.8900	-8	8.38.00	1.0
627	001 3.	3.A)	3	3	0	Ā	2320.6680	2369.2844	-13	4.25.01	.1.0
628	RQ(10.	1.E)	10		ī	Ē	2320.7475	2808.4266	6	6.39.00	1.0
629	901 2.	1.€)	. 5	5	ō	Ē	2320.8637	2347.0409	0	3.59.00	1.0
630	001 2.		2	ż	0	È	2320.8817	2345.4595	-18	1.53.01	1.0
631	RQ (5.	4.E)	5	5	ī	Ē	2320.8817	2445.8485	-7	1.62.01	1.0
632	RQ (7.	3.4)	7	4	i		2320.8817	2565.0805	-3	3.68+01	-1.0
633	901 1.	1.E)	i	1	ė	E	2321.0414	2329.4124	-9	7.48-00	1.0
634	·501 7.	1.€)	7	i	-1	È	2321.0792	2569.4801	5	6.38-03	1.0
635	.21100.	1.€)	12	ĩ	-i	È	2321.1105	3012.0393	-5	8.67-01	1.0
636	•NO (9.	3.41)	9	ò	0	42	2321.2184	2716.2054	-7	1.94+00	1.0
637	RQ(8.	2.6)	é	3	ĭ	Ē	2321.2184	2639.0283	-5	1.47.01	1.0
638	·501 9.		9	ž		Al	2321.2720	2720.9346	-2	1.95-03	1.0



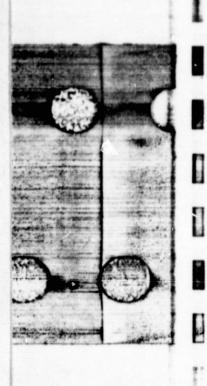
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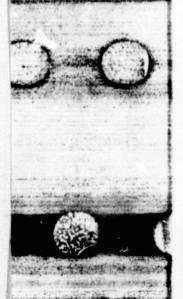
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TABLE 1-Continued

1)	(11			111)	(IV)	(V)	(VI)	(VII)	(VI
640	RQI 6.			4 1 4	2321.3826	2503.4054	-9	4.07.01	•1
641	PQ(12.	1.6)	12	0 1 6	2321.3826	3012.3114	0	4.10.00	i
642	RQ(9.	1.6)		2 1 6	2321.4977	2720.6418	10	9.46.00	i
643	*501 6.	1.E)		3 -i E	2321.6109	2507.8547	7	4.42-05	i
644	•NO(13.	5.E 1	13	3 0 6	2321.7751	3115.1822	27	8.57-01	i
645	RUI 5.	3.A 1	'5		1559.1252	2450.5107	-16	3.86.01	• i
646	RQ1 7.		7	3 1 6	2321.8221	2568.6492	-10	1.84.01	i
647	*NO(11.	4.E)	11	1 0 €	2321.8453	2899.0900	-5	1.11.00	i
648	.00111.	1.E 1	ii	1 -1 €	2321.8882	2906.8395	-3	6.94-01	i
649	PQ(13.	3.421	13	2 -1 A1	2321.9194	3123.4556	-1	2.82+00	i
650	.SO(A.			2 -1 A2	2322.0452	2641.9424	i	3.89-03	1
651	.00(13.		13	0 1 €	2322.0830	3126.1468	3	5.63-01	i
652	.501 5.	1.E)	5	3 -1 €	2322.0830	2455.0089	18	1.63-03	i
653	RQ(10.	0.A11	10	1 1 A2	2322.1109	2810.3053	12	1.76+01	1
054	RO(4.	3.A)	4	4 1 4	2322.1951	2406.4096	-55	2.68+01	
655	PQ(11.	1.E)	11	0 1 E	2322.1951	2907.1464	5	6.94+00	1
656	RQ (8.	1.E)		2 1 6	2322.1951	2641.5708	16	1.32.01	1
657	•NO (B.	3.421		0 0 A1	2322.2250	2637.4199	5	1.57+00	1
658	RG(6.	2.E)	6	3 1 E	2322.3499	2507.0126	-4	2.12.01	1
654	·50(4.	1 .E)	4	3 -1 €	2322.4721	2410.9377	13	4.27-03	1
990	P0(12.	5.E)	15	1 -1 €	2322.6421	3012.0453	0	3.31-00	1
661	·501 7.		7	2 -1 A1	2322.7331	2571.6581	3	4.47-03	1
662	HQ (9.	0.A2)	9	1 1 A1	2322.8121	2722.4747	1	2.47-01	1
663	RO(5.		5	3 1 €	2322.8121	2454.1510	-16	2.21.01	1
664	RO1 7.	1.E)	7	5 1 E	2322.6121	2571.2130	3	1.72-01	1
665	.00(12·	5.E)	15	0 1 €	2322.9038	3012.3071	-4	4.89-01	1
666	PG(10.	1.€	10	0 1 E	2322.9860	2810.6651		1.05-01	1
667	P0(12+	3.A1)	15	5 -1 WS	2323.0412	3009.8969	-10	4.46-00	1
668	•00: 9.	2.E)	7	5 1 E	2323.0647	2720.6518	50	1.11-01	1
670	•NG(7.	3.A1)	10	0 0 A2	2323.0990	2567.2978	-2	1.16.00	1
671	RQ (4.	2.6	4	1 0 E	2323.2125	2410.0860	-17	2.00.01	1
672	•NQ(12.		12	3 0 5	2323.3170	3001.9790	5	1.28.00	1
673		D.A1)	6	2 -1 A2	2323.3348	2510.1050	5	3.40-03	1
674	RU(6.	1.E)	6	2 1 E	2323.3746	2509.6184	5	2.08.01	i
675	PQ(11.	2.E)	11	1 -1 E	2323.4254	2906.8399	-3	5.79+00	i
676	HO! 8.	0.41)		1 1 42	2323.4676	2643.3647	0	3.28-01	i
677	PQ (3.	2.E)	3	3 1 €	2323.5303	2374.8126	-26	1.33-01	i
678	PUI 9.	1.€)	9	OIE	2323.7322	2722.8763	11	1.47-01	i
679	RQ1 5.	1.E)	5	2 1 6	2323.8539	2456.7748	-10	2.32.01	i
	.00(12.	3.421	12	1 1 41	2323.8539	3010.7097	-31	1.37.00	i
681	PQ(13.		13	3 -1 €	2324.0331	3122.0206	10	1.62.00	i
662	PG(11.	3.421	11	2 -1 A1	2324.0662	2904.9144	-35	6.72+00	i
683	RU(7.		7	1 1 41	2324.0662	2572.9912	5	4.11.01	1
684	PQ(10.		10	1 -1 E	2324.2012	2810.3330	3	9.04+00	i
685	RO1 4.	1.E)	4	2 1 E	2324.2755	2412.7411	-4	2.34+01	1
686	•NO (9.		9	1 0 €	2324.3074	2715.6433	0	1.36.00	1
687	PQ (8.	1.E 1	8	0 1 6	2324.4252	2643.8009	10	1.92+01	1
688	.NO! 5.	3.41)	5	0 0 42	2324.4634	2453.1519	5	3.88-01	1
689	.00(10.		10	0 1 E	2324.5331	2810.6649	9	3.23-01	1
640	RQ (3.	1.E)	3	2 1 €	2324.5895	2377.4680	-26	2.06.01	1
691	RO1 6.		6	1 1 A2	2324.5895	2511.3597	-4	4.84+01	1
692	-001 7.		7	1 -1 €	2324.7011	2573.1020	7	1.39-01	1
693	•NO(11.		11	5 0 E	2324.7525	2897.3460	-4	1.82.00	1
694	Pu(11.		11	2 -1 A2	2324.7944	2905.6426	-25	2.82.00	1
695	BO1 5.	1.6 1	5	5 1 E	2324.8488	2351.0260	-51	1.36-01	1
696	PO (9.		4	1 -1 €	2324.9394	2722.5265	3	1.29-01	1
697	PO(12.	3.421	16	2 -1 A1	2324.9394	3011.7951	-10	2.04.00	1



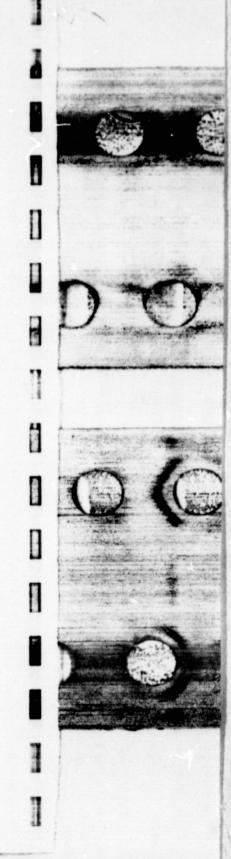


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TABLE I-Continued

(1)	(11			(111)		(IV)	(V)	(VI)	(VII)	(VIII
698	PQ(12.		12	3 -1		2325.0139	3008.2927	-14	3.03+00	1.0
699	A01 5.		'5		AL	2325.0500	2458.5043	-6	5.33.01	1.0
700	PQ(10.	3.41)	10	2 -i		2325.0500	2808.5979	-18	9.62.00	1.0
			7	0 i		2325.0500	2573.4509	-10	2.35.01	1.0
701	PQ (7.	1.€						34	2.34-02	
702	.00(15.	3.41)	15	1 1		2325.1477	3012.0034			1.0
703	.001 6.	1.E		1 -1		2325.2806	2511.5244	50	7.08-02	1.0
704	.001 9.		*	0 1		2325.2806	2722.8677	. 3	2.47-01	1.0
705	.MO! 8.	4.E)		1 0		2325.3554	2636.8784	15	1.28-00	1.0
706	801 4.	O.A1)		1 1		2325.4324	2414.4281	-14	5.42.01	1.0
707	PG (8.	5.E)		1 -1		2325.6304	2643403	0	1.72.01	1.0
708	PQ1 0.	1.E)	6	0 1	E	2325.6304	2511.8742	8	5.60.01	1.0
	.00:11.	3.41)	11	1 1		2325.6304	2906.4786	4	2.62.00	1.0
710	PG(10.	3.42)	10	5 -1		2325.6675	2809.2154	-4	5.65+00	1.0
711	RQ (3.	0.42)	3	1 1	41	2325.7450	2379.1550	-17	5.02.01	1.0
712	PG(11.	4.E)	11	3 -1	E	2325.9528	2903.1975	-20	4.98+00	1.0
713	PQ (9.	3.42)	9	2 -1	Al	2325.9528	2720.9398	5	1.30+01	1.0
714	RQ (2.	0 . A1)	5	1 1	42	2325.9852	2352.6949	-15	4.08+01	1.0
715	.NQ(12.	6.A)	12	3 0		2326.0598	2999.0517	63	3.97+00	.1.0
716	•NO(10.	5.E)	10	2 0	E	2326.0805	2801.3170	0	2.44+00	1.0
717	RO(1.	(SA.0	1	īi		2326.1322	2335.0365	-28	2.67.01	1.0
718	PQ (5.	1.€)	5	0 1		2326.1322	2459.0581	11	2.88.01	1.0
	-001 4.	1.E)	4	i -i		2326.1920	2414.6576	14	5.69-03	1.0
720	•NG (7.	4.E)	7	i o		2326.2487	2566.7567	9	1.03.00	1.0
721	PQ1 7.	2.6 1	7	i -i		2326.2769	2573.1040	9	2.13.01	1.0
722	·00(10.	3.42)	10	i i		2326.3236	2809.8715	o	2.55.00	1.0
723	PQ (9.	3.41)		2 -1		2326.4273	2721.4143	3	9.92.00	1.0
724	·001 3.	1.E)	3	î -i		2326.5131	2379.3916	-5	2.45-03	1.0
725		1.6	:	0 i		2326.5428	2415.0084	-1	2.86+01	1.0
726			7	0 1			2573.4570	5		
	*001 7.	2.E)				2326.6299		0	1.14-01	1.0
7. 7	PQ(12.		12			2326.6765	3005.3385	_	2.62+00	1.0
	.00(15.		15	5 1		2326.7198	3009.9986	-20	1.95-01	1.0
729	PQ(8.				AZ	2326.7485	2641.9434	. 5	1.67-01	1.0
730	PQ(10.	4.E)	10	3 -1		2326.826	2806.7457	-14	7.55+00	1.0
731	PQ (6.		0	1 -1		2326.8551	2511.5180	13	2.45+01	1.0
732	PU (3.	1.E)	3	0 1		2326.8812	2379.7597	-6	2.59.01	1.0
733	·00(1.	1.E)	1	1 -1		2326.9398	2335.3108	-11	1.59-04	1.0
734	.NO! 6.			1 0		2327.0229	2505.3383	13	6.67-01	1.0
735	.001 9.	3.41)	*	1 1		2327.0654	2722.0524	19	1.69+00	1.0
736	PQ (8.	3.421		5 -1		2327.0925	2642.2874	9	1.46+01	1.0
737	PO(2.	1.€)	2	0 1		2327.1436	2353.3208	-4	2.08.01	1.0
738	.001 6.	5.E)	6	0 1		2327.2108	2511.8737	H	6.68-02	1.0
739	. 401 9.	5.E)	*	5 0		2327.2834	2713.9064	3	3.04+00	1.0
740	PQ(1.	1.E)	1	0 1	E	2327.3158	2335.6868	-6	1.34.01	1.0
7+1	PO (5.	2.E)	5	1 -1	E	2327.3520	2458.6909	8	2.61.01	1.0
742	PQ(11.		11	5 -1	A	2327.4336	2894.3146	16	6.37.00	•1.
743	PG (7.	3.421	7	2 -1		2327.4576	2571.6564	5	2.01.01	1.0
744	.001 9.		9	1 1		2327.4919	2722.4789	5	5.00-02	1.0
745	PQ(9.			3 -1		2327.6186	2718.9545	-9	1.07.01	1.
746	PQ(11.		11	4 -1		2327.6186	2900.2121	-13	4.06.00	1.
747			ii	2 1		2327.6414	2904.8861	-6	2.48-01	i.
748	PQ (7.		7	2 -i		2327.6897	2571.8885	16	1.87+01	1.0
749	PO (4.		4		Ē	2327.7815	2414.6550	11	2.54.01	1.
750			6	2 -1		2328.0873	2510.1101	10	2.24.01	1.
751			3					5		
	PQ(3.					2328.1207	2379.4030		2.16.01	1.5
752	.001 8.			1 1		2328.1757	2643.3706	. 6	5.14-02	1.0
753	PG(6.		6		AL	2328.2148	2510.2376	11	2.16.01	1.5
754	PUI 8.	5.E)				2328.3380	2639.8610	16	3.45-00	1.0
							2435.1566			



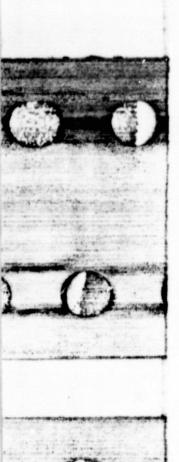
V, AND V3 OF PH3

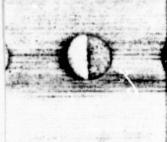
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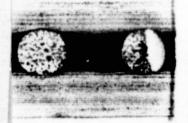
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TABLE 1-Continued

		*****	****	***	•••	***	***********	*********	*****	********	******
(1)	(11			11			(1)	(V)	(VI)	(VII)	(VIII
756			2	1	-1		2328.3735	2352.9513	-3	1.39+01	
757	PQ(10.		10	:	-;		2328.4803	2803.7167	-		1.0
758			10	2	-i		2328.5055	2808.4252	-16	5.91.00	1.0
759			12		-i		2328.5899			2.60-01	1.0
760				5				2994.8421	164	2.79-00	
761	PQ (5.		12	5	-1		2328.5899	3001.5818	35	3.48-00	*1.0
762	PQ (5.		5	2	-1		2328.6165	2457.3050	.7	2.28.01	1.0
763				5			2328.6962	2457.3847	51	2.24.01	1.0
764			10		-1		2328.6962	2798.1806	-6	9.77-00	*1.0
765			7	3	1		2328.7749	2572.9737	-14	4.60-02	1.0
766			4		-1		2328.9704	2569.4784	0	1.69-01	1.0
767			:	5	-!		2329.0656	2413.2802	12	2.01.01	1.0
768		3.A2		2	-!		2329.0908	2413.3053	9	2.00.01	1.0
	PQ (9.		i		-1		2329.2599	2715.8829	-10	8.05-00	1.0
770			3	5	0		2329.3171	2565.0609	6	3.42.00	1.0
771	PQ (3.			5	-1		2329.4250	2378.0414	. ?	2.64-01	*1.0
772	PQ (6.			3	-1	-	2329.5439	2507.8594	11	1.84-01	1.0
	•NO(11+	6.A)	11	3	0	-	2329.5690	2896.4500	-6	4.70-00	.1.0
773		3.41)	. 5	1	1		2329.6392	2458.3277	. 6	5.79-02	1.0
774	•NG(13.	7.E)	13	1	0		2329.7124	3110.8087	140	6.23-01	.0
775	P0(11.	7.E)	11	-	-1		2329.7669	2889.8575	47	4.39.00	1.0
776	PQ(9.	6.A)	4	5	-1		2324.8294	2710.6638	-19	1.41.01	.1.0
777	OH (0.	0.41)	1	0	0		2329.9466	2329.9466	3	5.32+00	1.0
778	PO1 8.	5.E)	8	4	-1		2329.9466	2636.7296	-9	1.02.01	1.0
779	PO 1 5.	4.E)	5	3	-1		2330.0401	2455.0069	16	1.72.01	1.0
	·001 4+	3.421	4	1		Al	2330.0954	2414.3044	-4	1.95-02	1.0
76	.NO! 6.	5.E)	6	5	0		2339.1457	2503.6754	5	2.71.00	1.0
782	.001 4.	3.41)	4	1	1	45	2330.2047	2414.4193	-23	1.11-02	1.0
783	PQ (4.	4.E)	*	3	-1		2330.4606	2410.9413	17	1.17.01	1.0
	•N0(10.	6.4	10	3	0		2330.5224	2800.0068	3	5.61.00	.1.0
785	PQ (7.	5.E)	7	4	-1	E	2330.5546	2566.2984	-3	1.17.01	1.0
786	·04(13.	6.42)	13		1	Al	2330.6468	3118.4289	18	1.65-01	1.0
787	·00(7.	4.E)	7	5	1	E	2330.7076	2571.2156		1.12-01	1.0
788	.00(10.	5.E 1	10	3	1	E	2330.7583	2805.9948	6	2.53-01	1.0
789	•00(13.	6.41)	13	4	1	42	2330.7583	3118.5404	-3	1.41-01	1.0
790	PO(10.	7.E 1	10	6	-1		2330.8383	2793.4846	-22	6.26.00	1.0
791	•NQ (5.	5.E)	5	5	0	E	2330.8383	2451.0006	-5	1.35+00	1.0
792	PU (8.	6.A)	8	5	-1	Ā	2330.8383	2631.7993	-9	1.84+01	.1.0
793	PO(12.	8.E)	12	7	-1	E	2330.9676	2989.3911	362	3.07.00	.0
794	PQ (6.	5.E)	6	4	-1		2331.0921	2504.6218	10	1.18-01	1.0
795	.001 6.	4.E 1	6	2	1	E	2331.3001	2509.6156	0	6.09-02	1.0
796	•NO (9.	6 . A)	9	3	0		2331.3963	2712.2307	5	5.66+00	.1.0
797	.51100	6.41)	12		1	42	2331.4953	3004.4872	-6	2.05-01	1.0
798	PO(5.	5.E)	5	4	-1		2331.5394	2451.7018	7	8.74-00	1.0
799	PQ (7.		7		-1		2331.6859	2561.5776	-12	2.03-01	•1.0
800	.00(12.		12	4		A1	2331.7239	3004.7158	6	1.36-01	1.0
801	PQ (9.	7.E)	9	6	-i		2331.8338	2705.7867	-24	7.90-00	1.0
802	PO(11.	8.E)	11	7	-i		2331.8936	2884.0961	119	4.42.00	.0
803	•NG(11.	7.E)	ii	4	0		2331.9643	2892.0548	0	7.14-01	1.0
804	•NQ (B.			3	o		2332.2105	2633.1715	15		1.0
805	•00(11.		11	4		AZ	2332.2437	2899.1247	9	1.85-01	-1.0
806	•00(11.		ii	-					-9		1.0
807				5		A1	2332.3242	2899.2052		2.25-01	1.0
808	PQ(6.		6		-1		2332.3933	2500.0443	-9	1.57.01	•1.0
	PO (8.	7.E)	. 6	6	-1	E	2332.6982	2626.7379	-24	8.39+00	1.0
809	PO(10.		10	7	-1	E	2332.8318	2787.5349	50	5.66.00	1.0
810	•001 7.		?	3	1	E	2332.8980	2568.6418	0	8.86-02	1.0
611	•NO (7.		7	3	0		2332.9584	2562.8501	19	2.48.00	.1.0
612	.00(10.		10	4	1		2333.1123	2802.5967	15	4.12-01	.1.0
813	PG(7.	7.E)	7	5	1	E	2333.4523	2556.3867	-13	6.38+00	1.0



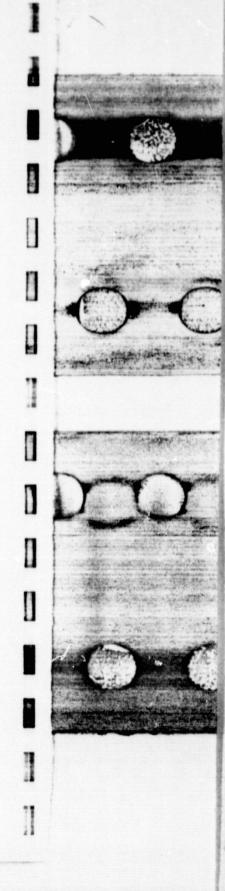




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TABLE I-Continued

(1)	(11)	(111)	(IV)	(V)	(AI)	(VII)	(VIII
	***************************************		2333.6361	2501.2871	14	7.38-01	
	*NG(6. 6.4)	6 3 0 A			14		*1.0
815	PG(9. 8.E)	9 7 -1 E	2333.7109	2699.6692	-17	6.16.00	1.0
	*001 9. 6.A)		2333.8654	2714.6948		3.41-01	*1.0
817	PQ(11. 9.A)	11 8 -1 A	2334.0063	2877.2003	216	7.62.00	
818	*00(12. 7.E)	12 5 1 €	2334.0558	3000.3080	0	1.54-01	1.0
819	•NO! 9. 7.E)	9 4 0 E	2334.1533	2708.1061	8	3.70-01	1.0
820	PG(8. 8.E)	8 7 -1 E	2334.5127	2620.5112	-53	4.82.00	1.0
821	*001 8. 6.A)	8 4 1 A	2334.5591	2635.5201	. 3	2.32-01	•1.0
822	PQ(10. 9.A)	10 8 -1 A	2334.8277	2780.4587	61	8.57-00	*1.0
823	*00(11. 7.E)	11 5 1 €	2334.8690	2894.9595	-12	1.68-01	1.0
824	PR(0. 0.41)	1 1 1 A2	2335.0351	2335.0351	-17	1.86-01	1.0
825	•NO! 8. 7.E)	8 4 0 E	2335.1245	2629.1642	-6	1.77-01	1.0
826	*00(7. 6.A)	7 4 1 4	2335.1951	2565.0868	5	1.18-01	*1.0
827	PQ(9. 9.A)	9 8 -1 A	2335.6441	2692.4716	3	6.96.00	*1.0
828	*00(6. 6.A)	6 4 1 4	2335.7674	2503.4184		3.77-02	•1.0
829	PQ(11+10+E)	11 9 -1 €	2336.2282	2869.2679	373	2.86.00	.0
	*001 9. 7.E)	9 5 1 E	2336.3711	2710.3239	!	1.20-01	1.0
831	PG(10.10.E)	10 9 -1 €	2336.9104	2772.3146	121	2.41.00	0
832	*00(8. 7.E)	8 5 1 6	2337.0411	2631.0807	11	7.07-02	1.0
833	*001 7. 7.E)	7 5 1 €	2337.6196	2560.5540	-8	2.54-02	1.0
834	*NO! 8. 8.E)	8 5 0 E	2338.1362	2624.1347	-1	5.65-03	1.0
835	PG(11-11-E)	11 10 -1 E	2338.5388	2860.2492	538	1.60.00	0
836	GR(1. 0.A2)	5 0 0 VI	2338.6585	2347.5628	0	1.03-01	1.0
837	QR(1. 1.E)	5 1 0 E	2338.6585	2347.0245	-11	7.96.00	1.0
838	.00(9. B.E)	9 6 1 6	2339.1105	2705.0688	16	5.95-02	1.0
839	PQ(12.12.A)	12 11 -1 A	2340.4931	2956.2200		2.05.00	0
840	RP(1. 1.E)	5 5 1 E	2342.6547	2351.0257	-51	2.67+01	1.0
841	RR(1. 0.A2)	5 1 1 VI	2343.7531	2352.6574	-14	2.67.01	1.0
842	PR(1. 1.E)	3 0 1 6	2344.9455	2353.3165	-8	4.38.00	1.0
843	OR(2. 0.41)	3 0 0 45	2347.2717	2373.9814	5	1.44.01	1.0
844	OR(2. 1.E)	3 1 0 E	2347.2717	2373.4489	-9	1.32.01	1.0
845	GR! 2. 2.E)	3 5 0 €	2347.3119	2371.8857	-8	9.21.00	1.0
646	*Si 1 2. 1.E)	3 3 -1 €	2349.4976	2375.6748	15	2.37-02	1.0
847	AR(2. 2.E)	3 3 1 €	2350.2392	2374.8130	-25	4.03-01	1.0
848	. TAL 3. 0.42)	4 3 0 A1	2351.0758	2404.4858	-15	2.37-02	1.0
849	481 2. 1.E)	3 2 1 E	2351.2984	2377.4756	-19	2.72.01	1.0
	*SR(2. 0.A1)	3 2 -1 A2	2351.3235	2378.0332	0	4.27-02	1.0
851	RP(2. 0.41)	3 1 1 A2	2352.3779	2379.0876	-7	3.25.01	1.0
852	*OR(2. 1.E)	3 1 -1 E	2353.2234	2379.4006	3	8.11-03	1.0
853	PRI 2. 1.E)	3 0 1 E	2353.5843	2379.7615	-4	7.85.00	1.0
854	PH (2. 2.E)	3 1 -1 E	2354.8167	2379.3945	-2	2.74.00	1.0
855	QR (3. 0.A2)	4 0 0 41	2355.7855	2409.1955	16	1.72.01	1.0
856	OR(3. 1.E)	4 1 0 E	2355.7855	2408.6640	6	1.67.01	1.0
857	UP (3. 2.E)	4 2 0 E	2355.7855	2407.0677	-10	1.42.01	1.0
858	OP (3. 3.A)	4 3 0 4	2355.8856	2404.5020	1	2.31.01	*1.0
859	*SP(3. 2.E)	4 4 -1 E	2356.3107	2407.5930	17	9.46-01	1.0
860	ARI 3. 3.4)	4 4 1 A	2357.7998	2406.4162	-15	9.57.01	.1.0
861	*SR(3. 1.E)	4 3 -1 E	2358.0645	2410.9430	19	1.38-01	1.0
862	HR (3. 2.E)	4 3 1 E	2358.8061	2410.0883	-15	3.60+01	1.0
863	*TR(4. 0.41)	5 3 0 A2	2359.4919	2448.4876	-6	6.38-02	1.0
864	RR(3. 1.E)	4 2 1 E	2359.8568	2412.7353	-10	2.68.01	1.0
865	*SR(3. 0.A2)	4 2 -1 A1	2359.9070	2413.3170	51	1.69-01	1.0
866	HP(3. 0.42)	4 1 1 A1	2360.9062	2414.3162	i	3.53+01	1.0
867	*QR(3. 1.E)	4 1 -1 E	2361.7751	2414.6536	10	2.64-02	1.0
868	PR(3. 1.E)	4 0 1 6	2362.1346	2415.0131	3	1.00-01	1.0
869	PR(3. 2.E)	4 1 -1 €	2363.3699	2414.6522		5.33.00	1.0
870	.OH (3. 5.E)	4 0 1 6	2363.7363	2415.0100		1.31-02	1.0



V, AND V3 OF PH3

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TABLE I-Continued

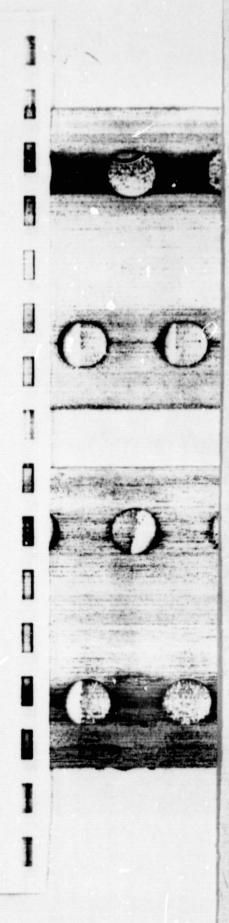
	33.	0.A1) 1.E 3.A 4.E 3.A1) 3.A2) 2.E 4.E 1 1.E 1 2.E 1 1.E 1 1 1.E 1 1 1 1	55554455555555	013422454332	0 A 2 0 E 0 A 0 E -1 A2 -1 A1 -1 E 1 A -1 E	2364.1525 2364.2855 2364.2855 2364.3119 2364.6648 2364.8336 2365.3770 2366.2998 2366.5341	2453.1482 2452.6181 2448.5000 2444.7926 2413.2812 2413.3050 2451.7071 2445.8577 2450.5143	-2 -2 -11 13 9 13	1.85.01 1.63.01 3.67.01 2.07.01 1.87.00 1.85.00 2.59.00 4.32.01	
	33.	1.E ; 3.A ; 4.E ; 3.A ; 1.E ; 2.E ; 1.E ;	555445555555	13422454332	0 E 0 A 0 E -1 A2 -1 A1 -1 E	2364.1525 2364.2855 2364.3119 2364.6648 2364.6886 2364.8336 2365.3770 2366.2998	2452.6181 2448.5000 2444.7926 2413.2612 2413.3050 2451.7071 2445.8577	-2 5 -11 13 9	1.63-01 3.67-01 2.07-01 1.87-00 1.65-00 2.59-00 4.32-01	
	3:	3.A) 4.E) 3.A1) 3.A2) 2.E) 4.E) 1.E) 0.A1) 0.A1)	55445555555	3422454332	0 A 0 E -1 A2 -1 A1 -1 E 1 A	2364.2855 2364.3119 2364.6648 2364.6886 2364.8336 2365.3770 2366.2998	2448.5000 2444.7926 2413.2812 2413.3050 2451.7071 2445.8577	-11 13 9	3.67.01 2.07.01 1.87.00 1.85.00 2.59.00 4.32.01	•
	3:	3.A1) 3.A2) 2.E) 4.E) 3.A) 1.E) 0.A1) 0.A1)	5 4 4 5 5 5 5 5 5 5	422454332	0 E -1 A2 -1 A1 -1 E 1 E -1 E	2364.3119 2364.6648 2364.6886 2364.8336 2365.3770 2366.2998	2444.7926 2413.2812 2413.3050 2451.7071 2445.8577	-11 13 9 13	2.07.01 1.87.00 1.85.00 2.59.00 4.32.01	1:
	3.	3.A1) 3.A2) 2.E 3.A 1.E 2.E 1.E 1.E 1.E 1.E 1.E 1.E 1.E 1.E 1.E 1	445555555	22454332	-1 A2 -1 A1 -1 E 1 E	2364.6686 2364.8336 2365.3770 2366.2998	2413.2812 2413.3050 2451.7071 2445.8577	13	1.87.00 1.85.00 2.59.00 4.32.01	1:
	4:	3.A2) 2.E) 4.E) 3.A) 1.E) 2.E) 1.E) 0.A1) 0.A1) 1.E)	45555555	2 4 5 4 3 3 2	-1 A1 -1 E	2364.6886 2364.8336 2365.3770 2366.2998	2413.3050 2451.7071 2445.8577	13	1.05.00 2.59.00 4.32.01	1:
	4:	2.E) 3.A) 1.E) 2.E) 1.E) 0.A1) 0.A1) 1.E)	5 5 5 5 5 5 5 5	4 5 4 3 3 2	-1 6	2364.8336 2365.3770 2366.2998	2451.7071	1	2.59.00	1:
	4:	1.E) 2.E) 1.E) 0.A1) 0.A1)	5 5 5 5 5	5 4 3 3 2	1 E	2365.3770	2445.8577	1	4.32.01	1.
# (4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4	4:	3.A) 1.E) 2.E) 1.E) 0.A1) 0.A1)	5 5 5 5	3 3 2	-1 E	2366.2998			2.00	
# (4 A A A A A A A A A A A A A A A A A A	4:	1.E) 1.E) 0.A1) 0.A1) 1.E)	5 5 5	3	-1 €				7.60.01	-1
R (4 A A A A A A A A A A A A A A A A A A	4.	2.E) 1.E) 0.A1) 1.E)	5	3			2454.9997	-12	3.98-01	1:
R(4 R(4 R(4 R(4	4:	1.E) 0.A1) 0.A1) 1.E)	5	5		2367.2759	2454.1494	-17	3.12.01	i:
R(4 R(4 R(4	4:	0.A1) 0.A1) 1.E	5		iξ	2368.3145	2456.7801	-9	2.52.01	i:
R(4 R(4 R(4	4:	1.E		2	-1 42	2368.3731	2457.3688	5	4.74-01	i.
R(4	4:	1.E)		i	1 42	2369.3323	2458.3280	6	3.50.01	i.
R(4	4:		5	i	-i E	2370.2280	2458.6936	11	8.16-02	i.
R(4	4.	1.E)	5	ô	iξ	2370.5838	2459.0494	. 5	1.08.01	i.
RI S		2.6)		ĭ	-1 E	2371.8154	2458.6889	6	7.00-00	i.
RI S		2.E)	5	ō	iè	2372.1744	2459.0479	0	4.81-02	i.
		2.E)	6	2	OE	2372.3329	2503.6718	1	1.59.01	1.
		0.A2)		0	0 41	2372.3964	2505.8507	-7	1.83-01	1.
		1.E)	6	i	0 E	2372.3964	2505.3223	-2	1.82.01	1.
		5.E)		6	1 €	2372.3964	2492.5587	-1	5.21.01	1.
		4.E)		4	0 E	2372.5430	2497.5048	1	2.85.01	1.
		3.A 1	6	3	0 4	2372.5905	2501.2791	6	4.22.01	.1.
R(6	6.	1.E)	7	4	0 E	2372.7367	2558.9805	11	6.96-02	1.
R(4	4.	3.421	5	2	-1 A1	2373.0914	2457.3059		3.72.00	1.
RI 4	4.	3.A1)	5	5	-1 AZ	2373.1728	2457.3873	24	3.65.00	1.
	5.	5.E)	6	5	0 E	2373.1726	2493.3351	-1	1.14.01	1.
		5 · E)	6	4	-1 E	2373.2784	2504.6173	6	3.99.00	1.
		3.A)		7	1 4	2373.6311	2617.8299	0	2.93-02	.1.
		4.E 1	6	5	1 €	2373.8512	2498.8180	0	2.84+01	1.
		3.451	5	1	1 41	2374.2784	2458.4930	-17	1.76-02	1.
		4.2)	5		-1 E	2374.5249	2455.0056	14	1.28.00	1.
		3.A 1	6	4	1 4	2374.7250	2503.4135	-1	5.80.01	.1.
		10)	6		-1 E	2374.9295	2507.8554	7	8.09-01	1.
		5.E)	6	3	1 €	2375.6735	2507.0124	-4	2.57.01	1.
		0 . Al)	7	3	0 WS	2376.0696	2562.8398		1.91-01	1.
		5.E)	7	7	-1 E	2376.5348	2550.0645	-26	2.00-02	1.
		1.E)	6	2	1 E	2376.6882	2509.6141	-1	5.53.01	1.
		0+A2)	6		-1 A1	2376.7866	2510.2409	14	1.07-00	1.
		154.0	6	1	1 41	2377.6655	2511.1198	14	3.19.01	1.
		4.E)	7		-1 E	2378.0599	2556.3754	-25	1.88-01	1.
	5+	1.E)	0		-1 E	2378.5848	2511.5107	6	1.55-01	1.
			6	5	1 €	2378.9452	2511.8711	-22	1.02.01	.1.
81 6					-1 4	2379.5449	2561.5677		4.33.00	:1.
		6.A)	!	?	1 4	2379.9533	2547.6043	10	1.16.02	.1.
			6	1	-1 E	2380.1759	2511.5148	10	7.56.00	1.
* (6		5.E)	,	5	OE	2380.3968	2565.0597	-5	4.47.01	1.
				5		2380.4369				1:
										i:
			7	:						i:
								-		·i.
RI										•1.
RIC										-i:
RIC								_		i.
RICE										i:
RI C			,					i		i.
		6.	6. 3.4) 6. 6.4) 5. 3.41) 5. 3.42) 6. 5.E)	6. 1.E) 7 6. 4.E) 7 6. 3.A) 7 6. 6.A) 7 5. 3.A1) 6 5. 3.A2) 6	6. 1.E) 7 1 6. 4.E) 7 4 6. 3.A) 7 3 6. 6.A) 7 6 5. 3.A1) 6 2 5. 3.A2) 6 2 6. 5.E) 7 6	6. 1.E) 7 1 0 E 6. 4.E) 7 4 0 E 6. 3.A) 7 3 0 A 6. 6.A) 7 6 0 A 5. 3.A1) 6 2 -1 A2 5. 3.A2) 6 2 -1 A1 6. 5.E) 7 6 1 E	6. 1.E) 7 1 0 E 2380.5223 6. 4.E) 7 4 0 E 2380.6568 6. 3.A) 7 3 0 A 2380.8196 6. 6.A) 7 6 0 A 2381.3855 5. 3.A1) 6 2 -1 A2 2381.4097 5. 3.A2) 6 2 -1 A1 2381.5526 6. 5.E) 7 6 1 E 2381.6409	6. 1.E) 7 1 0 E 2380.5223 2566.7661 6. 4.E) 7 4 0 E 2380.6568 2558.9723 6. 3.A) 7 3 0 A 2380.8196 2562.8424 6. 6.A) 7 6 0 A 2381.3855 2549.0365 5. 3.A1) 6 2 -1 A2 2381.4097 2510.0982 5. 3.A2) 6 2 -1 A1 2381.5526 2510.2411 6. 5.E) 7 6 1 E 2381.6409 2555.1706	6. 1.E) 7 1 0 E 2360.5223 2560.7661 18 6. 4.E) 7 4 0 E 2360.6568 2550.9723 3 6. 3.A) 7 3 0 A 2380.8196 2562.8424 11 6. 6.A) 7 6 0 A 2381.3855 2549.0365 8 5. 3.A1) 6 2 -1 A2 2381.4097 2510.0982 0 5. 3.A2) 6 2 -1 A1 2381.5526 2510.2411 15 6. 5.E) 7 6 1 E 2381.6409 2555.1706 7	6. 1.E) 7 1 0 E 2380.5223 2566.7661 18 1.68.01 6. 4.E) 7 4 0 E 2380.6568 2558.9723 3 2.94.01 6. 3.A) 7 3 0 A 2380.8196 2552.8824 11 4.06.01 6. 6.A) 7 6 0 A 2381.3855 2549.0365 8 1.55.00 5. 3.A1) 6 2 -1 A2 2381.4097 2510.0982 0 4.90.00 5. 3.A2) 6 2 -1 A1 2381.5526 2510.2411 15 4.70.00 6. 5.E) 7 6 1 E 2381.6409 2555.1706 7 8.59.00



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TABLE I-Continued

(1)		11)			11			(IV)	(٧)	(11)	(VII)	(VIII
			******	7				2302 2474				
930	RHI	6.	4.E)		5	1	€.	2382.2476	2560.5630	0	1.87+01	1.0
931	*0R(5.	3.421	6	1	1	41	2392.4297	2511-1182	13	8.94-02	1.0
932	*0R(5.	3.417		1	1	SA	2382.6778	2511.3663	1	4.16-02	1.0
933	PRI	5.	4.E !	7	3	-1	E	2382.8900	2507.8568	9	2.49-00	1.0
934	881	6.	3.A)			1	•	2383.0582	2565.0810	-3	4.24-01	*1.0
935	•SR (6.	1 .E)	7	3	-1	E	2383.2342	2569.4780	0	1.31-00	1.0
936	·NR (5.	5.E)		5	0	•	2383.5111	2503.6735	3	1.74-01	1.0
937	RR(6.	2.E)	7	3	1		2383.9759	2568.6388	-2	2.01.01	1.0
938	.TR	7.	0.42)		3	0	-1	2784.2346	2633.1596	3	2.43-01	1.0
939	PRI	5.	5.E)	6	*	-1	E	2384.4534	2504.6158	4	7.39-01	1.0
940	*5R1	7.	5.E 1	8	7	-1	E	2384.7533	2620.4971	-36	7.69-02	1.0
941	RR (6.	1.E)	7	5	1	E	2384.9711	2571.2149	5	1.87-01	1.0
942	·SR (6.	0.41)	,	2	-1	AZ	2385.1190	2571.8892	17	2.10.00	1.0
943	·NR (6.	3.41)	7	0	0	24	2385.2691	2567.2919	1	2.39-01	1.0
944	RR	6.	C.Al)	7	1	1	A2	2385.9050	2572.6752	50	2.66.01	1.0
945	·QR (6.	1.E)	7		-1	E	2386.8615	2573.1053	10	2.42-01	1.0
946	OR (5.	5.E)	6	3	1	E	2386.8615	2507.0239	6	7.01-03	1.0
947	PRI	6.	1.E)	7	0	1	E	2387.2158	2573.4596	7	8.89.00	1.0
948	RRI	7.	7.E)	8	8	1	E	2387.3411	2610.2755	53	5.16.01	1.0
949	·SR (7.	3.A)		5	-1		2387.5806	2631.7794	-29	7.19+00	*1.0
950	RRI	7.	6.A)	8	7	1		2387.9400	2617.8317	1	9.12.01	.1.0
951	GRI	7.	2.E)		2	0	E	2388.3143	2635.1414	0	1.27.01	1.0
952	QHI	7.	5.E 1		5	0	E	2388.3912	2624.1350	-1	3.63.01	1.0
953	PRI	6.	2.E)	7	1	-1		2388.4440	2573.1069	12	7.15.00	1.0
954	ORI	7.	1.E)	8	1	0		2388.4627	2636.8636	0	1.44+01	1.0
955	OP (7.	154.0	8	0	0	AI	2388.4952	2637.4202	5	1.45+01	1.0
956	GRI	7.	4.E)		4	0	€	2388.6691	2629.1771	6	2.63.01	1.0
957	.081	6.	2.E)	7	0	1	Ē	2388.7973	2573.4602	8	1.79-01	1.0
958	981	7.	3.A 1		3	0	Ä	2388.9663	2633.1651		3.45.01	*1.0
959	PRI	6.	3.421	7	ž	-1	Al	2389.6307	2571.6535	o	5.26.00	1.0
960	QR (7.	7.E)	8	7	o	Ē.	2389.6307	2612.5651	8	5.92-02	1.0
961	QRI	7.	6.A)	8	6	0	Ä	2389.7069	2619 5986	11	2.63.00	*1.0
962	PRI	6.	3.41)	7	2	-i	Ã2	2389.8612	2571.8840	iż	4.85.00	1.0
963	·SR(7.	2.6)		4	-1	E	2389.8975	2636.7246	-14	4.96.00	1.0
964	PR (7.	5.E)	8	6	i	È	2390.0003	2625.7441	-17	6.27.00	1.0
965	·PR (7:	2.6 1	8	i	ó	È	2390.0407	2636.8678	5	3.80-01	1.0
966	RRI	7.	4.E)	8	5	i	È	2390.5604	2631.0684	-1	1.21-01	1.0
967	OR (7	1	i		2390.6513	2572.6741	19	2.23-01	
968	OR (6.	3.A1)	'n			42			-7		1.0
969		6.	3.42)	7	1	-1	Al	2390.9577	2572.9805		7.17-02	1.0
970	PR (6.	4.E)	6	3		E	2391.1629	2569.4784	0	3.14.00	1.0
	RRI	7.	3.A)		3	1	*	2391.3165	2635.5153	-1	2.96.01	.1.0
971	•5R(7.	1.E)	7	3	-1		2391.4564	2639.8573	-1	1.76-00	1.0
972	-44	6.	5.E)	6	2	0	E	2391.5284	2565.0581	3	4.11-01	1.0
973	RRI	7.	2.E)		3	1	E	2392.1963	2639.0234	0	1.49-01	1.0
974	•TR(8.	0.AL)	9	3	0	AZ	2392.3249	2712.2220	-5	2.68-01	1.0
975	PHI	6.	5.E)	7		-1	E	2392.7745	2566.3042	5	1.37-00	1.0
976	•0R(6.	4.E !	. 7	5	1	3	2392.8979	2571.2133	3	6.27-02	1.0
977	•TH!	9.	SIE)	10		0	E	2393.0148	2790.6019	0	7.88-02	1.0
978	BP (7.	1.E)	8	5	1	E	2393.1605	2641.5614	7	1.48-01	1.0
979	•MB (7.	3.42)	8	0	0	Al	2393.2283	2637.4271	9	3.10-01	1.0
980	•5R(7.	0.A21	8	2	-1		2393.3632	2642.2882	9	3.74+00	1.0
981	PRI	6.	6.A)	7	5	-1		2393.9236	2561.5746	-15	1.13.00	*1.0
982	HH	7.	(SA.0		1	1	AL	2394.0588	2642.9838	20	1.97.01	1.0
983	RRI	8.	8.E)	9	9	1	E	2394.6553	2680.6538	37	4.28-01	1.0
984	·UR (8.	0.A1)	9		1	SA	2394.7975	2714.6946	2	2.34-01	1.0
985	·QR (7.	1.E)	8	1	-1	•	2395.0443	2643.4452	5	3.20-01	1.0
986	*OH (5.E)	7	3	i	Ē	2395.1004	2568.6301	-10	2.72-02	1.0
			7.E)		-	- 2	Ē	2345.2583	2689.2979	6	3.88+01	1.0



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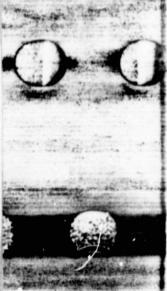
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TABLE 1—Continued

(1)	(11		(11			(1v)	(V)	(VI)	(VII)	(VIII
988	PR (7.		0	1		2395.4011	2643.8020	11	6.93-00	1.0
989	.SR (8.			-1		2395.4639	2710.6588	-24	9.13-00	•1.0
990	HR (B.	9		i		2395.8023	2696.7633	-1	6.81-01	•1.0
991	UR (8.			ō		2396.0910	2713.9009	-2	1.04.01	1.0
992	QP (B.	9		ō		2396.2443	2703.0273		2.79.01	1.0
993	GR (B.	9		ō		2396.2694	2715 6451	ĭ	1.18-01	1.0
994	QH (8.	·	ō	0		2396.3168	2716.2140	i	1.17-01	1.0
995	•NH (7.			0		2396.3595	2636.8675		2.95-01	1.0
996	90 6.	9	4	0		2396.5827	2708.1057		2.13.01	1.
997	Pt: (7.		1	-1	Ē	2396.6242	2643.4513	11	6.07.00	1.0
998	*OR (7.		ō	i	Ē	2396.9810	2643.8081	17	2.62-01	1.0
999	OR (B.	¥	3	ō	Ä	2397.0323	2712.2272	. 5	2.64.01	•1.
1000	•UR(10.	11	7	1		2397.0917	2880.6396	-6	9.96-02	•1.
1001	PR (7.	8	2	-1	42	2397.7427	2641.9415	0	4.95.00	1.0
1002	QR (8.	·	ě	0		2397.8302	2683.8287	18	4.95-03	1.
1003	Q# (8.		7	0	Ē	2397.8302	2691.8698	16	2.50-01	1.0
1004	.PP (8.	9	i	0	Ē	2397.8302	2715.6401	-3	2.24-01	1.0
1045	OR (8.	9	6	0	Ä	2397.9588	2698.9198	10	3.01.00	•1.
1006	•SP (8.			-1	ē	2398.0792	2715.8891	-4	4.92.40	î.
1007	PH (7.		2	-1	Al	2398.0792	2642.2780	0	4.19.00	1.
1008	RHI B.	9	6	î	Ē.	-398.2693	2705.0523	ō	4.43.00	i.
1009	.VR (9.	10	5	-i	A1	2398.5200	2798.1826	-4	1.38-01	i.
1010	PRI B.		5	i	Ē.	2398.7943	2710.3173	-5	7.72.00	i.
1011	•UP (9.	10	5	i	E	2399.1349	2798.2790	-19	9.41-02	i.
1012			ĩ	î	42	2399.1648	2643.3636	. 0	1.01-01	i.
1013	PR (7.		ŝ	-1		2399.3499	2639.8579	ő	3.18-00	i.
1014	*NR (7.	8	ž	o		2399.4004	2635.1442	3	5.86-01	1.
1015	RP (8:	9	-	ĩ		2399.5016	2714.6965		1.98.01	•1.
1016	*SH (8.	9	3	-1	Ē	2399.5805	2718.9562	-7	2.03.00	1.
1017	ARE B.	9	3	i	Ē	2400.3397	2718.1496	3	1.03-01	1.
1018	*TR(10.	11	5	ō	Ē	2400.7120	2886.8438	-10	8.18-02	1.
1019	PR (7.		4	-1	Ē	2400.9947	2636.7385	0	1.65.00	1.
1020	*NP (B.	4	0	ō	SA	2401.0205	2716.2154	ž	3.02-01	1.
1021	OR (7.	8	5	1	E	2401.0587	2641.5667	12	1.14-01	1.
1022	RP (8.	9	2	i	Ē	2401.2668	2720.6425	ii	1.09-01	1.
1023	*SP (8.	 ý	5	-i	SA	2401.5209	2721.4181	7	5.92.00	i.
1024	PH 1 7.		5	-1		2401.9053	2631.7970	-12	1.95.00	•1.
1025	RP (9.	10	10	i	Ā	2401.9053	2758.7328	44	6.76.01	•1.
1026	4H (B.	. 9	1	i	42	2402.1554	2722.0526	19	10.52.1	i.
1027	.SH (9.	10	6	-1	E	2402.1554	2793.4913	-16	1.32.00	i.
1028	BR (9.	 10	9	i	È	2402.4852	2768.4435	9	3.07.01	i.
1029	•UR (9.	10	4	i	AI	2402.9229	2802.5855		5.56-01	i.
1030	AP (9.	ic		i	Ē.	2403.0387	2776.9915	-2	2.77.01	i.
1031	•QR (8.	9	ĩ	-i	È	2403.1589	2722.5346	11	3.68-01	i.
1032	•SR 1 9.	10	5	-i	À	2403.1873	2798.1743	-13	9.33.00	•1.
1033	•NR (7.	8	3	ó	-	2403.2812	2633.1729	16	4.36-01	•1.
1034	*OR (7.		3	ĭ	È	2403.2812	2639.0250	2	6.15-02	1.
1035	PRI B.	9	ő	i	È	2403.5050	2722.8807	16	5.03.00	1.
1036	381 9.	10	7	i		2403.5425	2784.3769	-1	4.85.01	•1.
1037	OH (9.	10	ż	ò	ē	2403.7255	2801.3126	-3	8.09.00	1.
1038	PR (7.		6	-1	È	2403.8028	2626.7372	-25	4.19-01	1.
1039	OR (9.	10	ĩ	-0	È	2403.9200	2803.0641	-23	9.01.00	1.
1040	OH (9.	10	ò	ő	41	2403.9814	2803.6440	-12	8.61.00	
1041	OR (9.	10	5	0	Ē,	2403.9614	2790.6044	-12	2.03.01	1.
1042	ONE! B.		i	o	È	2404.1239		3		1.
1043	QR 1 9.	10	4	ő	È	2404.3945	2715.6469	-1	1.58-01	1.9
1044	PR: 8.	10	ī	-1	É	2404.7190	2722.5289	-1	4.67.00	1.5
1045	921 9.	10	3	- 6		2405.0092	2799.9962	-7	1.89-01	.1.9
		 	,		-	240010045	E	-,	1.04.01	•1.0



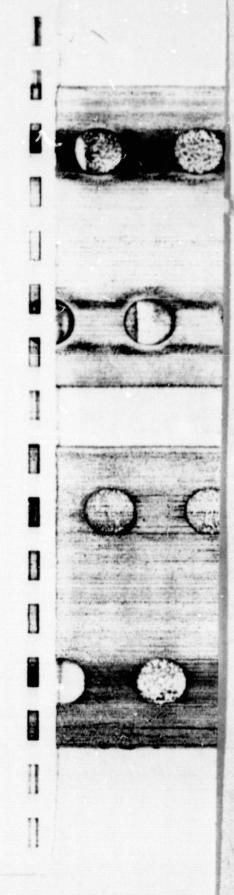


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TABLE 1-Continued

(1)	(11)			11			(IV)	(4)	(AI)	(VII)	(VIII
				•••		****	**********	THE STATE OF THE S			
		5.E 1	. *	0	1		2405.0659	2722.8758	11	3.37-01	1.0
1047	•PR(9.	5.6	10	1	0		2405.4724	2803.0595		2.83-01	1.5
1048	*OR (7.	6.4		*	1		2405.6334	2635.5251		5.39-02	•1.0
1049	PP (8.	3.45)	9	5	-1		2405.7364	2720.9313	-6	4.19.00	1.0
1050	QR (9.	8.E)	10		0	E	2405.9044	2771.8627	9	8.22-02	1.6
1051	GH (9.	9.4	10	9	0		2405.9536	2762.7811	53	3.15-03	.1.0
1052	OH! 4.	7.E)	10	7	0	E	2405.9536	2779.9065	0	4.56-01	1.0
1053	OR (9.	6.A)	10		0		2406.1242	2786.9586	-9	2.A2.00	.1.0
1054	•SH (9.	2.E)	10		-1	•	2406.1242	2803.7113	-21	4.24.00	1.0
1055	PR (8.	3.41)	9	5	-1	42	2406.2224	2721.4173	6	2.96.00	1.0
1056	BR (9.	5.E)	10	6	1		2406.4502	2793.0732	-14	3.02.00	1.0
1057	.OR (8.	3.A1)		ī	1	42	2406.8569	2722.0518	19	9.11-01	1.0
1058	RR (9.	4.E 1	10	5	i		2406.9500	2798.2859	-13	4.80.00	1.6
1059	•NP (8.	5.E)	9	2	ō		2407.1204	2713.9034	0	6.49-01	1.0
1000	*OR (8.	3.421	9	ī	ī	41	2407.2839	2722.4788	- 6	1.20-01	1.6
1061	PRI B.	4.E)		5	-i	£.	2407.4394	2718.9534	-10	2.77-00	1.0
1062	*SH 1 9.	1.€ 1	10	3	-1	è	2407.6027	2806.7468	-13	2.03.00	1.0
1063	RR(9.	3.4 1	10	-	-i		2407.6027	2802.5897	-13	1.30-01	•1.0
1064	*TR(10.	0.41)	ii	3	ò		2408.2520	2896.4464	-10	2.39-01	1.0
1065	AR (9.	2.6	10	3	ĭ	i.	2408.4011	2805.9882	-10	6.76.00	1.0
1066	•NR (9.	3.42)	10	ő	ò	41	2408.6623	2803.6493	-7	1.74-01	
1067	PRI B.	5.6	9	-	-1	£.					1.5
1068				-		-	2409.0924	2715.8755	-18	1.61.00	1.5
	9R(10+1		11	11	1	٤	2409.0924	2844.4966	39	2.54.01	1.5
1069	*OH (B.	4.E)		5	!	•	2409.1209	2720.6434	15	1,68-01	1.0
1070	RP (9.	1.5)	10	5	!	€.	2409.2819	2808.4260	6	7.62.00	1.0
1071	*SR(9.	0.AZ)	10	. 5	-1		2409.5568	2809.2194	0	7.45.00	1.0
1072	B# (10.	9.4	11	10	1		2409.6472	2855.2782	10	4.62.01	*1.6
1073	PR (8.	6.4)	. 0	5	-1	•	2409.7045	2710.6655	-17	2-14-00	•1.0
1674	•SR(10.	4.E)	11	6	-1	E	2409.9429	2889.8626	52	1.64.00	1.0
1075	MP(10.	8.E)	11	9	1		2410.1783	2864.9814	-5	5.01.01	1.0
1076	RK(9.	0.AZ)	10	1		41	2410.2135	2809.6761	4	5.76.00	1.0
1077	BR(10.	7.E)	11	1,	1	E	2410.6843	2873.3308	-4	1 48.01	1.0
1078	•5R(10+	3.A)	11	5	-1		2410.7600	2894.3079	9	8.21.00	*1.0
1079	*UH (10.	0.41)	11		1	42	2410.9306	2899.1250	9	1.67.00	1.0
1080	AR(10.	6.A)	11	7	1		2411.1587	2880.6431	-3	3.30+01	.1.6
1001	QR(10.	2.E)	11	5	0	•	2411.2099	2897.3417	-8	5.96+00	1.0
1082	•QH (9.	1.E)	10	1	-1	E	2411.2099	2810.3540	24	3.84-01	1.0
1083	*NR (8.	6.A)	9	3	0		2411.2720	2712.2330		7.79-01	.1.0
1084	.OR (8.	5.E)	9	3	1		2411.3710	2718.1541		1.05-01	1.0
1085	QR(10.	1.€)	11	ī	0	Ē	2411.4119	2899.0910	-4	6.52.00	1.0
1086	QR(10.	0.41)	11	0	0	42	2411.5102	2899.7046	-55	5.04+00	1.0
1087	PR (9.	1.E)	10	0	i		2411.5301	2810.6742	18	3.35+00	1.0
1088	QR(10.	5.E 1	11	5	ē	Ē	2411.6147	2886.8511	-3	1.40+01	1.0
1089	PR (8.	7.E)	9	6	-1	ē	2411.7409	2705.7805	-31	7.57-01	1.0
1090		4.E)	10	ĩ	ō	è	2411.7409	2803.0768	12	3.60-01	1.0
1091		4.E)	ii		ō	è	2412.1286	2892.0483	-6	1.07-01	1.0
1092	PP (9.	2.6)	10	i	-1	7	2412.7458	2810.3329	2	3.26.00	1.0
1093	QP(10.	3.4)	ii	3	ō		2412.9006	2896.4485	-7	1.28-01	•1.0
1094	*PR(10.	2.E)	ii	í	ő	è	2412.9570	2899.0888	-6	2.78-01	1.0
1095	*OP (9.	2.6	10	ò	ĭ	è	2413.0785	2810.6656	10	3.90-01	1.0
1096	•TR(11-	0.42)	12	3	ò	41	2413.5727	2999.0358	47	1.69-01	1.0
1097	PRI Ve	3.41)	10	ž	-1	42	2413.6175	2808.6045	-11	3.26.00	
1095	PP (8.	8.6		i	-1	ě	2413.6772	2699.6757	-11	2.81-01	1.9
1099	*OR (8.				-;				-11		1.0
100				:	0	•	2413.7405	2714.7015	3	1.10-01	•1.0
1101			11		0	E	2413.9376	2868.6409		2.04-01	1.0
	QR(10.	*** /	11				2413.9378	2859.5688	24	9.83-02	•1.0
1102	QR(10.1	0.E)	11	10	0		2414.0120	2849.4162	39	3.21-03	1.0

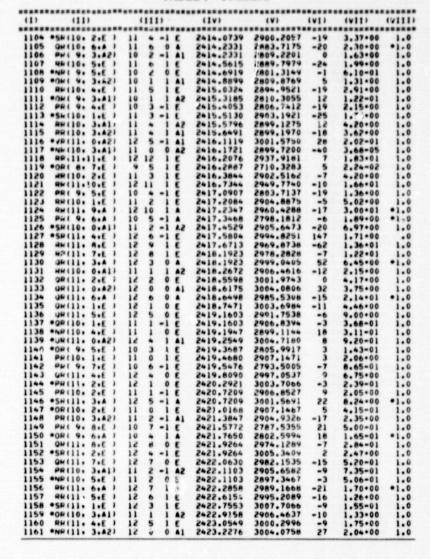


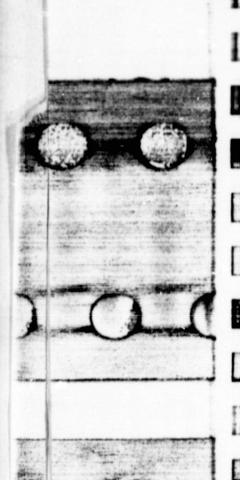
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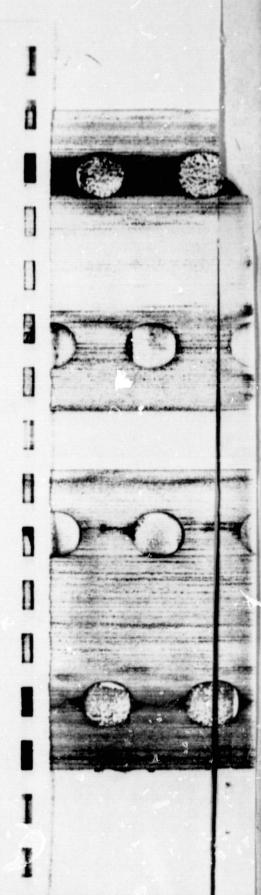


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TABLE I-Continued

(1)	(11)	-	111)		(IV)	(V)	(IV)	(VII)	(VIII)
1162	48(12.				A	2423.2653	3038.9922	-39	2.51.01	
1163	PR(10.		ii	3 -		2423.2653	2903.1850	-32	1.50.00	*1.0
	*SR(11.		iè			2423.3348	3008.2861	-20	1.27-00	1.0
1165	RR(11.		12		42	2423.6342	3004.4824	-	2.09.00	1.0
1166	PH (9.		10	i -		2423.6342	2740.4616	-11		1.0
1167	48(12.		13			2423.8023	3051.9672	-3	1.13.01	•1.0
1168	RP (12.		13	15 1	è	2424.3036	3063.7102	37	1.03-01	1.0
1169	RR(11.		iè	';		2424.3036	3007.7181	-	2.50.00	1.0
1170	*OR (9.		10	5 1		2430.3469	2798.2597	0	4.16-02	1.0
1171	BR(12.		13	10		2424.7571	3074.2405	52	1.86.01	•1.0
1172	PR(10.		ii	5 -1		2424.8331	2894.3175	19	1.47.00	•1.0
1173	PR (10.		ii	4 -1		2424.9718	2900.2083	-17	1.03-00	1.0
1174	RH(11.		iż	2 1		2425.0469	3009.9982	-21	3.16.00	1.0
1175	AP(12.		13	9 1		2425.1438	3083.5673	-6	8.37-00	
1176	RR(11.		12	1 1		2425.2313	3010.6944	-46	5.23-00	1.0
1177	98(12.	3.4)	13	3 6		2425.5094	3112.3651	137	4.62.00	1.0
1178	RR(12.		13			2425.5421	3091.7943	-40	7.49.00	1.0
1179	Q# (12.		13			2425.7638	3115.1671	11	2.76.00	1.0
1180	QR(12.		13	0 0		2425.8590	3117.2958	25	3.04.00	1.0
1181	QH (12.		13	i	-	2425.9457	3116.8745	-8	2.85.00	1.0
1182	QR(12.		13	6		2420.0372	3099.0291	-27	1.31.01	*1.0
1183	*S#(11.		12	2 -1		2426.3293	3011.7924	-13	7.31-01	1.0
1184	•NR(11.		12	1 6		2426.4575	3003.7022	-8	2.31-01	
1185	06(12.		13	5 6		2426.7017	3105.3637	64	5.40.00	1.0
1186	*NR(10.		ii	3 6		2426.9721	2896.4565	0	9.37-01	1.0
1187	*QR(11.	1.6)	iż	i -i		2427.0852	3012.0365	-7	3.43-01	
1188	PR(10.		ii	6 -1		2427.2056	2889.8561	46	7.78-01	1.0
1189	*QR(10.		ii	3 1		2427.2792	2902.5157	-7	1.61-01	1.0
1190		1.6)	iż	0 1		2427.3566	3012.3079	-3	1.13.00	1.6
1191	*PR(12.		13	i		2427.4742	3116.87 5	-5	1.90-01	1.0
1192	QR(12.		13	4 0		2427.5237	3110.8025	134	3.97.00	.0
1193	*SR(12.		13	5 -1		2428.5328	3115.3885	149	5.00-00	0
1194	PR(11.		iż	1 -1		2428.6229	3012.0374	-7	1.13.00	1.0
1195	*OR(11.		12	0 1		2428.8951	3012.3096	-1	4.20-01	1.0
1196	P# (11.		12	2 -1		2429.0392	3009.8874	-26	1.58-00	1.0
1197	PR(10.		ii	7 -1		2429.3954	2884.0985	121	5.65-01	
1198	*NR(11.		12	2 0		2429. 3954	3001.9889	15	3.80-01	1.0
1199	*0P!10.		11	4 1		2429.6458	2899.1302	14	8.72-02	1.0
1200	*SP112+		13	4 -1		2429.7176	3119.1208	82	1.66+00	1.0
1501	*OR(10.		11	4 1		2429.7176	2899.2020	-13	1.05-01	1.0
1202	96(12.		13	4 0		2429.7576	3079.2410	-24	3.71-01	•1.0
1203	OR (12.		13	8 0		2429.8890	3088.3125	-2	2.90-01	1.0
1204	*OR(11.		12	ii		2429.8890	3010.7372	-3	3.13-01	1.0
1205	OH (12.		13	7 0		2430.0951	3096.3473	35	4.22-01	1.0
1200	RR(12.		13	7 1		2430.3467	3163.3386	53	1.17.00	.1.0
1207	*NR(12.		13	0 0		2430.4444	3117.3001	29	9.53-01	1.0
1208	RR(12.		13	6 1		2430.6694	3109.3314	66	7.84-01	1.0
1209	PR(11.		12	2 -1		2430.9386	3011-7668	-18	1.05.00	1.0
1211	PR(11.		12	3 -1		2431.0516	301 1963	-10	9.46-01	1.0
1212	RR(12.		13	5 1		2431.0880	3114.3668	97	1.04+00	1.0
1213	*SR(12.	1.E)	13	3 -1		2431.0880	3122.0168	6	7.15-01	1.0
1214	*OR(11.	3.41)	12	i i		2431.1434	3011.9916	22	9.01-02	1.0
1215	AP(12.	3.42)	13	4 i		2431.5721	3118.4279	17	1.17.00	
1216	PR(10.		ii	ē -i		2431.5721	2877.2032	219		1.0
1217	AR(12.		13	. 1		2431.6974	3118.5531	6	6.26-01	0
1218	R# (12.		13	3 1		2432.1193		7	6.11-01	1.0
	•N#(11.		12	3 0		2432.1614	3121.5226	54	1.42.00	-1.0
1219										



V, AND V3 OF PH3

TABLE 1-Continued

(1)	(11)		-	111	1)		(IV)	(4)	(VI)	(VII)	IIIIV
	*******							*********		*****	********	******
1551	PH(11.	S.E	1	10		-1	E	2432.7485	3005.3420	4	7.09-01	1.0
1555	BF(12.	1 . €)	13	5	1	E	2432.8207	3123.7445	-15	1.91.00	1.0
1553	*NR(12+	4.E	,	13	1	0	E	2433.5978	3116.8766	-5	1.42-01	1.0
1224	PR(10.	10.E	1	11	9	-1	•	2433.8428	2869.2470	352	1.13-01	.0
1225	PH(11.)	12	5	-1		2434.6956	3001.5766	59	7.74-01	.1.0
1226	PP(11+	7.E)	15	6	-1	E	2434.7465	2994.8370	159	5.96-01	.0
1227	*OH(11.	5 . E	1	12	3	1	E	2435.1147	3007.7082	-8	1.50-01	1.0
1224	PH(12.	1.€	1	13	0	1	Ē	2435.2215	3126.1503	7	5.21-01	1.0
1229	*NP(12.	5.E)	13	2	0	E	2436.5142	3115.1762	21	2.59-01	1.0
1230	PR(12.	3.42	1	13	2	-1	41	2436.5763	3123.43.20	-25	9.98-01	1.0
1231	.OR(12.	3.€	1	13	0	1	E	2436.7418	3126.1491	1	4.09-01	1.0
1535	PH(11.	B.E)	12	7	-1	E	2437.1897	2989.3922	363	5.06-01	.0
1233	*OR(12+	3.41)	13	1	1	45	2437.5279	3124.3836	-49	1.34-01	1.0
1234	PH(12.	4.E	1	13	3	-1	5	2438.7388	3122.6176	7	5.15-01	1.0
1235	PR(12.	3.A1	1	13	2	-1	54	2438.9798	3125.8355	15	7.11-01	1.0
1236	*NR(12.	6 . A)	13	3	0		2439.3779	3112.3698	142	6.88-01	0
1237	PR(11.	9.4)	12	8	-1		2434.4508	29A2.6444	454	6.89-01	0
1236	PR(12.	5.E)	13	4	-1		2440.4513	3119.1133	74	4.53-01	1.0
1239	PR(12.	7 . E	,	13		-1	E	2442.1660	3108.4182	343	4.06-01	.0
1240	PH(12.	6.4)	13	5	-1		2442.3984	3115.3903	151	5.66-01	0
1241	.OH(12.	5.€)	13	3	1	Ε	2442.86#1	3121.5301	14	1.12-01	1.0
1242	.OR(11.	A.E	,	12	6	1	E	2442.9871	2995.1896	-35	3.24-02	1.0
1243	.0F(15.	6.42)	13	4	1	Al	2445.4431	3118.4350	24	9.61-02	1.0
1244	.OR(12.	6 . A 1)	13	4	1	84	2445.5578	3118.5497	5	8.36-02	1.0

RESULTS

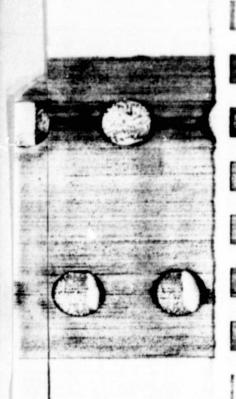
The transitions assigned to ν_1 and ν_3 are listed in Table I with the asterisk denoting "forbidden" transitions. The weights given in column VIII are equal to either 1.0 for transitions unambiguously assigned to ν_1 or ν_3 , or 0.0 for transitions which could be only tentatively assigned because of the Fermi interaction with $2\nu_4$. The weights noted with an asterisk correspond to doublets $A_1 \rightarrow A_2$, $A_2 \rightarrow A_1$ (with |K| = 3, 6, · · ·), unresolved experimentally and for which the expected splittings are small. The ground-state energy levels are labeled as (J, |K|, C), where $C = A_1, A_2$ or E gives the symmetry species (12). In the upper states, the levels can be again specified unambiguously as $(J', |K'|, l'_3 = 0, C')$ for $v_1 = 1$ and $(J', |K'|, l'_3 = \pm 1, C')$ for $v_3 = 1$ although the mixing in v_1, v_3, K , and l'_3 is large.

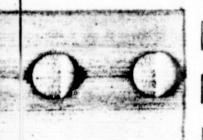
Twenty constants related to $v_1 = 1$ and $v_3 = 1$ were derived from the experimental data.² Six of them refer to $v_1 = 1$, ten to $v_3 = 1$, and four are interaction constants: $C_{11}^{(2)}$ and $C_{11}^{(3)}$ result from the first-order term T_1 , $C_{21}^{(2)}$ and $C_{21}^{(3)}$ arise from the second-order term T_2 ; the first-order constant

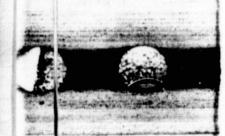
$$C_{11}^{(1)} = \frac{B_e \zeta_{13}}{2} \times \frac{\lambda_1^{1/2} + \lambda_3^{1/2}}{\lambda_1^{1/4} \lambda_3^{1/4}}$$

is insignificant here, consistent with the very small value predicted for ζ_{13} in Ref. (5), viz., 0.004.

The derived constants are reported in Table II. They were obtained in two steps: first, the fitting was performed within the third order of approximation, the distortion constants D being fixed to their ground-state values. The fourth-order constants, introduced as a second step, were found to be insignificant except for the vibrational corrections to the constants D which improved the fitting significantly.







² These were the 1186 data points retained with weight equal to 1.0.

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TABLE II Vibration-Rotation Constants for $v_1 = 1$ and $v_3 = 1$ States of PH₃

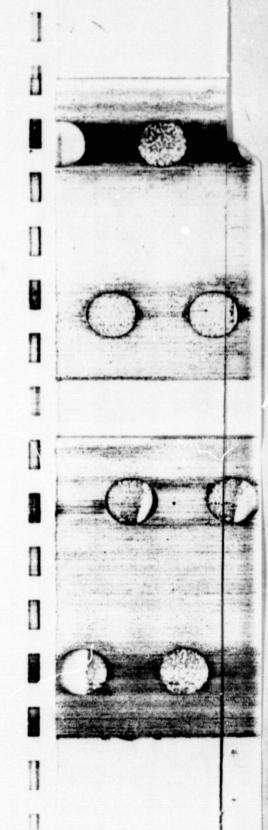
• _{v1}	2321.1314 (27)	• _{v3}	2326.8766 (23)
B ₁	4.406115(66)	*(C5)3	6.552 (30)×10 ⁻²
•c1	3.88468(13)	*B3	4.408738 (60)
₽ DJ	1.0876(42) 310-4	•c ₃	3.90537 (12)
P DJK	-9.97(14)×10 ⁻⁵	* D ^J ₃	1.3130 (35)x10 ⁻⁴
●DK	8.69 (15)×10 ⁻⁵	* DJX	-2.500 (15)x10 ⁻⁴
c(2)	Ex6.084(34)x10-3	* D ^K ₃	1.795 (15)×10 ⁻⁴
c(3)	-ex2.898(29)x10-4	± _n J ₃	1.524 (73)×10 ⁻⁴
c(2)	ε'x9.595(29)x10 ⁻³	nK n3	-8.49 (11)×10 ⁻⁴
c ₂₁	-e'x2.781(38)×10 ⁻⁴	*(q23)v	-1.628 (14)×10 ⁻³
constan	ts constrained $(H_1^J = H_3^J)$	- н ^Ј - 1.83	.10 ⁻⁸
to thei	r ground HJJK = 1		= -6.2.10 ⁻⁸
state v	alues (4) $H_1^{JKK} = 1$	JKK = HJKK	- 8.0.10 ⁻⁸

The errors quoted are one standard deviation

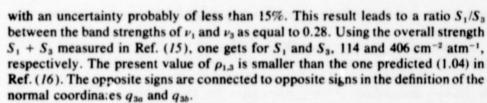
Also, the large deviations from their ground-state values, observed for these constants are, of course, related to strong vibrational interactions, especially with $2\nu_4$. The fourth-order distortion constants, H^J , H^{JJK} , and H^{JKK} were constrained for $v_1 = 1$ and $v_3 = 1$ to their ground-state values (4) recalled at the bottom of the table; no significant value of the constant, H^K , could be deduced from the present analysis. The 16 constants related to v_1 and v_3 are written in the usual notation (12), except that the asterisks point out that they do not involve exactly the same contributions as in the individual analyses of the bands. The set of constants of Table II reproduces the 1186 transitions assigned to v_1 and v_3 with an overall standard deviation of 0.017 cm⁻¹.

Concerning the intensities, the strengths of allowed and "forbidden" transitions could be deduced from measurement of equivalent widths (14), and compared together to adjust the ratio $\rho_{1.3}$ involved in the relative intensity calculations. In fact, the number of available data was very limited because of the intensity anomalies which appear early in the ν_1 and ν_3 bands as a result of the Fermi interaction with $2\nu_4$. Nevertheless, the comparison between experimental results and computed values of the relative intensities enabled us to approximate the coefficient $\rho_{1.3}$ to the value:

$$\rho_{1.3} = -0.75$$



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Although no special effort has been made to record spectra from which absolute values for line strengths could be determined, the quantities reported in column VII of Table I can be considered as reasonable values for the dimensionless quantities $(S_ipl)/d_i$ at 300 K, where S_i are the absolute line strengths, d_i are the Doppler halfwidths, p = 0.3 T and l = 100 cm were defined by the experimental conditions.

CONCLUSIONS

The present analysis allowed us to explain the main features of the absorption by PH_3 in the extended range 2184-2445 cm⁻¹, and to evaluate the ratio between the band strengths of the two fundamental bands ν_1 and ν_3 . Anomalies observed in frequencies and intensities pointed out that the Fermi interaction with $2\nu_4$ has to be taken into account for interpretation beyond J'=13. This interaction is especially strong in the component $Kl_3 < 0$ of the ν_3 band, giving rise to serious anomalies in intensity distribution between O- and S-type "forbidden" transitions. See for example Fig. 2.

Many unassigned transitions, observed near 2200 cm⁻¹ with intensities comparable to the ν_1 and ν_3 transitions, may be ascribed to the overtone band $2\nu_4$. The rotational structure of this band appears very complex at first sight; but a spectrum recorded at low temperature which would allow us to eliminate most of the ν_1 and ν_3

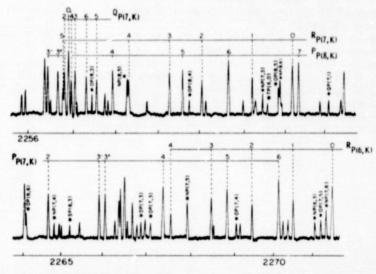
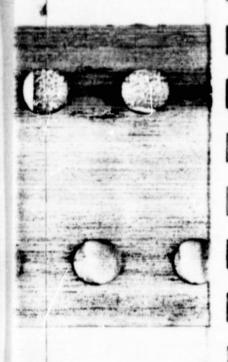
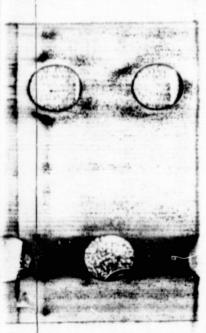


FIG. 2. Spectrum of PH₃ in the region 2256-2272 cm⁻¹. Pressure: 6 Torr; Pathlength: 1 m. The "forbidden" transitions are indicated by asterisks.





transitions, will probably be a good starting point for the analysis of this overtone band.

Finally, in the narrow region 2050-2129 cm⁻¹ which was convenient for the detection of PH₃ in Jupiter's spectrum at 5 μ m (7), the expected transitions for ν_1 and ν_3 correspond to high J values in the range 22-30, i.e., to ground-state energy levels very depopulated at low temperatures. So, the combination band $\nu_2 + \nu_4$ —instead of ν_1 or ν_3 —might be responsible for the PH₃ transitions observed in this region.

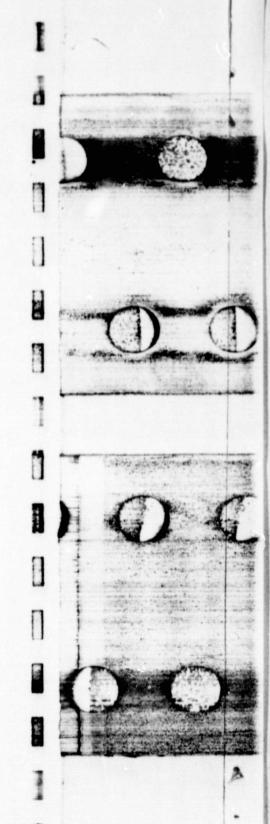
ACKNOWLEDGMENTS

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Diode Laser Measurements of Strengths of 16O12C32S Lines at 12 µm1

Objective and experimental method. This note describes a study of individual line strengths of some lines of the ν_1 band of $^{16}\text{O}^{12}\text{C}^{22}\text{S}$ at 12 μm . Values for the total band intensity, pure vibration transition moment, and vibration-rotation interaction constants were deduced from the measured data. Also, strengths of a few lines of the "hot" band $(\nu_1 + \nu_2) - \nu_2$ were measured from which the total band intensity is estimated. Self-broadening coefficients of a few high J transitions of the ν_1 band were also determined.

The diode layer spectrometer has been described in detail elsewhere (1, 2). A commercial sample of OCS with a minimum purity of 99.9% obtained from Matheson Gas Company was used without purification. A natural sample of OCS contains 93.72% of 16O12C32S. Pressures were measured with MKS Baratron gauges with 0-1 and 0-100 Torr pressure heads. The uncertainty of our measured pressures was estimated to be less than 1%. Pyrex cells of lengths 11.3, 21.7, 51.55, 101.6, and 201.6 cm fitted with KBr or KRS-5 windows were used. The cells were evacuated by a diffusion pump to less than 10-5 Torr. The pressures of OCS ranged from 0.040 to 0.280 Torr. All observations were taken within 1 h after filling the cells. Eight or more repetitive scans were made of each line at three or more pressure × pathlength values at room temperature (296-299 K). The cell temperatures were monitored during each scan by precision thermistor probes. The dispersion was calculated from the known fringe spacing of a one-inch germanium etalon. The analog records were digitized using a Bendix Datagrid Digitizer for processing.

Self-broadening parameters. The self-broadening parameter for OCS has been the subject of several experimental and theoretical investigations (3). However, they have all been done for J < 30. The theoretical self-broadening parameters calculated in Ref. (3) have been combined with the present high J measurements in order to determine this parameter for all values of J up to J = 76. For self-broadening measurements, a 11.3-cm cell was used and the pressures of OCS ranged from 33 to 55 Torr. At these pressures, the lines were essentially Lorentzian. From direct half-width measurements of P(67), P(72), and P(76), a mean value of (0.108 ± 0.005) cm⁻¹/atm. at 300 K was determined. There was no clear J dependence for these lines. In Table I, the self-broadening parameters obtained from this combined work are listed for J up to 76.

Strengths of lines. At pressures of less than 0.286 forr, self-broadening is negligible and thus the lines should be Doppler broadened. However, typical full widths at half height for the OCS lines were measured to be 0.0018 cm⁻¹, which is about 40% larger than the Doppler width. In previous publications from this laboratory (2, 4) it has been shown that this broadening is due to an instrumental function with Gaussian characteristics related mostly to the diode laser linewidth. The method of equivalent width is therefore used to determine line strengths.

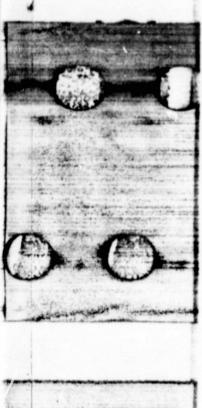
The equivalent width W is defined by

$$W = \int_{\text{line}} (1 - e^{-k(\nu)x}) d\nu, \tag{1}$$

where x is the measure of the number of absorbing molecules per unit area and $k(\nu)$ is the absorption coefficient which is related to the line strength S and the lineshape function $f(\nu - \nu_0)$ by the relation

$$k(\nu) = S \cdot f(\nu - \nu_0). \tag{2}$$

¹ Support extended this research by National Aeronautics and Space Administration is gratefully acknowledged.



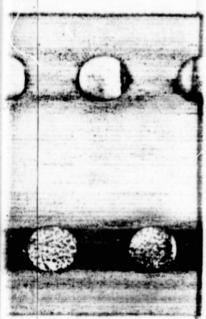


TABLE I

Strengths and Self-Broadening Parameters of Some v1 Band Lines of 16O12C22S at 300 K

LINE	Scalc. (cm ⁻² atm ⁻¹)	Sobs. (cm ⁻² atm ⁻¹)	(cm ⁻¹ atm ⁻¹)
P(76)	0.007	0.006	0.116
P(75)	0.008	0.007	0.117
P(72)	0.011	0.011	0.135
P(71)	0.013	0.014	0.137
P(67)	0.020	0.021	0.150
P(61)	0.037	0.037	0.163
P(55)	0.065	0.073	0.177
P(50)	0.096	0.103	0.190
P(45)	0.135	0.122	0.203
P(40)	0.180	0.170	0.215
P(35)	0.224	0.221	0.227
P(27)	0.277	0.312	0.258
P(20)	0.283	0.229	0.258
P(11)	0.206	0.210	0.235
R(7)	0.167	0.169	0.222
R(16)	0.297	0.303	0.252

A Voigt profile has been assumed for the OCS line, i.e.,

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$$f(\nu - \nu_0) = (P'a/\pi) \int_{-\infty}^{\infty} (e^{-\nu^2})/(a^2 + (\xi - y)^2) dy,$$
 (3)

where

$$P' = (1/b_{\rm D})\{(\ln 2)/\pi\}^{1/2},$$

$$a = (b_{\rm L}/b_{\rm D})(\ln 2)^{1/2},$$

$$\xi = \{(\nu - \nu_{\rm D})/b_{\rm D}\}(\ln 2)^{1/2}.$$
(4)

 $b_{\rm L}$ and $b_{\rm D}$ denote the Lorentzian and Doppler half widths, respectively. The measured values of W ranged from 0.0004 to 0.001 cm⁻¹ corresponding to Doppler lines on the linear part of the curve of growth.

The line strength S at a particular temperature was then calculated from W by iteration. From repeated measurements of line strengths at various pressure \times pathlength values, the experimental uncertainty was estimated to be about $\pm 5\%$. The line strength at a temperature T was converted to temperature T = 300 K by using the following expression,

$$S(T_0) = S(T) \frac{Q(T)}{Q(T_0)} \left(\frac{T}{T_0}\right) \exp \frac{hc\nu''}{k} \left(\frac{1}{T} - \frac{1}{T_0}\right)$$
, (5)

where Q is the internal partition function (calculated in this work), and ν'' , the energy of the lower level in cm⁻¹, evaluated from the molecular constants given in Ref. (6).

The line strength in a rovibrational band of OCS can be written as (5)

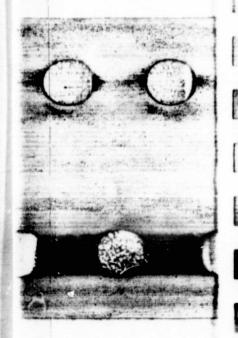
$$S(T) = \frac{8\pi^3 N_T}{3hcQ(T)} \nu_0 \exp\left(\frac{-hc\nu''}{kT}\right) \left(\frac{m^2 - l^2}{|m|}\right) |R_{elJ}^{eff}|^2 \left[1 - \exp\left(\frac{hc\nu_0}{kT}\right)\right] , \tag{6}$$

where N_T is the total number of molecules per unit volume per unit partial pressure of the absorbing gas at temperature T, $R_{vlJ}^{\nu P,T}$ is the matrix element of the dipole moment of the vibrational transition; m = J + 1 for the R branch and -J for the P branch, ν_0 is the wavenumber of the line. The square of the matrix element of the dipole moment of the vibrational transition $|R_{vlJ}^{\nu P,T}|^2$ can be expressed as

$$|R_{vU}^{v'FF}|^2 = |R_{vU}^{vFF}|^2 F_{vR}(m), \tag{7}$$

where $F_{vR}(m)$ is the factor that accounts for the vibration-rotation interaction (nonrigidity correction)





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TABLE II

Summary of Band Intensity Parameters for 16O12C32S

and $|R_{vR}^{ev}|$ is the matrix element of the pure vibrational transition. For a linear molecule, $F_{vR}(m)$ may be expressed as a function of m by

$$F_{vR}(m) = 1 + am + bm^2 + \cdots$$
 (8)

By introducing the measured value of S and all known quantities into Eq. (6), $\left|R_{el}^{\mu\nu}\right|^2$ was calculated for each line. These were then least-squares fitted to obtain a, b, and $\left|R_{el}^{\mu\nu}\right|^2$. The line intensities were then calculated from $\left|R_{el}^{\mu\nu}\right|^2$, a, and b. Table I presents these calculated values along with the mean line strengths of the measured lines.

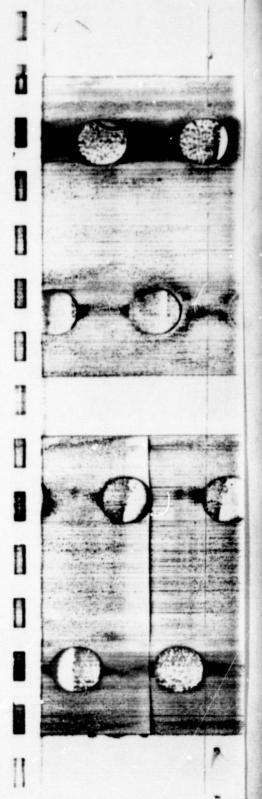
Band strength of ν_1 of OCS. In Table II, the band strength for the ν_1 band obtained by direct summation of Eq. (6) is presented along with $|R_{ii}^{\mu\nu}|^2$, a, and b. As in the case of parallel bands of other linear molecules like CO₂, the constants a and b were not well determined.

Line strengths of four lines belonging to the $(\nu_1 + \nu_2) - \nu_2$ band were also measured. This band is much weaker than the ν_1 band. The measured intensities of $P_{ee}(35)$, $P_{gg}(35)$, $P_{ee}(40)$, and $P_{gg}(40)$ are 0.0149, 0.0157, 0.0121, and 0.0131 cm⁻² atm⁻¹ at 300 K. The calculated values of $|R_{ed}^{egp}|^2$ and the band strength are also presented in Table II.

The value of 26.93 ± 1.23 cm⁻² atm⁻¹ for the band strength of the ν_1 band of $^{16}O^{12}C^{32}S$ at 300 K obtained in this work translates to an integrated band intensity of 34.9 cm⁻² atm⁻¹ for all isotopes and hot bands of OCS at $12 \mu m$.² This is in good agreement with the value of 33.4 cm⁻² atm⁻¹ obtained by Robinson (8) about 30 years ago. A value of (40.6 ± 4) cm⁻² atm⁻¹ has been recently reported by Foord and Whiffen (9) in 1973.

In conclusion, it may be mentioned that recently it was pointed out (10) that OCS could be a minor constituent in Io, the satellite of Jupiter. Estimates of its concentration were made from the strength of its ν_1 band.

 2 Dr. A. G. Maki of the National Bureau of Standards informed us that he and R. Kagann have measured the integrated intensity of the ν_1 band of OCS using a Fourier transform spectrometer and obtained 35.2 cm $^{-2}$ atm $^{-1}$ at 296 K (with an uncertainty of 15%) for the strength of this band. Oftentimes in the past, the band strengths determined at different times and in different laboratories differed by large amounts (see, for instance, Chapter IV on "Intensities in Infrared Spectra" in "Molecular Spectroscopy: Modern Research," Vol. II, (K. Narahari Rao, Ed.), Academic Press, New York, 1976). So, it is useful to note that the value of Maki and Kagann is close to the results obtained in the present investigation. We wish to thank Dr. Maki for bringing this to our attention and also for other useful comments.



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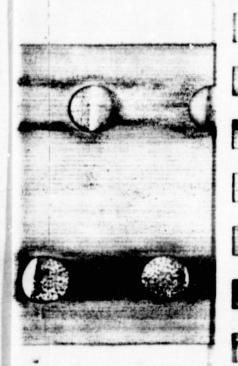
P. P. DAS,2

A. BANO.

K. NARAHARI RAO

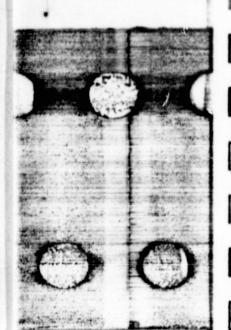
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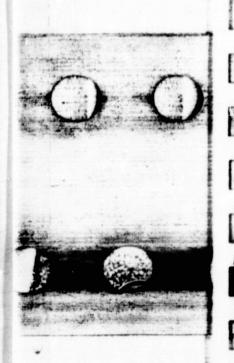
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Tunable Diode Laser Study of the v. Infrared Band of GeH.

The ν_4 infrared band of GeH₄, centered close to 820 cm⁻¹, has been studied by several authors. The most recent rotational analysis is due to Kattenberg *et al.* (1), but the resolution in this work was comparatively modest so that almost none of the fine structure expected from tetrahedral splitting and from the different isotopic species of Ge was clearly resolved. Preliminary accounts have been given of spectra recorded at 0.06-cm⁻¹ resolution, both using a natural sample of germane (2) and also with samples enriched in 70 GeH₄, 72 GeH₄, and 74 GeH₄ (3). Improved values of the ground-state constants B_0 , D_0 , and H_0 have been derived from the latter spectra, but the full assignment of the ν_2 and ν_4 bands has not yet been completed (4).

Here we report measurements on the ν_4 band carried out with a tunable diode laser spectrometer (5, 6). The diode laser provided wavenumber coverage from 818 to about 845 cm⁻¹. Assignments have been made to nearly 200 of the R(0) through R(10) transitions of the five naturally occurring species ⁷⁰GeH₄, ⁷²GeH₄, ⁷³GeH₄, ⁷⁴GeH₄, and ⁷⁶GeH₄. In addition, the high-wavenumber Q-branch head of ⁷⁰GeH₄ and a few high-Q-branch lines of ⁷²GeH₄ have also been assigned.

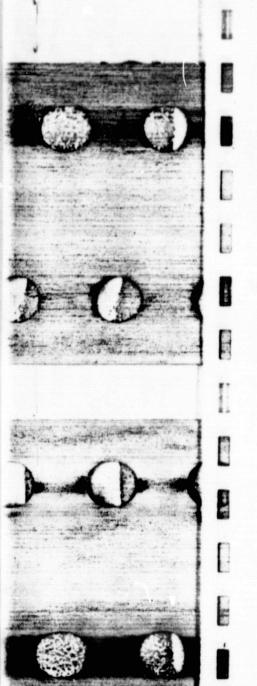
A commercial sample of germane (Matheson Gas Co., stated purity 99.9%) was used. Traces of H_2 were removed by freezing the sample at 77 K and pumping on the solir. Line positions were measured with ≤ 0.1 Torr of germane in a 1-m cell. ν_1 lines of OCS (7) were used for calibration. The absolute accuracy of measurement is believed to be about ± 0.002 cm⁻¹ (8). A typical scan, covering part of the overlapping R(6) multiplets of four isotopic species, is shown in Fig. 1. It is clear that the high resolution achieved with the diode laser allows both the tetrahedral splitting and the isotope effects to be determined despite the high density of lines.

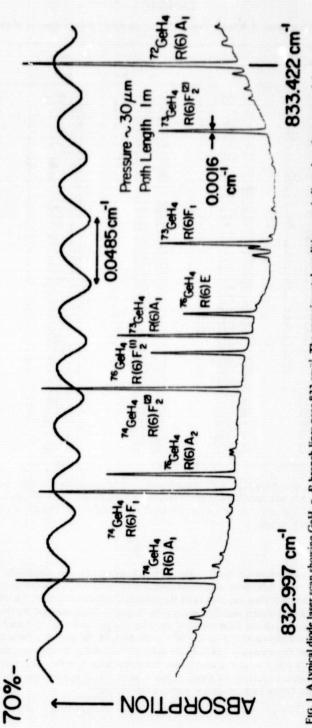
The results are summarized in Table I -III. Table I gives the R-branch assignments and Table II the Q-branch assignments. In both cases the comparison between experimental line positions and those calculated from molecular constants given in Table III is also presented. The molecular constants were obtained from least-squares fits to the data of Tables I and II, using the Hamiltonian of Gray, Robiette, and co-workers (9-11) in which the Coriolis-coupled ν_2 and ν_4 bands are treated together. The ground-state constants (3) and ν_2 constants (4) were taken from analyses of the 0.06-cm⁻¹ spectra and fixed during the fits of the diode laser data. Eventually, it should be possible to improve the constants by ν_2 combined analysis of the diode laser results and the 0.06 cm⁻¹ results with appropriate weighting of the two types of data.

In the case of 70 GeH₄, the observation of Q-branch transitions from Q(3) through Q(21) in addition to a complete set of transitions R(0) through R(9) allowed the refinement of nine parameters in the upper-state Hamiltonian of ν_4 . The Q-branch assignment was not straightforward, but the GeH₄ line identifications given in Table II have been conclusively established by absolute intensity measurements on both Q-branch and R-branch lines. The analysis of the intensity measurements and their interpretation in terms of the ν_4 band strength will be reported subsequently.

For the other isotopic species, where fewer transitions were recorded, the number of parameters refined was always smaller and the poorly determined constants were fixed at their ⁷⁰GeH₄ values. This procedure leads to a consistent set of molecular constants with a uniform quality of fit for the five isotopic species. A few lines, noted in Table I, were given zero weight in the final series of refinements. These lines are less accurate because of relatively poor diode laser performance where they appeared.

The constants of Table III are in satisfactory agreement with theoretical predictions of the isotope shifts. From an approximate harmonic force field adjusted to fit the ν_4 constants and the four fundamentals of 70 GeH₄ (uncorrected for anharmonicity), we have calculated the isotope shifts in the band origin ν_4 and the Coriolis coefficient $2B\zeta_4$. These shifts are compared with experiment in the last four lines of Table III. The trends are well reproduced, especially when the uncertainties in the experimental shifts





NOTES

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Fig. 1. A typical diode laser scan showing GeH₄ v₄ R-branch lines near 833 cm⁻¹. The experimental conditions are indicated on the trace, and the assignments correspond to those in Table 1.

NOTES

TABLE I

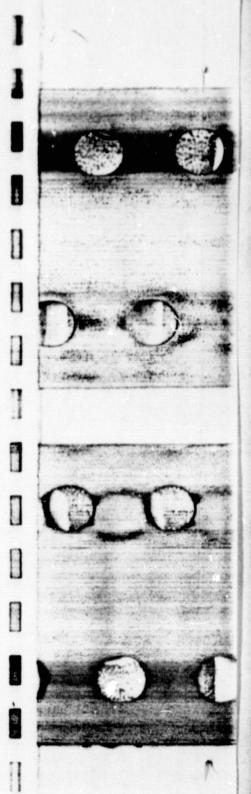
Observed and Calculated R-Branch Transitions of the Five Isotopic Species of Germane

	70 GeH		72 GeH .		73GeH		74 GeH		76 GeH	
Transition	v/cm ⁻¹	o-c	v/cm ⁻¹	o-c	v/cm ⁻¹	0-c	√/cm ⁻¹	0-c	v/cm ⁻¹	0=0
R(0) A1	820.9128	-3	(820.4764)		(820.2672	,	(820.0455)		(919.4722)	
*(1) F1	823.2212	-23	822.7849		822.5742		(822.3700)		(821.9743)	
(2) F2	825.4814	30	825.0357	-2	824.8241		824.6194		(824.2211)	
(2) €	825.5120	27	825.0460	0	824.8564	2	824.6515		(824.2531)	
(3) A2	827.6124	-9	827.1489	-5	826.9571		826.7494		826.3437	
(3) F2	827.6991	-12	827.2545		827.0438		826.8363	-1	826.4341	-24
	827.7706	-19	627.3290	11	827.1153		824.9087	-7	824.5049	-10
(4) F2	829.7676	-:	829.3214 829.4838	-7	829.1075 829.2484	-13	828.8995	-7	(828.4987)	
(4) F1	829.9895	-	829.5405	-12	829.3244	-13	829.1188	-3	(828.7149)	
(4) AI	830.0728	21	829.6232	-3	829.4076	-15	829.1988	-18	(828.7981)	
(5) F1(2)	831.8652	-10	831.4181		831.2031	-5	830.9948	1	(830.5914)	
(5) E	831.9129	-21	831.4440	-	831.2509	-5	831.0422	-2	(830.4391)	
(5) F2	832.1728		831.7228		831.5043	-16	031.2975	-7	830.8932	-
(5) F1(1)	832.2809	-2	831.8305	-11	831.6148	-13	831.4045	-14	831.0013	
(6) A1	833.8703	-15	833.4217	-4	833.2045		832.9971	13	832.5920	- 1
(6) F1	833.9462	1	833.4975	-7	833.2822		833.0733	17	832.6676	- 1
1(6) F2(2)	834.0374	10	833.5852	-10	833.3698	1	833.1604	10	832.7539	
(6) A2	834.3703	-10	833.9200	0	833.4984	(-50)	833.4919	-4	833.0674	1
(6) F2(1)	834.4769	5	834.0262	15	833.8040	(-42)	833.5957	-10	833.1904	
(6) E	834.5115	5	834.0612	21	833.8394	-32	833.6292	-18	833.2244	*
(7) F1(2)	835.9094	,	835.4587	17	835.2398	3	835.0301	16	834.6258	3
(7) E	834.0311	21	835.5782	10	835.3603		835.1490		(834.7419)	
(7) F2(2)	836.1118	17	835.4582	2	835.4418	13	835.2290	0	(834.8223)	
(7) A2	834.2645	. 5	835.81.35	.1	835.5957	-1	835.3843	•	834.9773	
(7) F2(1)	836.6798	-19	836.2291	14	836.0113	11	835.7977	:	835.3886 835.4474	
(7) F1(1)	836.7389	11	834.2867		836.0683	.1	935.8555 (837.0114)	,		
(8) F2(2)	837.8950 837.9490	-1	837.4422 837.4956	:	837.2247 837.2782	13	(837,0444)		836.6041 836.6567	
(8) F1(2)	838.1248	•	837.4736	10	837.4557	16	837.2439	23	(834.8334)	
(8) F2(1)	838.3427	15	837.8855	-11	837.6679	-3	837.4557	7	(837.0463)	
(8) E(1)	838.9253	-2	838.4679	-11	838.2528	23	838.0369	11	837.6263	٠,
(8) F1(1)	838.9501	2	838.4925	- 0	838.2726	-22	838.0408		837.6499	
(8) A1	838.9908	-6	838.5332	-16	838.3178	15	838.1019		837.6917	1
(9) A2	839.8115	-2	(839.3567)		839.1388	13	838.9253	7	838.5153	-
(9) F2(2)	839.8757	-8	(839.4214)		839.2041	20	838.9908	17	(838.5801)	
(9) F1(3)	839.9605	3	839.5053		(839.2856		839.0725	1	(838.6632)	
(9) A1	840.1626	1	839.7083	17	839.4887	13	(839.2737)		838.8642	
(9) F1(2)	840.3864	-17	839.9312	-4	839.7108	-15	839.4988		839.0878	
₹(9) E	840.4437	-5	839.9881		839.7675	-8	839.5545	7	839.1440	
R(9) F2(1)	841.1709	,	840.7108	-5	840.4923		840.2753	-2	839.8633	3
(9) F1(1)	841.2045		840.7458	10	840.5261		840.3078	-11	839.8955	-
R(10) F2(3)	(841.7644)		841.3060	-15	841.0852	-23	(840.8737)		840.4631	
(10) E(2)	(841.8528)		841.3948	-9	841.17	-20	840.9581	-36	840.5529	1
R(10) F1(2)	(841.9329)		(841.4756)		841.2539	-17	841.0397	-17	840.6302	*
R(10) A1	(842.1800)		(841.7220)		(841.5021		841.2852	-21	(840.8762)	
R(10) F1(1)	(842.4478)		(841.9892)		(841.7692		(841.5537)		841.1395	-2
R(10) F2(2)	(842.5483)		(842.0894)		(841.8694		(841.6536)		841.2390	-24
R(10) A2	(843.4028)		(842.9414)		(842.7214		(842.5033)		(842.0888)	
R(10) F2(1)	(843.4266)		(842.9631)		(842.7451		(842.5269)		(842.1124)	
R(10) E(1)	(843.4376)		(842.9761)		(842.7560	,	(842.5378)		(842.1233)	,

Note. The transitions are labeled by their ground-state symmetry species. Columns headed o – c are the residuals (observed – calculated) from the least-squares fits, in units of 10⁻⁴ cm⁻¹. Bracketed line positions are calculated values for transitions which could not be measured. Bracketed residuals indicate transitions omitted from the fits.

are considered. The constants other than ν_4 and $2B\zeta_4$ are, as anticipated, essentially independent of isotopic species so far as we have been able to determine them. It should be noted that the upper-state constants of the ν_2 band have been assumed to be the same for all isotopic species, but that the strength of the ν_2/ν_4 Coriolis ccupling varies slightly from species to species. We have used the theoretical isotope dependence of ζ_{24} to determine the fixed values of the coupling parameter R_{24} (9) used in the analysis.

In conclusion, we have shown that a wealth of precise data on the isotope effects in GeH_4 can be obtained by diode laser spectroscopy, and that the data can be readily interpreted using the Coriolis-coupled Hamiltonian in which ν_2 and ν_4 are treated simultaneously. The data discussed here establish for the first time the most important constants of the ν_4 band for all five isotopic species of germane, and provide a basis for future high-resolution work on this molecule.



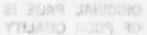


TABLE II

Observed and Calculated Q-Branch Transitions of ⁷⁰GeH₄ and ⁷²GeH₄. Listed in Order of Wavenumber

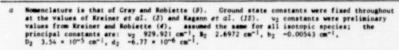
70GeH4	72 GeH.	v/cm ⁻¹	70 _{Ge}	72 Ge	70GeH4	72GeH.	v/cm ⁻¹	70 _{Ge}	72 Ge
Transition	Transition	V/CM	0-0	0-c	Transition	Transition	V/CM	0-6	0-6
0(8) F2(2)		018.6341	-11			0(19) F1(5)	010.9028		-
D(0) F2(2) D(7) F1(2) D(3) A2					0(17) E(3)	0(10) A1(2)	818.9202		-
0(7) F1(2) 0(3) A2		018.6475 018.6570	-1					-:	
0(12) F1(3)		W18.4751			0(14) F2(4) 0(12) A1(2)		010.9219		
12, **						0/21) F2/5)	019.9547		
	Q(15) A2(2)	\$18.4900 \$18.4941		-1		0(21) F2(5) 0(20) F2(5)	818.9780		
(14) E(2)		@18.7224	11,			a.ze, .z.s.	017.0476	-2	
101 61			**		0(19) E(3) 0(16) F2(4)		017,0001		
(10) F2(3) (11) F1(3)		010.7401	.,		0110, 1214,			-1 -2 17 -3	
P(11) F1(3)			10		0(15) A2(1)		019.1132 019.1132	- 11	
1(19) F2(5)	8(20) E(4)	010.7402		10	0(10) F1(4)			**	
0(16) E(3) 0(21) A1(2) 0(9) A2		010.7703	3		0(17) F1(5)		019.1432	-3	
D(21) A1(2)		010.7866				G(21) A2(2)	817.1665		-1
9(9) A2		818.8067	0		0(20) E(4)		019.1792	-6	
9(18) F2(5) 9(20) F1(5)		818.830 2	-3		0(19) F1(5)		819.3197	-2	
2(20) F1(5)		818.8511	-3		0(18) A1(2)		819.3384 819.3937	-	
0(15) F2(4)		818.8492	-1		0(16) A1(2) 0(20) F2(5)		819.3937		
0(13) F1(4)		010.0015	-1						

Note. The transitions are labeled by their ground-state symmetry species. Columns headed o-c are the residuals (observed – calculated) from the least-squares fits, in units of 10^{-4} cm⁻¹.

TABLE III

Molecular Constants (cm⁻¹) for the Five Isotopic Species of Germane, Derived from the Analysis of
Diode Leser Measurements on the P4 Band

Constant	70 _{GeH4}	72Ga H4	73 _{GeR4}	74GeH4	76GeH4
V.	821.5392(12)	821,1084 (23)	820.9025(11)	820.7046(10)	820.3166(12)
	2.69569(8)	2.69570(5)	b	b	b
2864	3.0010(7)	3.0041(7)	3.0056(2)	3.0074(2)	3.0101(2)
a220	9.45(12) × 10-3	9.45(6) * 10-3	ь	b	,
0224	-8.53(15) × 10 ⁻⁴	-8.53(5) × 10 ⁻⁴	-8.63(4) × 10 ⁻⁴	-F.58(4) × 10 ⁻⁴	-8.60(4) × 10"
D.O.	3.06(2) × 10 ⁻⁵	3.07(2) × 10 ⁻⁵	b	b	b
¥110	8.2(6) × 10 ⁻⁵	ь	b	ь	
F134	6.0(9) × 10 ⁻⁵	b	ь	ь	b
G224	6.5(5) × 10 ⁻⁷	6.4(5) × 10 ⁻⁷		b	b
R2.d	2.5431	2.5413	2.5404	2.5395	2.5379
.*	0,0012	0.0010	0.0013	0.0012	0.0012
Av., obs	0.4306	0.206	0.1979	0.388	
Ave, calef	0.438	0.211	0.205	0.394	
4(28C4),obs	-0.0031	-0.0015	-0.0018	-0.0027	
A(2BC4), calef	-0.0035	-0.0017	-0.0016	-0.0031	

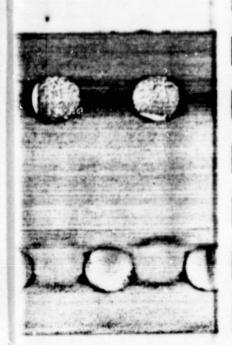


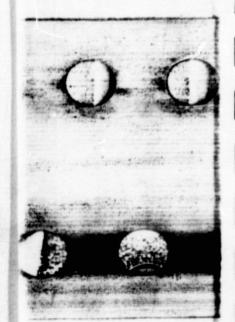
- b Constant fixed at 70GeH, value.
- σ All up- ϵ r state quartic and sextic centrifugal distortion constants other than D constrained to ground state values.
- $d=v_2/v_s$. Coriolis coupling parameter (θ); fixed at values calculated from force field predictions of ζ_{2s} for the five isotopic species.
- e Standard deviation of the fit.
- f Calculated value from approximate harmonic force field (see text).

Note. Errors in parentheses are one standard deviation in the last figure quoted.









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NOTES

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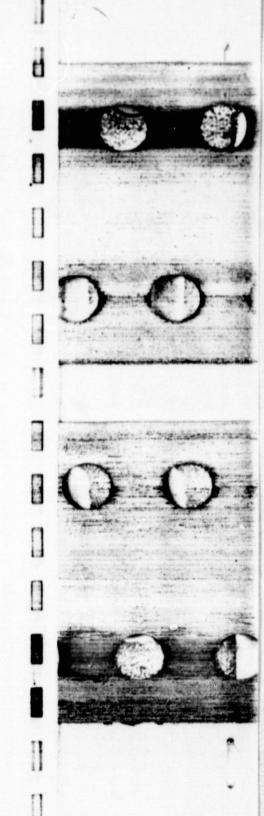
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NOTE

Convolution of a Doppler Line by a Gaussian Instrument Function

In this note a simple method is derived for assessing the distortion of a Doppler line by a Gaussian instrument function. It has immediate applicability for diode laser measurements, and may be used whenever a Gaussian instrument function is a reasonable approximation to reality. The notation is similar to that of Penner (1). Define the absorptance

$$A(x) = 1 - \tau(x) = 1 - \exp[-P(x)],$$
 (1)

where

$$P(x) = P' \exp[-(x/b_D)^2(\ln 2)], \tag{2}$$

$$P' = S/b_n * (\ln 2/\pi)^{1/2} p/.$$
(3)

and

$$x = \nu - \nu_0$$

Let the absorptance A(x) be convoluted by a normalized Gaussian instrument function

$$f(x - x') = \frac{1}{\delta \pi^{1/2}} \exp[-(x - x')^2/\delta^2], \tag{4}$$

where $\delta = d/(\ln 2)^{1/2}$, and d is the half-width half-height (HWHH) of the convoluting function.

The convolution is

$$A'(x) = \int_{-x}^{x} A(x')f(x - x')dx'.$$
 (5)

If we expand A(x) in a series,

$$A(x) = \sum_{n=1}^{\infty} \frac{(-1)^{n+1} P^{/n}}{n!} \cdot \exp[-x^2/(\beta^2/n)],$$
 (6)

where $\beta = b_D/(\ln 2)^{1/2}$, Eq. (5) becomes

$$A'(x) = \sum_{n=1}^{\infty} \frac{(-1)^{n+1} P'^*}{\delta \pi^{1/2} n!} \int_{-\infty}^{\infty} \exp[-x'^2/(\beta^2/n)] \cdot \exp[-(x-x')^2/\delta^2] dx'. \tag{7}$$

The form of Eq. (7) has been deliberately chosen to exhibit the fact that each term of the series for A'(x) contains the convolution of the Gaussian instrument function with a Gaussian of 1/e half-width $\beta/n^{1/2}$.

For n = 1, the integration gives a Gaussian of 1/e half-width $(\beta^2 + \delta^2)^{1/2}$. This result was discussed by Van DeHulst and Reesinck (2). The other terms in the series can be written immediately, giving

$$A'(x) = \sum_{n=1}^{\infty} \frac{(-1)^{n+1} P'^{*}}{n! n^{1/2}} \exp \left[\frac{[-x^2/\beta^2 (1/n + r^2)]}{(1/n + r^2)^{1/2}} \right], \tag{8}$$

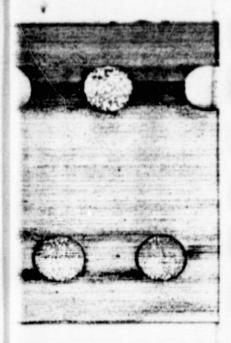
where $r = \delta/\beta = d/b_D$. If Eq. (6) or Eq. (8) is used to compute the equivalent width,

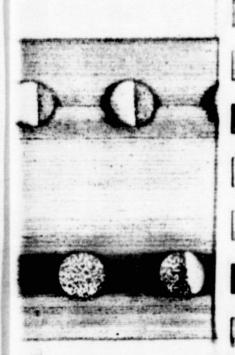
$$W = \int_{-\infty}^{\infty} A(x)dx = \int_{-\infty}^{\infty} A'(x)dx, \qquad (9)$$

we get the result of Ladenburg (3),

$$W = Spl \sum_{n=0}^{\infty} \frac{(Spl/\beta \pi^{1/2})^n (-1)^n}{(n+1)!(n+1)!^2}.$$
 (10)

The form of Eq. (8) permits simple rapid calculation using a desktop programmable calculator.





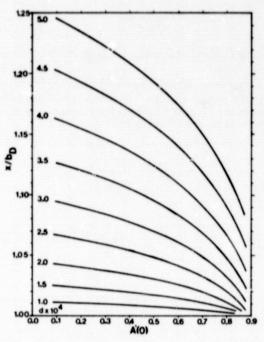


Fig. 1. Instrumental distortion, x/b_D , versus maximum observed absorptance, A'(0), for various values of instrument HWHH, d.

The convergence of Eq. (8), as well as Eq. (10), is absolute. The rate of convergence depends mostly on the value of P'. For an accuracy of 10^{-5} , P' = 0.2 requires 5 terms, while P' = 10 requires 35 terms in the series. However, even the longer series is computed in a few seconds on a programmable calculator.

If there were no instrumental distortion we would measure the HWHH of a spectral line, b_D , at that point for which $\pi(x) = (\pi(0))^{1/2}$, or where $A(x) = 1 - \{1 - A(0)\}^{1/2}$. To find the error in this procedure for a convoluted line we iterated on Eq. (8) until we found an x such that

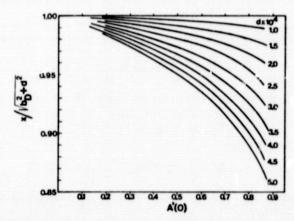
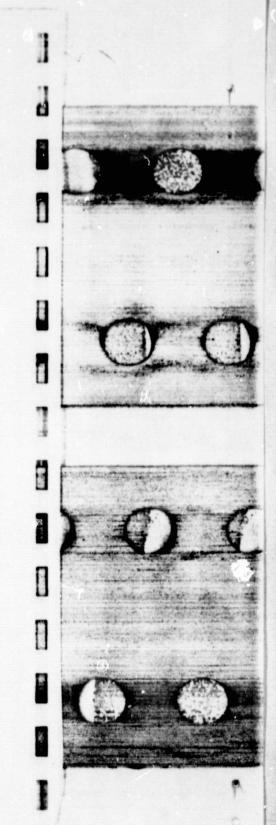


Fig. 2. The error in the first-order approximation to the instrumental distortion, $x/(b_0^2 + d^2)^{1/2}$, versus maximum observed absorptance for various values of instrument HWHH, d.



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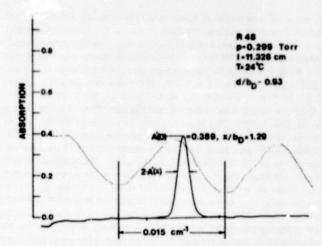


Fig. 3. Reproduction of laser diode spectrometer scan taken at OSU. Measurement of $d/b_{\rm D}$ is from Fig. 4.

$$A'(x) = 1 - [1 - A'(0)]^{1/2}$$
 (11)

This value, x (or x/b_D) is a measure of the broadening of the absorptance, or transmittance.

Figure 1 was prepared by choosing values for P', b_D , and d and computing A'(0); then with this A'(0) we compute A'(x) to satisfy Eq. (11). Values of $d < b_D = 0.00066$ cm⁻¹ were chosen to correspond to anticipated conditions for measurement in the 15 μ m CO₂ bands. The results of these computations

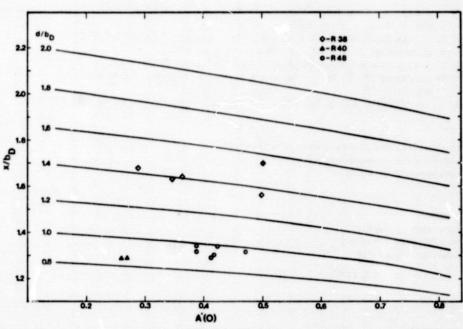
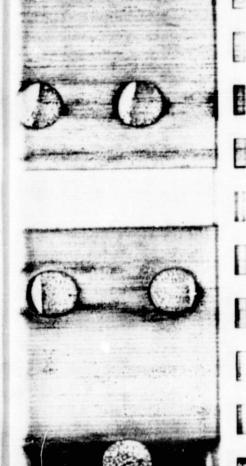


Fig. 4. Chart similar to Fig. 1 produced for larger values of instrument HWHH, shown as the ratio d/b_D.



agree perfectly with the published results of Kostkowski and Bass (4), who calculated the convolution, on a large computer by digital integration, for three values of the peak absorption.

The appearance of these curves shows clearly that for the same instrument function, and using the standard method of measuring HWHH for diode laser measurements (see Planet et al. (5,6)), the distortion is greater for weak lines than for the stronger ones. If we assume that the Doppler measurements of Ref. (5) had A'(0) = 0.5, then $x = 1.08b_D$, 8% wider than Doppler, indicates an instrument function of about 3.2×10^{-4} cm⁻¹ (HWHH). If A'(0) had been 0.3, Fig. 1 still shows $d = 2.9 \times 10^{-4}$ cm⁻¹.

Figure 2 shows the error incurred by assuming that the convoluted half-width is given by the first-order correction alone, $(b_B^2 + d^2)^{1/2}$. This assumption is seen to be better for weaker lines.

An example of the use of Fig. 1, or any similar chart, is demonstrated in Fig. 3. This is a reproduction of a measurement made at OSU on a diode laser spectrometer in the ν_2 15- μ m band of $^{12}C^{16}O_2$. A simultaneous etalon trace serves as the wavenumber (cm⁻¹) marker. We measured A'(0) = 0.389, and the HWHH $x = 1.29b_D$. A new chart, Fig. 4, was prepared, because the instrument function was too large for Fig. 1. Using this figure we estimate $d/b_D = 0.93$ ($d = 6.1 \times 10^{-4}$ cm⁻¹).

The results of several other measurements in the R branch made at OSU are also shown in Fig. 4. The instrument function clearly changes with wavenumber, i.e., with diode laser mode.

The method of computation presented here is simple and direct. It suggests that a close approximation to the width of a Gaussian instrument function, or an almost Gaussian function may be found by measuring a line with a Doppler absorption coefficient. Charts for specific spectral regions and expected instrument widths can be easily prepared by the use of Eq. (8).

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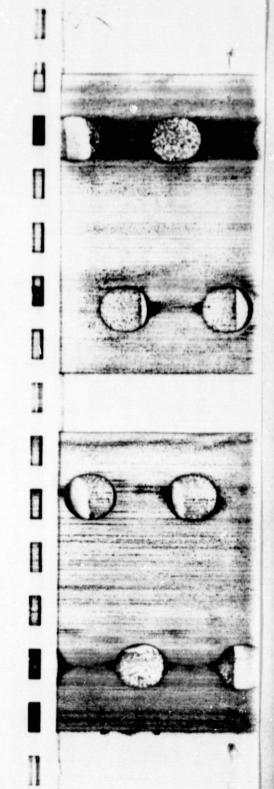
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Intercomparison of CO₂ and HCN wave numbers at 800-600 cm⁻¹ with a diode laser spectrometer

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This investigation is based on two recent publications.^{1,2} The high spectral resolution and sensitivity obtained with diode laser spectrometers are being increasingly used in procuring basic observational data for determination of molecular parameters. With diode lasers, typically we are able to scan spectral intervals of about 0.5-1 cm⁻¹ at any one time, and quite frequently these spectral intervals do not have a continuity in them. Obviously, meaningful spectroscopic constants can be determined only when there is internal consistency in the wave-number data available for such disjointed

spectral intervals. At the longer wavelengths of 12–17 μ m where much useful information can be obtained by studying low-frequency fundamental vibration-rotation bands of polyatomic molecules, the band spectra of CO₂ and HCN seem to provide a set of convenient wave-number standards. The HCN data presented in Ref. 1 were generated from available rotational constants for its 0110 and 0000 levels combined with the ν_2 band center evaluated from grating measurements.³ On the other hand, the CO₂ wave numbers of Ref. 2 were obtained with a high-resolution Fourier spectrometer, and their wave-number precision is quoted to be better than 0,001 cm⁻¹ by relating the Fourier data for pure rotational water vapor lines to the values calculated from water vapor energy levels. We have intercompared these HCN and CO₂ data by making measurements at Doppler-limited spectral resolution with a diode laser spectrometer and find that there is internal consistency in the wave-number values of Refs. 1 and 2 within about ± 0.0005 cm⁻¹.

The diode laser assembly described by Reddy et al. has been used for the intercomparison of the measurements of rememts with the Fourier Data of Ref. 2

R(56) of \(\nu_2 \) of \(\frac{12C^{16}O_2}{2} \) relative to \(Q(12) \) From of \(\nu_2 \) of \(\frac{12C^{14}N}{2} \) as standard Fourier

713.1348

713.1350 Av. Value
713.1359

Table I. Comparison of the Present Measurements with the Fourier Data of Ref. 2

$R(76)$ of ν_2 of $^{12}\text{C}^{16}\text{O}_2$ relative to $R(5)$ of ν_2 of $H^{12}\text{C}^{14}\text{N}$ as standard	From Fourier	$R(56)$ of ν_2 of $^{12}\text{C}^{16}\text{O}_2$ relative to Q of ν_2 of $H^{12}\text{C}^{14}\text{N}$ as standard	(12) From Fourier
729.6847		713.1348	
729.6844 Av. value 729.6847 729.6846	729,6847	713,1350 Av. Value 713.1352 713.1350	713,1348

Table II. Wavenumbers (vac. cm⁻¹) of the *Q* Branches of th/2 ν_1 - ν_2 Bands of ¹²C¹⁶O₂ and ¹³C¹⁶O₂

		¹² C ¹⁶ O ₂		13C16O2			
Q(J)	Present obs. valuea	High-resolution Fourier ^b	Diff. × 10⁴	Present obs. value ^c	High-resolution Fourier ^b	Diff. × 10 ⁴	
Q(2)	720,7998	720,7994	4	721.5752	721.5749	3	
$\dot{\mathbf{Q}}(4)$	720.7846	720.7845	1	721.5539	721.5536	3.	
Q(6)	720,7608	720.7610	-2	721.5206	721,5201	5	
Q(8)	720.7292	720,7290	2	721.4745	721,4744	1	
Q(10)	720.6890	720.6886	4	721.4165	721,4166	-1	
$\dot{Q}(12)$				721.3460	721.3466	6	
Q(14)				721.2644	721.2644	0	
Q(16)				721.1692	721.1700	-8	
Q(18)				721.0638	721,0635	3	

 $^{^{}a}$ $^{12}\mathrm{C}^{16}\mathrm{O}_2$ lines were measured relative to R(2) of ν_2 of H $^{12}\mathrm{C}^{14}\mathrm{N}$.

Table III. Wavenumbers (vac. cm⁻¹) of the Q Branch Lines of the ν_1 - ν_2 Band of $^{12}\text{C}^{16}\text{O}^{17}$ and the ν_2 Band of $\text{H}^{13}\text{C}^{14}\text{N}$

		12C16O1	16O17O	170			ñï-3C¹4N	
Q(J)	v calc.ª	$(0-C) \times 10^4$	Q(J)	v calc. a	$(O-C) \times 10^4$	Q(J)	v calc. b	$(O-C) \times 10^4$
Q(1)	711.2981	2	Q(16)	711.0743	-11	Q(4)	706,0999	-3
Q(4)	711,2831	1	Q(17)	711.0462	-6	Q(5)	706.1672	-5
Q(5)	711.2748	0	Q(18)	711.0166	-1	Q(6)	706,2479	-5
Q(6)	711.2648	1	$\dot{Q}(19)$	710.9854	2	$\mathbf{Q}(7)$	706.3421	-1
Q(7)	711.2532	3	Q(20)	710,9526	4	Q(8)	706,4498	13
Q(8)	711.2399	3	Q(21)	710.9182	3	Q(9)	706,5709	16
Q(9)	711.2249	-1	Q(22)	710.8822	4	Q(10)	706.7055	-3
Q(10)	711,2083	3	Q(23)	710.8446	. 5	Q(11)	706.8535	-10
$\hat{Q}(11)$	711.1900	3	Q(24)	710.8055	7	Q(12)	707.0150	-2
Q(12)	711.1701	-3	Q(25)	710.7649	0	Q(13)	707.1899	0
Q(13)	711.1486	-4	Q(26)	710.7227	-3	• • • • • • • • • • • • • • • • • • • •		
Q(14)	711.1255	-5	$\dot{Q}(27)$	710.6789	-4			
Q(15)	711,1007	-1	Q(28)	710.6337	-2			

^a ν calc, = 711.29978 - 0.0008336 $J(J+1)+1.64\times 10^{-8}$ J^2 $(J+1)^2$. The measurements were made relative to P(13) of ν_1 - ν_2 of 12 C¹⁶O₂.

^b Ref. 4; R. Antilla, U. Oulu, Finland; private communication.

c $^{13}\text{C}^{16}\text{O}_2$ lines were measured relative to R(66) of the ν_2 of $^{12}\text{C}^{16}\text{O}_2$.

 $^{^{}b} \nu$ calc. = 705.96535 + 0.0067283 J(J+1). The measurements were made relative to R(48) of ν_2 of $^{12}C^{16}O_2$.

 $^{12}\mathrm{C}^{16}\mathrm{O}_2$ lines relative to H¹²C¹⁴N lines at 14 µm. It may be recalled that with a setup of this type, Fabry-Perot fringes of about 0.015 cm⁻¹ apart produced with an air-spaced étalon are recorded simultaneously with the scanning of diode laser IR spectra. The absolute wave number on the fringe scale is obtained by recording an available absorption standard. By sdopting this procedure, R(76) and R(56) lines of the ν_2 band of $^{12}\mathrm{C}^{16}\mathrm{O}_2$ have been measured, the first one by using an R branch line and the second one by using a Q branch line of the ν_2 band of H¹²C¹⁴N as standards. The data presented in Table I show the extent of agreement between our measurements and the Fourier data² and also demonstrate that the reproducibility of diode laser measurements is excelent

Another test for internal consistency between the R- and Q-branch wavenumbers of Ref. 1 was possible because we could measure the R(1) of ν_2 of $H^{12}C^{14}N$ relative to the Q(28) line of the same band. It has been found that such a measurement for R(1) came out to be 717.8932 cm⁻¹ as compared to the value 717.8927 cm⁻¹ calculated in Ref. 1.

Table II compares the diode laser measurements for the ν_1 - ν_2 bands of $^{12}\mathrm{C}^{16}\mathrm{O}_2$ and $^{13}\mathrm{C}^{16}\mathrm{O}_2$ with the Fourier spectroscopic data of Paso et al.⁴ The agreement between the two sets of data is again within a few units in the fourth decimal place.

While performing the above described measurements, we have observed the ν_2 of $H^{13}C^{14}N$ and ν_1 - ν_2 of $H^{12}C^{16}O^{17}O$ in the natural sample. The wave numbers of the bands of these two molecules are given in Table III.

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NO₂
$$v_2$$
 & $2v_2$; H₂O (8050-9730 cm⁻¹); O₃ v_2 ; C₂H₂ v_5 & 6680-6460 cm⁻¹
CH₃D v_3 & v_6 ; CH₃I v_6 ; NH₃ v_2 , $2v_2$, v_4 ; ND₃ v_2 ; 13 C¹⁶O₂ v_3 ;
PH₃ v_1 & v_3 ; OCS v_1 ; GeH₄ v_4 .

Much of the work was done by using the Doppler-limited resolution provided by diode lasers. The diode laser was used as a source to a grating spectrometer which has been used earlier for high resolution studies. This technique provided many advantages. Wherever possible, the studies have been directed to intensity determinations of infrared bands.

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